

Support Information For

SiC₃N₃ Monolayer as a Universal Anode for Alkali Metal-ion Batteries

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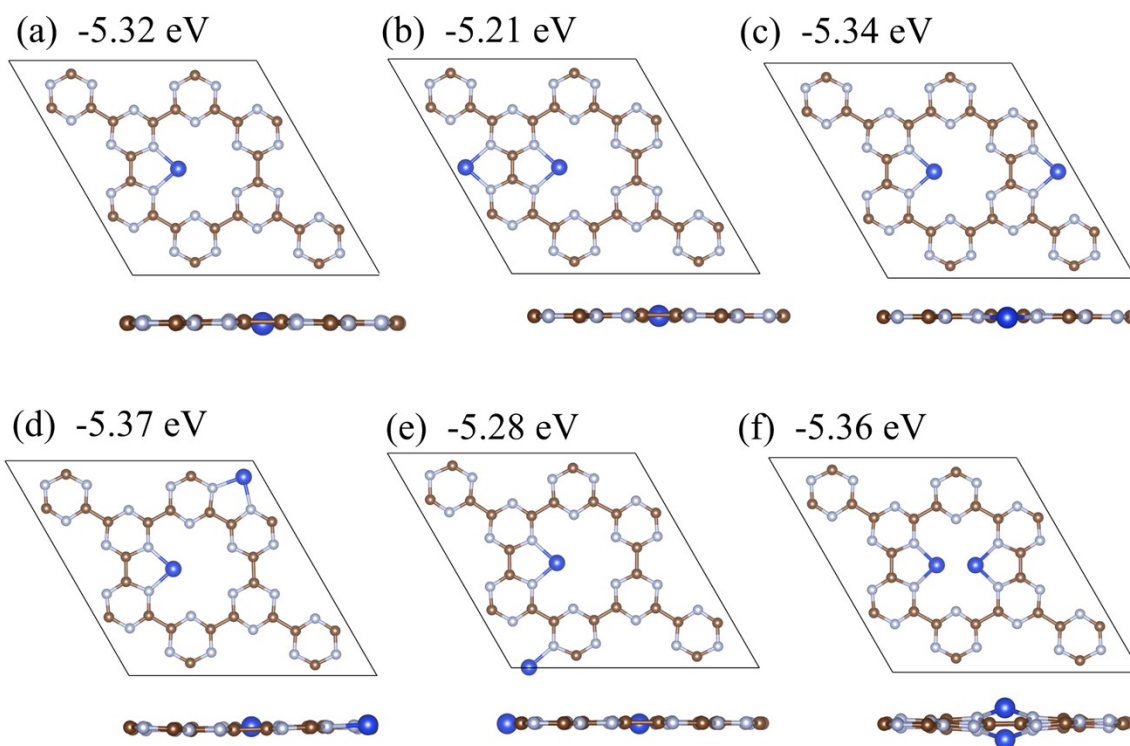


Fig. S1 The highly symmetric configurations of (a) $\text{SiC}_{24}\text{N}_{24}$ and (c-f) $\text{SiC}_{12}\text{N}_{12}$ and their corresponding formation energies.

