

Supporting Information

Rational design of efficient photosensitizers based on cyclometalated iridium(III) complexes with 2-arylbenzimidazole and aromatic 1,3-diketone ligands

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1. Complexes studied in this work
2. NMR and mass spectra.
3. X-ray data.
4. Redox and optical properties
5. Computational details
6. Photovoltaic measurements

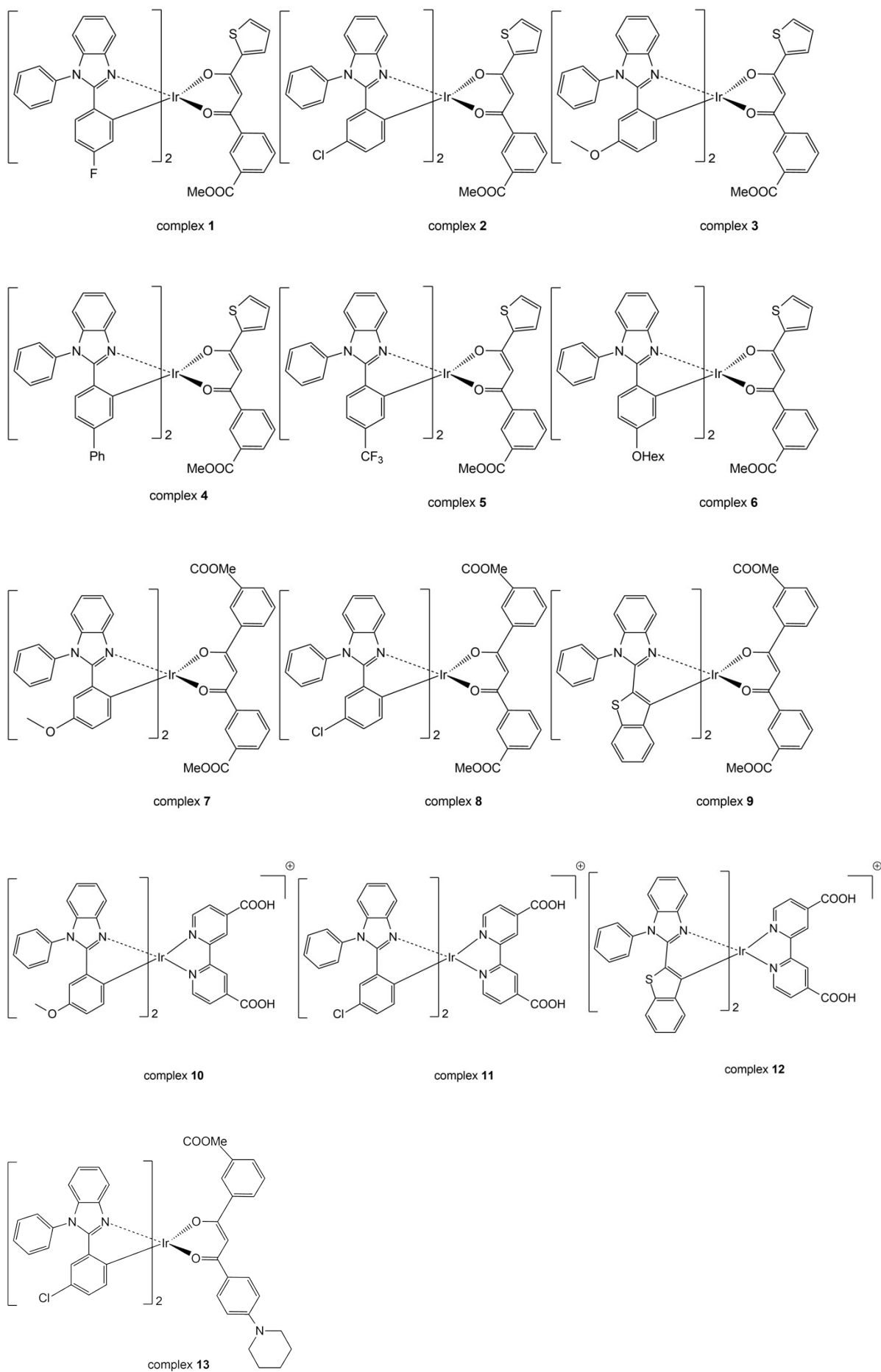


Figure S1. Synthesized complexes 1–13

1. NMR and mass spectra.

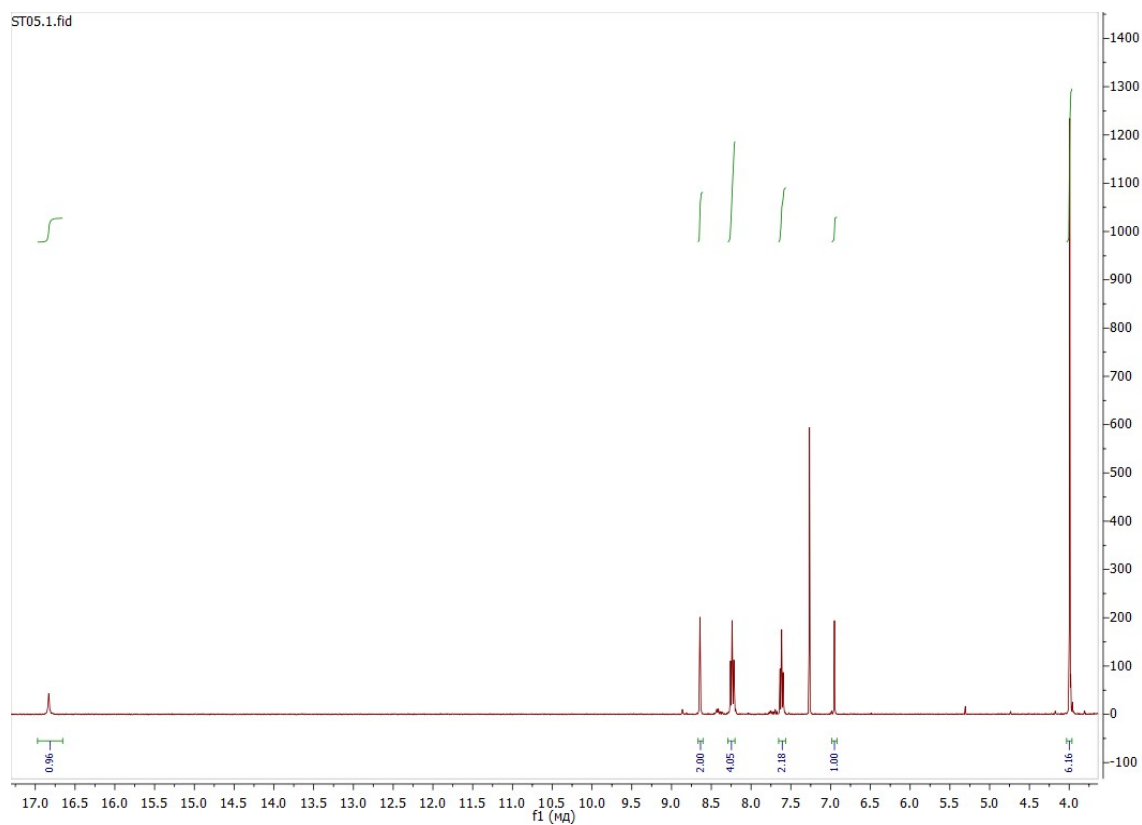


Figure S2. ^1H NMR spectrum of L_2 (400 MHz, 298K, CDCl_3).

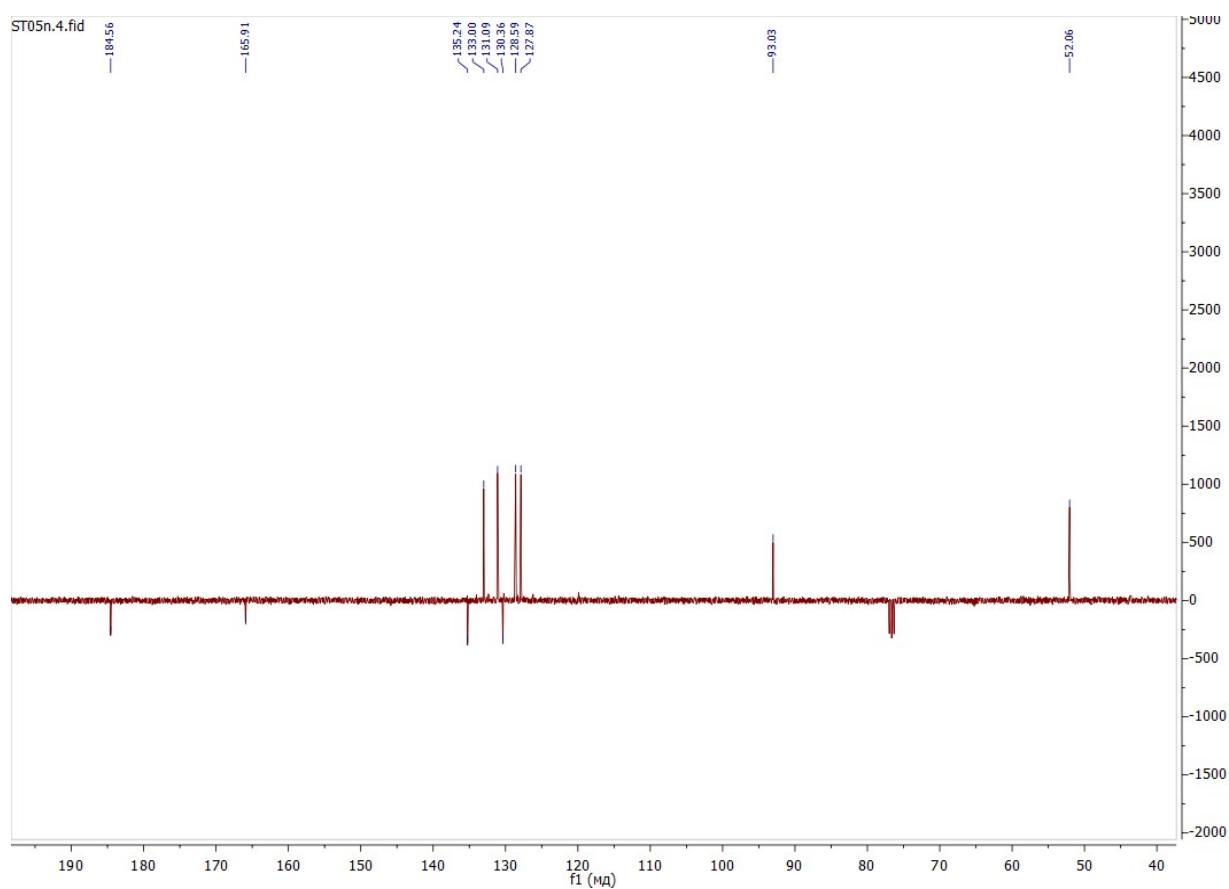


Figure S3. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (APT) of L_2 (101 MHz, 298K, CDCl_3).

