Supporting Information

A novel broadband Ba₃Ca₄(BO₃)₃(SiO₄)Cl:Mn⁴⁺ near-infrared

phosphor with special pseudo-octahedron Mn⁴⁺ coordination structure

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Host	Emission range (nm)	Emission peak (nm)	FWHM (nm)	IQE (%)	Ref.
Ba ₃ Ca ₄ (BO ₃) ₃ (SiO ₄)Cl	650-1100	~765	~90	69.7	This work
Li ₃ Mg ₂ NbO ₆	600-780	~670	~50	28.6	[1]
Gd_2ZnTiO_6	650-720	670, 685, 690 and 704	<50	-	[2]
Li ₂ MgTiO ₄	630-850	~676	~46	32	[3]
Na ₄ Mg(WO ₄) ₃	560-850	682	_	_	[4]
$BaLaMgNbO_6$	650-800	700	~36	52	[5]
SrGdAlO ₄	650-800	709 and 725	<40	23	[6]
NaLaMgWO ₆	650-800	~700	~38	60	[7]

Table S1. Luminescent properties of BCBSC:0.01Mn⁴⁺ phosphor and some reported Mn⁴⁺-doped oxide phosphors in recent years.

Ion	Valences	CN	IR (Å)
Mn ⁴⁺	4	6	0.53
Ba ²⁺	2	11	1.57
Ca^{2+}	2	9	1.18
Ca	2	10	1.23
Si^{4+}	4	4	0.26
B ³⁺	3	3	0.01

Table S2. Valences and effective ionic radius for related ions.⁸

CN: Coordination number; IR: Effective ionic radius.

Atom	Site	X	у	Z	Occupancy	U _{iso}
Ba1	12d	0.1476	0.2953	0.3479	1.0000	0.0125
Cal	6c	0.4764	0.5235	-0.0054	1.0000	0.0125
Ca2	2b	0.3333	0.6667	0.6247	1.0000	0.0166
Si1	2b	0.3333	0.6667	0.2094	1.0000	0.028
B1	6c	0.8063	0.6125	0.2927	1.0000	0.0114
Cl1	2a	0	0	0.1203	1.0000	0.0195
01	2b	0.3333	0.6667	-0.028	1.0000	0.0259
02	12d	0.6825	0.5842	0.1971	1.0000	0.0213
03	6c	0.4146	0.8291	0.3018	1.0000	0.0177
O4	6c	0.8355	0.6711	0.4699	1.0000	0.0410

 Table S3. Atomic positions of BCBSC.

