

Supporting Information for

Rational design, crystal structure, and frustrated magnetism of the Ge-containing YbFe₂O₄-type layered oxides In₂Zn_{3-x}Co_xGeO₈ (0 ≤ x ≤ 3)

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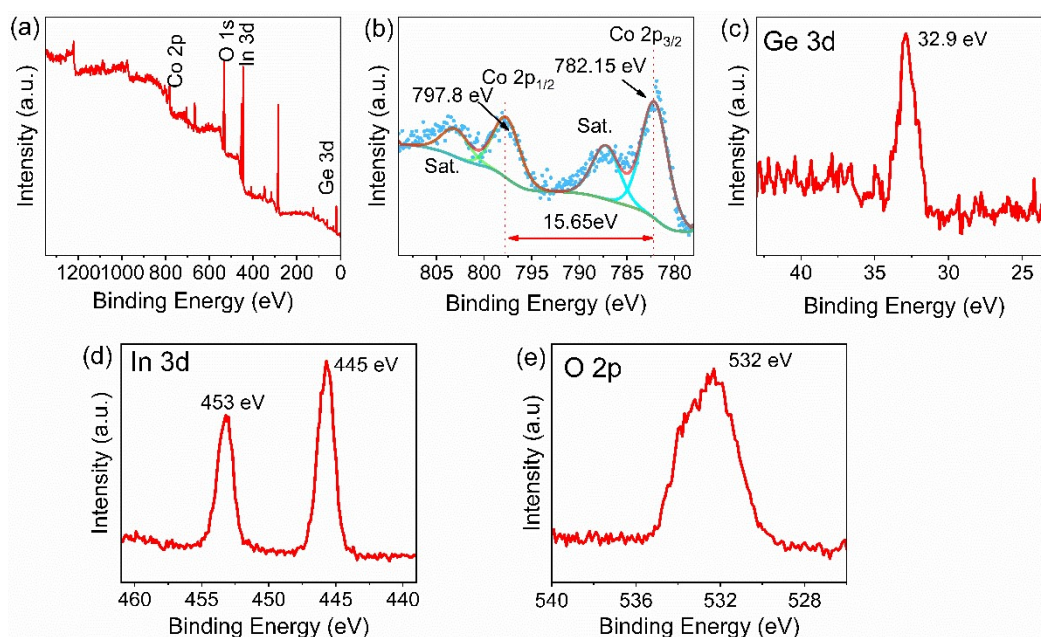


Fig. S1 (a) The survey XPS of In₂Co₃GeO₈. XPS spectra of Co 2p (b), Ge 3d (c), In 3d (d), and O 2p (e).

Table S1. Crystallographic data for $\text{In}_3\text{Zn}_{4.5}\text{Ge}_{1.5}\text{O}_{12}$, $\text{In}_3\text{Zn}_3\text{Co}_{1.5}\text{Ge}_{1.5}\text{O}_{12}$, $\text{In}_3\text{Zn}_{1.5}\text{Co}_3\text{Ge}_{1.5}\text{O}_{12}$ and $\text{In}_3\text{Co}_{4.5}\text{Ge}_{1.5}\text{O}_{12}$, corresponding to $\text{In}_2\text{Zn}_{3-x}\text{Co}_x\text{GeO}_8$ with $x = 0, 1, 2,$ and $3,$ respectively.

