

*Electronic Supporting Information (ESI)*

**What do we know about bifunctional cage-like T<sub>8</sub> silsesquioxanes?  
Theory versus lab routine**

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**Table of contents**

**1. Characterization of phenylhepta(isobutyl)-POSS T<sub>8</sub>(i-Bu)<sub>7</sub>(Ph) (1)**

<b>Figure S1.</b>	<sup>1</sup> H NMR (500 MHz, CDCl <sub>3</sub> , 300 K) spectrum of <b>1</b> .....	3
<b>Figure S2.</b>	<sup>13</sup> C NMR (500 MHz, CDCl <sub>3</sub> , 300 K) spectrum of <b>1</b> .....	3
<b>Figure S3.</b>	<sup>29</sup> Si NMR (600 MHz, CDCl <sub>3</sub> , 300 K) spectrum of <b>1</b> .....	4
<b>Figure S4.</b>	FT-IR spectrum (KBr pellet) of <b>1</b> .....	4
<b>Figure S5.</b>	MALDI-MS ([M + Na] <sup>+</sup> ) of <b>1</b> .....	5

**2. Characterization of partially condensed trisilanol phenylhexa(isobutyl)-POSS  
T<sub>7</sub>(i-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub> (2)**

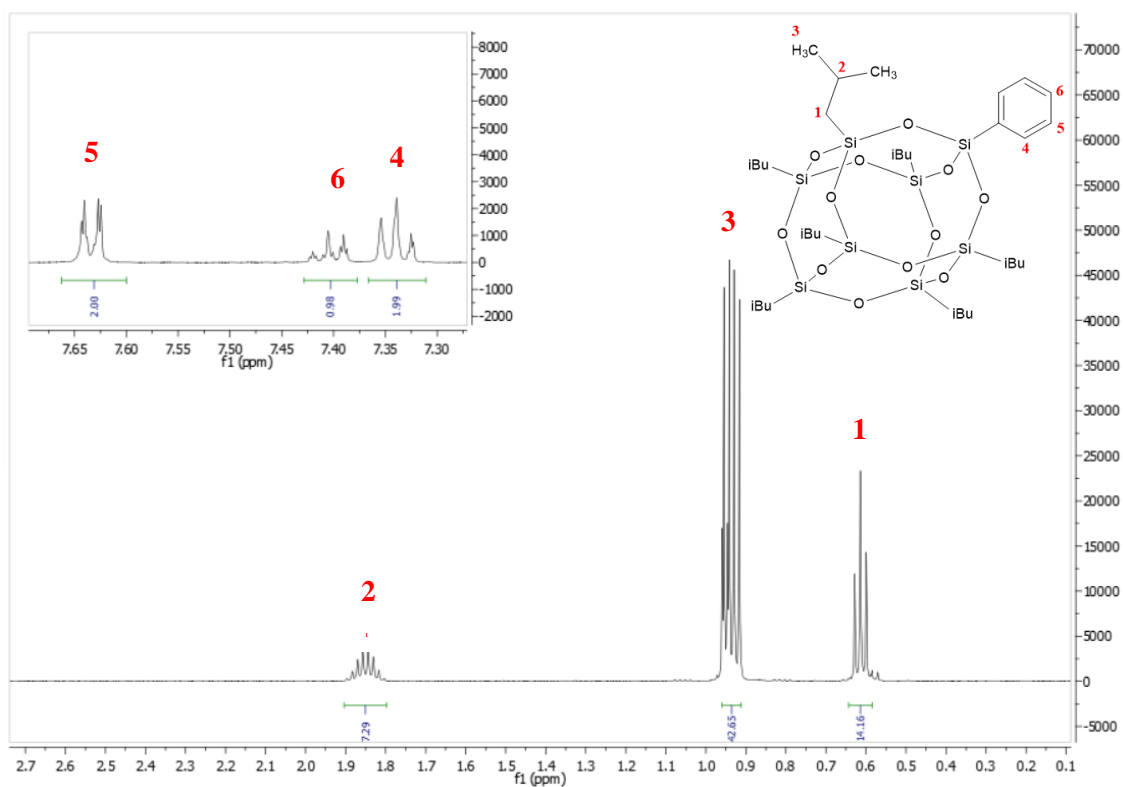
<b>Figure S6.</b>	<sup>1</sup> H NMR (600 MHz, CDCl <sub>3</sub> , 300 K) spectrum of <b>2</b> .....	5
<b>Figure S7.</b>	<sup>13</sup> C NMR (500 MHz, CDCl <sub>3</sub> , 300 K) spectrum of <b>2</b> .....	6
<b>Figure S8.</b>	<sup>29</sup> Si NMR (600 MHz, CDCl <sub>3</sub> , 300 K) spectrum of <b>2</b> .....	6
<b>Figure S9.</b>	FT-IR spectrum (KBr pellet) of <b>2</b> .....	7
<b>Figure S10.</b>	MALDI-MS ([M + Na] <sup>+</sup> ) of <b>2</b> .....	7
<b>Figure S11.</b>	DOSY NMR of <b>2</b> .....	8

**3. Diphenylhexa(isobutyl)-POSS  $T_8(i-Bu)_6(Ph)_2$  synthesis attempt (instead – characterization of trisilanol hepta(isobutyl)-POSS  $T_7(i-Bu)_7(OH)_3$  (3))**

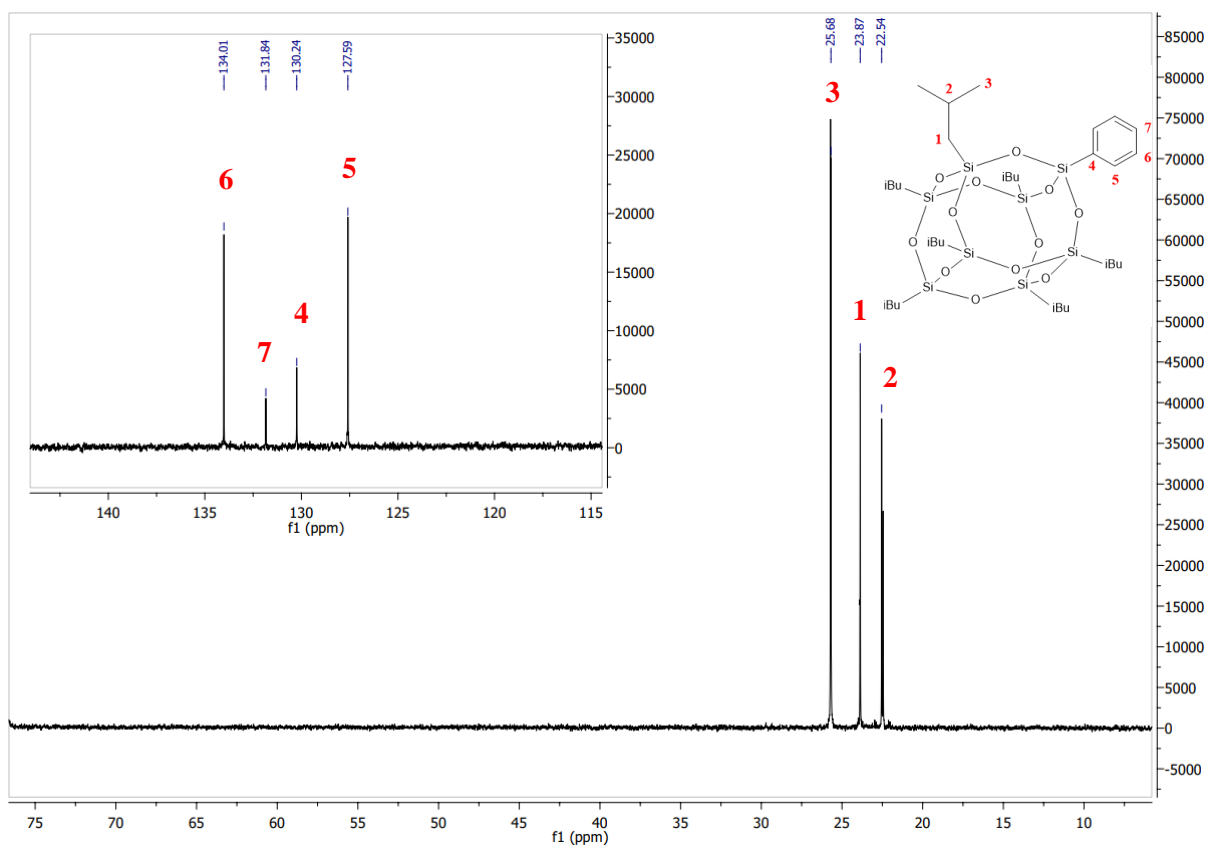
<b>Figure S12.</b>	$^1H$ NMR (500 MHz, $CDCl_3$ , 300 K) spectrum of <b>3</b> .....	8
<b>Figure S13.</b>	$^{13}C$ NMR (500 MHz, $CDCl_3$ , 300 K) spectrum of <b>3</b> .....	9
<b>Figure S14.</b>	$^{29}Si$ NMR (600 MHz, $CDCl_3$ , 300 K) spectrum of <b>3</b> .....	9
<b>Figure S15.</b>	MALDI-MS ( $[M + Na]^+$ ) of <b>3</b> .....	10
<b>Figure S16.</b>	DOSY NMR of <b>3</b> .....	10
<b>Figure S17.</b>	Molecular structure of <b>3</b> .....	11