BCBSC				BCBSC:0.1Mn ⁴⁺			
Bone	d length	Bond	angle	Bond length		Bond	angle
Ba1-Cl1	3.15198(10)	O1-Ca1-O2	126.480(1)	Ba1-Cl1	3.15284(9)	O1-Ca1-O2	126.481(0)
Ba1-Cl1	3.31221(10)	O1-Ca1-O2	76.441(0)	Ba1-Cl1	3.31308(9)	O1-Ca1-O2	76.441(0)
Ba1-O2	2.87748(9)	O1-Ca1-O2	126.509(1)	Ba1-O2	2.87812(8)	O1-Ca1-O2	126.510(0)
Ba1-O2	2.95102(10)	O1-Ca1-O2	76.431(0)	Ba1-O2	2.95185(10)	O1-Ca1-O2	76.431(0)
Ba1-O2	2.95202(10)	O1-Ca1-O3	143.803(2)	Ba1-O2	2.95286(10)	O1-Ca1-O3	143.807(2)
Ba1-O2	2.87704(9)	O1-Ca1-O3	64.446(2)	Ba1-O2	2.87768(8)	O1-Ca1-O3	64.443(1)
Ba1-O3	3.01498(12)	O2-Ca1-O2	154.625(0)	Ba1-O3	3.01587(11)	O2-Ca1-O2	154.625(0)
Ba1-O3	3.01438(12)	O2-Ca1-O2	73.147(2)	Ba1-O3	3.01527(11)	O2-Ca1-O2	73.150(1)
Ba1-O4	3.03131(11)	O2-Ca1-O2	103.128(2)	Ba1-O4	3.03219(10)	O2-Ca1-O2	103.124(2)
Ba1-O4	2.58928(11)	O2-Ca1-O3	80.706(2)	Ba1-O4	2.58977(10)	O2-Ca1-O3	80.702(2)
Ba1-O4	3.02953(11)	O2-Ca1-O3	76.686(3)	Ba1-O4	3.03041(10)	O2-Ca1-O3	76.691(2)
Cal-O1	2.66661(11)	O2-Ca1-O2	103.165(2)	Ca1-O1	2.66740(10)	O2-Ca1-O2	103.160(2)
Ca1-O2	2.40386(7)	O2-Ca1-O2	69.129(3)	Ca1-O2	2.40323(7)	O2-Ca1-O2	69.135(3)
Cal-O2	2.52421(3)	O2-Ca1-O3	73.937(2)	Ca1-O2	2.52339(8)	O2-Ca1-O3	73.941(2)
Ca1-O2	2.40331(3)	O2-Ca1-O3	127.667(2)	Ca1-O2	2.40259(7)	O2-Ca1-O3	127.663(2)
Ca1-O2	2.52458(7)	O2-Ca1-O2	154.628(0)	Ca1-O2	2.52400(8)	O2-Ca1-O2	154.628(0)
Cal-O3	2.41448(3)	O2-Ca1-O3	80.735(2)	Ca1-O3	2.41409(7)	O2-Ca1-O3	80.730(2)
Cal-O3	2.38563(4)	O2-Ca1-O3	76.682(3)	Ca1-O3	2.38512(8)	O2-Ca1-O3	76.687(2)
Ca1-O4	2.93457(12)	O2-Ca1-O3	73.911(2)	Ca1-O4	2.93545(4)	O2-Ca1-O3	73.915(2)
Cal-O4	2.93629(4)	O2-Ca1-O3	127.656(2)	Ca1-O4	2.93728(11)	O2-Ca1-O3	127.652(2)
Ca2-O1	2.36160(10)	O3-Ca1-O3	151.751(0)	Ca2-O1	2.36205(9)	O3-Ca1-O3	151.751(0)
Ca2-O2	2.65815(10)			Ca2-O2	2.65893(9)		
Ca2-O2	2.65920(10)			Ca2-O2	2.65998(9)		
Ca2-O2	2.65873(10)			Ca2-O2	2.65951(9)		
Ca2-O2	2.65815(10)			Ca2-O2	2.65893(9)		
Ca2-O2	2.65873(10)			Ca2-O2	2.65951(9)		
Ca2-O2	2.65920(10)			Ca2-O2	2.65998(9)		
Ca2-O3	2.66484(9)			Ca2-O3	2.66544(8)		
Ca2-O3	2.66484(9)			Ca2-O3	2.66544(8)		
Ca2-O3	2.66537(9)			Ca2-O3	2.66597(8)		
Si1-O1	1.61429(7)			Si1-O1	1.61460(6)		
Si1-O3	1.63558(6)			Si1-O3	1.63604(5)		
Sil-O3	1.63558(6)			Si1-O3	1.63604(5)		
Si1-O3	1.63644(6)			Si1-O3	1.63690(5)		
B1-O2	1.37037(4)			B1-O2	1.37074(4)		
B1-O2	1.36945(4)			B1-O2	1.36982(4)		
B1-O4	1.32242(5)			B1-O4	1.32269(4)		

Table S4. Bond length (Å) and O-Ca1-O bond angle (°) of BCBSC and BCBSC: $0.1Mn^{4+}$.

Host	Space group	$\langle \lambda \rangle$	σ^2	Ref.
Ba ₃ Ca ₄ (BO ₃) ₃ (SiO ₄)Cl	Hexagonal P63mc	1.12	449.48	This work
Na ₂ WO ₂ F ₄	Orthorhombic C_{2v} - <i>pbcn</i>	1.02485	78.5316	[9]
Rb ₂ GeF ₆	Hexagonal C ₃ -p6 ₃ mc	1.00340	35.3698	[10]
K ₂ SiF ₆	Cubic $O_{\rm h}$ - Fm_3m	1.00000	0.0000	[11]
La ₄ Ti ₃ O ₁₂	Rhombohedral C ₃ -R	1.00370	56.9296	[12]

Table S5. Polyhedral distortion indices in different Mn⁴⁺-activated hosts.