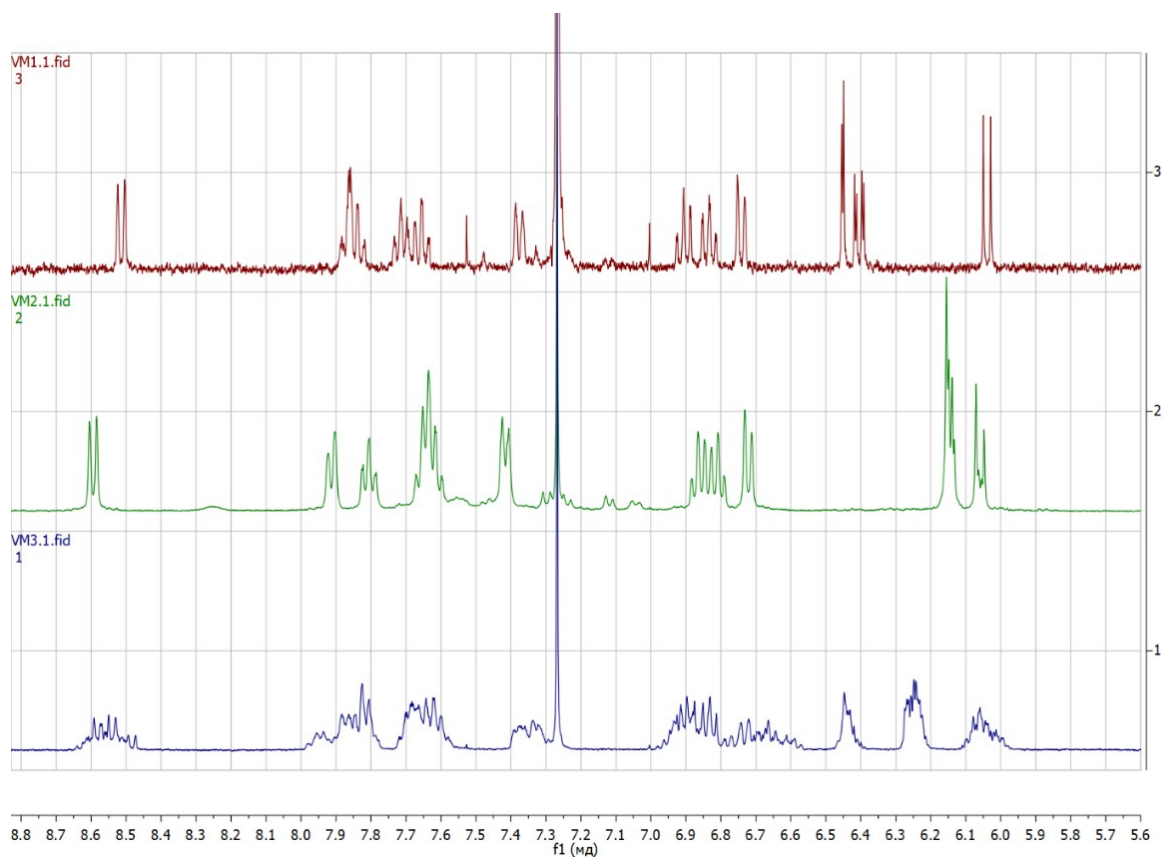


Fig.S4. ¹H NMR spectra of dimeric compounds with 3cbi (red), 3mbi (green) and 3fbi (blue) (400 MHz, 298K, CDCl₃).

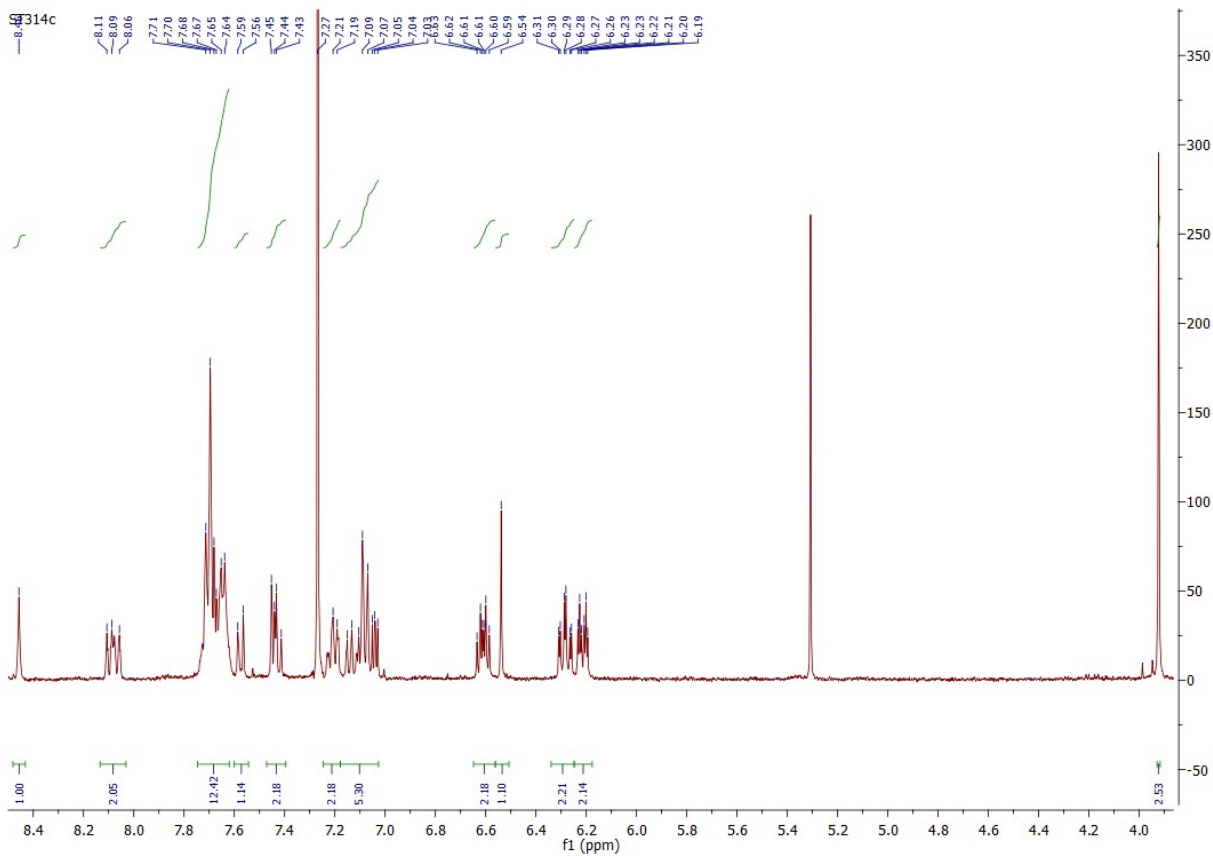


Figure S5. ¹H NMR spectrum of 1 (400 MHz, 298K, CDCl₃).

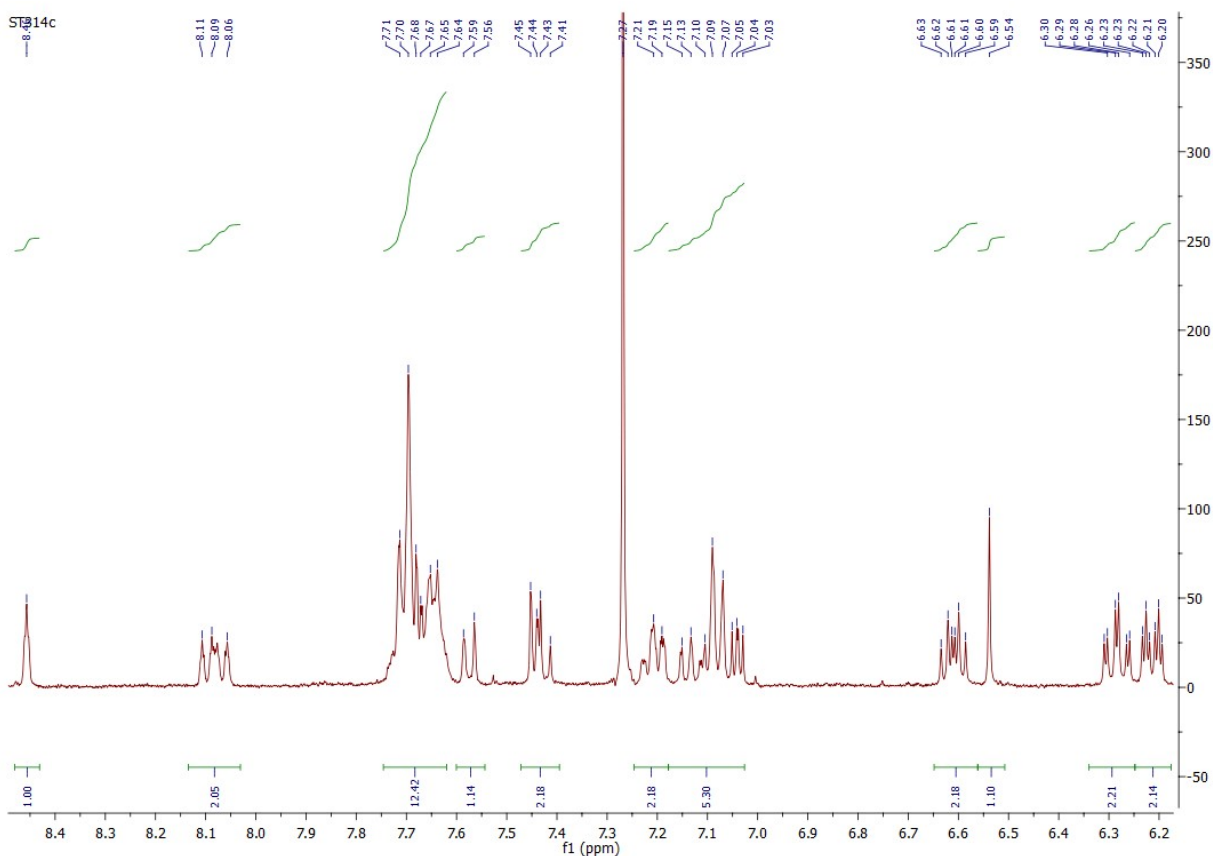


Figure S6. Aromatic region of ^1H NMR spectrum of 1 (400 MHz, 298K, CDCl_3).

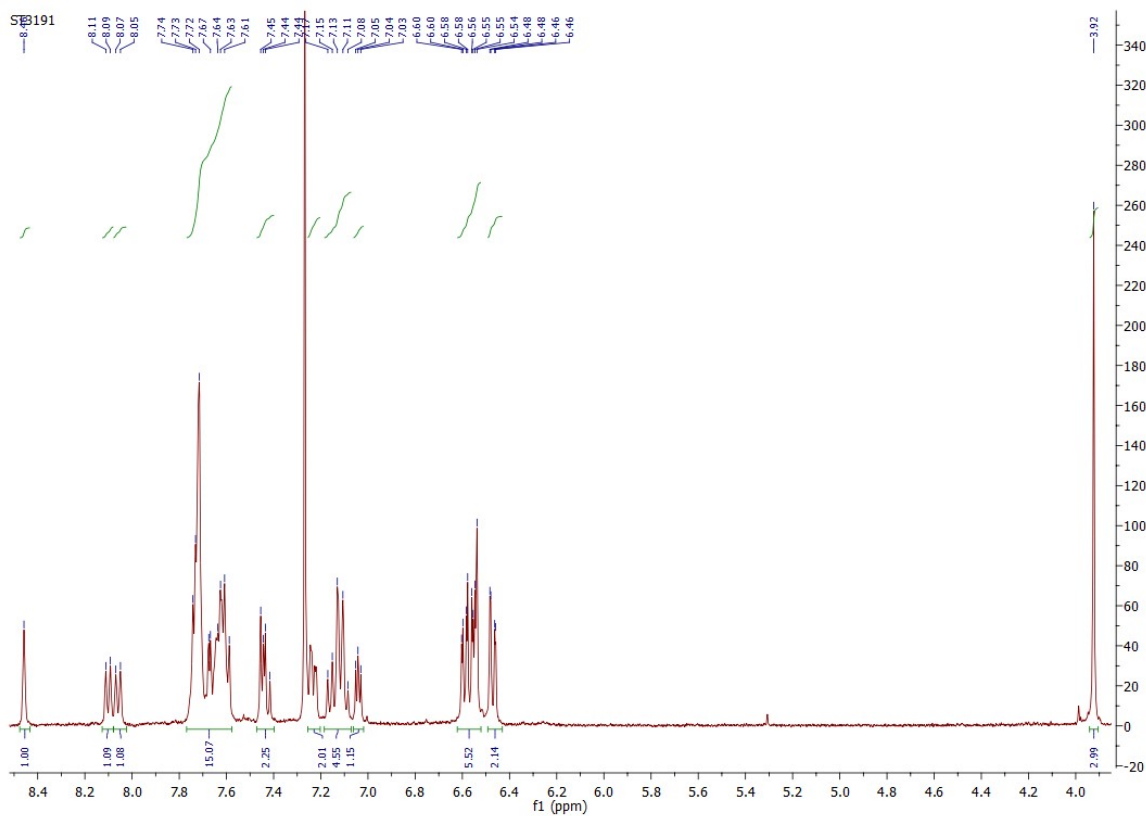


Figure S7. ^1H NMR spectrum of 2 (400 MHz, 298K, CDCl_3).