**4. Quantum calculations**

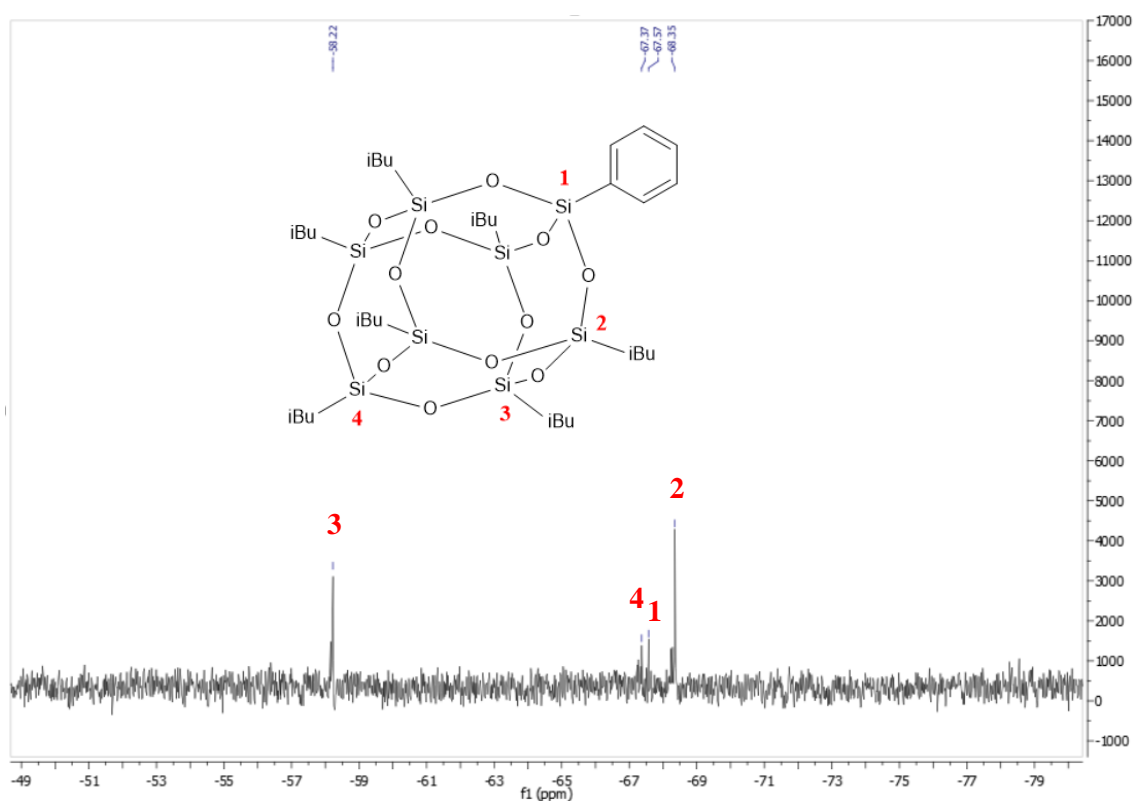
<b>Figure S18.</b>	Relative Gibbs free energies (kcal/mol) for DFT optimized POSS structures .....	12
<b>Table S1.</b>	The calculated HOMO-LUMO energy gaps and quantum chemical properties of POSS complexes .....	15
<b>Figure S19.</b>	The calculated HOMO and LUMO energy levels and HOMO-LUMO energy gaps of optimized POSS models .....	17
<b>Table S2.</b>	Energies of calculated POSS models .....	19



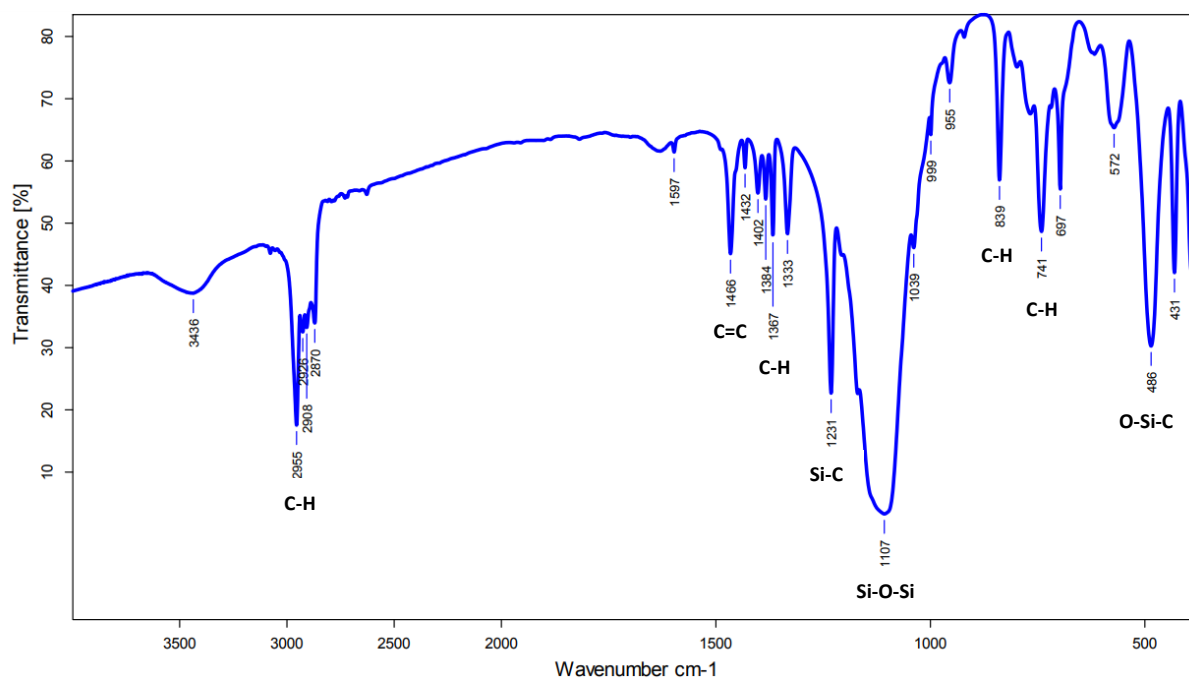
**Figure S1.**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ , 300 K) spectrum of **1**.



**Figure S2.**  $^{13}\text{C}$  NMR (500 MHz,  $\text{CDCl}_3$ , 300 K) spectrum of **1**.



**Figure S3.**  $^{29}\text{Si}$  NMR (600 MHz,  $\text{CDCl}_3$ , 300 K) spectrum of **1**.



**Figure S4.** FT-IR spectrum (KBr pellet) of **1**.

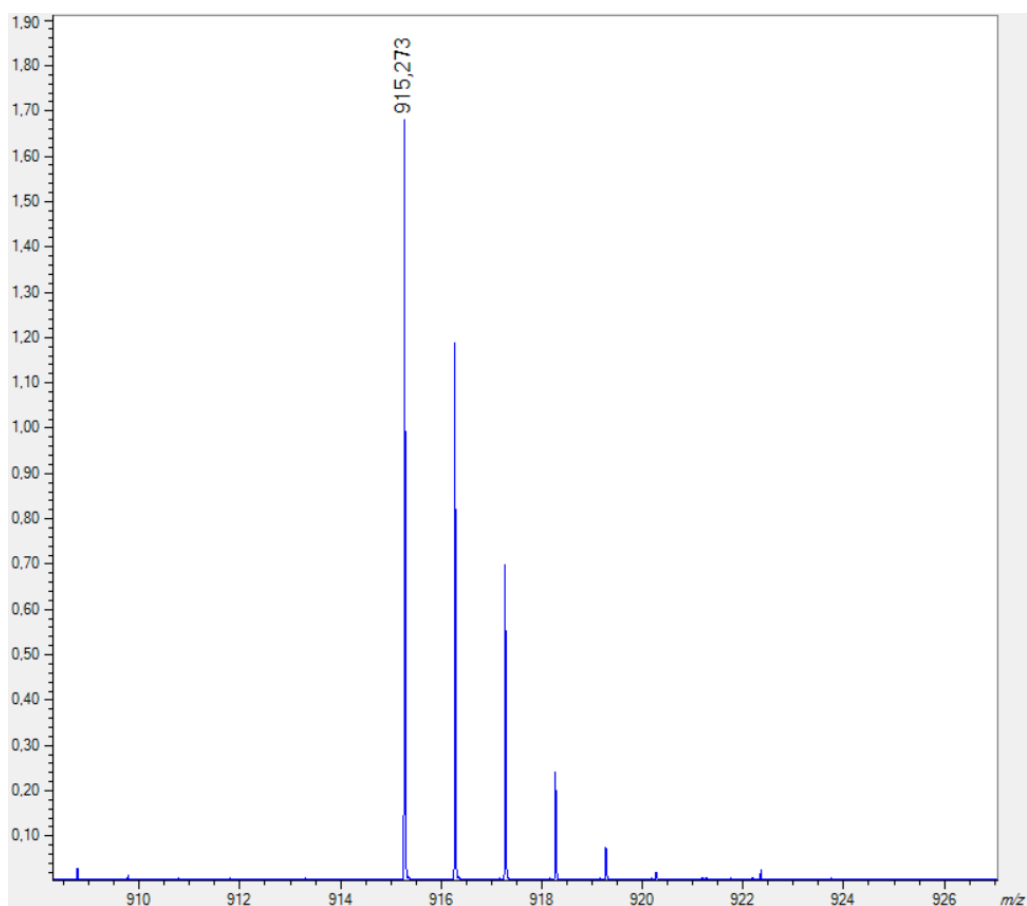


Figure S5. MALDI-MS of 1.