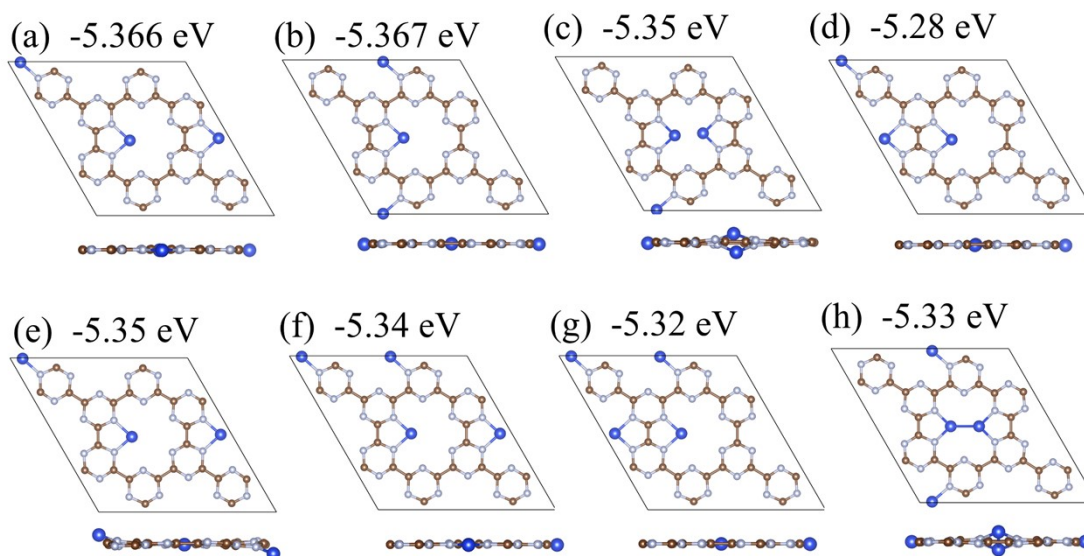


Fig. S2 The highly symmetric configurations of (a-e) SiC_8N_8 and (f-h) SiC_6N_6 and their corresponding formation energies.

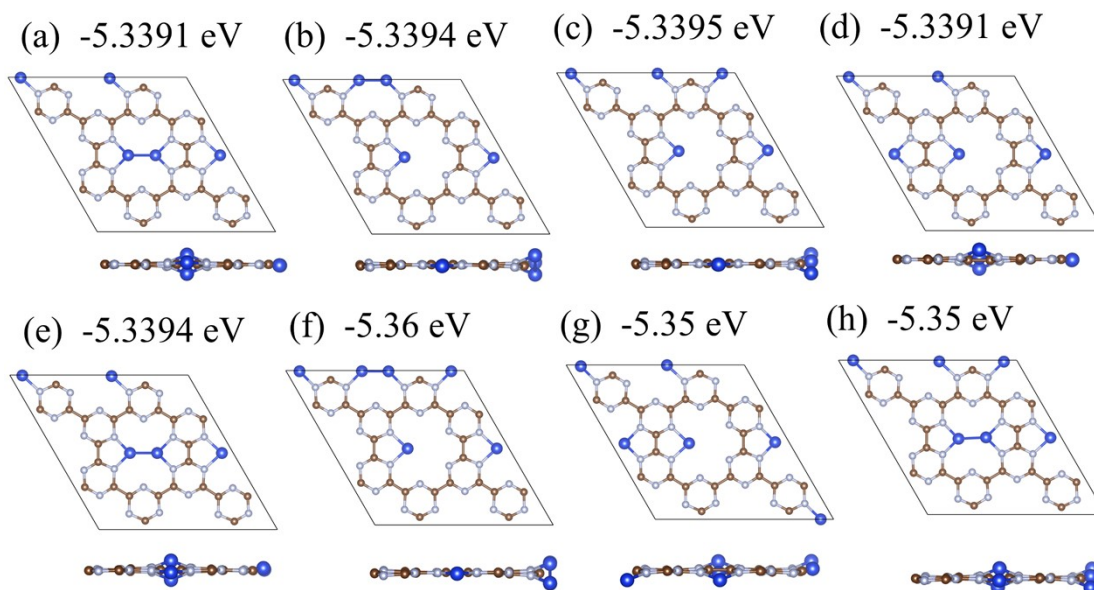


Fig. S3 The highly symmetric configurations of (a-e) $\text{Si}_5\text{C}_{24}\text{N}_{24}$ and (f-h) SiC_4N_4 and their corresponding formation energies.