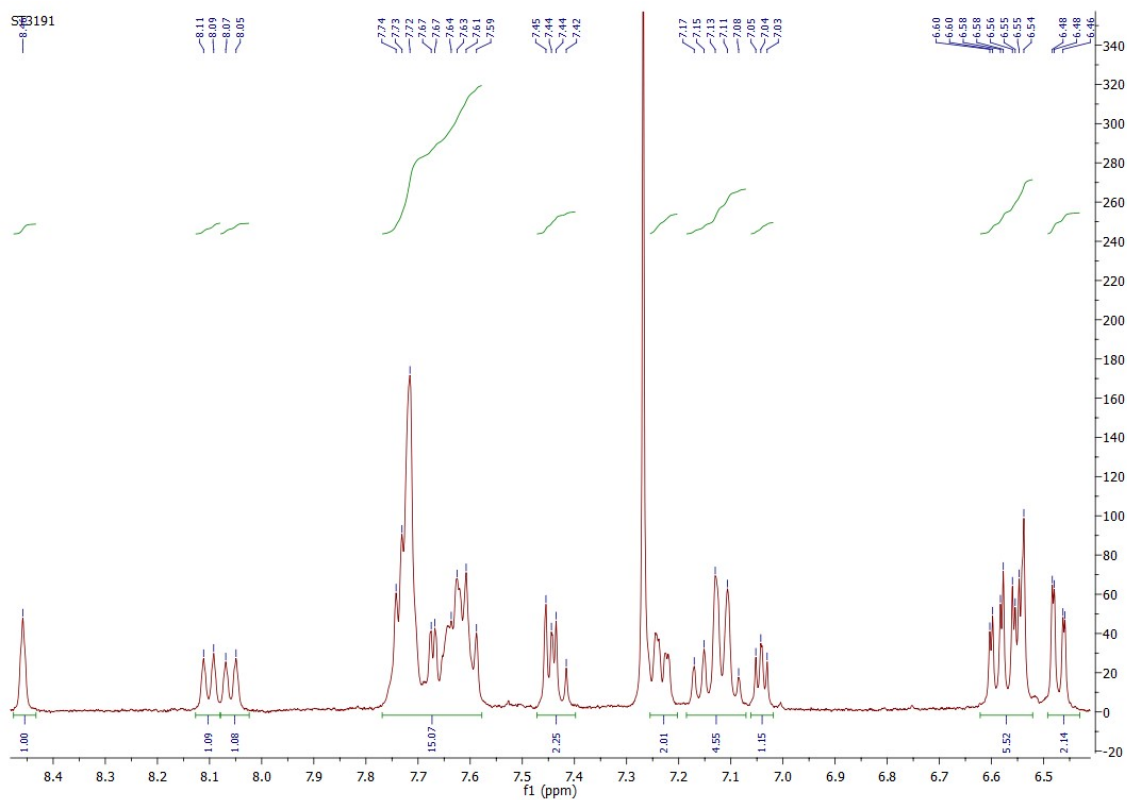


Figure S8. Aromatic region of ^1H NMR spectrum of **2** (400 MHz, 298K, CDCl_3).

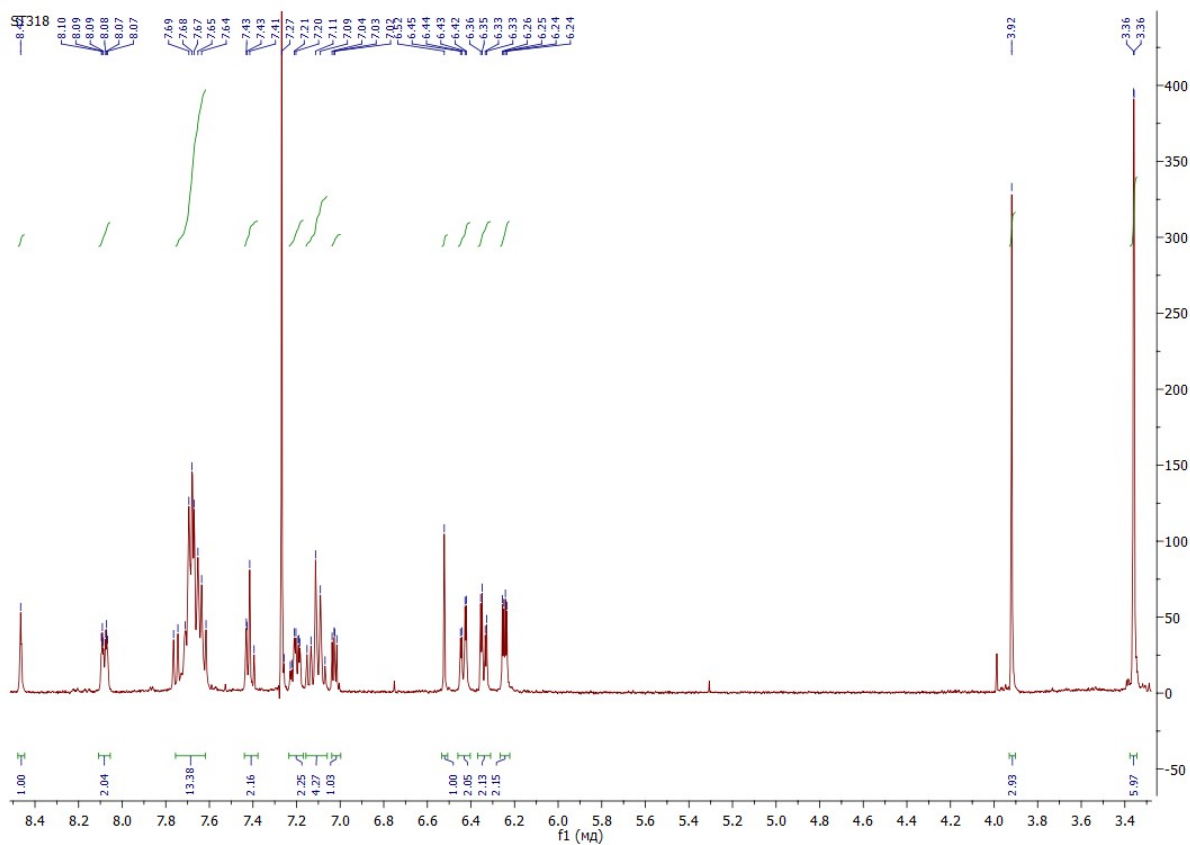


Figure S9. ^1H NMR spectrum of **3** (400 MHz, 298K, CDCl_3).

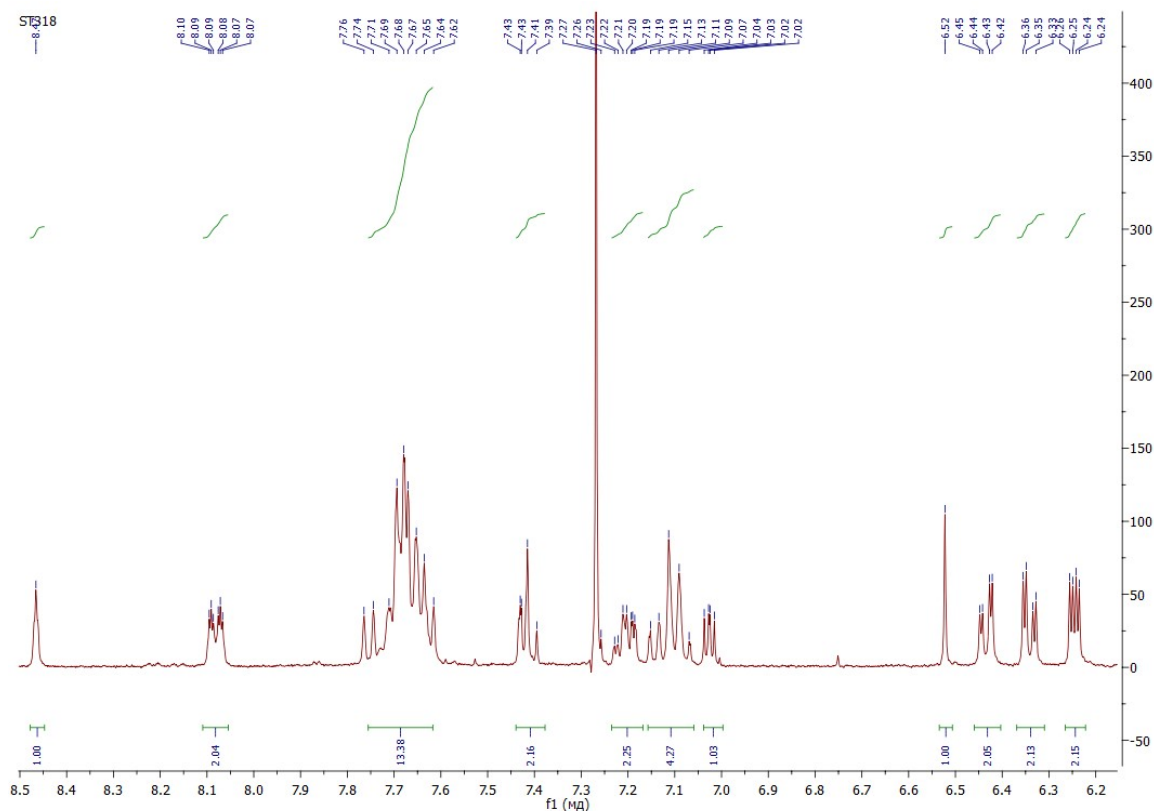


Figure S10. Aromatic region of ^1H NMR spectrum of **3** (400 MHz, 298K, CDCl_3).

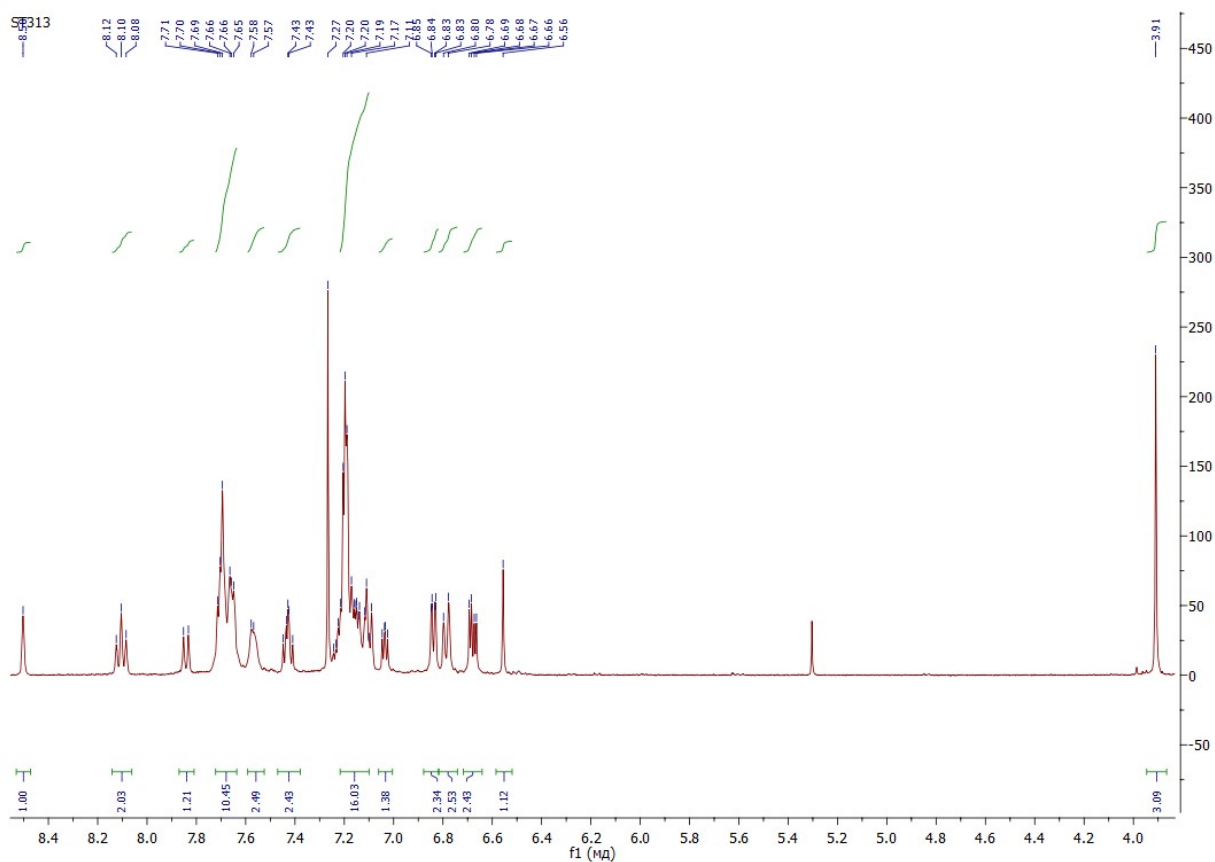


Figure S11. ^1H NMR spectrum of **4** (400 MHz, 298K, CDCl_3).