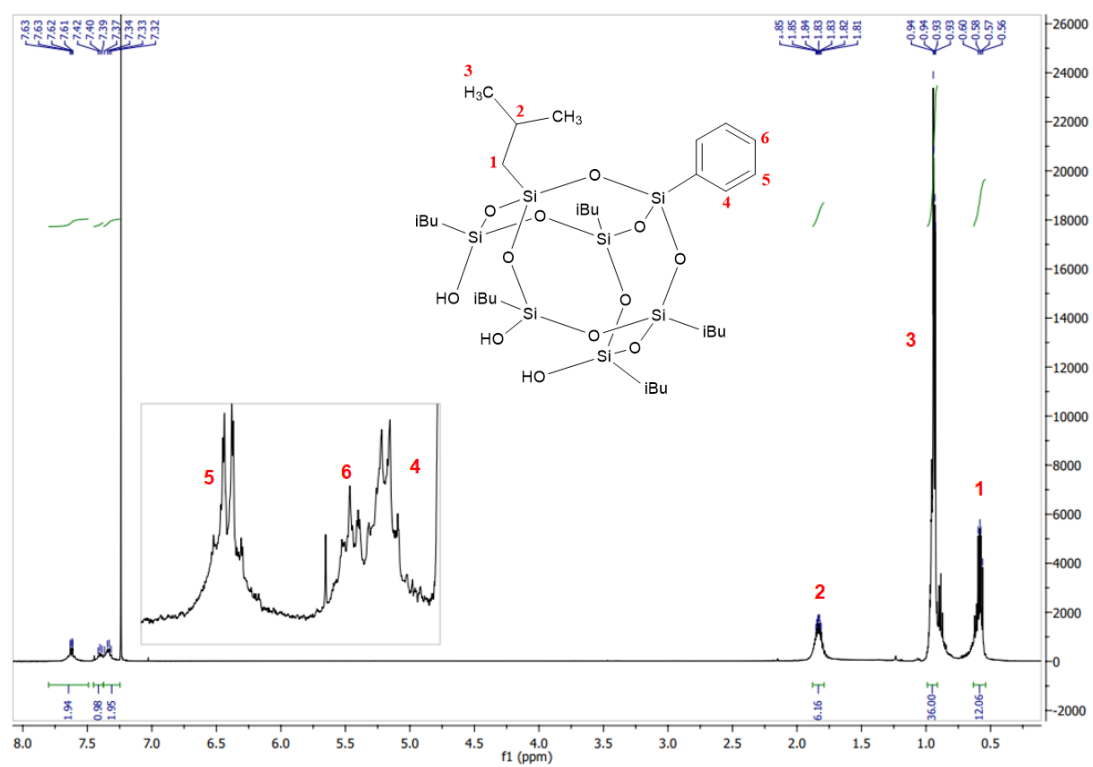
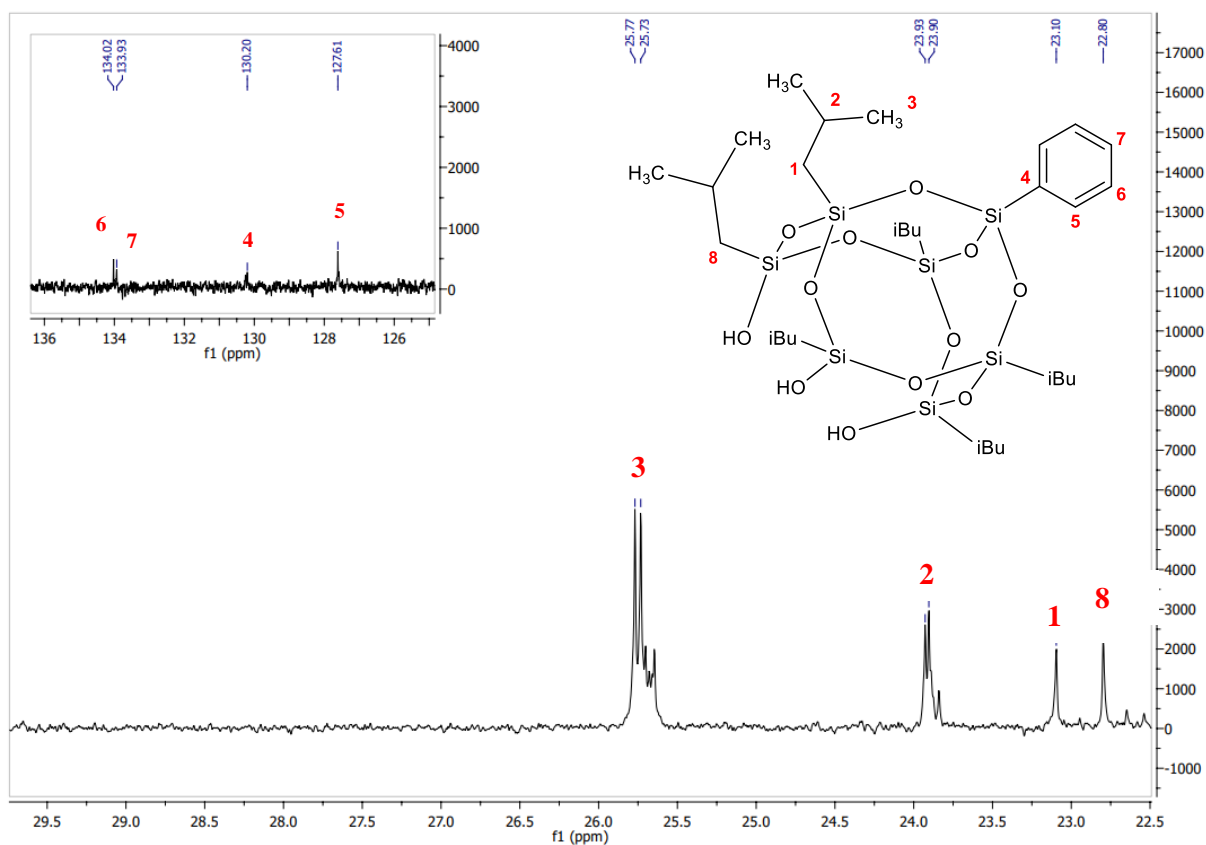
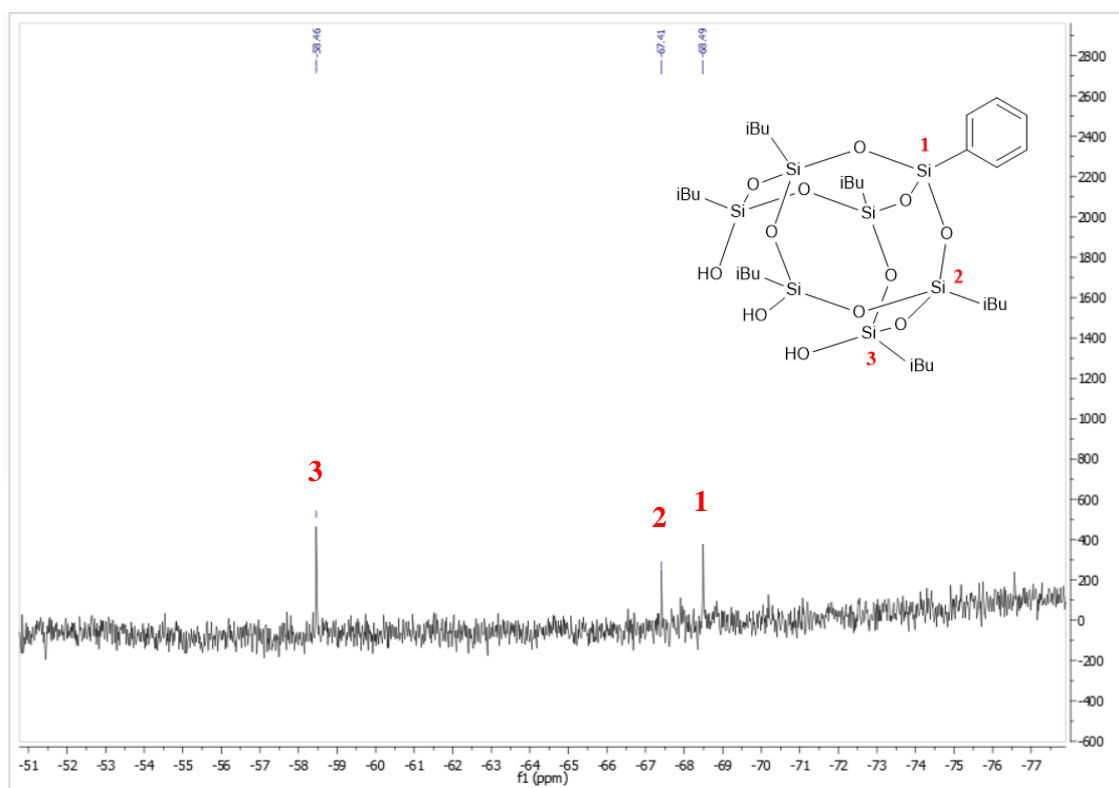


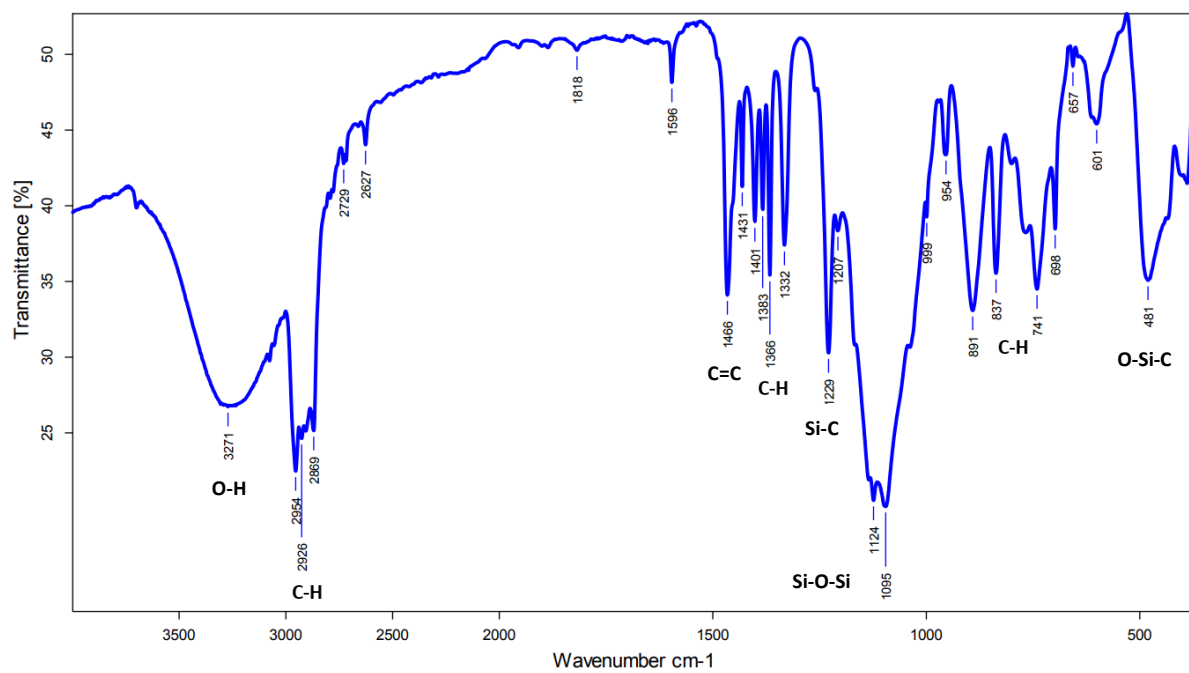
Figure S6.  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ , 300 K) spectrum of 2.



