

Electronic Supporting Information

Tetracobalt(II) cluster with two vertex truncated dicubane topology endogenously supported by carboxylate-based (2-pyridyl)methylamine ligands: magneto-structural and DFT study

Shashi Kant, Sayan Saha, Francesc Lloret, Joan Cano* and Rabindranath Mukherjee*

Figures

Fig. S1 IR spectrum of $[\text{Co}^{\text{II}}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ (**1**).

Fig. S2 ^1H NMR spectrum of **1** in CD_3OD .

Fig. S3 Crystal packing diagram, showing $\text{O-H}\cdots\text{O}$, $\text{C-H}\cdots\text{O}$, $\text{C-H}\cdots\text{Cl}$ and $\text{C-H}\cdots\pi$ interactions of $[\text{Co}^{\text{II}}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ (**1**).

Fig. S4 XPRD spectra of $[\text{Co}^{\text{II}}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ (**1**).

Fig. S5 UV-VIS spectrum of $[\text{Co}^{\text{II}}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ (**1**) in $\text{CH}_3\text{CN-MeOH}$ (2:1; v/v).

Tables

Table S1. Data Collection and Structure Refinement Parameters for $[\text{Co}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ (**1**)

Table S2. Hydrogen-bonding parameters in **1**

Table S3. Calculated spin configurations and their relative energies as a function of different J_i constants

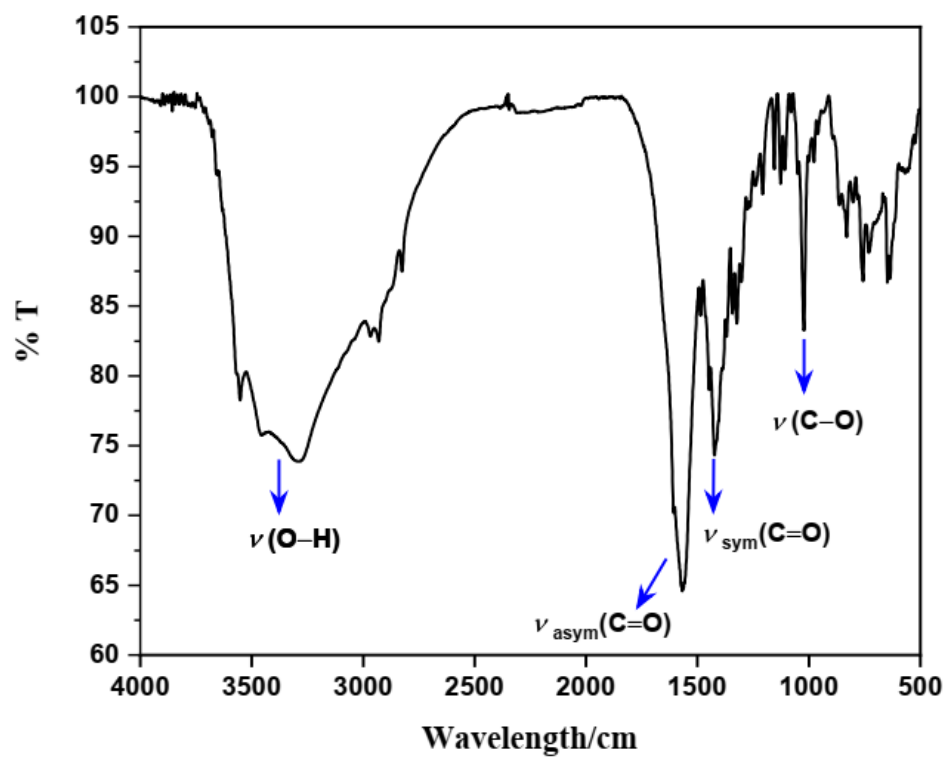


Fig. S1 FT-IR spectrum of $[\text{Co}^{\text{II}}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ (**1**).