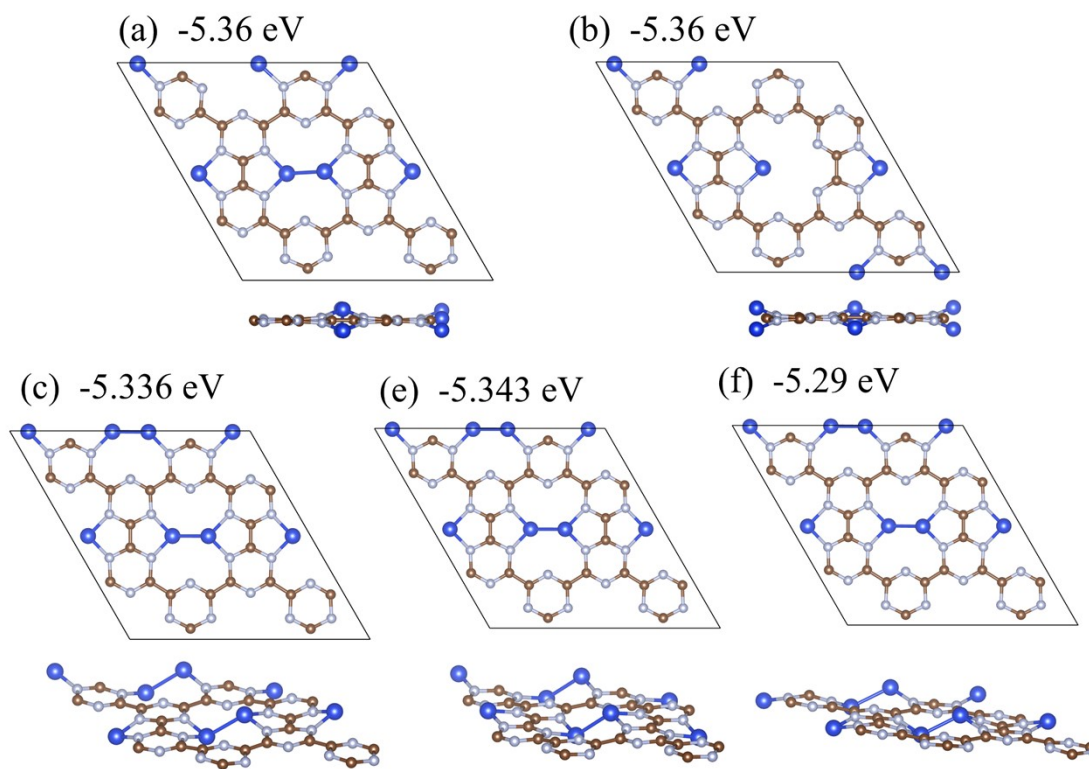


Fig. S4 The highly symmetric configurations of (a-b) $\text{Si}_7\text{C}_{24}\text{N}_{24}$ and (c-f) SiC_3N_3 and their corresponding formation energies.

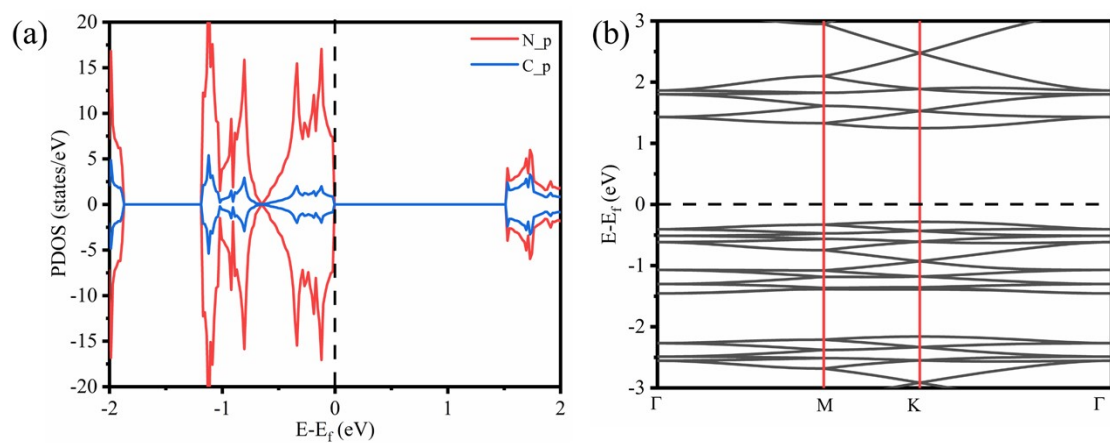


Fig. S5 (a) Partial density of states and (b) energy bands of g-CN.