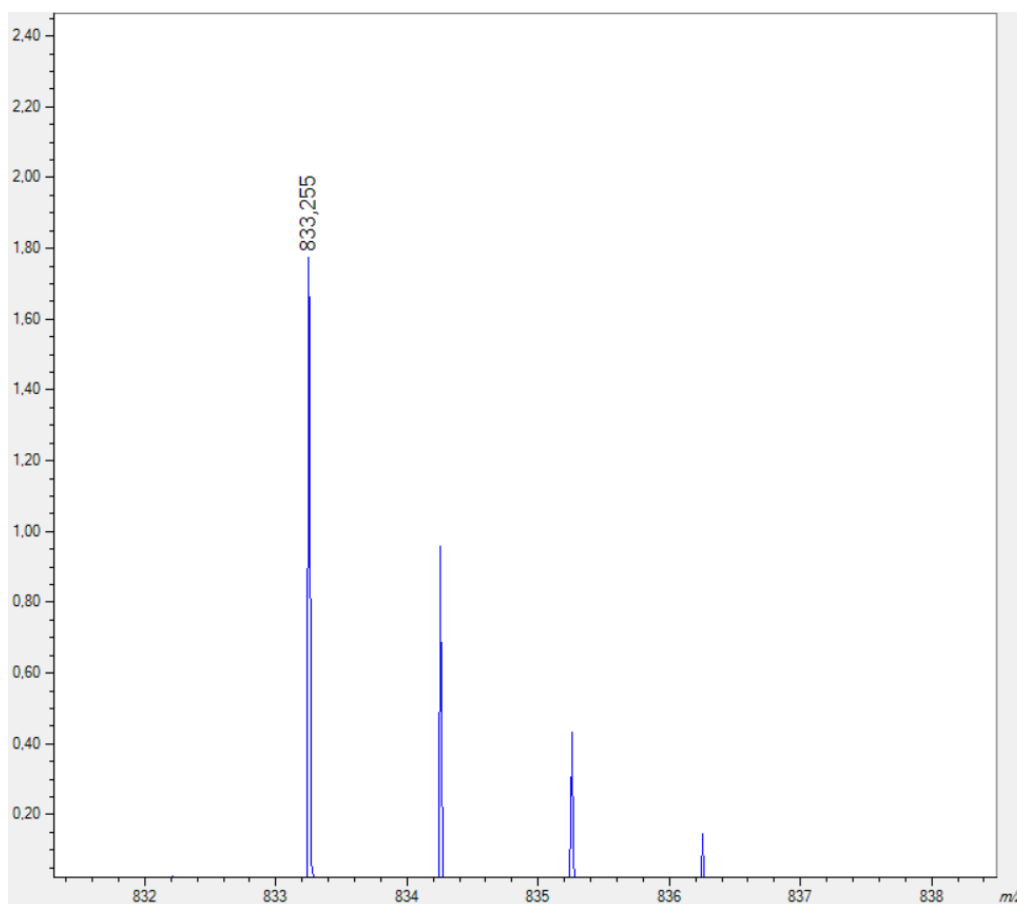
**Figure S7.**  $^{13}\text{C}$  NMR (500 MHz,  $\text{CDCl}_3$ , 300 K) spectrum of **2**.



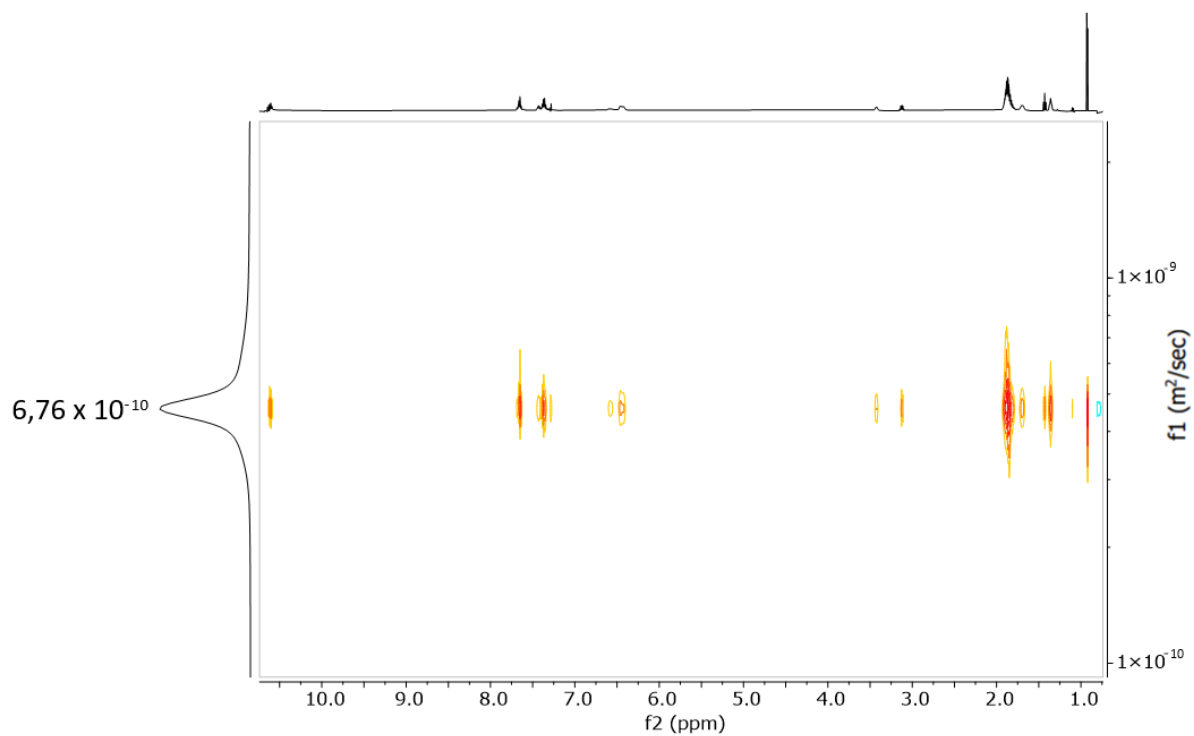
**Figure S8.**  $^{29}\text{Si}$  NMR (600 MHz,  $\text{CDCl}_3$ , 300K) spectrum of **2**.



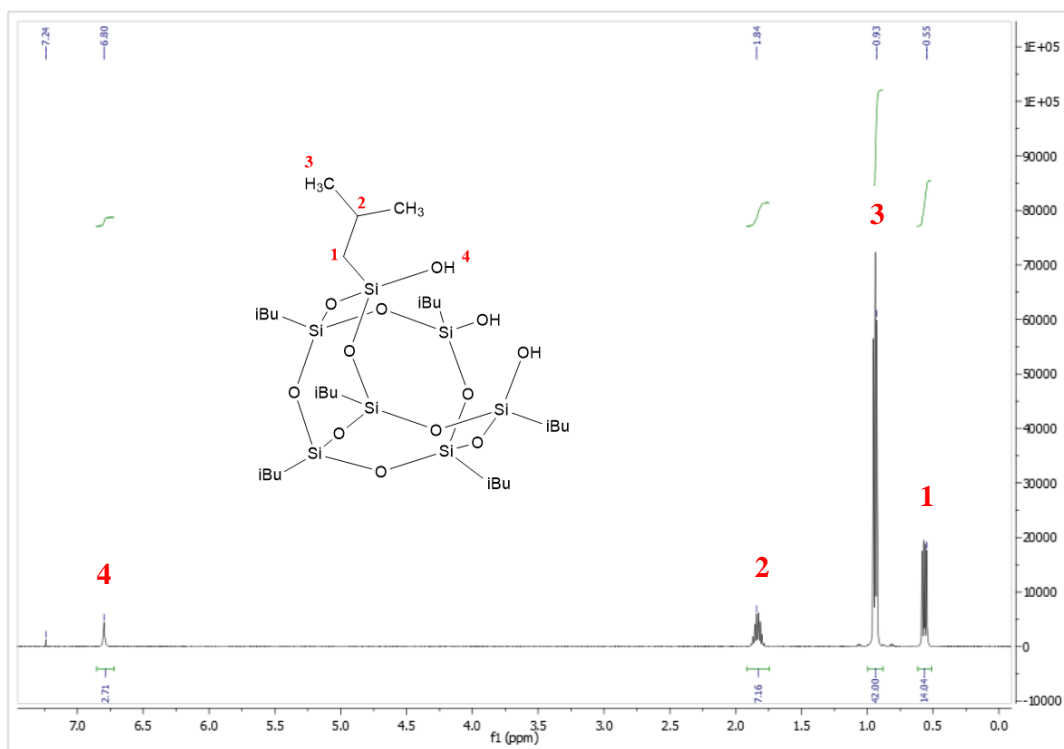
**Figure S9.** FT-IR spectrum (KBr pellet) of **2**.



**Figure S10.** MALDI-MS of **2**.



**Figure S11.** DOSY NMR of **2**.



**Figure S12.**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ , 300 K) spectrum of **3**.



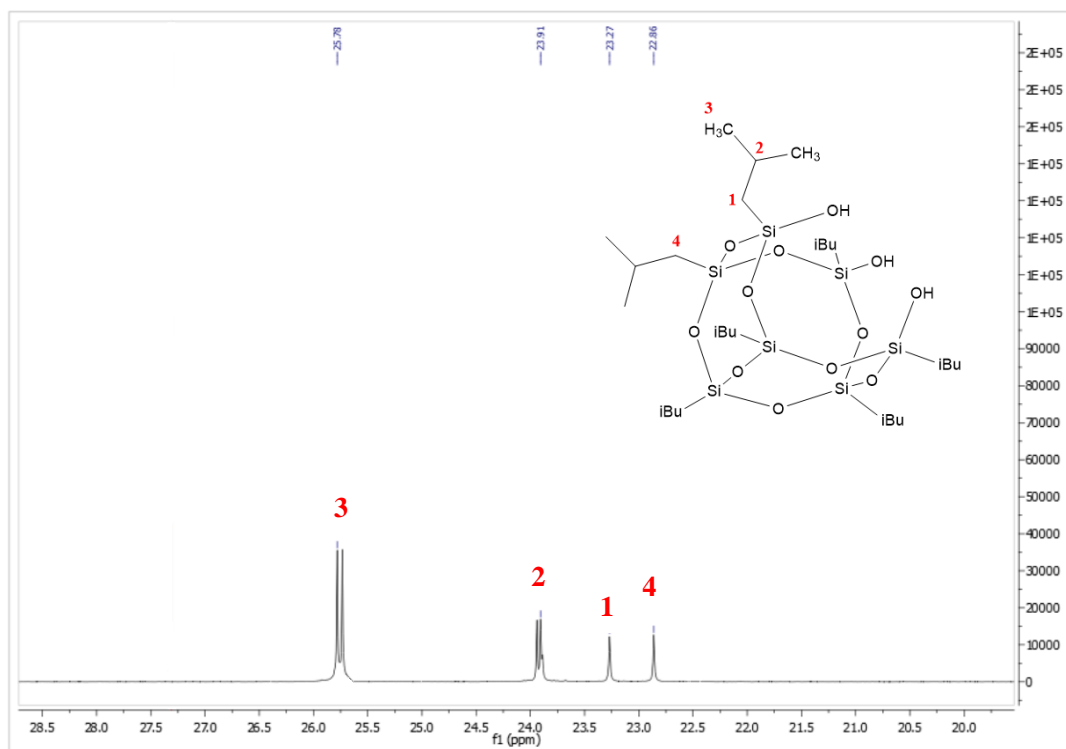


Figure S13.  $^{13}\text{C}$  NMR (500 MHz,  $\text{CDCl}_3$ , 300 K) spectrum of **3**.