Co Tetramer

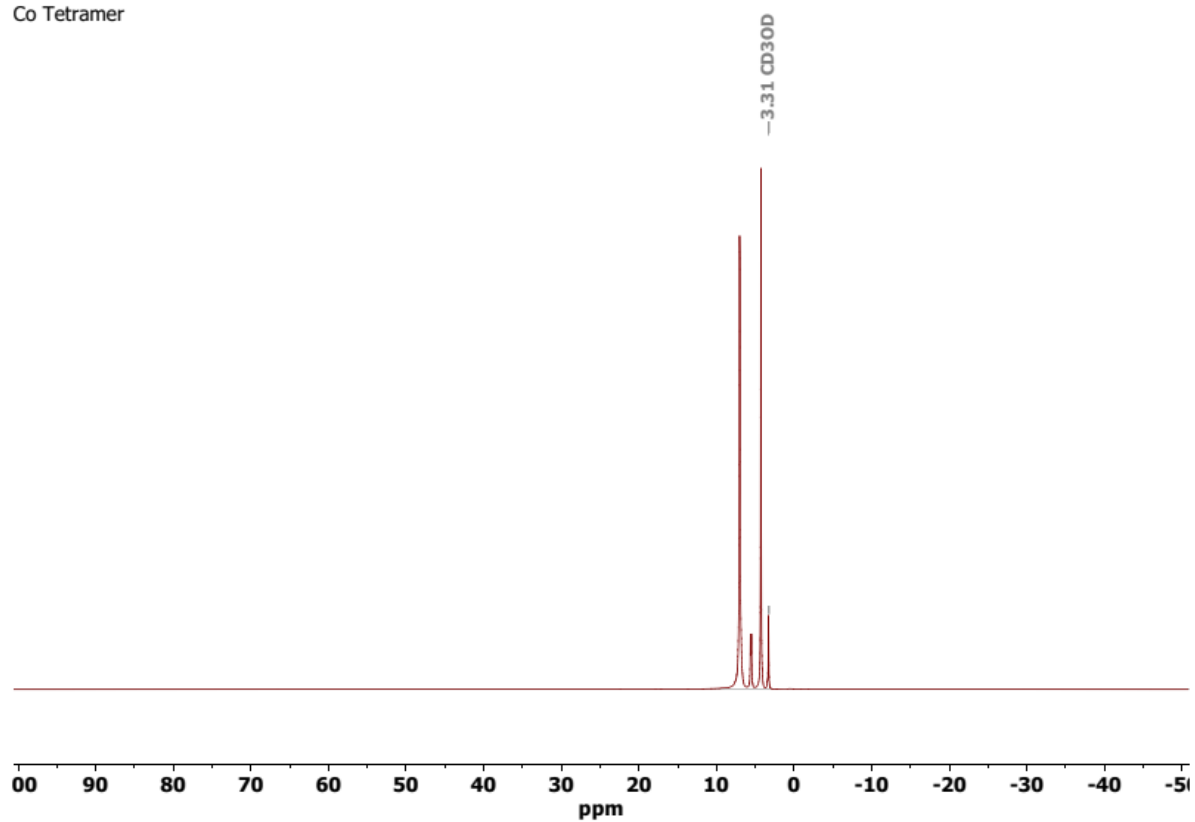
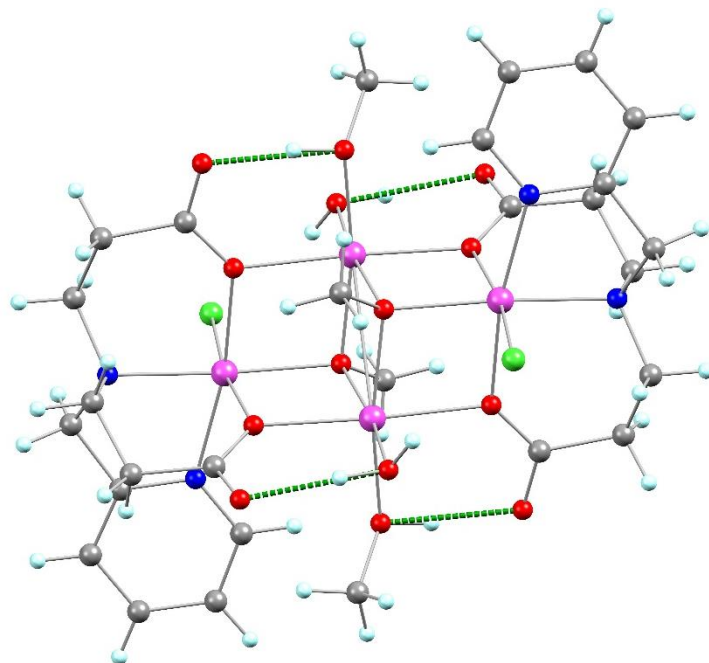


