Electronic Supporting Information for

A Palladium complex of a macrocyclic selenium ligand: Catalyst for dehydroxymethylation of dihydroxy compounds

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EXPERIMENTAL SECTION

Crystallography: A DCM : EtOAC (4 mL, 2:2) solution of C1 was slowly concentrated over a period of 10 days to afford suitable orange colored block like crystals. The crystals were carefully picked under a polarizing microscope and mounted on loop with the help of paraffin oil. The single-crystal X-ray data collection was carried out at Bruker D8 Quest PHOTON II diffractometer with monochromatic Mo K α radiation ($\lambda = 0.71073$ Å) at 296(2) K operating at 50 kV voltage and 20 mA current by using ω and ϕ scan. The diffraction profiles were integrated with the SAINT program^{s1} and the obtained data sets were reduced using the APEX3 software. Numerical absorption corrections (multi scan) were performed with SADABS program.^{s2} The structure was solved and refined by the full matrix least-square method on F² using SHELXL-2016^{s3} present in the WINGX package of programs (Version 2018-3).^{s4} One can find B-level alerts are due to the significant disorder in the compounds C1, for which no further modelling was attempted. All the non-hydrogen atomic sites in the present structure were located successfully in different Fourier maps and finally refined with anisotropic displacement parameters. Hydrogen atoms were fixed geometrically in calculated positions and refined using the riding model. The details about the crystal structure solution and final refinement parameters for compound C1 are listed in Table s1 along with data of L1. Selected bond distances and angles are given in Table s2. The crystallographic files can be obtained for the compounds L1, and C1 (2192408 and 2192410), from the Cambridge Crystallographic Center (CCDC) Data via www.ccdc.cam.ac.uk/datarequest/cif. Selected bond lengths around the square planer geometry of palladium are shown in figure s1.

	L1	C1
empirical formula	C ₁₁ H ₁₃ OSe _{0.50}	$C_{44}H_{52}Cl_2O_4PdSe_2$
formula weight	200.69	980.07
temperature [K]	293(2)	296(2)
	Rigaku Oxford	Bruker D8 Quest
Diffractometer	2018	PHOTON II
wavelength [Å]	1.54184	0.71073
crystal system	Orthorhombic	Monoclinic
space group	P n m a	P 21/c
unit cell dimensions:		
<i>a</i> [Å]	18.412(2)	24.3388(17)
<i>b</i> [Å]	23.1159(13)	10.2891(7)
<i>c</i> [Å]	4.7170(4)	18.3315(12)
α [°]	90.00	90
β[°]	90.00	109.150(2)
⊁[°]	90.00	90
<i>V</i> [Å ³]	2007.6(3)	4336.6(5)
Ζ	8	4
$\rho_{\text{calc}}[\text{Mg/m}^3]$	1.328	1.501
μ[mm ⁻¹]	2.611	2.269
F(000)	832	1984
crystal size [mm ³]	$0.9 \times 0.547 \times 0.501$	$0.30\times0.250\times0.180$
θ limit[°]	4.7990 to 64.6250	3.132 to 25.000
index range (h, k, l)	-16, 21, -20, 27, -4, 5	-28, 28, -12, 12, -21, 21
reflections collected	3480	75216
independent reflections	1802	7581
<i>R</i> (int)	0.0503	0.0284
max. and min. transmission	1.0000 and 0.18744	0.685 and 0.549
data/restraints/parameters	1802/0/ 115	7581/0/478
goodness-of-fit on F^2	1.056	1.034
\overline{R} indices (final) $[I > 2\sigma(I)]$		
R_1	0.0999	0.0284
wR_1	0.1345	0.0665
R indices (all data)		
R_1	0.2731	0.0368
wR_2	0.3115	0.0692
largest diff. peak and hole [eÅ ⁻³]	2.254 and -1.060	0.635 and -0.575

 Table s1.
 Summary of crystallographic data.

C1				
Bond	Length (Å)			
Pd(1)—Se(1)	2.4320(3)			
Pd(1)—Se(2)	2.4306(4)			
Pd(1)— $Cl(1)$	2.2979(8)			
Pd(1)—Cl(2)	2.2930(8)			
Se(1)—Cl(1)	1.9770(3)			
Se(2)—Cl(2)	1.9820(3)			
C1				
Bond	Angle (deg.)			
Se(2)—Pd(1)— $Se(1)$	169.20(1)			
Cl(1)— $Pd(1)$ — $Se(2)$	94.16(2)			
Cl(2)— $Pd(1)$ — $Se(2)$	86.12(2)			
Cl(1) - Pd(1) - Se(1)	87.29(2)			
Cl(2) - Pd(1) - Se(1)	93.22(2)			
Cl(1)— $Pd(1)$ — $Cl(2)$	175.75(4)			

Table s2. Key crystallographic distances [Å] and angles [°].