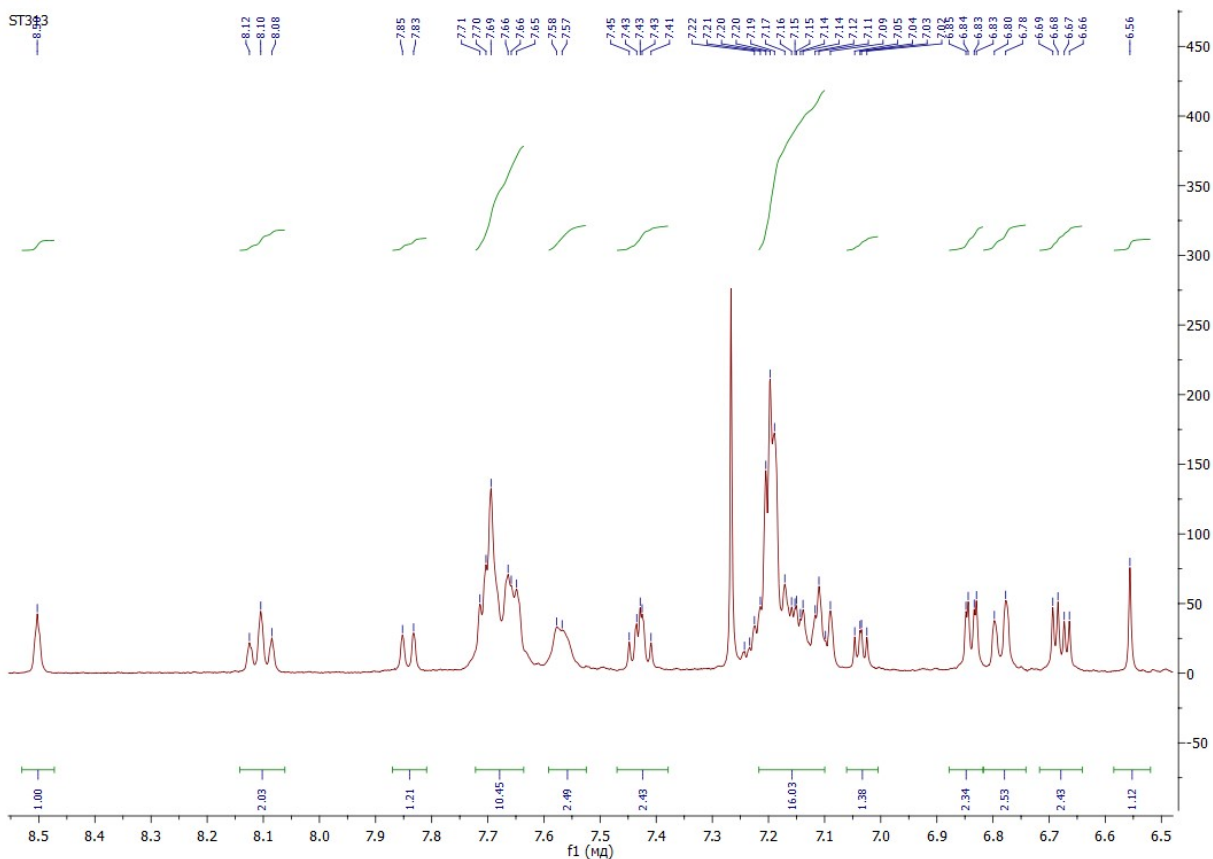


Figure S12. Aromatic region of ^1H NMR spectrum of **4** (400 MHz, 298K, CDCl_3).

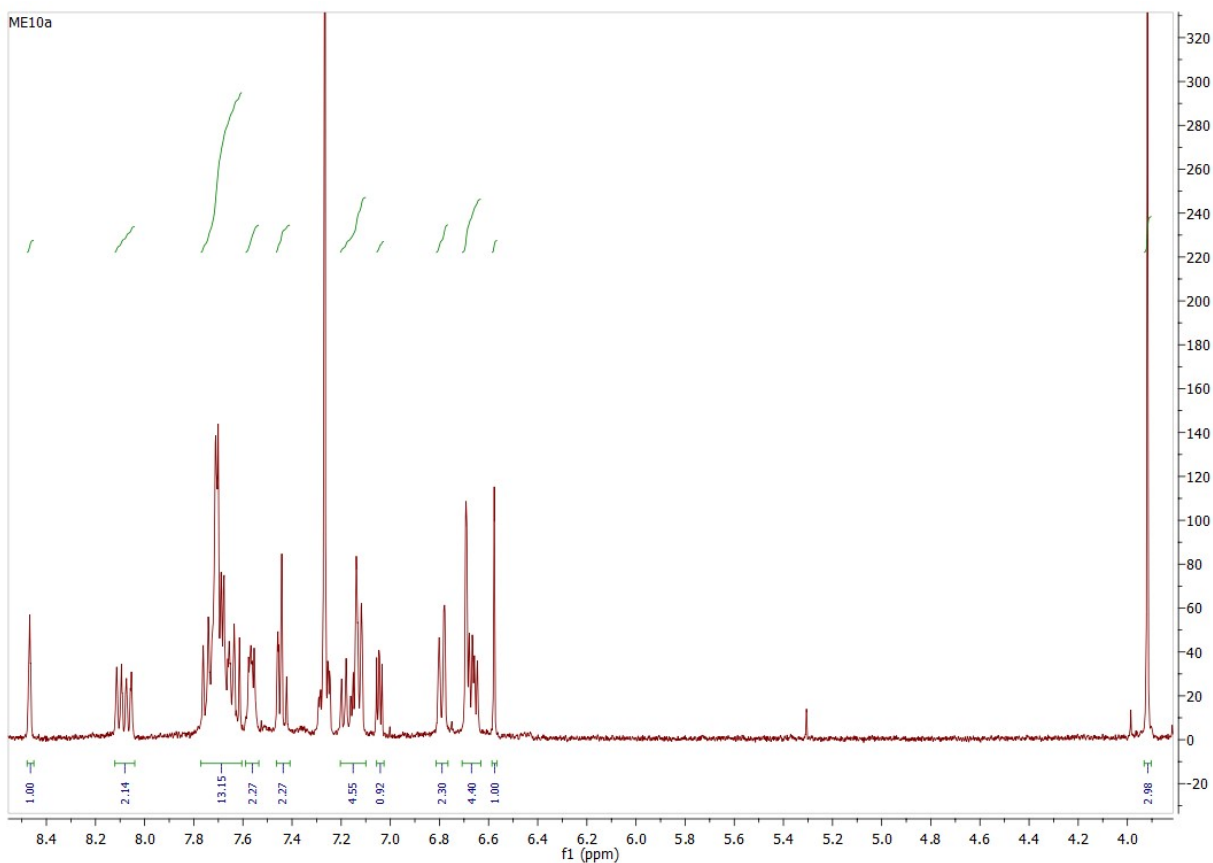


Figure S13. ^1H NMR spectrum of **5** (400 MHz, 298K, CDCl_3).

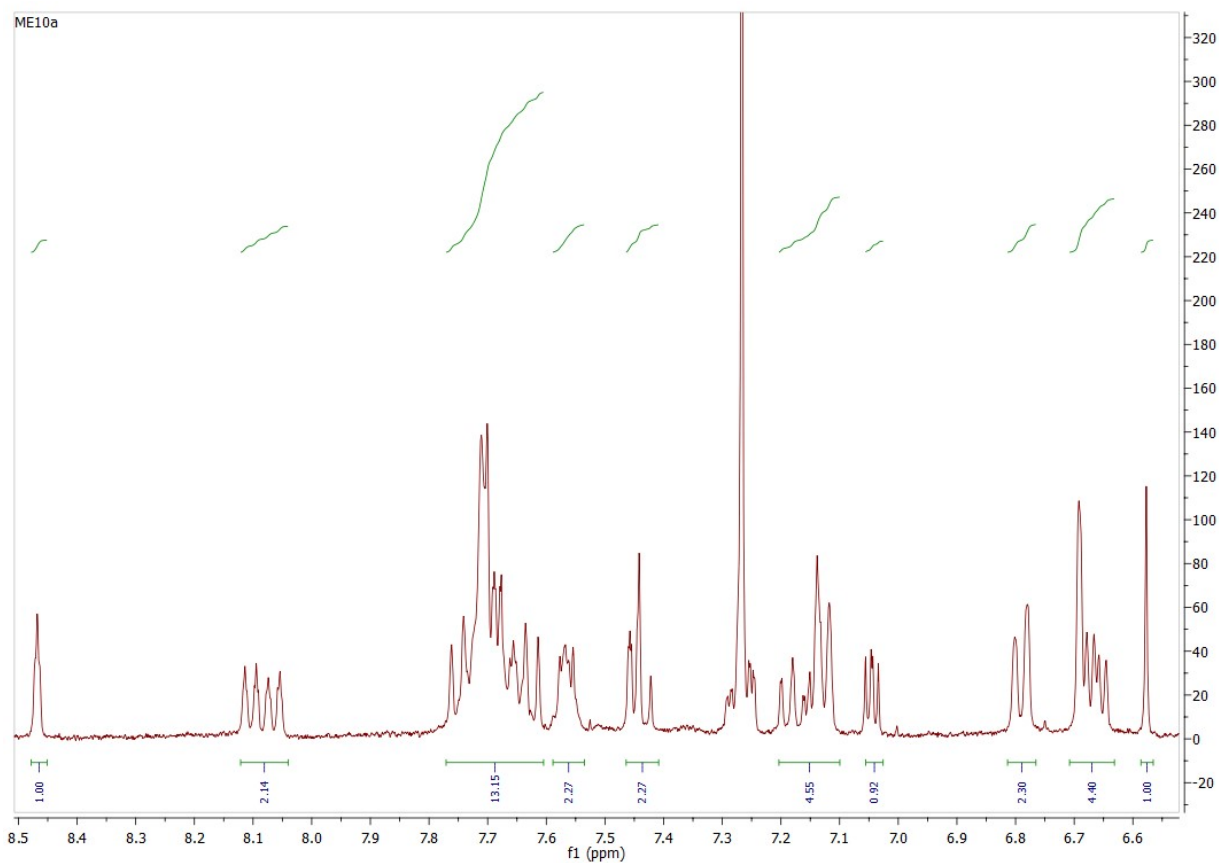


Figure S14. Aromatic region of ^1H NMR spectrum of **5** (400 MHz, 298K, CDCl_3).