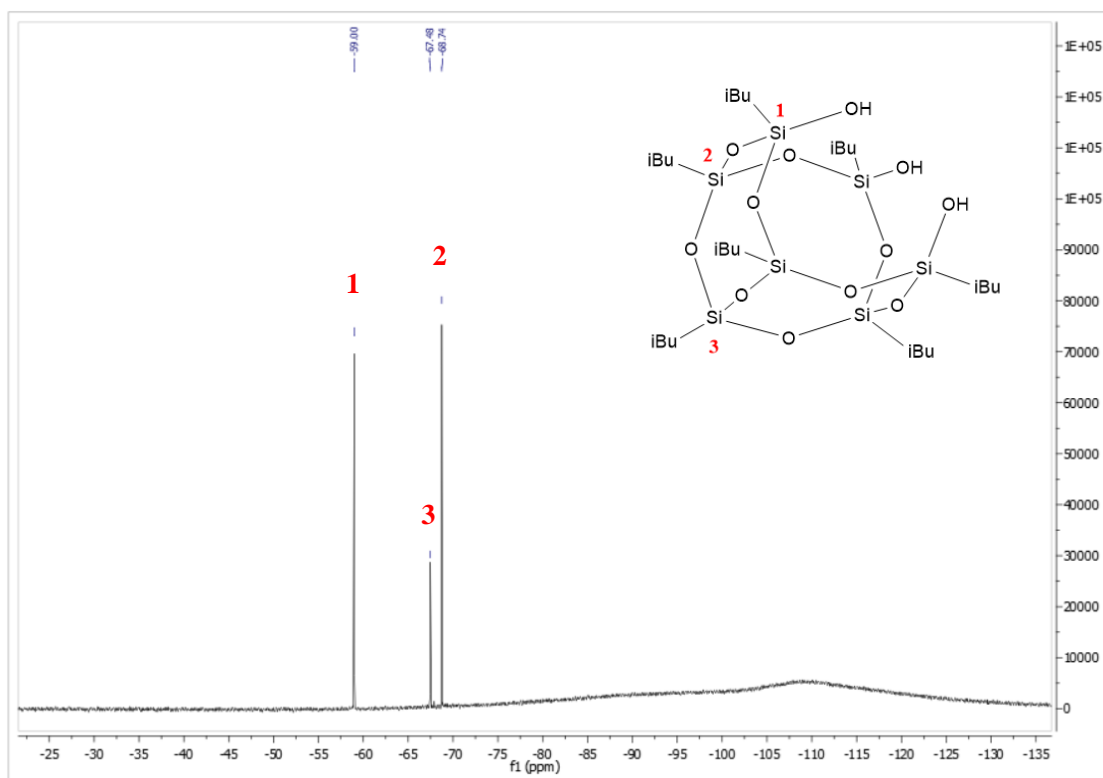
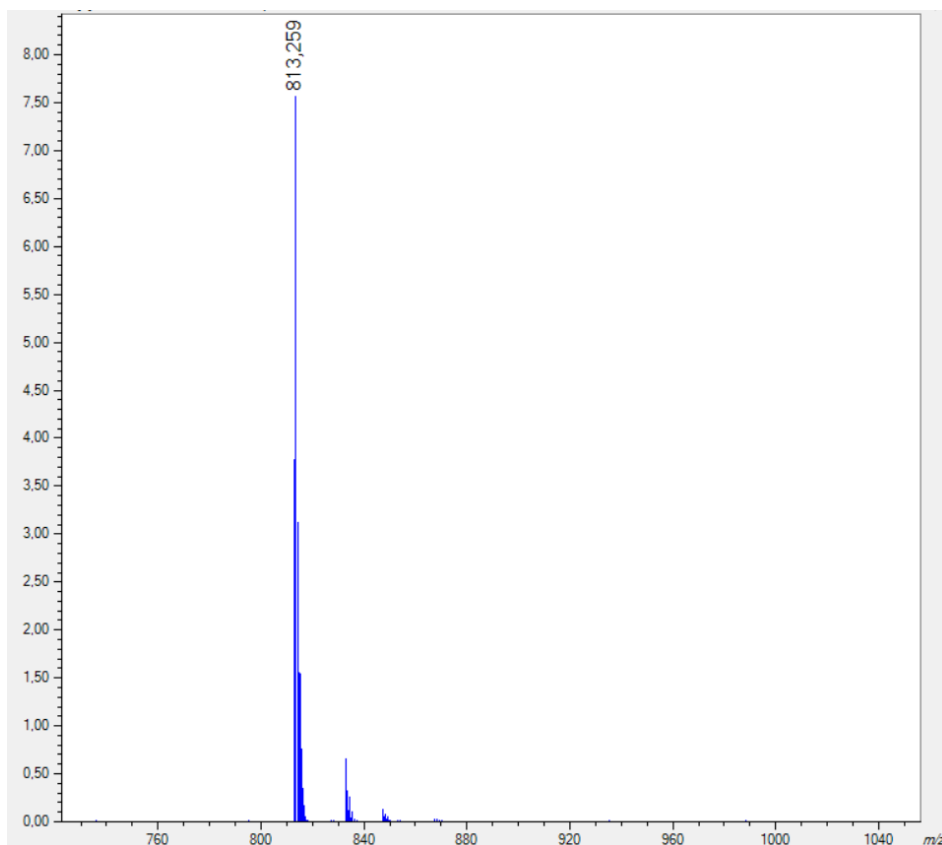
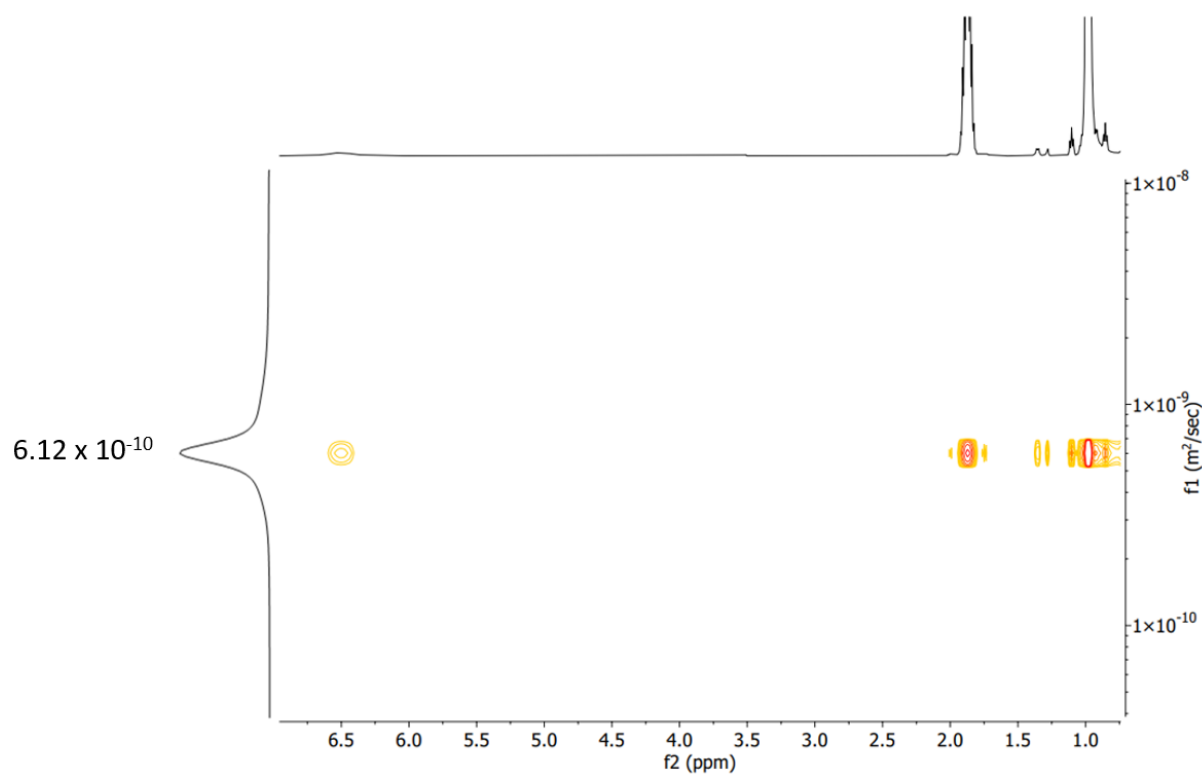


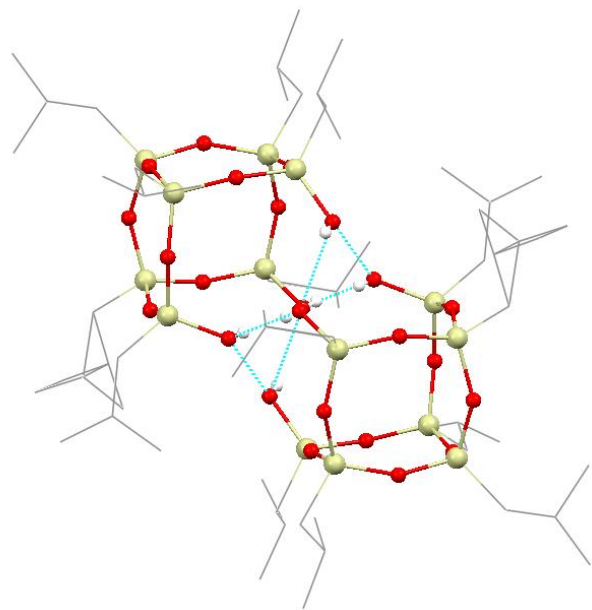
Figure S14.  $^{29}\text{Si}$  NMR (600 MHz,  $\text{CDCl}_3$ , 300K) spectrum of **3**.