Figure s1. Selected bond lengths around the square planer geometry of palladium in C1.

Computational Studies of Selenium Inversion: To understand the selenium inversion process in the complex **C1**, we have calculated the energy barrier for the inversion process by employing the density functional method B3LYP.^{s5-s8} We have employed a mixed basis set consisting of LANL2DZ^{s8-s11} for heavy atoms (Pd, Se, Cl) to account for the relativistic effects and a 6-31G (D, P)^{s12-s14} basis set for the lighter atoms (H, C, O). All the calculations are performed with Gaussian

program package.^{s15} To the end, we have optimized geometries **C1-R**, **TS** (pyramidal inversion) and **C1-P** (see Fig. 6). The **C1-R** and **C1-P** are local minima on the account of all positive frequencies and **TS** is a first order saddle point with one negative frequency of magnetude -118 cm⁻¹. The activation energy required for the pyramidal inversion is ~18.81 kcal/mol. This activation energy can easily be achieved at the elevated temperatures used in the variable temperature NMR (see main manuscript).

Table s3.	Comparison	on earlier	reported	dehydroxy	methylation	of mono-	-alcohols	with	current
protocol.									

Catalyst	mol%	Time (h)	Temp. (°C)	Yield	Reference
				(%)	
[To ^M Rh(CO) ₂]	0.009	24	450 W Hg lamp.	>95	s16
[Ir(coe) ₂ Cl] ₂ , Ligand- rac-BINAP	2.5 and 5	8	164	61	s17
$Pd(OAc)_2$	8 - 16	24-48	130	93	s18
[Ir(coe) ₂ Cl] ₂ Ligand_rac-BINAP, LiCl	2.5, 5, and 10	8	164	66-97	s19
Ru(COD)Cl ₂ Ligand- $P(o-tolyl)_3$	5 and 12.5	16	177	90-93	s20
Pd/CeO ₂ .	3	5	180	>90	s21
Ni/CeO ₂ -NaNaph	10	24	150	67	s22
Trans-PdCl ₂ (SeR ₂) ₂	5	48 h	130	64-91	Current work

Spectroscopic Data of Starting Materials:

The starting materials (**3a-3t**) were synthesized by using an earlier reported procedure.^{s23} The spectroscopic data of starting materials **3a-3l** are in agreement with the reported data.^{s23}



((Octane-1,8-diylbis(oxy))bis(5-methyl-2,1-phenylene))dimethanol) (3m): Appeared as offwhite solid, (89%). ¹H NMR (500 MHz CDCl₃, δ/ppm): 7.08 (s, 2H), 7.03 (d, *J* = 8 Hz, 2H), 6.76 (d, *J* = 8 Hz, 2H), 4.65 (s, 4H), 3.98 (t, *J* = 6 Hz, 4H, OC*H*₂), 2.94 (s, 1H), 2.88 (s, 1H), 2.28 (s, 6H), 1.79 – 1.81 (m, 4H), 1.40 – 1.48 (m, 8H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 154.7 (s), 129.6 (s), 129.3 (s), 128.9 (s), 128.8 (s), 111.0 (s), 68.0 (s), 62.0 (s), 29.2 (s), 29.2(s), 26.0 (s), 20.3 (s).



((Hexane-1,6-diylbis(oxy))bis(5-methyl-2,1-phenylene))dimethanol) (3n): Appeared as offwhite solid, (92%). ¹H NMR (500 MHz CDCl₃, δ/ppm): 7.08 (s, 2H), 7.03 (d, *J* = 8 Hz, 2H), 6.76 (d, *J* = 8.5 Hz, 2H), 4.63 (s, 4H), 4.00 (t, *J* = 6 Hz, 4H, OC*H*₂), 2.95 (s, 1H, OH), 2.88 (s, 1H, OH), 2.28 (s, 6H), 1.83 - 1.85 (m, 4H), 1.54 - 1.57 (m, 4H); ¹³C {¹H} NMR (125 MHz, CDCl₃) δ 154.7 (s), 129.8 (s), 129.4 (s), 128.9 (s), 111.1 (s), 67.9 (s), 62.0 (s), 29.2 (s), 25.9 (s), 20.3 (s).