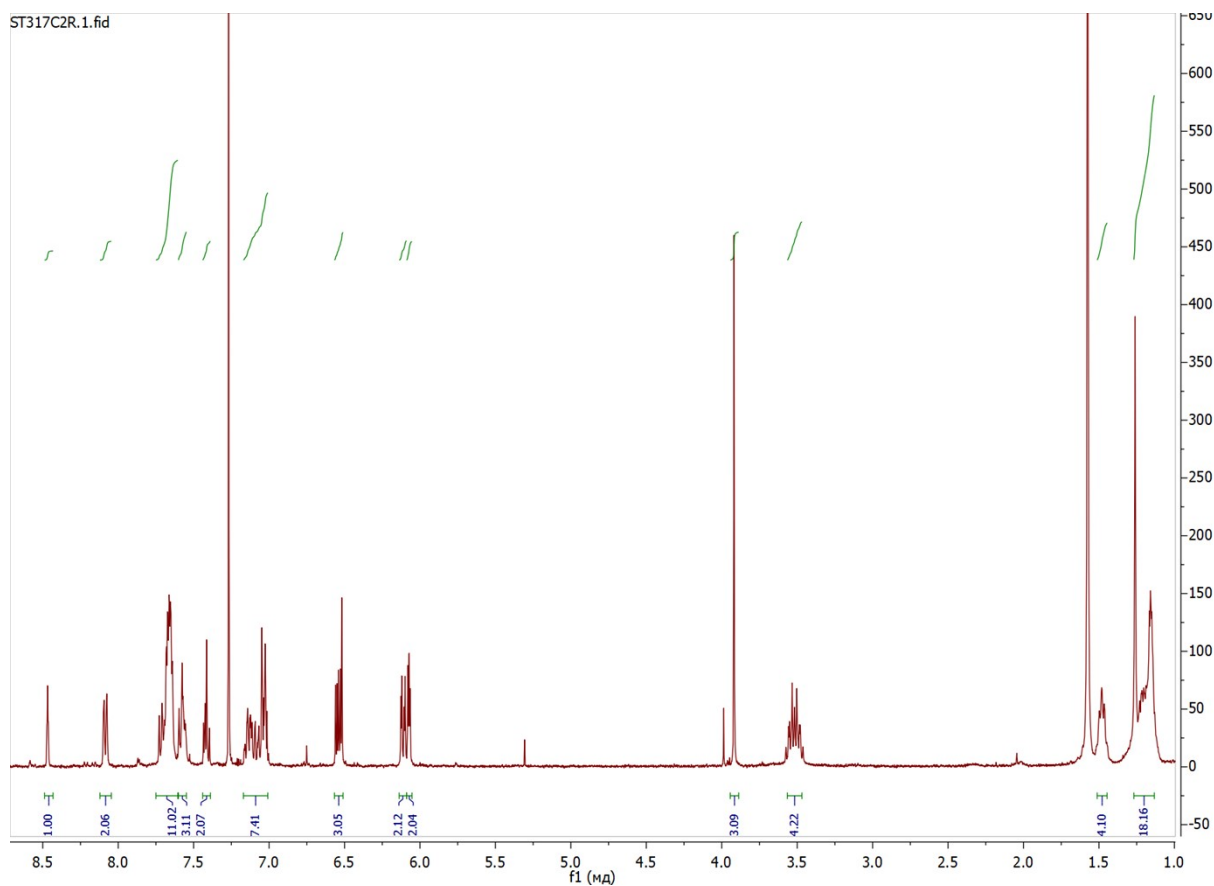
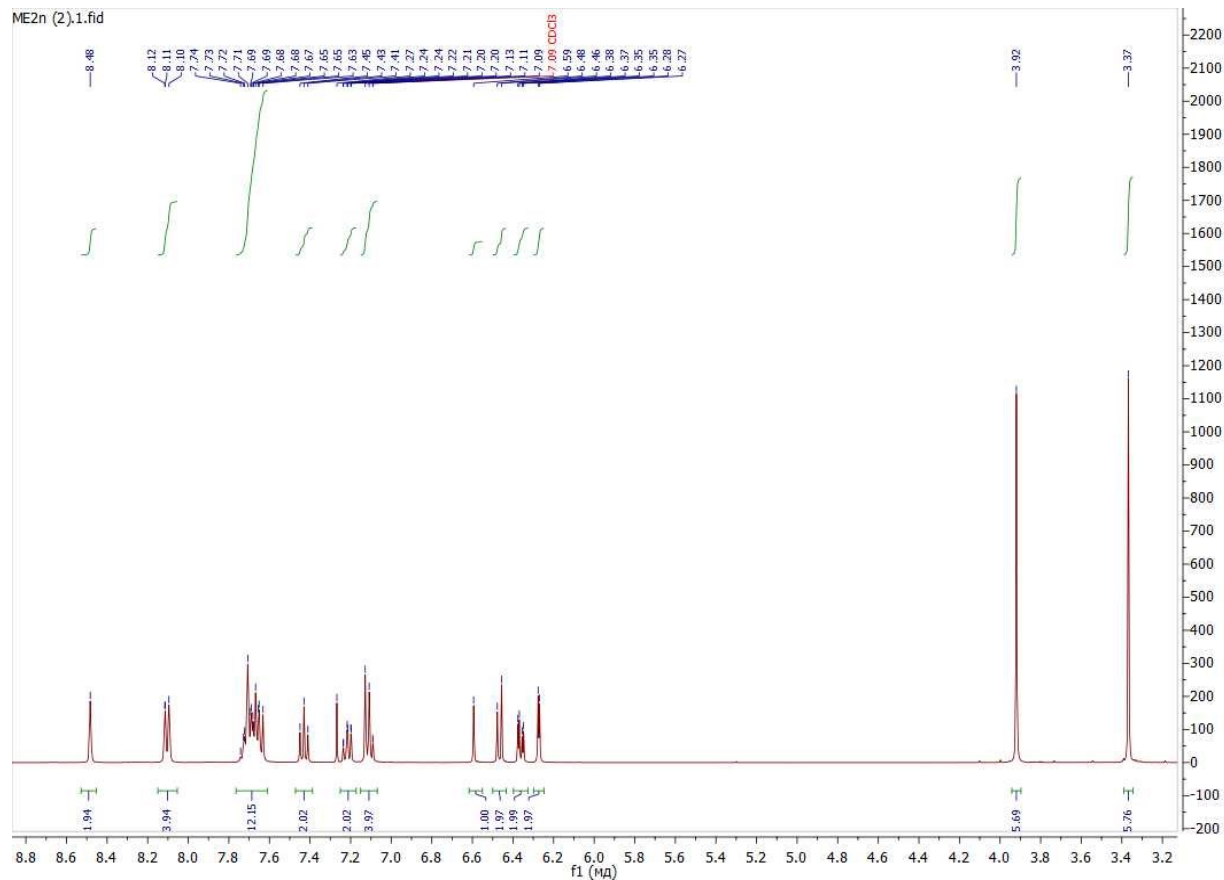
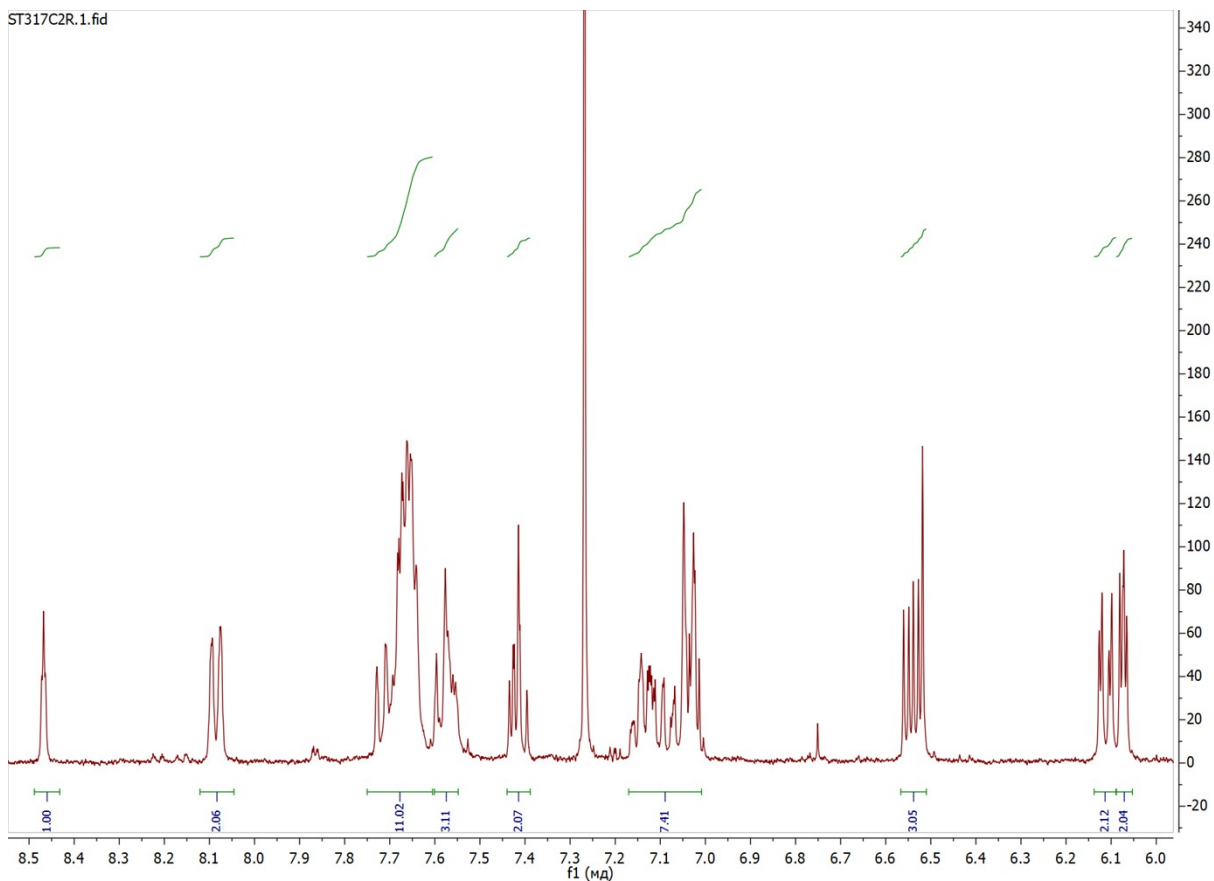


Figure S15. ^1H NMR spectrum of **6** (400 MHz, 298K, CDCl_3).



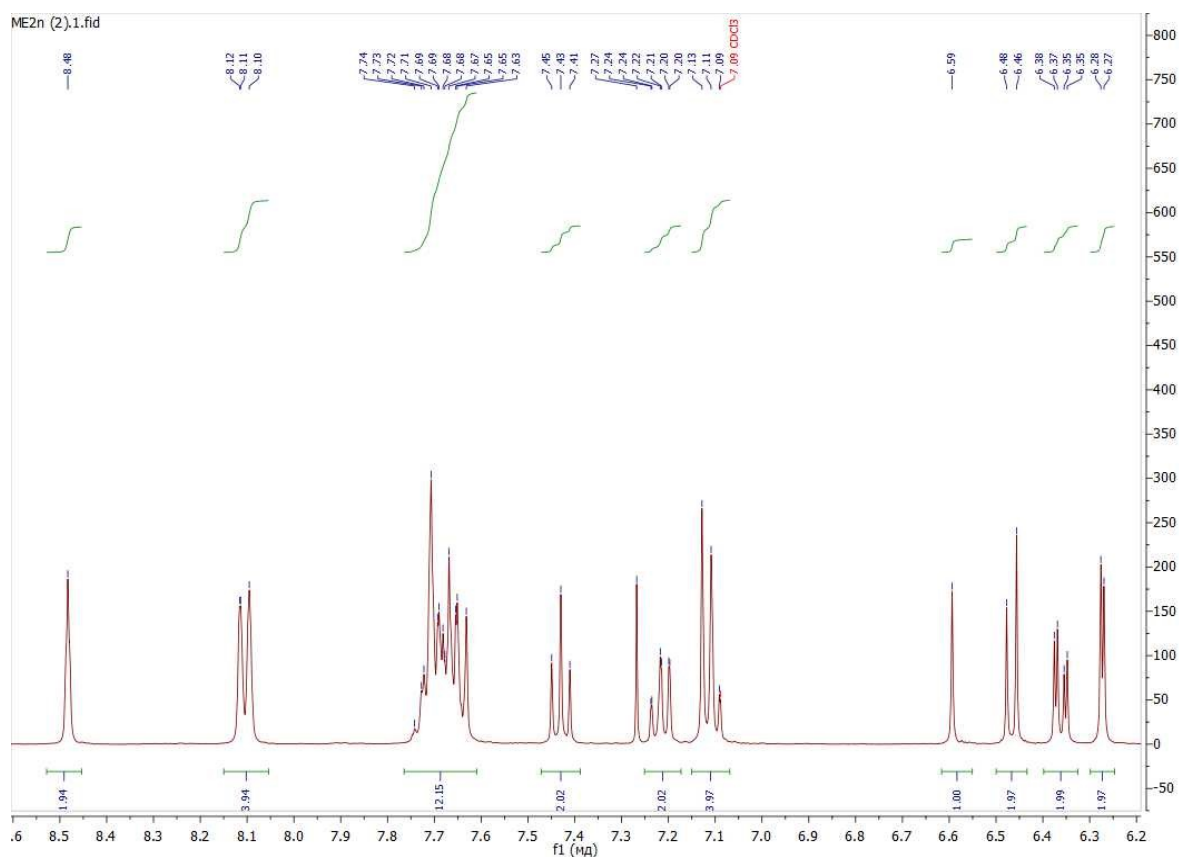


Figure S18. Aromatic region of ¹H NMR spectrum of **7** (400 MHz, 298K, CDCl₃).

