Supplementary information for

## Exclusive confinement of Bi<sup>3+</sup>-activators in the triangular prism enabling efficient and thermally stable green emission in tridymite-type phosphor CaBaGa<sub>4</sub>O<sub>8</sub>:Bi<sup>3+</sup>

Jie Qin,<sup>a</sup> Pengfei Jiang,<sup>\*,a</sup> Rihong Cong and Tao Yang<sup>\*,a</sup>

<sup>a</sup> College of Chemistry and Chemical Engineering, Chongqing University, Chongqing 401331, China

\*Corresponding authors: pengfeijiang@cqu.edu.cn, taoyang@cqu.edu.cn.



**Fig. S1** Plots of lattice parameters of CaBaGa<sub>4</sub>O<sub>8</sub>: $xBi^{3+}$  as a function of  $Bi^{3+}$ -content (*x*).



Fig. S2 PLE spectra of CaBaGa<sub>4</sub>O<sub>8</sub>: $xBi^{3+}$  recorded with  $\lambda_{em} = 530$  nm.



Fig. S3 XRPD patterns for  $Ca_{1-p}Sr_pBaGa_4O_8:0.007Bi^{3+}$  ( $C_{1-p}S_pBGO:Bi^{3+}$ ) (a) and  $CaBaGa_{4-q}A_qO_8:0.007Bi^{3+}$  ( $CBG_{4-q}A_qO:Bi^{3+}$ ) (b).



Fig. S4 PLE and PL spectra for  $C_{1-p}S_pBGO:Bi^{3+}$  (a, b) and  $CBG_{4-q}A_qO:Bi^{3+}$  (c, d).



Fig. S5 PLE spectra of CBGO:Bi<sup>3+</sup>,  $yZn^{2+}$  recorded with  $\lambda_{em} = 530$  nm.



**Fig. S6** The optical spectra of CBGO: $0.007Bi^{3+}$ ,  $yZn^{2+}$  with y = 0 and 0.003 phosphors recorded by the quantum efficiency system under the excitation wavelength of 340 nm. The black lines in the figures represents the reflectance of the BaSO<sub>4</sub> standard.

**Table S1** Atomic coordinates, site occupancy factor (*sof.*), isotropic thermal displacement factorsfor CaBaGa<sub>4</sub>O<sub>8</sub>:0.01Bi<sup>3+</sup> and CaBaGa<sub>4</sub>O<sub>8</sub>:0.007Bi<sup>3+</sup>, 0.003Zn<sup>2+</sup> obtained from Rietveld refinementsagainst high-resolution Cu K<sub> $\alpha$ 1</sub> XRPD data.

$CaBaGa_4O_8:0.01Bi^{3+}$								
Atom	site	x	у	Ζ	sof.	Beq. (Å <sup>2</sup> )		
Ca	2a	0	0.5	0.4258(5)	0.99	1.3(1)		
Bi	2a	0	0.5	0.4258(5)	0.01	1.3(1)		
Ba	2a	0	0	0	1	0.98(3)		
Ga	8e	0.29176(9)	0.16730(9)	0.4516(3)	1	0.64(2)		
01	8e	0.2957(6)	0.1655(6)	0.0924(9)	1	0.7(1)		
O2	4d	0	0.3288(7)	0.077(1)	1	0.4(1)		
03	4c	0.1891(9)	0	0.566(1)	1	1.0(2)		
$CaBaGa_4O_8{:}0.007Bi^{3+}{,}0.003Zn^{2+}$								
Atom	site	x	у	Ζ	sof.	Beq. (Å <sup>2</sup> )		
Ca	2a	0	0.5	0.4261(6)	0.993	1.2(1)		
Bi	2a	0	0.5	0.4261(6)	0.007	1.2(1)		
Ba	2a	0	0	0	1	0.97(3)		
Ga/Zn	8e	0.2919(1)	0.1671(1)	0.4519(3)	0.997/0.003	0.68(3)		
01	8e	0.2954(7)	0.1674(7)	0.092(1)	1	0.5(1)		
O2	4d	0	0.3289(9)	0.078(2)	1	0.3(1)		
03	4c	0.191(1)	0	0.564(2)	1	0.7(2)		

Table	<b>S2</b>	Selected	interatomic	distances	in	CaBaGa <sub>4</sub> O <sub>8</sub> :0.01Bi <sup>3+</sup>	and
CaBaGa	4O8:0.00	07Bi <sup>3+</sup> ,0.003Z	n <sup>2+</sup> obtained from	m Rietveld refi	nement	ts against high-resolution C	Cu K <sub>α1</sub>
XRPD d	ata.						

CaBaGa <sub>4</sub> O <sub>8</sub> :0.01Bi <sup>3+</sup>						
bond	length (Å)	bond	length (Å)	bond	length (Å)	
Ga–O2	1.837(3)	$Ca-O2 \times 2$	2.393(7)	Ba–O3 × 2	2.732(7)	
Ga-O1	1.849(5)	Ca–O1 × 4	2.430(5)	Ba–O1 × 4	2.917(5)	
Ga-O1	1.854(5)			$Ba-O2 \times 4$	3.053(6)	
Ga-O3	1.854(4)			Ba–O3 × 4	3.309(7)	
CaBaGa <sub>4</sub> O <sub>8</sub> :0.007Bi <sup>3+</sup> ,0.003Zn <sup>2+</sup>						
bond	length (Å)	bond	length (Å)	bond	length (Å)	
Ga–O2	1.835(6)	Ca $-O2 \times 2$	2.389(8)	Ba–O3 × 2	2.749(9)	
Ga-O1	1.837(3)	Ca–O1 × 4	2.442(6)	Ba–O1 × 4	2.923(6)	
Ga-O1	1.857(6)			Ba–O2 × 4	3.055(8)	
Ga–O3	1.842(5)			Ba–O3 $\times$ 2	3.309(9)	