

## Supporting Information

### Efficient and zero thermal quenching Sm<sup>3+</sup>-activated

### Li<sub>2</sub>LaTiTaO<sub>7</sub> phosphor for white LEDs

Qi Wang<sup>a,b,c</sup>, Shangwei Wang<sup>b,c</sup>, Ran Pang<sup>c\*</sup>, Taixing Tan<sup>b</sup>, Herui Wen<sup>a</sup>, Su Zhang<sup>c</sup>,  
Hongpeng You<sup>a,b,c\*</sup>, Chengyu Li<sup>a,b,c</sup>, Hongjie Zhang<sup>c</sup>

*\* Corresponding author.*

*E-mail addresses:* pangran@ciac.ac.cn (R. Pang), hpyou@ciac.ac.cn (H. You)

<sup>a</sup> *Faculty of Materials Metallurgy and Chemistry, Jiangxi University of Science and Technology, Ganzhou, Jiangxi 341000, P. R. China*

<sup>b</sup> *Ganjiang Innovation Academy, Chinese Academy of Sciences, Ganzhou 341000, P. R. China*

<sup>c</sup> *State key Laboratory of Rare Earth Resource Utilization, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, Changchun 130022, P. R. China*

## Figures and Figure Captions

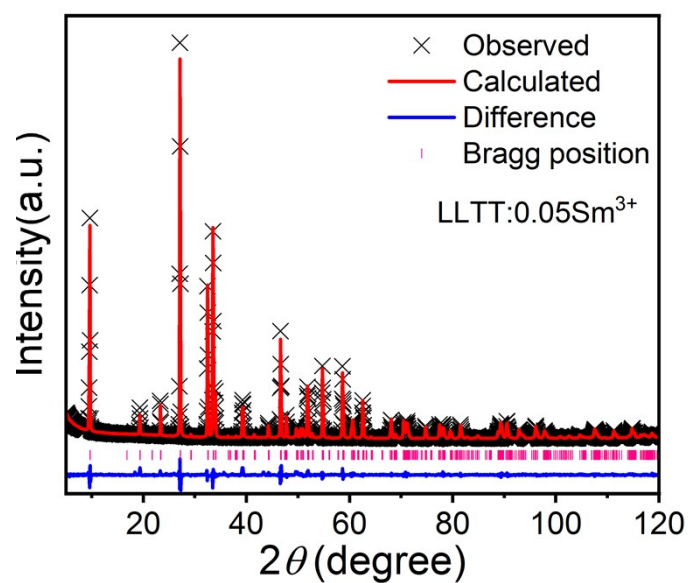


Fig. S1 Rietveld refinement of LLTT:0.05Sm<sup>3+</sup> sample.

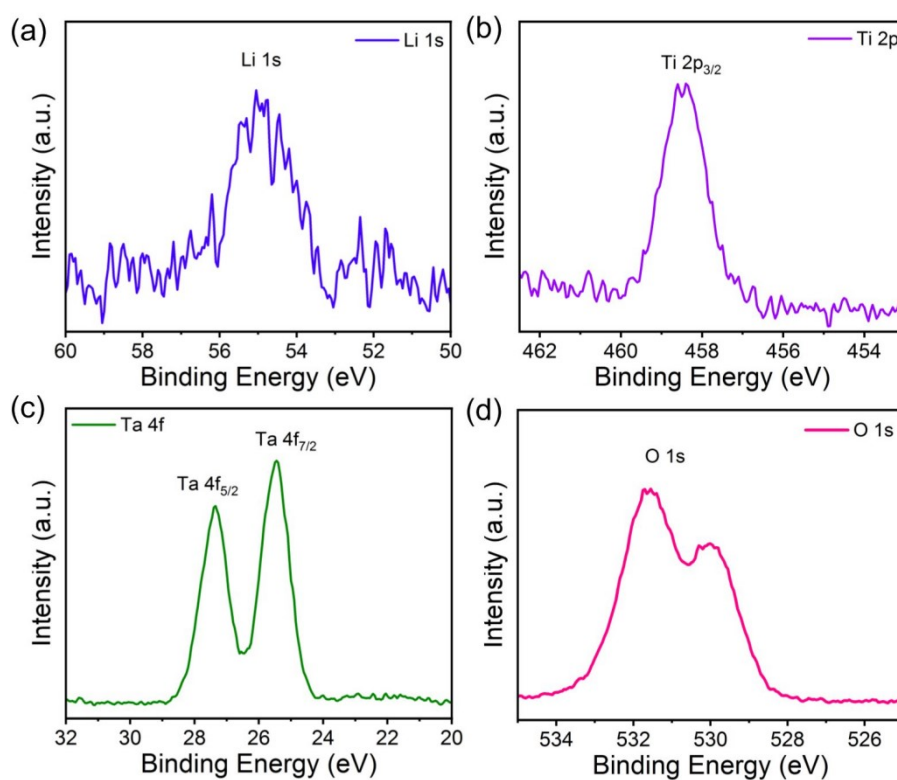


Fig. S2 (a)-(d) high-resolution XPS spectra of Li 1s, Ti 2p, Ta 4f, and O 1s, respectively.