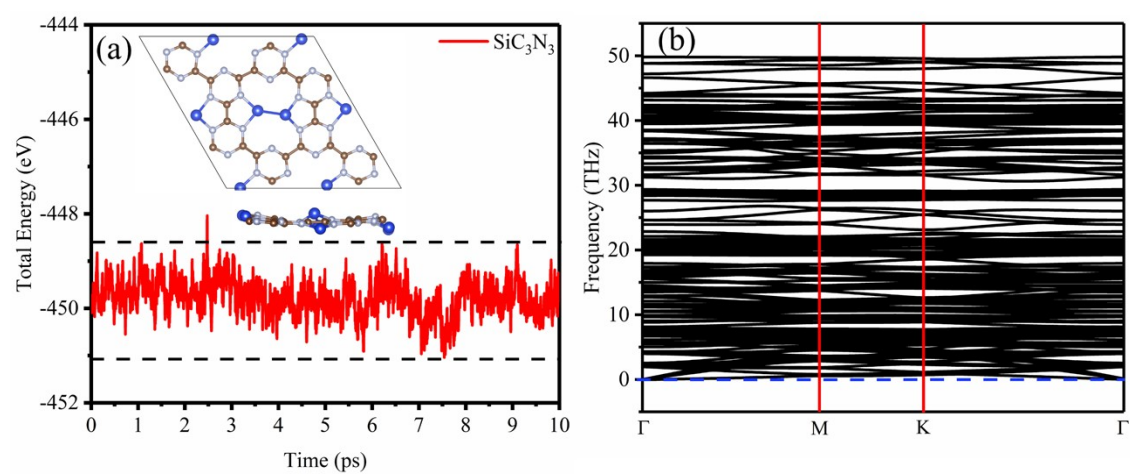


Fig. S6 (a) The fluctuation of total-energies, top and side-snapshots of SiC_3N_3 monolayer after AIMD simulation for 10 ps at 300 K. (b) Computed phonon dispersions of SiC_3N_3 .

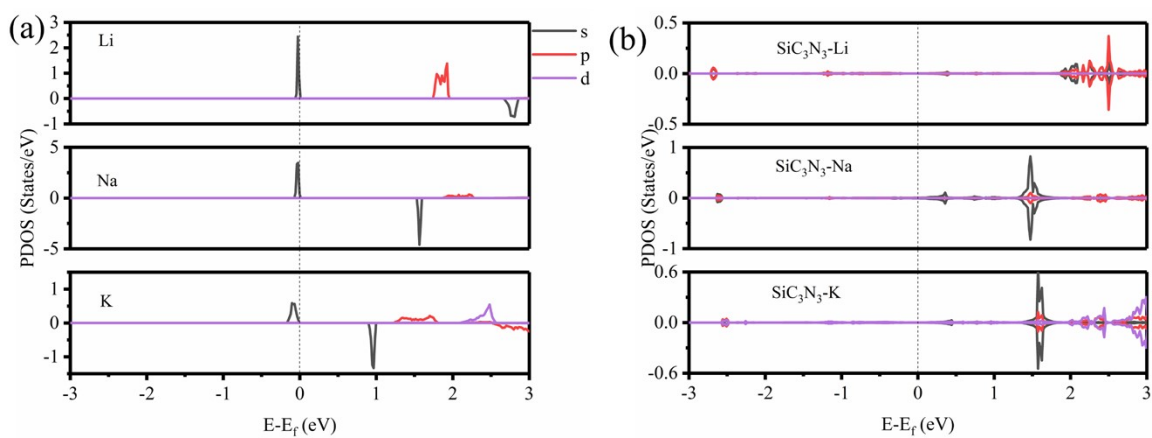


Fig. S7 The partial density of states of isolated Li, Na, and K. (b) The partial density of states of one Li, Na, and K adsorbed on SiC_3N_3 .