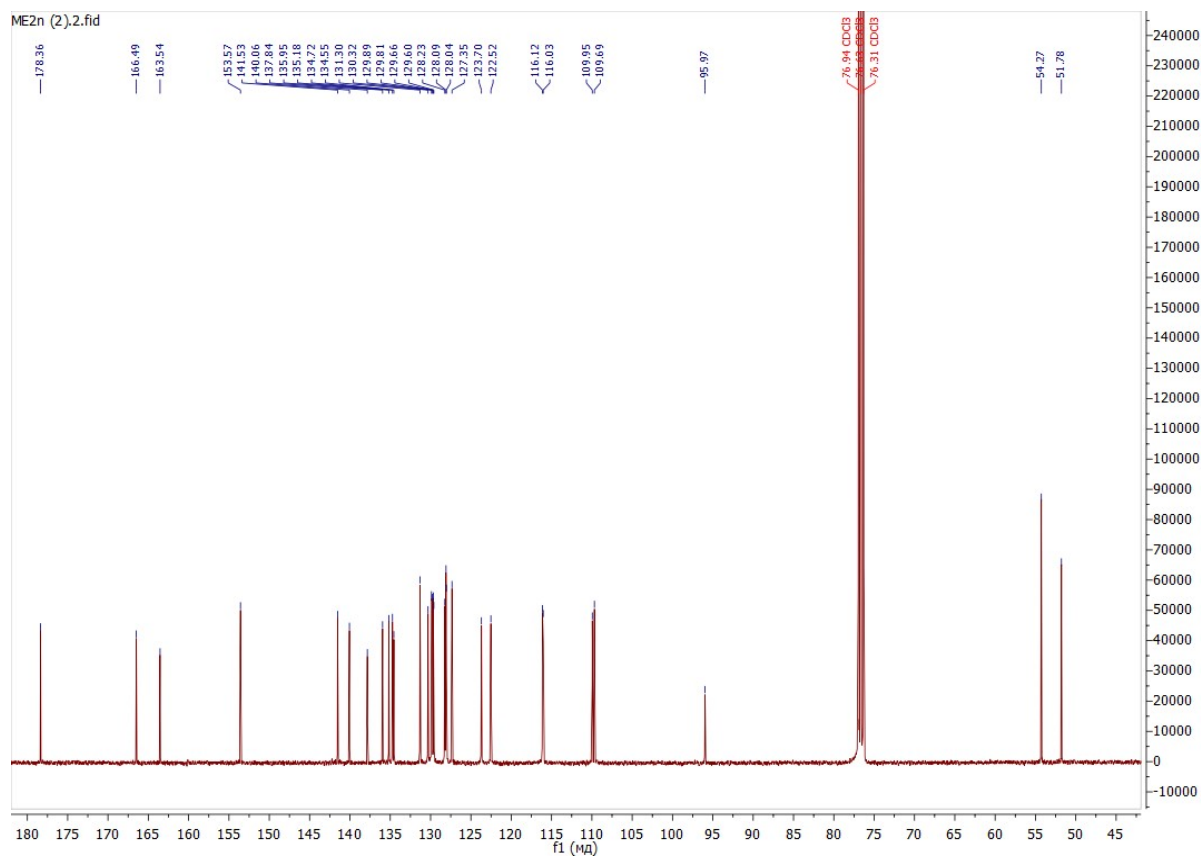


Figure S19. ¹³C {¹H} NMR spectrum of **7** (101 MHz, 298K, CDCl₃).

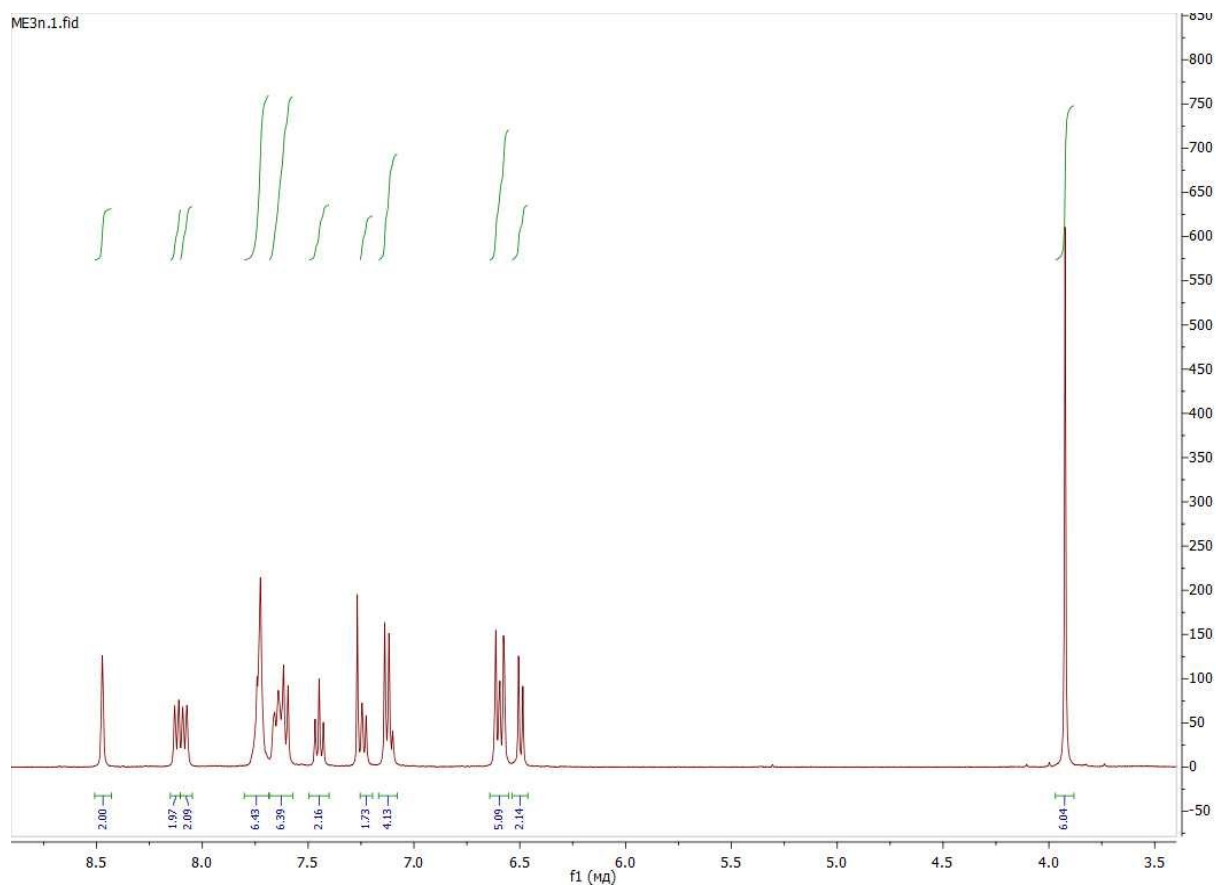


Figure S20. ^1H NMR spectrum of **8** (400 MHz, 298K, CDCl_3).

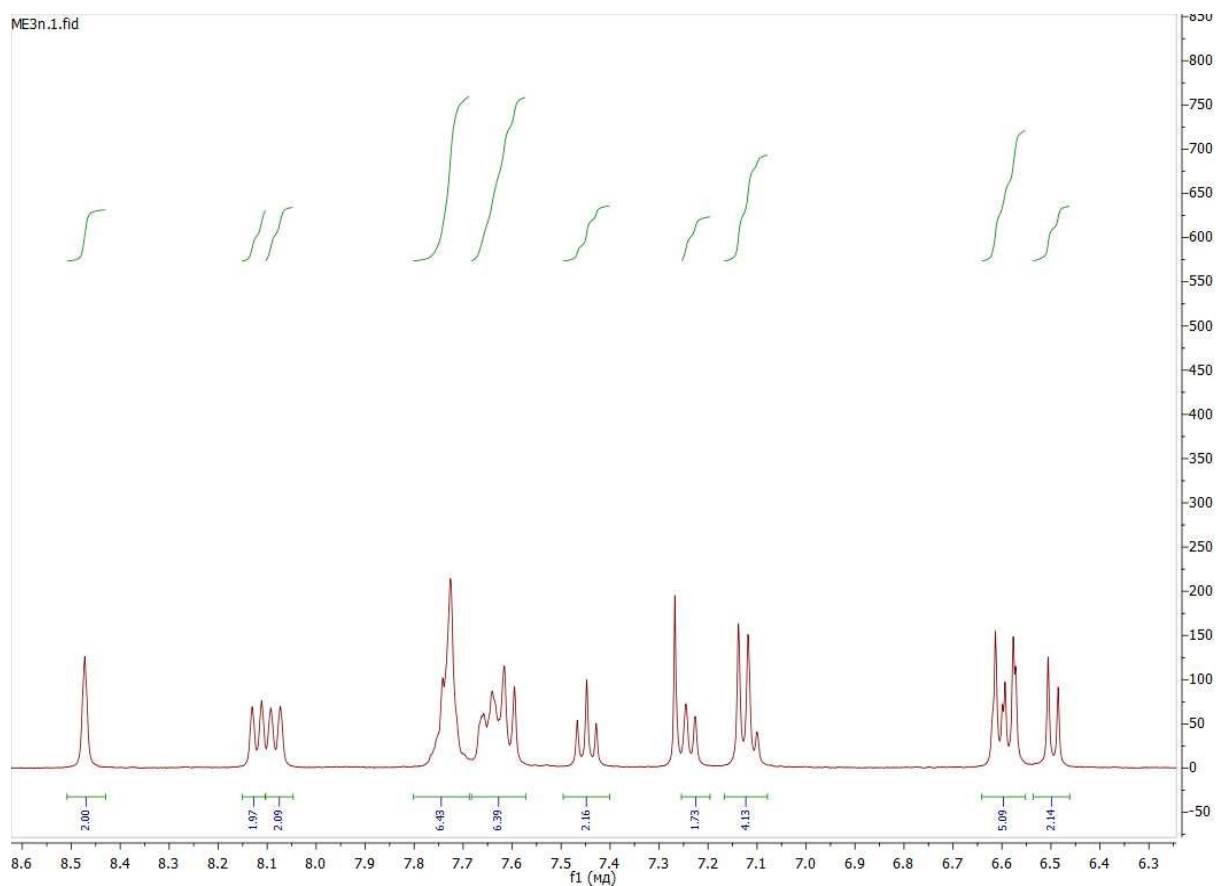


Figure S21. Aromatic region of ^1H NMR spectrum of **8** (400 MHz, 298K, CDCl_3).