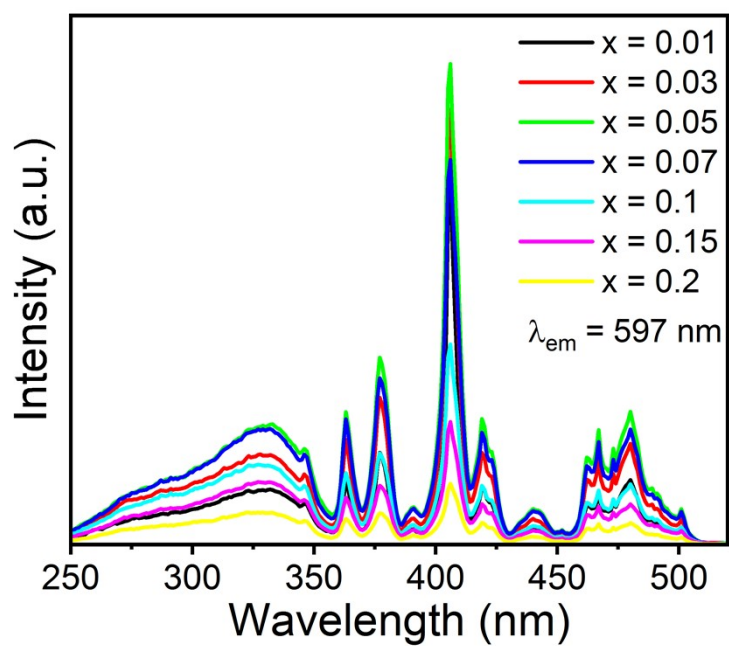


Fig. S3 (a) PLE spectra of LLTT: $x\text{Sm}^{3+}$  samples with varied  $x$  values monitored at 597 nm.

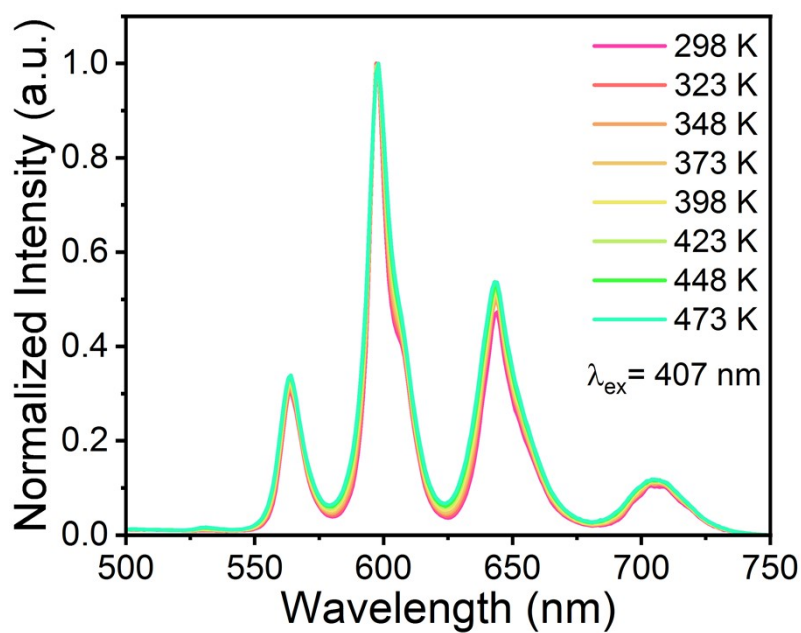


Fig. S4 Temperature-dependent Normalized PL spectra of the LLTT:0.05 $\text{Sm}^{3+}$  sample (298-473 K).

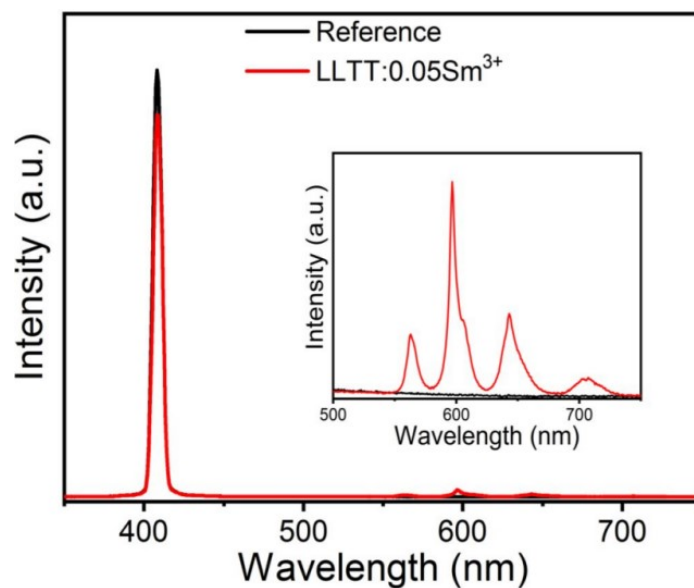


Fig. S5 The spectra of the LLTT:0.05Sm<sup>3+</sup> sample and the reference sample ( $\lambda_{\text{exc}} = 407 \text{ nm}$ ) in quantum efficiency measurements.