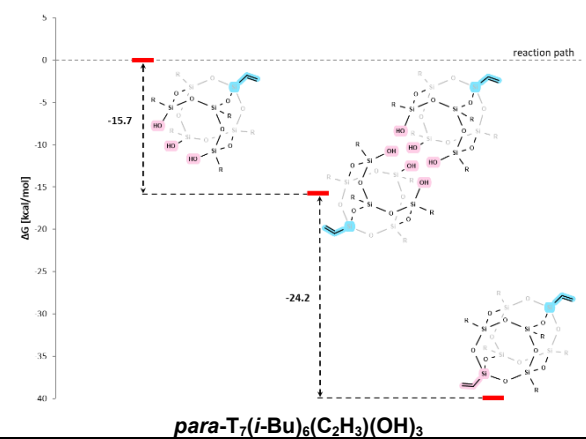
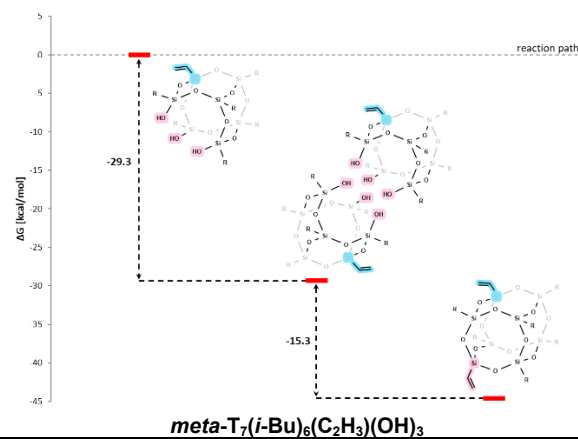
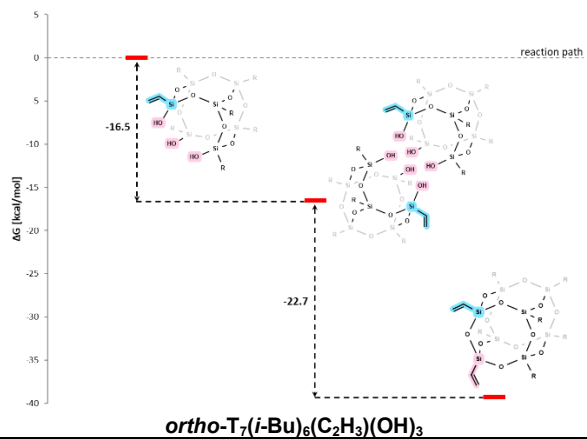
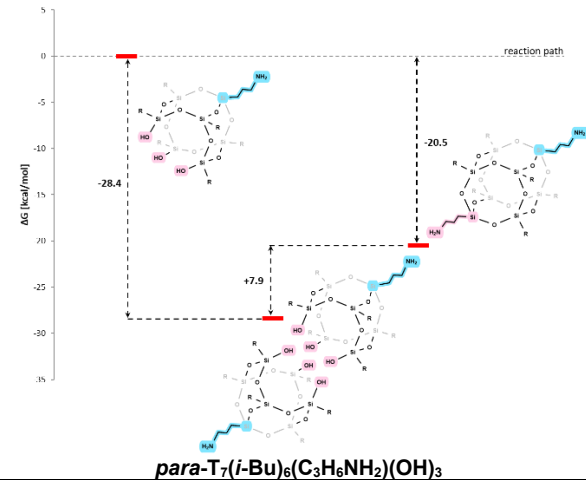
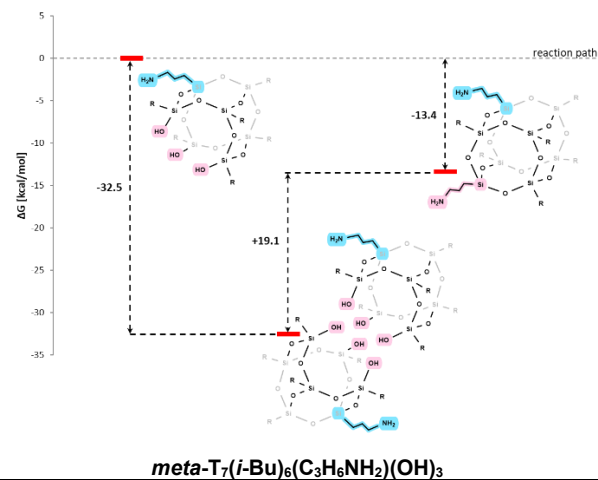
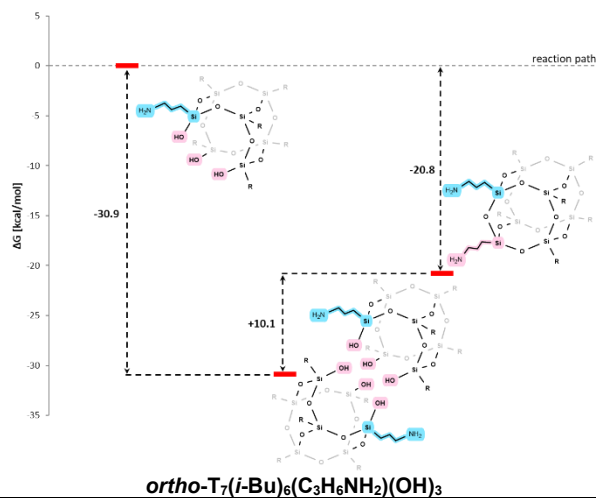
**Figure S15.** MALDI-MS of **3**.

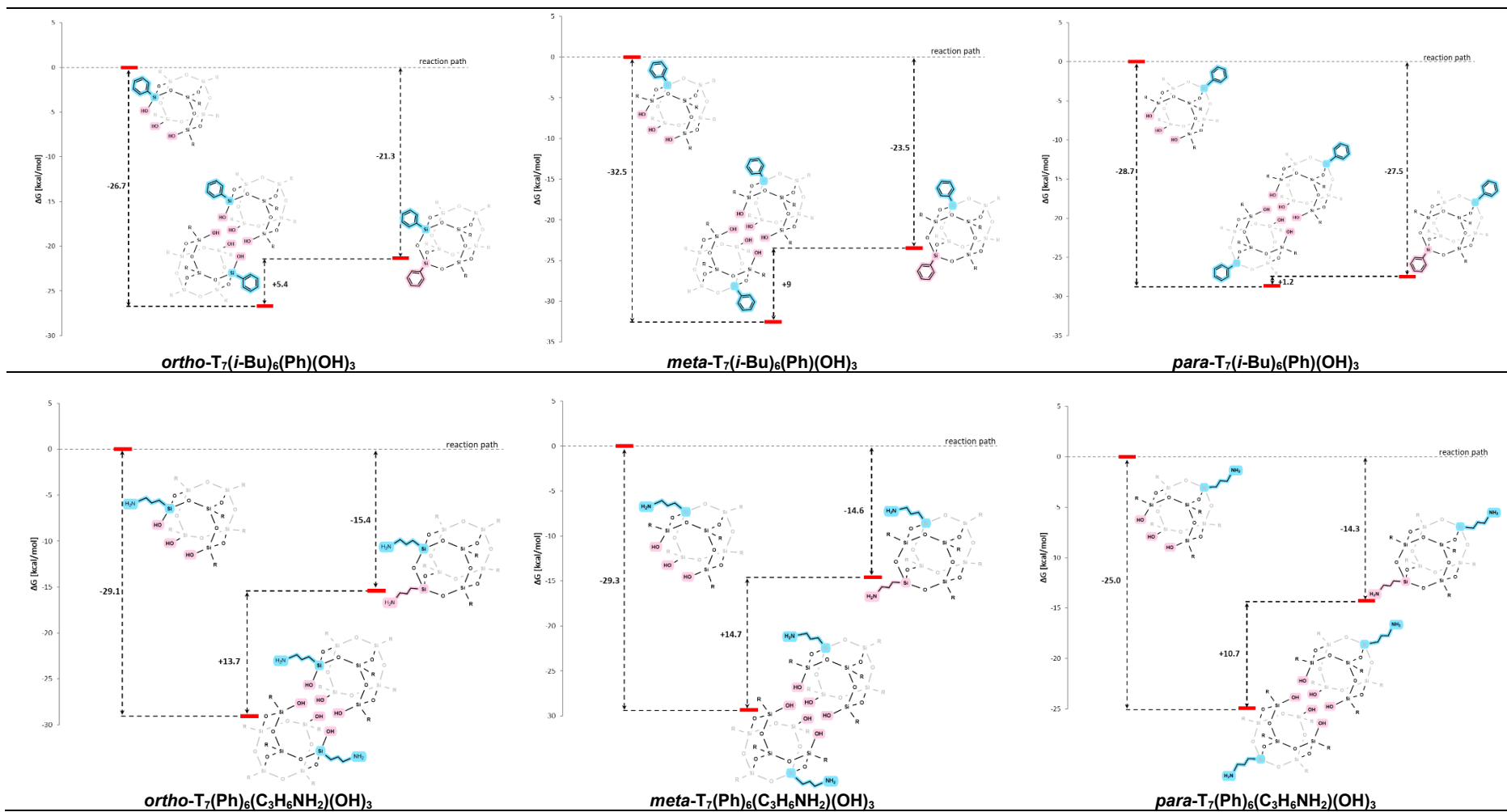


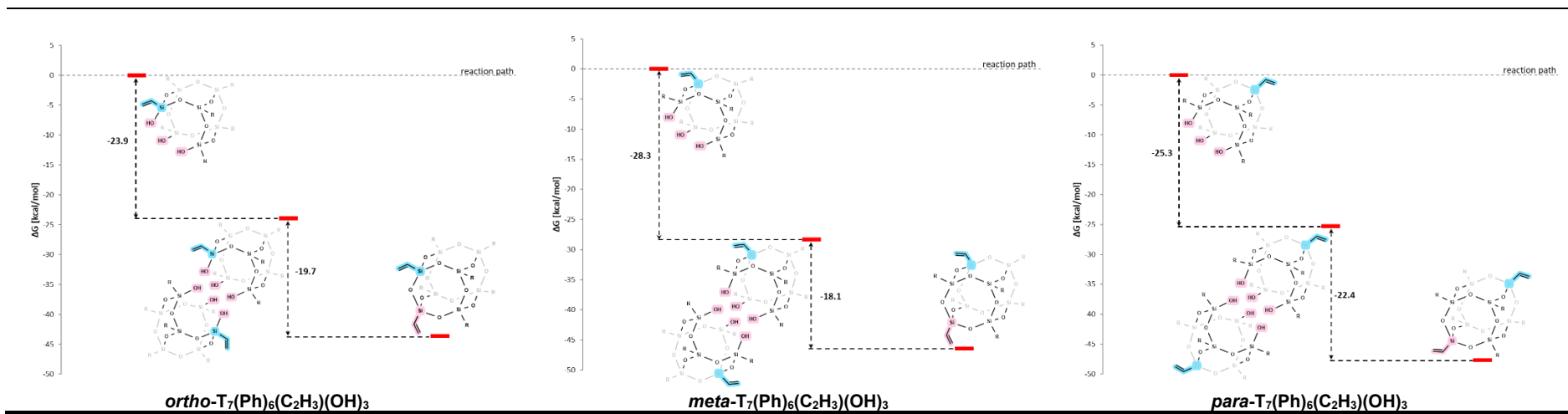
**Figure S16.** DOSY NMR of **3**.