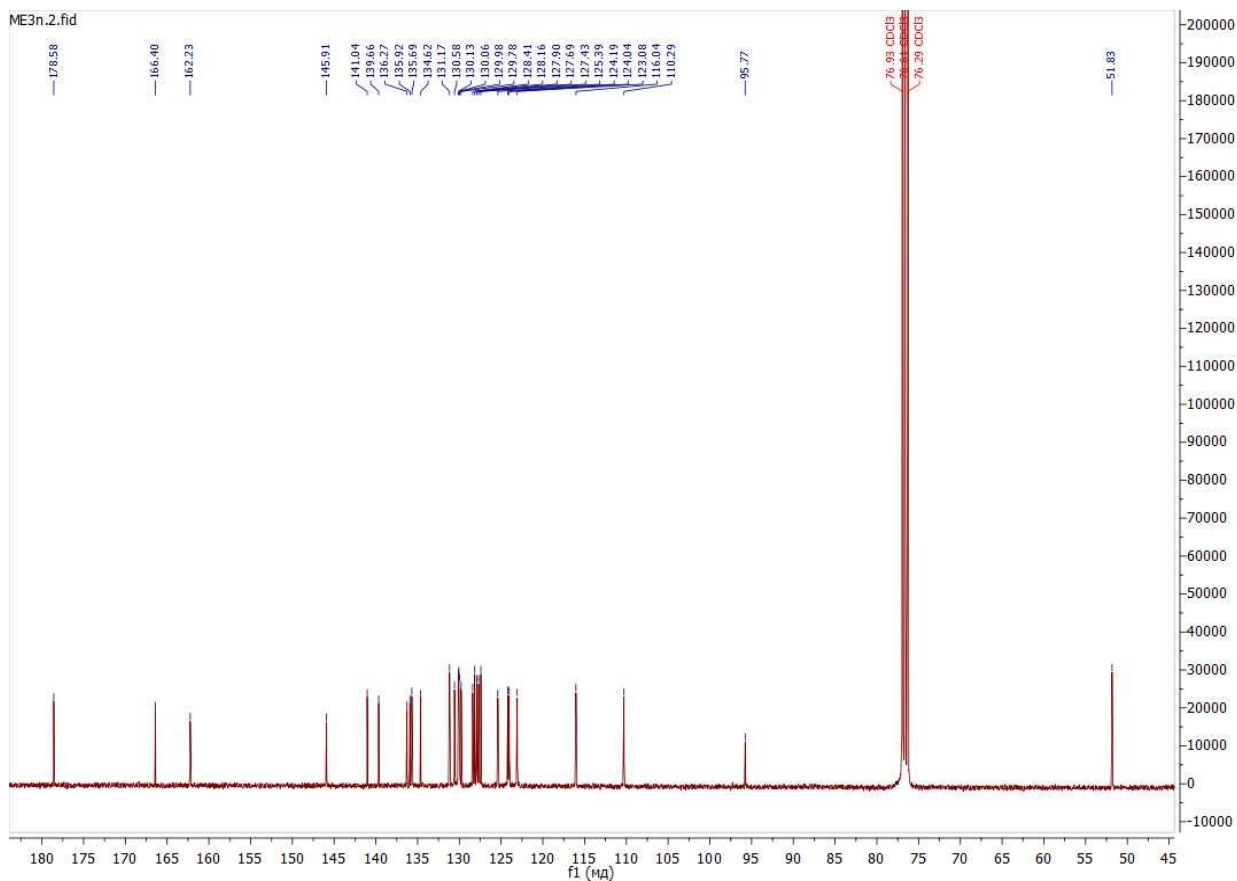


Figure S22. ^{13}C $\{^1\text{H}\}$ NMR spectrum of **8** (101 MHz, 298K, CDCl_3).

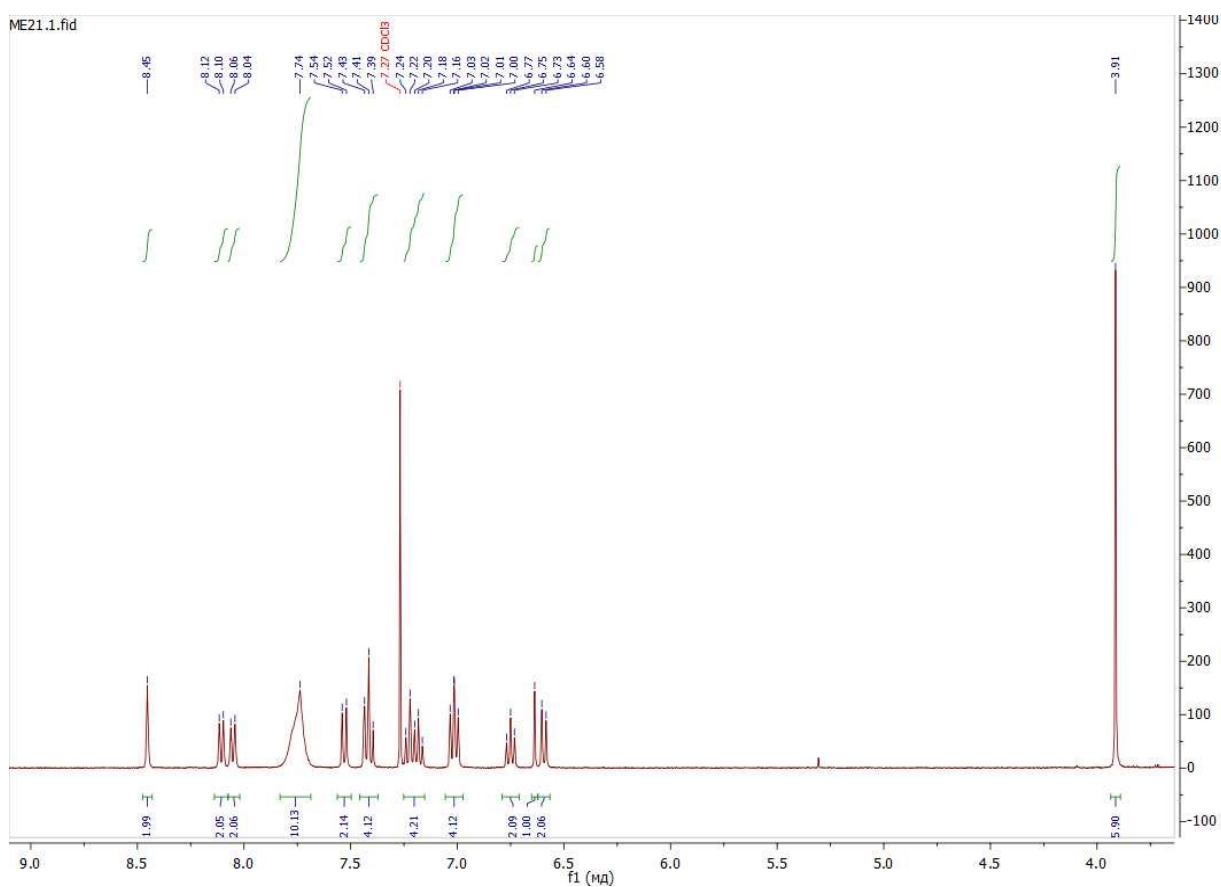


Figure S23. ^1H NMR spectrum of **9** (400 MHz, 298K, CDCl_3).

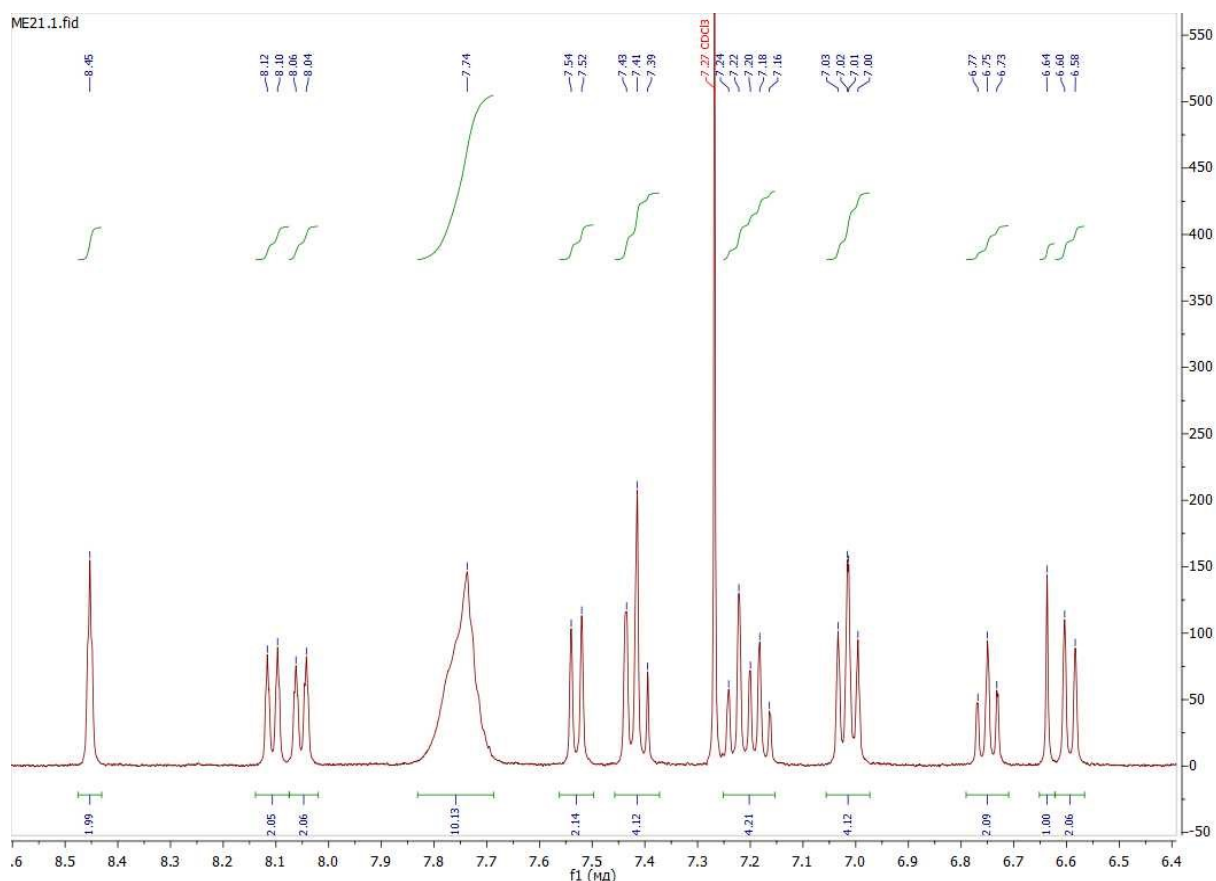


Figure S24. Aromatic region of ¹H NMR spectrum of **9** (400 MHz, 298K, CDCl₃).

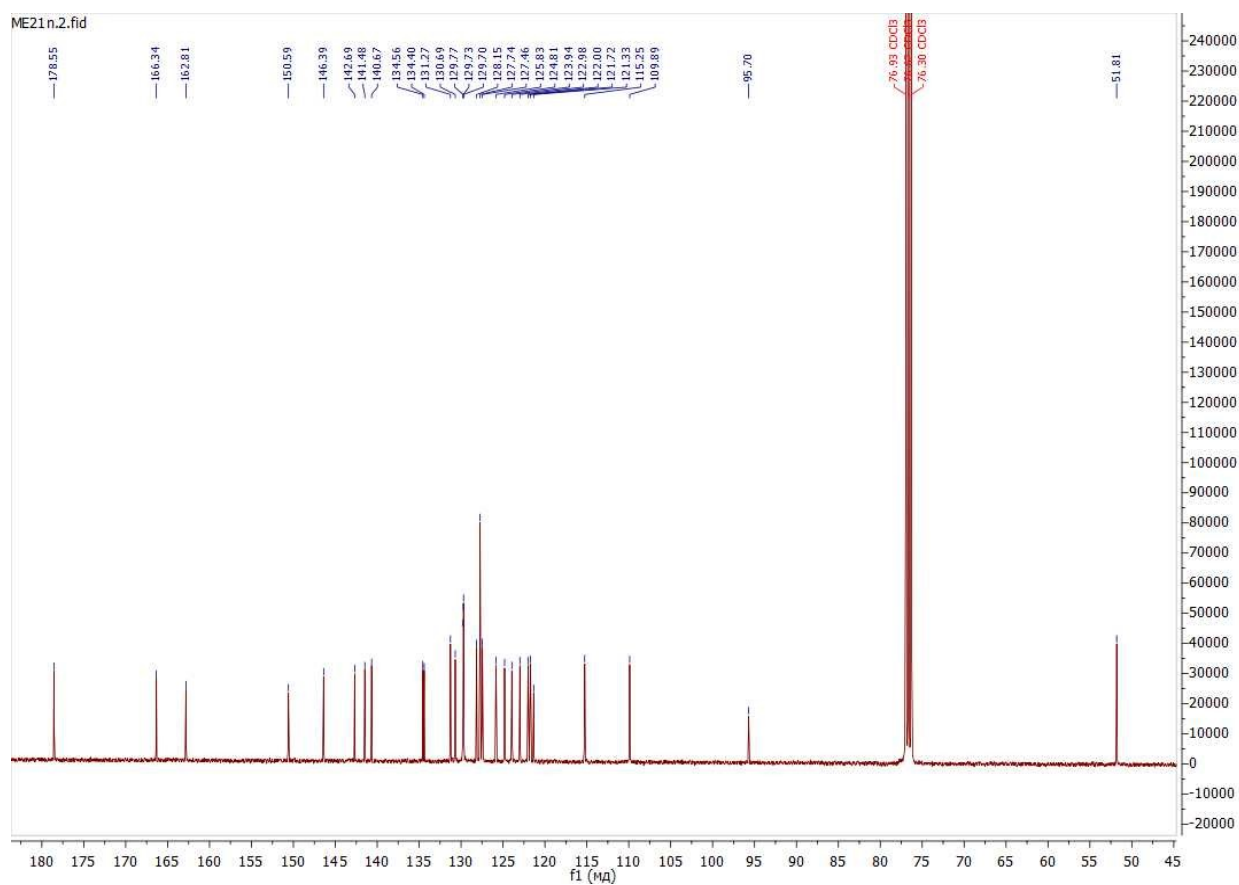


Figure S25. ¹³C {¹H} NMR spectrum of **9** (101 MHz, 298K, CDCl₃).

