

[Electronic supplementary information (ESI)]

Structural diversity of copper(I) ethylene complexes with 2,4-bis(2-pyridyl)pyrimidine directed by anions

Masahiko Maekawa,^{a,*} Terumasa Hayashi,^b Kuniyoshi Sugimoto,^b Takashi Okubo,^b Takayoshi Kuroda-Sowa^b

^a Research Institute for Science and Technology, Kindai University, Kowakae, Higashi-Osaka, Osaka 577-8502, Japan. E-mail: maekawa@rist.kindai.ac.jp

^b Department of Chemistry, Kindai University, Kowakae, Higashi-Osaka, Osaka 577-8502, Japan

Table S1 Crystallographic data for Cu(I)-bpprd/C₂H₄ complexes 1–4

	[Cu ₂ (bpprd)(η ² -C ₂ H ₄) ₂ (ClO ₄) ₂] (1)	[Cu ₄ (bpprd) ₂ (η ² -C ₂ H ₄) ₄ (μ-NO ₃) ₂](NO ₃) ₂ (2)	[Cu ₂ (bpprd)(η ² -C ₂ H ₄) ₂ (BF ₄)]BF ₄ (3)	{[Cu ₆ (bpprd) ₄ (η ² -C ₂ H ₄) ₂ (μ-η ² :η ² -C ₂ H ₄)(μ-BF ₄) ₂](BF ₄) ₄] _n (4)
Formula	C ₁₈ H ₁₈ Cl ₂ Cu ₂ N ₄ O ₈	C ₁₈ H ₁₈ Cu ₂ N ₆ O ₆	C ₁₈ H ₁₈ B ₂ Cu ₂ F ₈ N ₄	C ₆₂ H ₅₂ B ₆ Cu ₆ F ₂₄ N ₁₆
Formula Weight	616.34	541.46	591.06	1923.29
Crystal System	monoclinic	triclinic	monoclinic	monoclinic
Space Group	<i>P</i> 2 ₁ / <i>c</i> (No. 14)	<i>P</i> 1(-) (No. 2)	<i>P</i> 2 ₁ / <i>c</i> (No. 14)	<i>C</i> 2 ₁ / <i>c</i> (No. 15)
<i>a</i> /Å	11.3225(3)	9.3576(3)	11.1296(4)	22.7929(6)
<i>b</i> /Å	13.0447(4)	10.9852(6)	12.9401(4)	11.3730(2)
<i>c</i> /Å	15.7577(5)	11.9635(7)	15.7337(7)	29.0946(7)
<i>α</i> /°	90	64.609(5)	90	90
<i>β</i> /°	109.276(3)	67.153(4)	109.055(4)	109.594(3)
<i>γ</i> /°	90	84.922(4)	90	90
<i>V</i> /Å ³	2196.91(12)	1019.33(10)	2141.78(15)	7105.3(3)
<i>Z</i>	4	2	4	4
<i>D</i> _{calc} /gcm ⁻³	1.863	1.764	1.833	1.798
<i>F</i> (000)	1240.0	548.0	1176.0	3824.0
<i>μ</i> (MoKα) /mm ⁻¹	2.235	2.137	2.070	1.881
Temperature /K	100.15	100.15	97.15	97.15
Observed reflections	26013 (<i>R</i> _{int} = 0.0640)	19629 (<i>R</i> _{int} = 0.0577)	31668 (<i>R</i> _{int} = 0.0556)	50700 (<i>R</i> _{int} = 0.0281)
Refined reflections	6603 (all data); 5613 (<i>I</i> > 2σ(<i>I</i>))	6011 (all data); 5130 (<i>I</i> > 2σ(<i>I</i>))	6502 (all data); 5816 (<i>I</i> > 2σ(<i>I</i>))	10738 (all data); 9477 (<i>I</i> > 2σ(<i>I</i>))
<i>R</i>	0.0462 (all data)	0.0396 (all data)	0.0355 (all data)	0.0408 (all data)
<i>R</i> ₁	0.0381 (<i>I</i> > 2σ(<i>I</i>))	0.0325 (<i>I</i> > 2σ(<i>I</i>))	0.0306 (<i>I</i> > 2σ(<i>I</i>))	0.0356 (<i>I</i> > 2σ(<i>I</i>))
<i>wR</i> ₂	0.1080 (all data)	0.0855 (all data)	0.0776 (all data)	0.0944 (all data)
GOF	1.072	1.048	1.031	1.041

$$R = \sum ||F_o| - |F_c|| / \sum |F_o|. R_1 = \sum ||F_o| - |F_c|| / \sum |F_o| (F_o > 4\sigma(F_o)). wR_2 = [\sum (w(F_o^2 - F_c^2))^2 / \sum w(F_o^2)]^{1/2}.$$

Table S2 Selected bond lengths (Å) and bond angles (°) of Cu(I)–bpprd/C₂H₄ complexes **1–4**