Host	Dq (cm ⁻¹)	B (cm ⁻¹)	C (cm ⁻¹)	β_1	$E(^{2}E_{g})$ (cm ⁻¹)	Ref.
$\overline{3.5MgO \cdot 0.5MgF_2 \cdot GeO_2}$	2381	700	3416	0.997	15,151	[13]
BaSiF ₆	2128	568	3879	1.025	16,048	[14]
Na_2SiF_6	1970	775	3490	1.051	16,210	[15]
K_2GeF_6	2123	590	3831	1.025	16,050	[16]
K_2SiF_6	2030	770	3470	1.044	15,873	[17]
LiAl ₅ O ₈	2014	725	2900	0.92	13,977	[18]
NaLaMgTeO ₆	2008	790	2949	0.966	14,676	[19]
LiGa ₅ O ₈	_	653	3289	0.949	14,724	[20]
SrTiO ₃	1818	719	2839	0.905	13,827	[21]
Ba ₂ LaNbO ₆	1780	670	3290	0.958	14,679	[22,23]
LaAlO ₃	2123	700	2941	0.912	14,045	[24]
Zn_2TiO_4	2156	691	3198	0.952	14,815	[25]
Mg ₂ TiO ₄	2096	700	3348	0.985	15,267	[26]
CaZrO ₃	1850	754	3173	0.983	15,054	[27]
$Sr_4Al_{14}O_{25}$	2222	790	3192	1.007	15,337	[28]
CaAl ₁₂ O ₁₉	2146	750	3245	0.993	15,243	[29]
SrMgAl ₁₀ O ₁₇	2237	791	3084	0.989	15,152	[30]
$Y_2Sn_2O_7$	2100	700	3515	1.016	15,563	[31]
$Y_3Ga_5O_{12}$	1922	699	3197	0.957	14,859	[25]
$Mg_{14}Ge_5O_{24}$	2375	709	3263	0.974	15,175	[25]
YAl ₃ (BO ₃) ₄	1890	755	3015	0.956	14,620	[32]
Li ₂ MgTiO ₄	2101	724	3122	0.957	14,793	[33]
SrGe ₄ O ₉	2362	832	3024	1.004	15,267	[25]
Gd_2ZnTiO_6	2041	794	2783	0.924	14,205	[34]
$Mg_3Ga_2SnO_8$	2033	847	2846	0.99	14,706	[35]
BaLaMgNbO ₆	1923	736	2944	0.93	14,286	[36]
Ba ₃ Ca ₄ (BO ₃) ₃ (SiO ₄)Cl	2128	567	2958	0.843	13,228	This work

Table S6. Spectroscopic parameters and calculated β_1 values of Mn⁴⁺ ions in various hosts.

Current (mA)	Voltage (V)	Input power (mW)	Output power (mW)
10	3.192	31.92	0.6813
20	3.234	64.36	1.665
30	3.261	97.51	2.667
40	3.282	130.9	3.604
50	3.3	164.6	4.481
60	3.314	198.5	5.253
70	3.327	232.5	5.911
80	3.339	266.7	6.531
90	3.349	301.1	6.998
100	3.359	335.5	7.408
110	3.369	370.3	7.818
120	3.377	404.9	8.018
130	3.385	439.8	8.203

 Table S7. The device performance of BCBSC:0.01Mn⁴⁺ NIR pc-LED excited by 365 nm n-UV chip.

Current (mA)	Voltage (V)	Input power (mW)	Output power (mW)	
10	2.623	26.23	1.801	
20	2.667	53.07	3.391	
30	2.702	80.78	4.764	
40	2.732	109	5.904	
50	2.76	137.7	6.827	
60	2.786	166.9	7.624	
70	2.811	196.5	8.215	
80	2.836	226.6	8.706	
90	2.859	257	9.021	
100	2.881	287.8	9.236	
110	2.904	319.1	9.389	
120	2.925	350.7	9.419	
130	2.947	382.8	9.429	

 Table S8. The device performance of BCBSC:0.01Mn⁴⁺ NIR pc-LED excited by 460 nm blue chip.



Fig. S1. SEM images of BCBSC: $0.1Mn^{4+}$ phosphor in different magnification of (a) ×1000, (b) ×2000, (c) ×3000, (d) ×5000, (e) ×5000, and (f) ×10000.



Fig. S2. (a) XPS survey spectrum of MnO_2 standard sample. (b) the magnified Mn 2p spectrum.



Fig. S3. The local magnification PDOS of BCBSC.



Fig. S4. The PL spectrum of the BCBSC: $0.1Mn^{4+}$ phosphor measured at 425 K.



Fig. S5. Fitting results of square of full width at half maximum (FWHM²) as a function of 2kT.



Fig. S6. The integral area of the PL spectrum of the BCBSC:0.1Mn⁴⁺ phosphor in the range of 600–960 nm and 960–1200 nm, respectively.

Supplementary Note

Computational Detail

Based on density functional theory (DFT), the band structure and electronic structure were calculated by Vienna ab initio simulation package (VASP) codes. The exchange correlation potential was approximated by generalized gradient approximation (GGA) using the Perdew–Burke–Ernzerhof (PBE) formulation. We used the projected augmented wave (PAW) potentials to describe the ionic cores. The plane wave basis set with a kinetic energy cutoff of 450 eV and we used $2\times2\times2$ for the Monkhorst-Pack k-point for sampling the Brillouin zone during the whole computations. Partial occupancies of the Kohn–Sham orbitals were allowed using the Gaussian smearing method and a width of 0.05 eV. The electronic energy was considered to be self-consistent when the energy change was less than 10^{-5} eV. The geometric optimization was considered convergent when the force change was less than 0.02 eV/Å.

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