((Butane-1,4-diylbis(oxy))bis(5-methyl-2,1-phenylene))dimethanol) (3o): Appeared as offwhite solid, (84%). ¹H NMR (500 MHz CDCl₃, δ /ppm): 7.10 (s, 2H), 7.05 (d, *J* = 8 Hz, 2H), 6.77 (d, *J* = 5 Hz, 2H), 4.65 (s, 4H), 4.07 (t, *J* = 6 Hz, 4H, OCH₂), 2.95 (s, 1H, OH), 2.88 (s, 1H, OH), 2.50 (s, 2H), 2.29 (s, 6H), 2.00 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 154.5 (s), 129.8 (s), 129.5 (s), 129.0 (s), 128.9 (s), 111.1 (s), 67.5 (s), 61.9 (s), 26.2 (s), 20.3 (s).



((Octane-1,8-diylbis(oxy))bis(5-methoxy-2,1-phenylene))dimethanol) (3p): Appeared as offwhite solid, (83%). ¹H NMR (500 MHz CDCl₃, δ/ppm): 6.83 (s, 6H), 3.91 (t, *J* = 6.5 Hz, 4H, OC*H*₂), 3.77 (s, 6H), 1.73 – 1.79 (m, 4H), 1.45 – 1.48 (m, 4H), 1.38 – 1.40 (s, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 153.6 (s), 151.0 (s), 130.4 (s), 114.6 (s), 113.1 (s), 112.3 (s), 68.6 (s), 62.1 (s), 55.8 (s), 29.3 (s), 29.2(s), 26.1 (s).



((Octane-1,8-diylbis(oxy))bis(5-chloro-2,1-phenylene))dimethanol) (3q): Appeared as white solid, (84%). ¹H NMR (500 MHz CDCl₃, δ /ppm): 7.27 (d, *J* = 2.5 Hz, 2H), 7.18 (dd, *J* = 6.5 Hz, *J* = 2 Hz, 2H), 6.76 (d, *J* = 8.5 Hz, 2H), 4.64 (s, 4H), 3.97 (t, *J* = 6.5 Hz, 4H, OC*H*₂), 2.40 (s, 2H), 1.77 - 1.82 (m, 4H), 1.45 - 1.48 (m, 4H), 1.36 - 1.41 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 155.1 (s), 130.9 (s), 128.1 (s), 128.1 (s), 125.3 (s), 112.1 (s), 68.3 (s), 61.2 (s), 29.1 (s), 29.0 (s), 25.9 (s).



((Octane-1,8-diylbis(oxy)bis(3-bromo-4,1-phenylene))dimethanol) (3r) Appeared as off-white solid, (85%). ¹H NMR (500 MHz CDCl₃, δ /ppm): 7.53 (d, *J* = 2.5 Hz, 2H), 7.21 (dd, *J* = 6.5 Hz, *J* = 2Hz, 2H), 6.85 (d, *J* = 8.5 Hz, 2H), 4.57 (s, 4H), 4.01 (t, *J* = 6.5 Hz, 4H, OCH₂), 1.91 (s, 2H), 1.81 – 1.84 (m, 4H), 1.50 – 1.53 (m, 4H), 1.40 – 1.42 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 154.9 (s), 134.3 (s), 132.1 (s), 127.1 (s), 113.1 (s), 112.2 (s), 69.2 (s), 64.2 (s), 29.1 (s), 28.9(s), 25.8 (s).



(((1,4-Phenylenebis(methylene))bis(oxy))bis(2,-1-phenylene))dimethanol) (3s): Appeared as off-white solid, (83%). ¹H NMR (500 MHz CDCl₃, δ/ppm): 7.42 (s, 4H), 7.26 (dd, *J* = 22.5 Hz, *J* = 7 Hz, 4H), 6.92 - 6.97 (m, 4H), 5.11 (s, 4H), 4.72 (s, 4H), 1.23 (s, 2H, OH); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 156.4 (s), 136.7 (s), 129.5 (s), 128.9 (s), 128.8 (s),127.6 (s), 121.1 (s), 111.6 (s), 69.7 (s), 62.0 (s).