### Tables and Table Captions

Table S1 Rietveld refinement and crystallographic data for the LLTT and LLTT:0.05Sm<sup>3+</sup>.

Formula	LLTT	LLTT:0.05Sm <sup>3+</sup>
Space group	<i>Cmcm</i> (No.63)	<i>Cmcm</i> (No.63)
Symmetry	Orthorhombic	Orthorhombic
Units, <i>Z</i>	4	4
<i>a</i> (Å)	18.3154(3)	18.3343(1)
<i>b</i> (Å)	5.5141(1)	5.5066(2)
<i>c</i> (Å)	5.5057(1)	5.5043(3)
<i>V</i> (Å <sup>3</sup> )	556.04(1)	555.72(1)
<i>R</i> <sub>p</sub> (%)	4.8	5.23
<i>R</i> <sub>wp</sub> (%)	6.54	7.13
$\chi^2$	3.573	3.763

Table S2 Structural parameters of the LLTT determined by Rietveld refinement of the corresponding XRD pattern.

Atom	Wyck	$x/a$	$y/b$	$z/c$	Occ	$U_{iso}$
La1	4c	0.11656(8)	0.7548(8)	1/4	1.0	0.0200
Ta1	8g	0.11656(8)	0.7548(8)	1/4	0.5	0.0187
Ti1	8g	0.2824(22)	0	0	0.5	0.0187
Li1	8e	0	0.6970(34)	1/4	1.0	0.0220
O1	4c	0.2126(6)	0.7852(33)	1/4	1.0	0.0013
O2	8g	0.6149(8)	0	0	1.0	0.0129
O3	8e	0.0964(8)	0	0	1.0	0.0171
O4	8e	0.11656(8)	0.7548(8)	1/4	1.0	0.0338

Table S3 Structural parameters of the LLTT:0.05Sm<sup>3+</sup> determined by Rietveld refinement of the corresponding XRD pattern.

Atom	Wyck	$x/a$	$y/b$	$z/c$	Occ	$U_{iso}$
La1	4c	0	0.2598(10)	1/4	0.95	0.0166
Ta1	8g	0.11691(9)	0.7542(8)	1/4	0.5	0.0143
Ti1	8g	0.11691(9)	0.7542(8)	1/4	0.5	0.0143
Li1	8e	0.2908(22)	0	0	1.0	0.0042
O1	4c	0	0.713(6)	1/4	1.0	0.0173
O2	8g	0.2141(6)	0.786(4)	1/4	1.0	0.0061
O3	8e	0.6190(8)	0	0	1.0	0.0101
O4	8e	0.1015(10)	0	0	1.0	0.0185
Sm1	4c	0	0.2598(10)	1/4	0.05	0.0166

Table S4 Average lifetime ( $\tau$ ) of Sm<sup>3+</sup> 597 nm emission in LLTT: xSm<sup>3+</sup> phosphors  
(under 407 nm excitation)

Sm <sup>3+</sup> (x)	$A_1$	$\tau_1$ (ms)	$A_2$	$\tau_2$ (ms)	$\tau_{av}$ (ms)
0.01	3956.21	1.79	981.84	0.9	1.69
0.03	3374.84	1.54	1436.92	0.61	1.41
0.05	2912.57	1.36	2056.93	0.54	1.18
0.07	2529.63	1.19	2415.97	0.45	0.99
0.1	1845.5	1.06	3219.24	0.42	0.8
0.15	1614.02	0.76	3273.32	0.31	0.56
0.2	2094.91	0.5	2662.12	0.2	0.4

## Supplementary Notes

### Supplementary Note 1:

The percentage difference in radius ( $D_r$ ) should be explored to further ensure that the Sm<sup>3+</sup> ion may be readily doped into the LLTT host, which could be calculated using the following formula:<sup>1</sup>

$$D_r = \frac{R_l(\text{CN}) - R_2(\text{CN})}{R_l(\text{CN})} \quad (\text{S1})$$

where the radius of the activated Sm<sup>3+</sup> ions ( $R_2$  (CN)) and the potential superseded La<sup>3+</sup> ions ( $R_l$  (CN)) are 1.24 and 1.36 Å, respectively, when the coordination number (CN) is 12.

## Supplementary References

1. M. Peng, Z. Pei, G. Hong and Q. Su, *J. Mater. Chem.*, 2003, **13**, 1202-1205.