Fig. S2 ^1H NMR spectrum of **1** in CD_3OD .

(a)



(b)

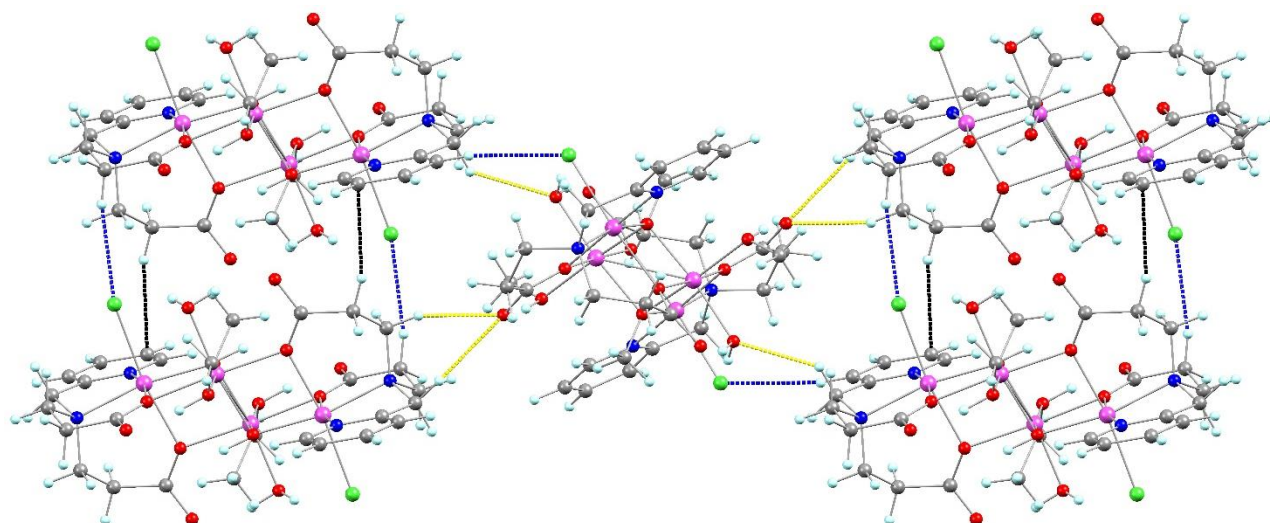


Fig. S3 Crystal packing diagram, (a) showing intramolecular O–H...O (in green) and (b) intermolecular C–H...O (in yellow) and C–H...Cl (in blue) C–H... π (in black) interactions of $[\text{Co}^{\text{II}}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ (**1**).