**Figure S17.** Molecular structure of **3**.







**Figure S18.** Relative Gibbs free energies (kcal/mol) for DFT optimized POSS structures.

**Table S1.** The calculated HOMO-LUMO energy gaps and quantum chemical properties of POSS complexes. Data are given in eV.

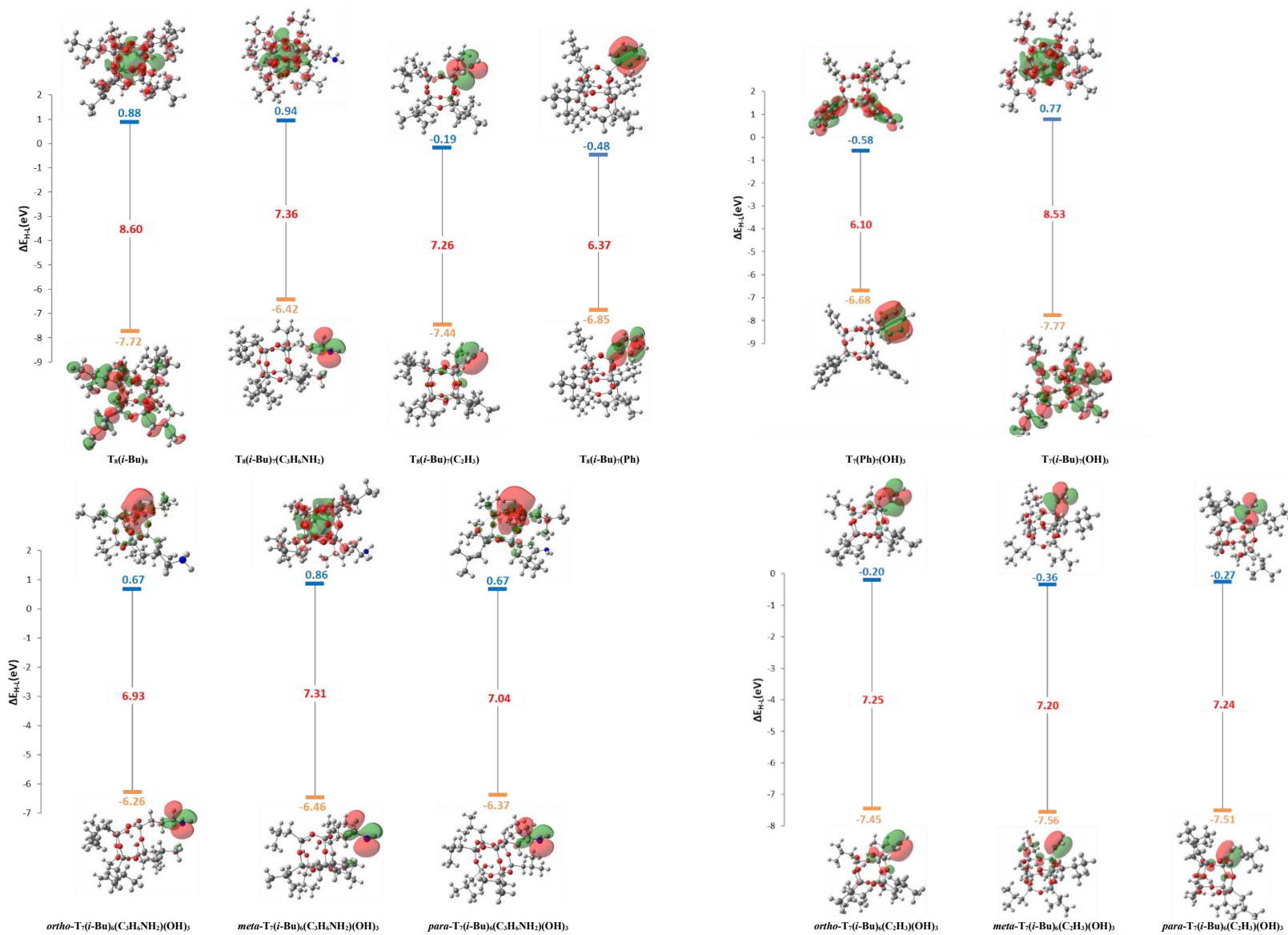
	Molecular Orbitals	Energy	Energy gap	Ionization potential (I)	Electron affinity (A)	Global hardness ( $\eta$ )	Electronegativity ( $\chi$ )	Chemical potential ( $\mu_e$ )	Global softness ( $\sigma^a$ )	Global electrophilicity ( $\omega$ )
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>8</sub></b>	HOMO	-7.72	8.60	7.72	-0.88	4.30	3.42	-3.42	0.12	1.36
	LUMO	0.88								
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>8</sub></b>	HOMO	-7.77	8.53	7.77	-0.77	4.27	3.50	-3.50	0.12	1.44
	LUMO	0.77								
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>7</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)</b>	HOMO	-6.42	7.36	6.42	-0.94	3.68	2.74	-2.74	0.14	1.02
	LUMO	0.94								
<b><i>ortho</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	HOMO	-6.26	6.93	6.26	-0.67	3.47	2.80	-2.80	0.14	1.13
	LUMO	0.67								
<b><i>meta</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	HOMO	-6.46	7.31	6.46	-0.86	3.66	2.80	-2.80	0.14	1.07
	LUMO	0.86								
<b><i>para</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	HOMO	-6.37	7.04	6.37	-0.67	3.52	2.85	-2.85	0.14	1.15
	LUMO	0.67								
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>7</sub>(C<sub>2</sub>H<sub>3</sub>)</b>	HOMO	-7.44	7.26	7.44	0.19	3.63	3.82	-3.82	0.14	2.01
	LUMO	-0.19								
<b><i>ortho</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	HOMO	-7.45	7.25	7.45	0.20	3.63	3.82	-3.82	0.14	2.02
	LUMO	-0.20								
<b><i>meta</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	HOMO	-7.56	7.20	7.56	0.36	3.60	3.96	-3.96	0.14	2.17
	LUMO	-0.36								
<b><i>para</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	HOMO	-7.51	7.24	7.51	0.27	3.62	3.89	-3.89	0.14	2.09
	LUMO	-0.27								
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>7</sub>(Ph)</b>	HOMO	-6.85	6.37	6.85	0.48	3.19	3.66	-3.66	0.16	2.10
	LUMO	-0.48								
<b><i>ortho</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub></b>	HOMO	-6.81	6.39	6.81	0.42	3.19	3.61	-3.61	0.16	2.04
	LUMO	-0.42								
<b><i>meta</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub></b>	HOMO	-6.92	6.37	6.92	0.55	3.18	3.73	-3.73	0.16	2.19
	LUMO	-0.55								
<b><i>para</i>-T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub></b>	HOMO	-6.86	6.38	6.86	0.48	3.19	3.67	-3.67	0.16	2.12
	LUMO	-0.48								
<b>T<sub>8</sub>(Ph)<sub>8</sub></b>	HOMO	-6.69	6.12	6.69	0.56	3.06	3.62	-3.62	0.16	2.14
	LUMO	-0.56								
<b>T<sub>7</sub>(Ph)<sub>7</sub>(H)<sub>3</sub></b>	HOMO	-6.68	6.10	6.68	0.58	3.05	3.63	-3.63	0.16	2.17
	LUMO	-0.58								

