# Strongly interfacial interaction NiCoSe<sub>x</sub>/CG heterostructure with rapid diffusion kinetics as flexible anode for high-rate sodium storage

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Fig. S1 Photographs of CoSe<sub>2</sub>/CG and NiSe<sub>2</sub>/CG in the tiled and curled state.



Fig. S2 Cross sectional SEM pictures of CoSe<sub>2</sub>/CG (a), NiSe<sub>2</sub>/CG (b) and NiCoSe<sub>x</sub>/CG (c) films.

### Calculation of the relative content of CoSe2 and carbon in CoSe2/CG.

As shown in Figure 1c, the weight loss of  $CoSe_2/CG$  is 77.8 wt%. After heating to 500 °C in the air, the final product is  $Co_3O_4$ . According to the reaction equation (1), the weight loss from pure  $CoSe_2$  to  $Co_3O_4$  is 63.0 wt%. The weight reduction of composite is consisted of two parts: weight

loss of the variation from  $CoSe_2$  to  $Co_3O_4$  and the weight loss of carbon oxidation, which can be illustrated as the equation (2):

$$3CoSe_2 + 8O_2 \rightarrow Co_3O_4 + 6SeO_2 \uparrow \tag{1}$$

$$W \times 63.0\% + 100\% - W = 77.8\%$$
 (2)

W represents the loading content of  $CoSe_2$ , and the loading content of carbon is (100%-W). Thus, the content of  $CoSe_2$  in  $CoSe_2/CG$  can be calculated as 60.0 wt%.

#### Calculation of the relative content of NiSe<sub>2</sub> and carbon in NiSe<sub>2</sub>/CG.

As shown in Figure 1c, the weight loss of NiSe<sub>2</sub>/CG is 91.0 wt%. After heating to 500 °C in the air, the final product is Ni<sub>2</sub>O<sub>3</sub>. According to the reaction equation (1), the weight loss from pure NiSe<sub>2</sub> to Ni<sub>2</sub>O<sub>3</sub> is 61.7 wt%. The weight reduction of composite is consisted of two parts: weight loss of the variation from NiSe<sub>2</sub> to Ni<sub>2</sub>O<sub>3</sub> and the weight loss of carbon oxidation, which can be illustrated as the equation (2):

$$4NiSe_2 + 11O_2 \rightarrow 2Ni_2O_3 + 8SeO_2 \uparrow$$
(3)

$$W \times 61.7\% + 100\% - W = 91.0\%$$
(4)

W represents the loading content of NiSe<sub>2</sub>, and the loading content of carbon is (100%-W). Thus, the content of NiSe<sub>2</sub> in NiSe<sub>2</sub>/CG can be calculated as 23.5 wt%.

#### Calculation of the relative content of CoSe<sub>2</sub>, NiSe<sub>2</sub> and carbon in NiCoSe<sub>x</sub>/CG.

As shown in Figure 1c, the weight loss of  $NiCoSe_x/CG$  is 84.1 wt%. The weight reduction of composite is consisted of three parts: weight loss of the variation from  $CoSe_2$  to  $Co_3O_4$ ,  $NiSe_2$  to  $Ni_2O_3$  and the weight loss of carbon oxidation.

W represents the loading content of  $CoSe_2$ . Because the ratio of cobalt to nickel in the raw material is controlled to be 1:1, the loading content of NiSe<sub>2</sub> is also W, and the loading content of carbon is (100%-2W). Thus, according to the equation (5), it can be calculated that the content of  $CoSe_2$  in NiCoSe<sub>x</sub>/CG is 21.1 wt%, the content of NiSe<sub>2</sub> in NiCoSe<sub>x</sub>/CG is 21.1 wt%. W × 63.0% + W × 61.7% + 100% - W = 84.1% (5)



Fig. S3 Initial three discharge/charge voltage curves at 0.1 A  $g^{-1}$  of CoSe<sub>2</sub>/CG.



Fig. S4 Initial three discharge/charge voltage curves of  $NiSe_2/CG$  at 0.1 A g<sup>-1</sup>.



Fig. S5 The initial three discharge/charge curves of NiCoSe<sub>x</sub>/CG at different current densities.



Fig. S6 Cyclic performance of NiCoSe<sub>x</sub>/CG at different voltage ranges.



Fig. S7 Cyclic performance of NiCoSe<sub>x</sub>/CG with different cobalt to nickel at 0.1 A g<sup>-1</sup>.



Fig. S8 Rate performance of  $NiCoSe_x/CG$  with different cobalt to nickel at various current density from 0.1 to 5 A g<sup>-1</sup>.

**Table S1.** Randles equivalent circuit and the dynamical parameters of the  $CoSe_2/CG$ ,NiSe\_2/CG and NiCoSe\_x/CG after three cycles at 100 mA/g.

	Cr Re Re Rf		<b>—</b> 0
	$R_e/\Omega$	$R_{f}\!/\Omega$	$R_{ct}/\Omega$
CoSe <sub>2</sub> /CG	13.4	58.5	43.5
NiSe <sub>2</sub> /CG	11.8	27.0	29.2
NiCoSe <sub>x</sub> /CG	11.6	13.9	11.1

**Table S2.** A comparison of this work with previously reported performances of flexible

 anode materials for SIBs.

	Initial	Capacity	Rate capability	Capacity	
	roversible	retention —	1 5	<ul> <li>retention</li> </ul>	
Samples	Teversible				Ref

	capacity/mAh	at a low	mAh·g <sup>-1</sup>	mA·g <sup>-1</sup>	at a high	
	$\cdot g^{-1}$	current			current	
		density			density	
Bi <sub>4</sub> Se <sub>3</sub> /Bi <sub>2</sub> O <sub>2</sub> Se- CNTs-rGO	345.6		345.6	0.1		
		68.1% at	235.3	0.2	84% at 1	[1]
		0.1 A/g	155	0.5	A/g after	
		after 100	120.5	1	1000	
		cycles	101.5	2	cycles	
			85.4	5		
			242.7	0.05		
		88% at	215.5	0.1	null	[2]
P-doped carbon	197	0.2 A/g	186.8	0.2		
cloth	180	after 600	152.7	0.5		
		cycles	123.1	1		
			87	2		
			183	0.1	000/ 1	
			150	0.2	89% at 1	
Crumpled graphene	190	null	120	0.5	A/g after	[3]
			100	1	500	
			80	2	cycles	
	145	020/	163	0.1		
		83% at	143	0.3		
1 / 1		0.3 A/g	120	0.5		[4]
graphene stacks		atter	100	1	null	נין
		1000	82	2		
		cycles	52	5		
		91% at	110	0.1		
		05 A/g	82	0.2		
Ni-CNTs	101	after	78	0.5	null	[5]
		1800	72	1		
		cycles	67	2		
	695		400	0.05		[6]
		33.4% at	210	0.1	null	
Bismuth Selenide		0.05 A/g	160	0.2		
/Graphene		after 50	70	0.5		
		cycles	30	1		
			20	2		
NiCoSe <sub>x</sub> /CG	338		338	0.1	66% at 2 A/g after 1000	This work
		61% at	285	0.2		
		0.1 A/g	225	0.5		
		after 100	206	1		
		cycles	183	2		
		-	140	5	cycles	

## References

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