((([1,1'-Biphenyl]-4,4'-diylbis(methylene))bis(oxy))bis(2,-1-phenylene))dimethanol) (3t): Appeared as off-white solid, (89%). ¹H NMR (500 MHz CDCl₃, δ /ppm): 7.63 (d, *J* = 8 Hz, 4H), 7.5a (d, *J* = 8 Hz, 4H), 7.33 (d, *J* = 8 Hz, 2H), 7.27 – 7.30 (m, 2H), 6.97-7.00 (m, 4H), 5.18 (s, 4H), 4.77 (s, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 156.5 (s), 140.5 (s), 135.9 (s), 129.5 (s), 128.9 (s),128.8 (s), 127.8 (s), 127.4 (s), 121.1 (s), 111.7 (s), 69.8 (s), 62.1 (s).

Spectroscopic Data of Dehydroxymethylated Product:



1,8-Diphenoxyoctane (4a-4c): Appeared as white solid, ¹H NMR (500 MHz CDCl₃, δ/ppm): 7.28 (t, *J* = 7.5 Hz, 4H), 6.93 (t, *J* = 7.5 Hz, 2H), 6.90 (d, *J* = 8 Hz, 4H), 3.95 (t, *J* = 6.5 Hz, 4H OC*H*₂), 1.76 – 1.82 (m, 4H), 1.46 – 1.49 (m, 4H), 1.39 – 1.41 (m, 4H); ¹³C {¹H} NMR (125 MHz, CDCl₃)

δ 159.1 (s), 129.3 (s), 120.4 (s), 114.5 (s), 67.8 (s, OCH₂), 29.7 (s), 29.2 (s), 26.0 (s).



1,6-Diphenoxyhexane (4d-4f): Appeared as white solid, ¹H NMR (500 MHz CDCl₃, δ /ppm): 7.29 (t, *J* = 7.5 Hz, 4H), 6.95 (t, *J* = 7.5 Hz, 2H), 6.91 (d, *J* = 8 Hz, 4H), 3.99 (t, *J* = 6 Hz, 4H, OCH₂), 1.81 – 1.85 (m, 4H), 1.55 – 1.58 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 159.0 (s), 129.4 (s), 120.4 (s), 114.5 (s), 67.6 (s, OCH₂), 29.2 (s), 25.8 (s).



1,5-Diphenoxpentane (4g-4i): Appeared as off-white solid, **NMR** (CDCl₃, δ/ppm): ¹H (500 MHz): 7.32 (t, *J* = 8.5 Hz, 4H), 6.98 (t, *J* = 7 Hz, 2H), 6.94 (d, *J* = 8 Hz, 4H), 4.02 (t, *J* = 6.5 Hz, 4H OC*H*₂), 1.88 – 1.93 (m, 4H), 1.68 – 1.73 (m, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 159.0 (s), 129.3 (s), 120.5 (s), 114.4 (s), 67.5 (s, OC*H*₂), 29.0 (s), 22.7 (s).



1,4-Diphenoxybutane (4j-4l): Appeared as White Solid, ¹H NMR (500 MHz CDCl₃, δ/ppm):

7.28 (t, J = 7.5 Hz, 4H), 6.94 (t, J = 7.5 Hz, 2H), 6.90 (d, J = 8 Hz, 4H), 4.04 (t, J = 5.5 Hz, 4H, OCH₂), 1.98 – 2.00 (m, 4H),; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 158.9 (s), 129.4 (s), 120.6 (s), 114.5 (s), 67.3 (s, OCH₂), 26.0 (s).



1,8-Bis(p-tolyloxy)octane (4m) : Appeared as white solid, (88%). ¹H NMR (500 MHz CDCl₃, δ/ppm): 7.07 (d, *J* = 8 Hz, 4H), 6.80 (d, *J* = 8 Hz, 4H), 3.93 (t, *J* = 6.5 Hz, 4H, OC*H*₂), 2.29 (s, 6H), 1.74 - 1.80 (m, 4H), 1.45 - 1.48 (m, 4H), 1.38 - 1.40 (m, 4H); ¹³C {¹H} NMR (125 MHz, CDCl₃) δ 157.0 (s), 129.8 (s), 129.6 (s), 114.3 (s), 67.9 (s, OC*H*₂), 29.3 (s), 25.9 (s), 20.4 (s).



1,6-Bis(p-tolyloxy)hexane (4n): Appeared as white solid, (82%). ¹H NMR (500 MHz CDCl₃, δ/ppm): 7.07 (d, *J* = 8.5 Hz, 4H), 6.80 (d, *J* = 8.5 Hz, 4H), 3.94 (t, *J* = 8.5 Hz, 4H, OC*H*₂), 2.28 (s, 6H), 1.79 - 1.81 (m, 4H), 1.50 - 1.55 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 156.9 (s), 129.8 (s), 129.6 (s), 114.3 (s), 67.9 (s, OC*H*₂), 29.2 (s), 25.8 (s), 20.4 (s).