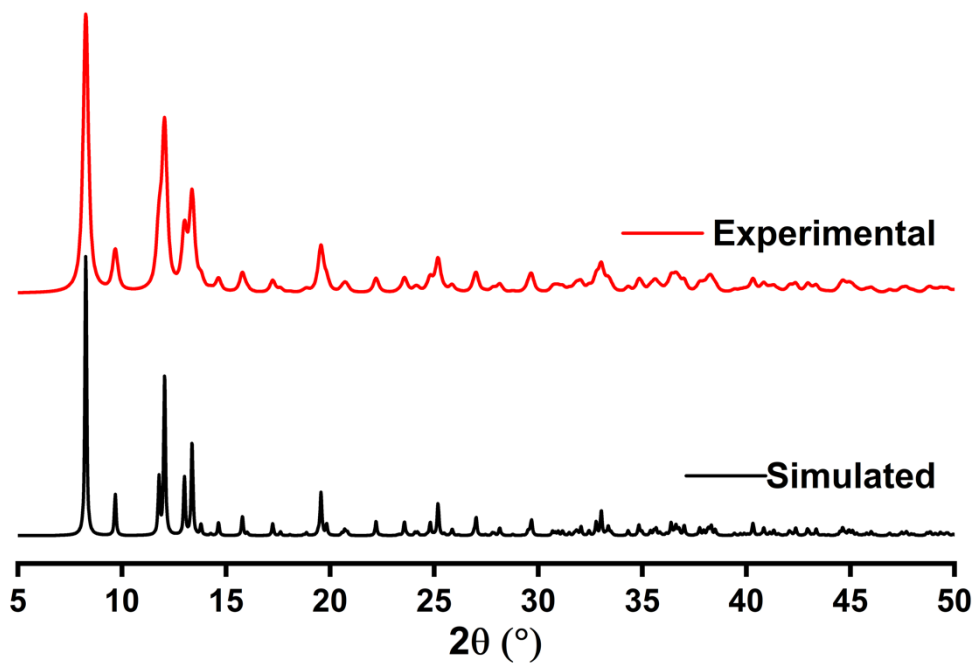


Fig. S4 XPRD spectra of $[\text{Co}^{\text{II}}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ (**1**).

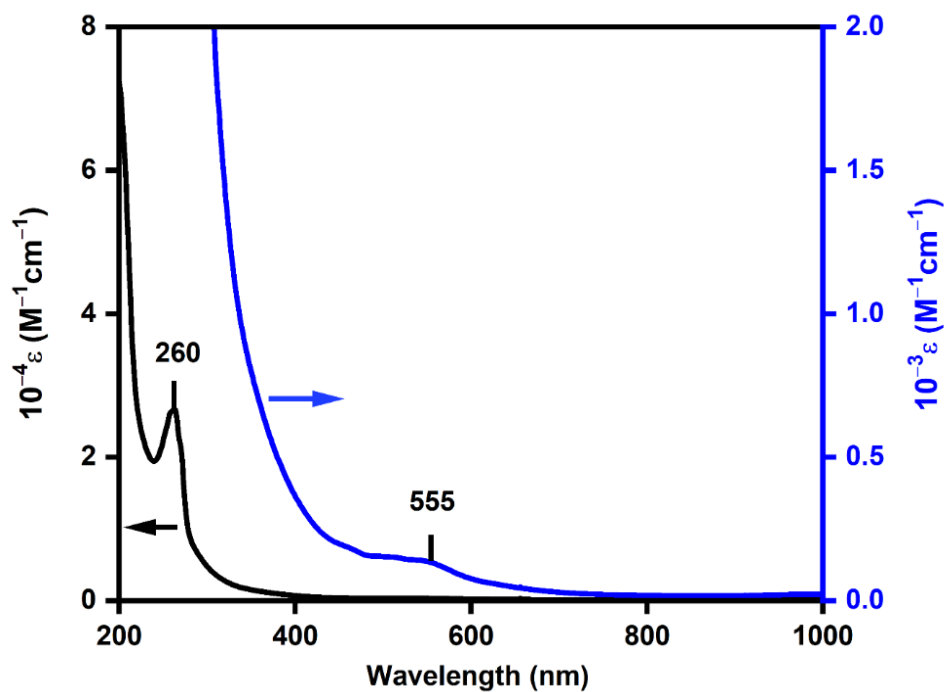


Fig. S5 UV-Vis spectrum of $[\text{Co}^{\text{II}}_4\{\text{L}^3\text{-(CO}_2^-)_2\}_2(\mu_3\text{-OCH}_3)_2(\text{CH}_3\text{OH})_2(\text{H}_2\text{O})_2\text{Cl}_2]$ **1** in $\text{CH}_3\text{CN-CH}_3\text{OH}$ (2:1; v/v).