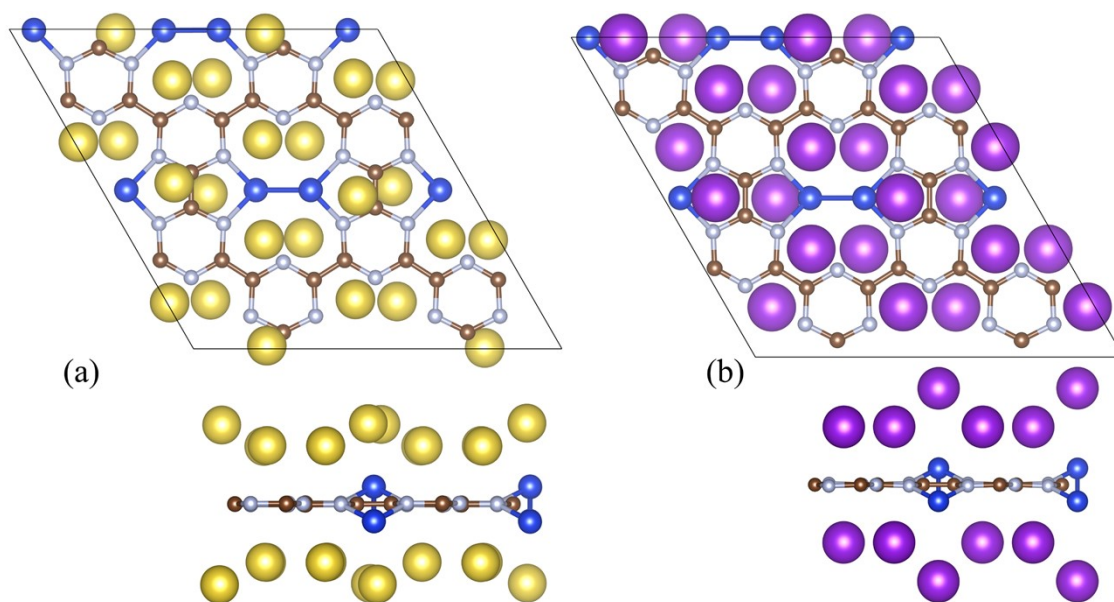


Fig. S8 The most stable configurations of (a) $\text{Na}_3\text{SiC}_3\text{N}_3$ and (b) $\text{K}_3\text{C}_3\text{N}_3$.

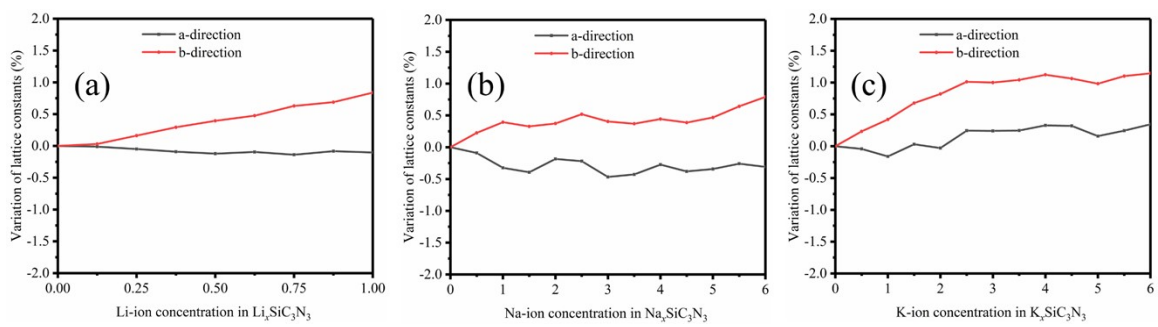


Fig. S9 Variation of lattice constants at different (a) Li, (b) Na and (c) K adsorption concentrations.