1,4-Bis(p-tolyloxy)butane (40): Appeared as white solid, (74%). ¹H NMR (500 MHz CDCl₃, δ /ppm): 7.07 (d, J = 8.5 Hz, 4H), 6.80 (d, J = 8.5 Hz, 4H), 4.00 (t, J = 5 Hz, 4H, OCH₂), 2.29 (s, 6H), 1.94 - 1.97 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 156.8 (s), 129.8 (s), 129.7 (s), 114.3 (s), 67.5 (s, OCH₂), 29.0 (s), 20.4 (s).



1,8-Bis(4-methoxyphenoxy)octane (4p): Appeared as off-white solid. ¹H NMR (500 MHz CDCl₃, δ /ppm): 6.83 (s, 8H), 3.91 (t, J = 6.5 Hz, 4H, OCH₂), 3.77 (s, 6H), 1.74 – 1.79 (m, 4H), 1.45 – 1.48 (m, 4H), 1.38 – 1.40 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 153.7 (s), 153.3 (s), 115.4 (s), 114.6 (s), 68.6 (s), 55.7 (s), 29.3 (s), 29.3 (s), 25.9 (s).



1,8-bis(4-chlorophenoxy)octane (4q): Appeared as white solid. ¹H NMR (500 MHz CDCl₃, δ/ppm: 7.29 (d, *J* = 6.5 Hz, 2H), 7.17 (t, *J* = 7.5 Hz, 2H), 6.87 (t, *J* = 7.5 Hz, 2H), 6.83 (d, *J* = 8 Hz, 2H), 4.03 (t, *J* = 5 Hz, 4H, OCH₂), 1.76 - 1.83 (m, 4H), 1.69 - 1.71 (m, 4H), 1.50 - 1.51 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 156.7 (s), 130.4 (s), 127.8 (s), 127.7 (s), 120.3 (s), 111.3 (s), 67.7 (s), 29.0 (s), 27.7 (s), 25.1 (s). 23.2 (s).



1,8-bis(ortho-bromophenoxy)octane (4r): Appeared as white solid. ¹H NMR (500 MHz CDCl₃, δ/ppm): 7.53 (dd, *J* = 12 Hz, *J* = 2 Hz, 2H), 7.22 – 7.26 (m, 2H), 6.88 (dd, *J* = 10.5 Hz, 1.5 Hz, 2H), 6.81 (td, *J* = 9 Hz, *J* = 2 Hz, 2H), 4.02 (t, *J* = 8 Hz, 4H, OCH₂), 1.81- 1.88 (m, 4H), 1.49 - 1.55 (m, 4H), 1.40 - 1.44 (m, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 155.4 (s), 133.2 (s), 128.3 (s), 121.5 (s), 113.2 (s), 112.2 (s), 69.0 (s, OCH₂), 29.1 (s), 29.0 (s), 25.9 (s).



1,4-bis(phenoxymethyl)benzene (4s): Appeared as white solid. ¹H NMR (500 MHz CDCl₃, δ /ppm):7.46 (s, 4H), 7.28 - 7.31 (m, 5H), 6.96 - 6.99 (m, 5H), 5.08 (s, 4H, *OCH*₂); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 158.7 (s), 136.8 (s), 129.4 (s), 127.6 (s), 121.0 (s), 114.9 (s), 69.6 (s, OCH₂).



4,4'-bis(phenoxymethyl)-1,1'-biphenyl (4t): Appeared as white solid. ¹H NMR (500 MHz CDCl₃, δ /ppm): 7.63 (d, *J* = 8 Hz, 4H), 7.53 (d, *J* = 8 Hz, 4H), 7.33 (t, *J* = 7.5 Hz, 4H), 7.03 (d, *J* = 8 Hz, 4H), 7.00 (t, *J* = 7.5 Hz, 2H), 5.13 (s, 4H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 158.7 (s), 140.4 (s), 136.2 (s), 129.4 (s), 127.9 (s), 127.2(s), 120.9 (s), 114.8 (s), 69.6 (s).

Copies of spectral data







Figure s4. ¹H NMR spectrum of C1.



Figure s6. HRMS of C1.



Figure s8. ¹H NMR spectrum of 3m.































































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Optimized cartesian coordinates (in angstrom units):

C1-R	R (E= -2186.5	707735 au)	
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