Table S1. Data Collection and Structure Refinement Parameters for [Co₄{L³-(CO₂⁻)₂}(μ₃-OCH₃)₂(CH₃OH)₂(H₂O)₂Cl₂] (**1**)

1	
Formula	C ₂₈ H ₄₆ Cl ₂ Co ₄ N ₄ O ₁₄
Formula weight	969.31
Crystal colour, habit	red, block
<i>T</i> / K	100(2)
Crystal system	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>n</i> (no. 14)
<i>a</i> /Å	7.694(5)
<i>b</i> /Å	13.188(5)
<i>c</i> /Å	18.365(5)
<i>α</i> /°	90.0
<i>β</i> /°	96.099(5)
<i>γ</i> /°	90.0
<i>V</i> /Å ³	1852.9(15)
<i>Z</i>	2
<i>D_c</i> /g cm ⁻³	1.737
<i>μ</i> /mm ⁻¹	1.973
Reflections measured	9843
Unique reflections/ <i>R</i> _{int}	3436/0.0383
Reflections used <i>I</i> > 2σ(<i>I</i>)	2712
<i>R</i> ₁ ^{<i>a</i>} , <i>wR</i> ₂ ^{<i>b</i>} [<i>I</i> > 2σ(<i>I</i>)	<i>R</i> ₁ = 0.0404 ^{<i>a</i>} <i>wR</i> ₂ = 0.0975 ^{<i>b</i>}
<i>R</i> ₁ ^{<i>a</i>} , <i>wR</i> ₂ ^{<i>b</i>} (all data)	<i>R</i> ₁ = 0.0546 ^{<i>a</i>} <i>wR</i> ₂ = 0.1075 ^{<i>b</i>}
GOF on <i>F</i> ²	1.037

$${}^a R_1 = \Sigma(|F_o| - |F_c|)/\Sigma|F_o|. \quad {}^b wR_2 = \{\Sigma[w(|F_o|^2 - |F_c|^2)^2] / \Sigma[w(|F_o|^2)^2]\}^{1/2}$$

Table S2. Hydrogen-bonding parameters in **1**

D–H \cdots A	H \cdots A, Å	D \cdots A, Å
O7–H7 \cdots O4	1.646	2.563
O6–H6D \cdots O2	1.820	2.690
C11–H11B \cdots Cl1	2.852	3.816
C4–H4 \cdots Cl1	2.834	3.736
C6–H6A \cdots O6	2.539	3.273
C7–H7A \cdots O4	2.437	3.382
C6–H6B \cdots O4	2.690	3.542
C8–H8B \cdots π	2.753	3.715

Table S3 Calculated spin configurations and their relative energies as a function of different J_i constants. The spin configuration used as a reference is that with the maximum multiplicity generated from the parallel alignment of all local spin moments of the Co^{II} ions. Only the centres with an antiparallel (negative) alignment of their spin moment are noted.

Spin conf.	S	J_1	J_2	J_3	Energy ^a
{Co2,Co2 [#] }	0	12	12	0	–52.53
{Co1 [#] ,Co2}	0	12	0	6	+78.94
{Co1 [#] ,Co2 [#] }	0	0	12	6	–1.55
{Co2 [#] }	3	6	6	6	+38.57
{Co1}	3	6	6	0	–26.10

^aIn cm^{–1}.