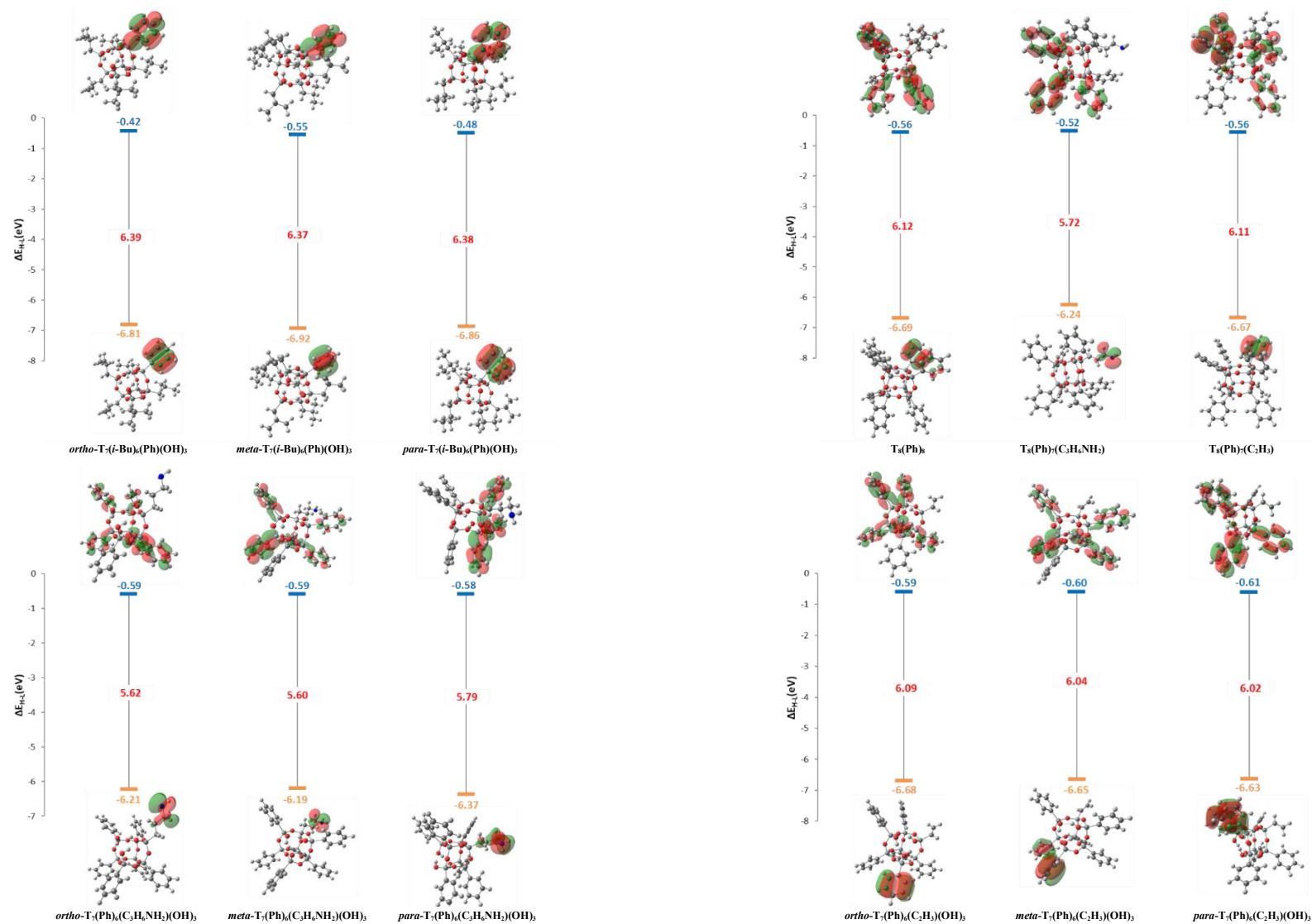
<b>T<sub>8</sub>(Ph)<sub>7</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)</b>	HOMO	-6.24	5.72	6.24	0.52	2.86	3.38	-3.38	0.17	2.00
	LUMO	-0.52								
<i>ortho</i> -T <sub>7</sub> (Ph) <sub>6</sub> (C <sub>3</sub> H <sub>6</sub> NH <sub>2</sub> )(OH) <sub>3</sub>	HOMO	-6.21	5.62	6.21	0.59	2.81	3.40	-3.40	0.18	2.06
	LUMO	-0.59								
<i>meta</i> -T <sub>7</sub> (Ph) <sub>6</sub> (C <sub>3</sub> H <sub>6</sub> NH <sub>2</sub> )(OH) <sub>3</sub>	HOMO	-6.19	5.60	6.19	0.59	2.80	3.39	-3.39	0.18	2.05
	LUMO	-0.59								
<i>para</i> -T <sub>7</sub> (Ph) <sub>6</sub> (C <sub>3</sub> H <sub>6</sub> NH <sub>2</sub> )(OH) <sub>3</sub>	HOMO	-6.37	5.79	6.37	0.58	2.89	3.47	-3.47	0.17	2.09
	LUMO	-0.58								
<b>T<sub>8</sub>(Ph)<sub>7</sub>(C<sub>2</sub>H<sub>3</sub>)</b>	HOMO	-6.67	6.11	6.67	0.56	3.06	3.61	-3.61	0.16	2.13
	LUMO	-0.56								
<i>ortho</i> -T <sub>7</sub> (Ph) <sub>6</sub> (C <sub>2</sub> H <sub>3</sub> )(OH) <sub>3</sub>	HOMO	-6.68	6.09	6.68	0.59	3.04	3.64	-3.64	0.16	2.17
	LUMO	-0.59								
<i>meta</i> -T <sub>7</sub> (Ph) <sub>6</sub> (C <sub>2</sub> H <sub>3</sub> )(OH) <sub>3</sub>	HOMO	-6.65	6.04	6.65	0.60	3.02	3.62	-3.62	0.17	2.17
	LUMO	-0.60								
<i>para</i> -T <sub>7</sub> (Ph) <sub>6</sub> (C <sub>2</sub> H <sub>3</sub> )(OH) <sub>3</sub>	HOMO	-6.63	6.02	6.63	0.61	3.01	3.62	-3.62	0.17	2.18
	LUMO	-0.61								

<sup>a</sup> data are given in eV<sup>-1</sup>

$$I = -E_{HOMO}; A = -E_{LUMO}; \eta = \frac{I-A}{2}; \chi = \frac{I+A}{2}; \mu_c = -\chi; \sigma = \frac{1}{2\eta}; \omega = \frac{\mu^2}{2\eta}$$







**Figure S19.** The calculated HOMO (orange) and LUMO (blue) energy levels and their visualization, and HOMO-LUMO energy gaps (red) of optimized POSS models. HOMO – orange; LUMO – blue;  $\Delta E_{H-L}$  – red

**Table S2.** Energies of calculated POSS models. Data are given in kcal/mol.

	$E_{\text{tot}}$	$E_{\text{tot}} + \text{ZPE}$	H	G	$S^a$	
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>8</sub></b>	-2813122	-2812459	-2812416	-2812524	0.365	
<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>7</sub>(OH)<sub>3</sub></b>	-2533419	-2532815	-2532775	-2532877	0.344	
<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>7</sub>(OH)<sub>3</sub>]<sub>2</sub></b>	-5066879	-5065669	-5065589	-5065776	0.625	
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>7</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)</b>	-2823175	-2822518	-2822475	-2822584	0.365	
<i>ortho</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	-2543470	-2542872	-2542833	-2542935	0.342
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5086989	-5085792	-5085712	-5085901	0.631
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)<sub>2</sub></b>	-2833231	-2832580	-2832537	-2832646	0.364
<i>meta</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	-2543473	-2542875	-2542836	-2542938	0.344
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5086997	-5085800	-5085721	-5085909	0.631
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)<sub>2</sub></b>	-2833229	-2832579	-2832536	-2832646	0.368
<i>para</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	-2543470	-2542873	-2542833	-2542935	0.340
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5086989	-5085792	-5085713	-5085898	0.619
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)<sub>2</sub></b>	-2833229	-2832579	-2832536	-2832645	0.367
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>7</sub>(C<sub>2</sub>H<sub>3</sub>)</b>	-2762997	-2762384	-2762343	-2762449	0.358	
<i>ortho</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	-2483295	-2482742	-2482704	-2482805	0.338
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-4966633	-4965524	-4965448	-4965626	0.595
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)<sub>2</sub></b>	-2712874	-2712312	-2712273	-2712373	0.337
<i>meta</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	-2483297	-2482743	-2482706	-2482803	0.326
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-4966644	-4965534	-4965460	-4965635	0.589
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)<sub>2</sub></b>	-2712875	-2712312	-2712273	-2712374	0.339
<i>para</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	-2483297	-2482744	-2482706	-2482804	0.330
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-4966635	-4965525	-4965450	-4965625	0.586
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)<sub>2</sub></b>	-2712874	-2712312	-2712273	-2712373	0.338
<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>7</sub>(Ph)</b>	-2859426	-2858782	-2858740	-2858849	0.365	
<i>ortho</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub></b>	-2579723	-2579139	-2579100	-2579202	0.344
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub>]<sub>2</sub></b>	-5159491	-5158322	-5158243	-5158431	0.631
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)<sub>2</sub></b>	-2905731	-2905108	-2905066	-2905175	0.364

<i>meta</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub></b>	-2579723	-2579139	-2579100	-2579203	0.344
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub>]<sub>2</sub></b>	-5159503	-5158333	-5158255	-5158438	0.613
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)<sub>2</sub></b>	-2905731	-2905108	-2905066	-2905176	0.369
<i>para</i>	<b>T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub></b>	-2579718	-2579135	-2579095	-2579198	0.346
	<b>[T<sub>7</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)(OH)<sub>3</sub>]<sub>2</sub></b>	-5159491	-5158320	-5158243	-5158425	0.613
	<b>T<sub>8</sub>(<i>i</i>-Bu)<sub>6</sub>(Ph)<sub>2</sub></b>	-2905731	-2905107	-2905065	-2905174	0.363
	<b>T<sub>8</sub>(Ph)<sub>8</sub></b>	-3183560	-3183058	-3183018	-3183126	0.363
	<b>T<sub>7</sub>(Ph)<sub>7</sub>(OH)<sub>3</sub></b>	-2857554	-2857091	-2857054	-2857156	0.339
	<b>[T<sub>7</sub>(Ph)<sub>7</sub>(OH)<sub>3</sub>]<sub>2</sub></b>	-5715163	-5714235	-5714163	-5714342	0.599
	<b>T<sub>8</sub>(Ph)<sub>7</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)</b>	-3147310	-3146794	-3146755	-3146863	0.363
<i>ortho</i>	<b>T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	-2821304	-2820827	-2820790	-2820892	0.341
	<b>[T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5642666	-5641710	-5641637	-5641813	0.590
	<b>T<sub>8</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)<sub>2</sub></b>	-3111059	-3110530	-3110489	-3110600	0.372
<i>meta</i>	<b>T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	-2821304	-2820827	-2820790	-2820891	0.339
	<b>[T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5642662	-5641706	-5641633	-5641812	0.599
	<b>T<sub>8</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)<sub>2</sub></b>	-3111062	-3110532	-3110492	-3110599	0.358
<i>para</i>	<b>T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub></b>	-2821305	-2820828	-2820791	-2820891	0.336
	<b>[T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5642655	-5641700	-5641627	-5641808	0.608
	<b>T<sub>8</sub>(Ph)<sub>6</sub>(C<sub>3</sub>H<sub>6</sub>NH<sub>2</sub>)<sub>2</sub></b>	-3111060	-3110531	-3110491	-3110599	0.363
	<b>T<sub>8</sub>(Ph)<sub>7</sub>(C<sub>2</sub>H<sub>3</sub>)</b>	-3087132	-3086660	-3086623	-3086726	0.348
<i>ortho</i>	<b>T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	-2761127	-2760694	-2760660	-2760756	0.323
	<b>[T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5522304	-5521437	-5521368	-5521535	0.563
	<b>T<sub>8</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)<sub>2</sub></b>	-2990705	-2990263	-2990227	-2990327	0.333
<i>meta</i>	<b>T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	-2761127	-2760694	-2760659	-2760755	0.321
	<b>[T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5522300	-5521433	-5521364	-5521538	0.583
	<b>T<sub>8</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)<sub>2</sub></b>	-2990705	-2990263	-2990227	-2990327	0.335
<i>para</i>	<b>T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub></b>	-2761127	-2760694	-2760660	-2760754	0.317
	<b>[T<sub>7</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)(OH)<sub>3</sub>]<sub>2</sub></b>	-5522300	-5521433	-5521363	-5521534	0.573
	<b>T<sub>8</sub>(Ph)<sub>6</sub>(C<sub>2</sub>H<sub>3</sub>)<sub>2</sub></b>	-2990705	-2990263	-2990227	-2990327	0.336

<sup>a</sup> data are given in kcal/mol