(a) [Cu₂(bpprd)(η²-C₂H₄)₂(ClO₄)₂] (1)			
Cu(1)-N(1)	1.9705(18)	Cu(1)-N(2)	2.0573(17)
Cu(1)-C(15)	2.010(2)	Cu(1)-C(16)	2.027(2)
Cu(1)-O(1)	2.573(2)	Cu(2)-N(3)	1.9988(19)
Cu(2)-N(4)	1.9990(17)	Cu(2)-C(17)	1.998(2)
Cu(2)-C(18)	2.010(2)	Cu(2)-O(7)	2.507(2)
C(15)=C(16)	1.361(3)	C(17)=C(18)	1.370(4)
N(1)-Cu(1)-N(2)	82.10(7)	N(1)-Cu(1)-C(15)	113.08(9)
N(1)-Cu(1)-C(16)	152.19(9)	N(1)-Cu(1)-O(1)	92.00(7)
N(2)-Cu(1)-C(15)	158.88(8)	N(2)-Cu(1)-C(16)	123.28(8)
N(2)-Cu(1)-O(1)	96.06(7)	C(15)-Cu(1)-C(16)	39.41(10)
C(15)-Cu(1)-O(1)	97.88(8)	C(16)-Cu(1)-O(1)	95.85(8)
N(3)-Cu(2)-N(4)	83.15(7)	N(3)-Cu(2)-C(18)	155.41(9)
N(3)-Cu(2)-C(17)	116.28(9)	N(3)-Cu(2)-O(7)	87.26(7)
N(4)-Cu(2)-C(17)	159.12(9)	N(4)-Cu(2)-C(18)	119.64(9)
N(4)-Cu(2)-O(7)	90.32(7)	C(17)-Cu(2)-C(18)	39.97(11)
C(17)-Cu(2)-O(7)	97.81(9)	C(18)-Cu(2)-O(7)	100.78(9)
(b) [Cu₄(bpprd)₂(η²-C₂H₄)₄(μ-NO₃)₂](NO₃)₂ (2)			
Cu(1)-N(1)	2.0016(14)	Cu(1)-N(2)	2.2922(13)
Cu(1)-C(15)	2.0347(16)	Cu(1)-C(16)	2.0374(17)
Cu(1)-O(1)	2.0615(12)	Cu(2)-N(3)	1.9826(14)
Cu(2)-N(4)	2.0151(13)	Cu(2)-C(17)	2.0166(17)
Cu(2)-C(18)	2.0169(17)	Cu(2)-O(2')	2.444(1)
C(15)=C(16)	1.364(3)	C(17)=C(18)	1.373(3)
N(1)-Cu(1)-N(2)	77.93(5)	N(1)-Cu(1)-C(15)	149.59(7)
N(1)-Cu(1)-C(16)	110.84(6)	N(1)-Cu(1)-O(1)	98.62(5)
N(2)-C(15)-C(16)	111.97(6)	N(2)-Cu(1)-C(16)	108.30(6)
N(2)-Cu(1)-O(1)	95.09(5)	C(15)-Cu(1)-C(16)	39.15(7)
C(15)-Cu(1)-O(1)	108.59(7)	C(16)-Cu(1)-O(1)	145.33(6)
N(3)-Cu(2)-N(4)	83.02(6)	N(3)-Cu(2)-C(17)	113.25(6)
N(3)-Cu(2)-C(18)	149.59(7)	N(3)-Cu(2)-O(2')	101.58(6)
N(4)-Cu(2)-C(17)	156.16(7)	N(4)-Cu(2)-C(18)	118.86(7)
N(4)-Cu(2)-O(2')	103.33(6)	C(17)-Cu(2)-C(18)	39.81(7)
C(17)-Cu(2)-O(2')	90.76(7)	C(18)-Cu(2)-O(2')	93.87(7)
(c) [Cu₂(bpprd)(η²-C₂H₄)₂(BF₄)]BF₄ (3)			
Cu(1)-N(1)	1.9664(14)	Cu(1)-N(2)	2.0551(13)
Cu(1)-C(15)	2.0025(16)	Cu(1)-C(16)	2.0192(17)
Cu(2)-N(3)	.9894(14)	Cu(2)-N(4)	1.9967(13) . ?
Cu(2)-C(17)	1.9976(17)	Cu(2)-C(18)	2.0002(17)
Cu(2)-F(8)	2.586(2)	C(15)=C(16)	1.370(2)
C(17)=C(18)	1.368(3)		
N(1)-Cu(1)-N(2)	82.35(5)	N(1)-Cu(1)-C(15)	112.19(7)

N(1)-Cu(1)-C(16)	151.94(7)	N(2)-Cu(1)-C(15)	161.96(6)
N(2)-Cu(1)-C(16)	124.66(6)	C(15)-Cu(1)-C(16)	39.82(7)
N(3)-Cu(2)-N(4)	83.54(5)	N(3)-Cu(2)-C(17)	116.70(7)
N(3)-Cu(2)-C(18)	156.25(7)	N(3)-Cu(2)-F(8)	83.20(5)
N(4)-Cu(2)-C(17)	159.05(7)	N(4)-Cu(2)-C(18)	119.23(7)
N(4)-Cu(2)-F(8)	85.58(5)	C(17)-Cu(2)-C(18)	40.04(8)
C(17)-Cu(2)-F(8)	101.42(5)	C(18)-Cu(2)-F(8)	103.96(7)