parameters				
temperature (K)	300			
formula	$\text{In}_3\text{Zn}_{4.5}\text{Ge}_{1.5}\text{O}_{12}$	$\text{In}_3\text{Zn}_3\text{Co}_{1.5}\text{Ge}_{1.5}\text{O}_{12}$	$\text{In}_3\text{Zn}_{1.5}\text{Co}_3\text{Ge}_{1.5}\text{O}_{12}$	$\text{In}_3\text{Co}_{4.5}\text{Ge}_{1.5}\text{O}_{12}$
formula weight (g mol ⁻¹)	939.747	930.034	920.320	910.606
crystal system	Trigonal			
Space group (No.)	$R\bar{3}m$ (166)			
wavelength (Å)	1.54059			
d spacing range	0.82–8.8	0.82–8.8	0.82–8.8	0.82–8.8
density (g cm ⁻³)	6.300(4)	6.216(3)	6.154(4)	6.113(6)
unit cell dimensions	$a = b =$ 3.2989(1) $c = 26.3198(7)$	$a = b =$ 3.30826(7) $c = 26.2143(6)$	$a = b =$ 3.31488(6) $c = 26.0974(5)$	$a = b =$ 3.3159 (1) $c = 25.9718(5)$
V	248.06(2)	248.47(1)	248.47(1)	247.30(2)
Z	1	1	1	1
no. of reflections	85	85	85	85
no. of parameters	36	42	42	36
no. of constrains	6	10	10	6
R_{wp}	7.95%	2.68%	2.12%	2.23
R_{p}	5.89%	1.94%	1.26%	1.75
goodness of fit	1.78	1.40	2.36	1.12

Table S2. Atomic coordinates, site occupancy factors (*sof.*), and isotropic thermal displacement factors for $\text{In}_2\text{Zn}_{3-x}\text{Co}_x\text{GeO}_8$ ($x = 1$ and 2) obtained from Rietveld refinements against PXRD data.

$x = 1$	Site	x	y	z	<i>sof.</i>	$B_{\text{iso.}}$ (\AA^2)
In1	3a	0	0	0	0.897(1)	0.14(1)
Zn1	3a	0	0	0	0.069(1)	0.14(1)
Co1	3a	0	0	0	0.034(1)	0.14(1)
Zn2	6c	2/3	1/3	0.11563(2)	0.4656(5)	0.82(2)
In2	6c	2/3	1/3	0.11563(2)	0.0516(7)	0.82(2)
Co2	6c	2/3	1/3	0.11563(2)	0.2328(2)	0.82(2)
Ge1	6c	2/3	1/3	0.11563(2)	0.25	0.72(5)
O1	6c	0	0	0.13035(9)	1	1.30(8)
O2	6c	2/3	1/3	0.0388(1)	1	0.47(6)
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$x = 2$						
In1	3a	0	0	0	0.935(2)	0.27(2)
Zn1	3a	0	0	0	0.022(1)	0.27(2)
Co1	3a	0	0	0	0.043(2)	0.27(2)
Co2	6c	2/3	1/3	0.11598(3)	0.4784(5)	0.58(3)
In2	6c	2/3	1/3	0.11598(3)	0.0326(8)	0.58(3)
Zn2	6c	2/3	1/3	0.11598(3)	0.2390(3)	0.58(3)
Ge1	6c	2/3	1/3	0.11598(3)	0.25	0.58(3)
O1	6c	0	0	0.1308(1)	1	1.3(1)
O2	6c	2/3	1/3	0.0403(2)	1	0.30(9)

Table S3. Selected interatomic bond lengths of $\text{In}_2\text{Zn}_{3-x}\text{Co}_x\text{GeO}_8$ ($0 \leq x \leq 3$).

$\text{In}_2\text{Zn}_3\text{GeO}_8$			
In-O2 ($\times 6$)	2.170(2)	<i>M</i> -O2	1.997(3)
<i>M</i> -O1 ($\times 3$)	1.9358(6)	<i>M</i> -O1	2.356(3)
$\text{In}_2\text{Zn}_2\text{CoGeO}_8$			
In-O2 ($\times 6$)	2.164(2)	<i>M</i> -O2	2.014(3)
<i>M</i> -O1 ($\times 3$)	1.9486(5)	<i>M</i> -O1	2.290(2)
$\text{In}_2\text{ZnCo}_2\text{GeO}_8$			
In-O2 ($\times 6$)	2.184(2)	<i>M</i> -O2	1.975(4)
<i>M</i> -O1 ($\times 3$)	1.9523(7)	<i>M</i> -O1	2.260(3)
$\text{In}_2\text{Co}_3\text{GeO}_8$			
In-O2 ($\times 6$)	2.205(4)	<i>M</i> -O2	1.917(8)
<i>M</i> -O1 ($\times 3$)	1.964(1)	<i>M</i> -O1	2.194(5)