(d) $\{[\text{Cu}_6(\text{bpprd})_4(\eta^2\text{-C}_2\text{H}_4)_2(\mu\text{-}\eta^2\text{:}\eta^2\text{-C}_2\text{H}_4)(\mu\text{-BF}_4)_2](\text{BF}_4)_4\}_n$ (**4**)

Cu(1)-N(1)	1.9697(15)	Cu(1)-N(2)	2.1450(15)
Cu(2)-N(3)	1.9774(15)	Cu(2)-N(4)	1.9873(15)
Cu(2)-F(2)	2.543(1)	Cu(2)-C(29)	2.0130(18)
Cu(3)-N(5)	1.9222(16)	Cu(3)-N(6)	2.3470(17)
Cu(4)-N(7)	1.982(2)	Cu(4)-N(8)	2.0020(19)
Cu(4)-F(1)	2.591(1)	Cu(4)-C(30)	1.997(3)
Cu(4)-C(31)	2.004(2)	C(29)=C(29'')	1.436(2)
C(30)=C(31)	1.365(5)		
N(1)-Cu(1)-N(2)	80.77(6)	N(1)-Cu(1)-N(1')	156.61(9)
N(1)-Cu(1)-N(2')	109.96(6)	N(2)-Cu(1)-N(2')	126.97(8)
N(3)-Cu(2)-N(4)	83.99(6)	N(3)-Cu(2)-C(29'')	160.73(7)
N(3)-Cu(2)-C(29)	119.11(7)	N(4)-Cu(2)-C(29'')	115.00(7)
N(4)-Cu(2)-C(29)	156.80(7)	F(2)-Cu(2)-C(29)	96.71(6)
F(2)-Cu(2)-C(29'')	92.19(6)	F(2)-Cu(2)-N(3)	87.31(6)
F(2)-Cu(2)-N(4)	81.32(6)	C(29)-Cu(2)-C(29'')	41.82(9)
N(5)-Cu(3)-N(6)	79.91(6)	N(5)-Cu(3)-N(5')	160.82(10)
N(5)-Cu(3)-N(6')	107.68(6)	N(6)-Cu(3)-N(6')	134.67(8)
N(7)-Cu(4)-N(8)	82.53(8)	N(7)-Cu(4)-C(30)	160.82(12)
N(7)-Cu(4)-C(31)	120.93(11)	N(8)-Cu(4)-C(30)	116.56(12)
N(8)-Cu(4)-C(31)	156.04(11)	F(1)-Cu(4)-C(30)	101.3(1)
F(1)-Cu(4)-C(31)	96.97(9)	F(1)-Cu(4)-N(7)	78.52(6)
F(1)-Cu(4)-N(8)	82.52(6)	C(30)-Cu(4)-C(31)	39.89(14)

Symmetry codes. Complex **1**: (i): x, y, z. Complex **2**: (i): x, y, z; (ii(')): 1-x, 1-y, 1-z. Complex **3**: (i): x, y, z. Complex **4**: (i): x, y, z; (ii(')): 1-x, y, 3/2-z; (iii(')): 1-x, 1-y, 1-z.

Table S3 The selected dihedral angles (°) of Cu(I)–bpprd/C₂H₄ complexes **1–4**

Complexes	Plane 1 ^a	Plane 2 ^a	Dihedral angles (°)
[Cu ₂ (bpprd)(η ² -C ₂ H ₄) ₂ (ClO ₄) ₂] (1)	A	B	13.31
	B	C	20.40
[Cu ₄ (bpprd) ₂ (η ² -C ₂ H ₄) ₄ (μ-NO ₃) ₂](NO ₃) ₂ (2)	A	B	18.43
	B	C	10.76
[Cu ₂ (bpprd)(η ² -C ₂ H ₄) ₂ (BF ₄)]BF ₄ (3)	A	B	15.42
	B	C	20.39
{[Cu ₆ (bpprd) ₄ (η ² -C ₂ H ₄) ₂ (μ-η ² :η ² -C ₂ H ₄)(μ-BF ₄) ₂](BF ₄) ₄] _n (4)	A	B	13.90 ^b
	B	C	24.34 ^b
	A	B	15.84 ^c
	B	C	25.29 ^c

^a The terminal pyridyl rings (A) and (C), and the central pyrimidine ring (B) are shown in Figs **1-4**, respectively.

^b The [Cu₃(bpprd)₂(η²-C₂H₄)₂]³⁺ cation moiety.

^c The [Cu₃(bpprd)₂(μ-η²:η²-C₂H₄)]³⁺ cation moiety.