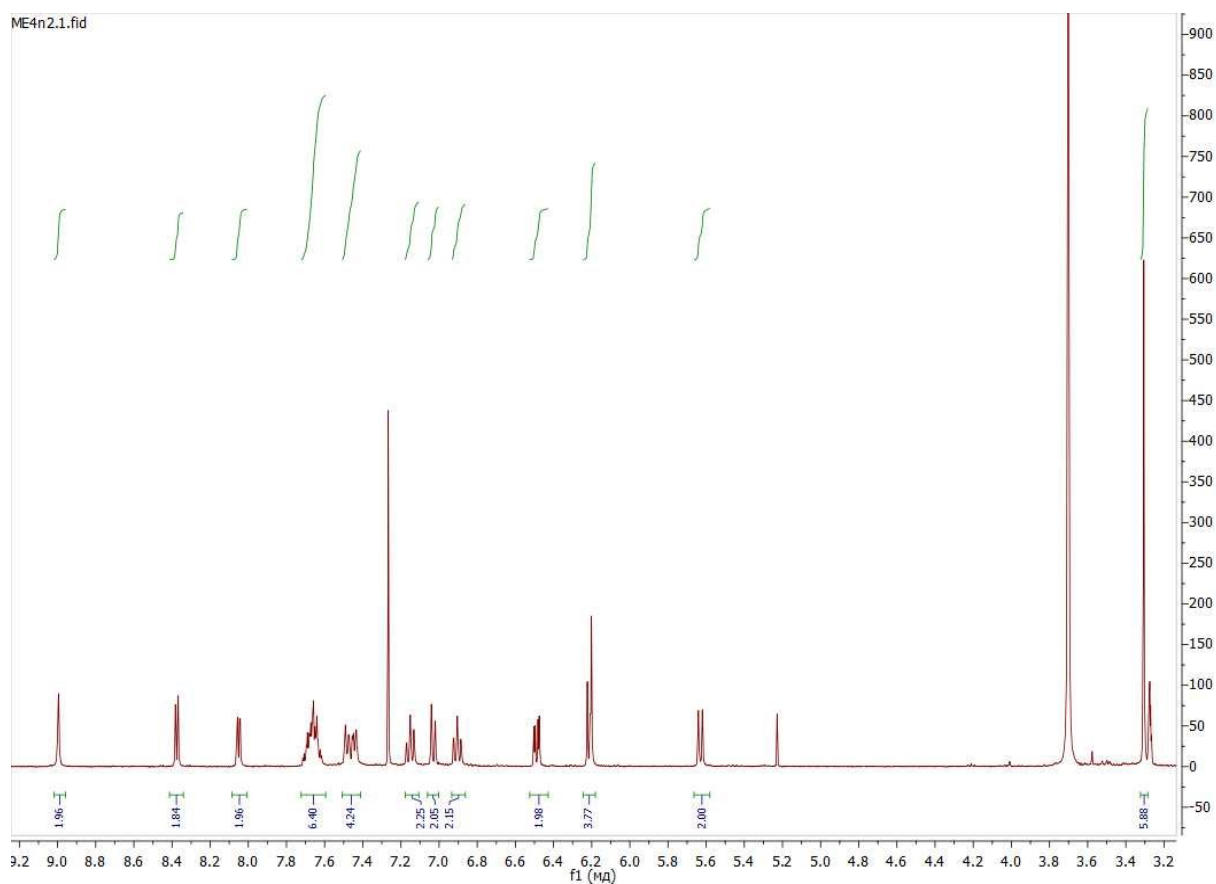


Figure S26. ^1H NMR spectrum of **10** (400 MHz, 298K, $\text{CDCl}_3:\text{CD}_3\text{OD}$ 6:1).

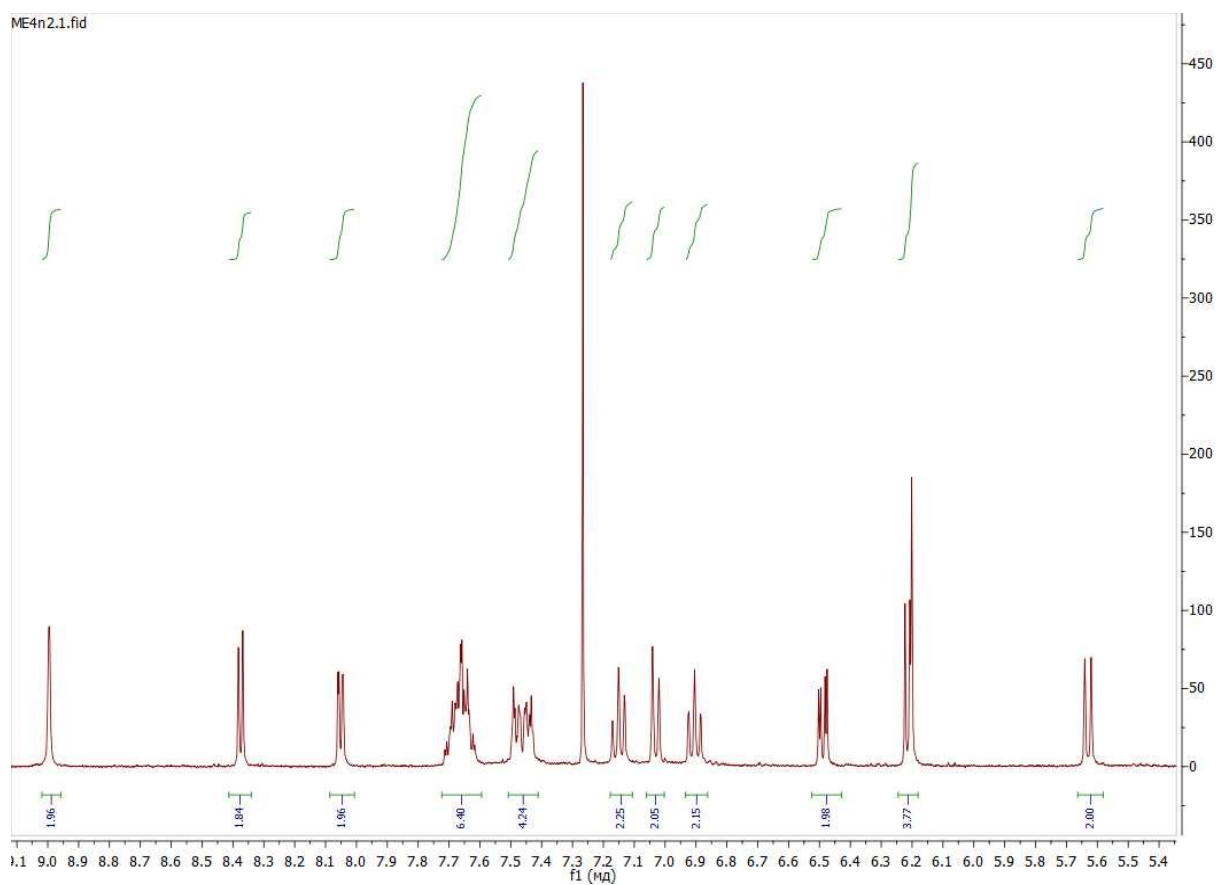
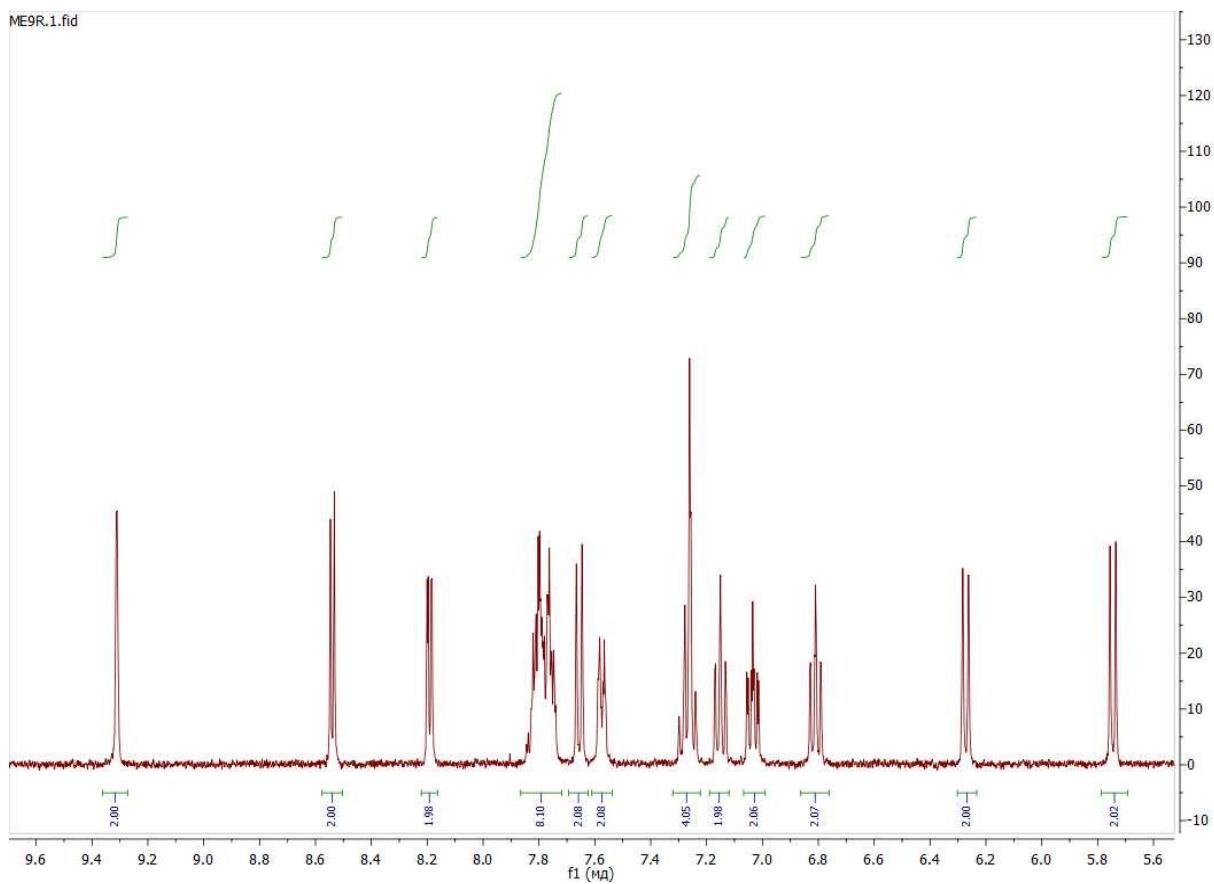
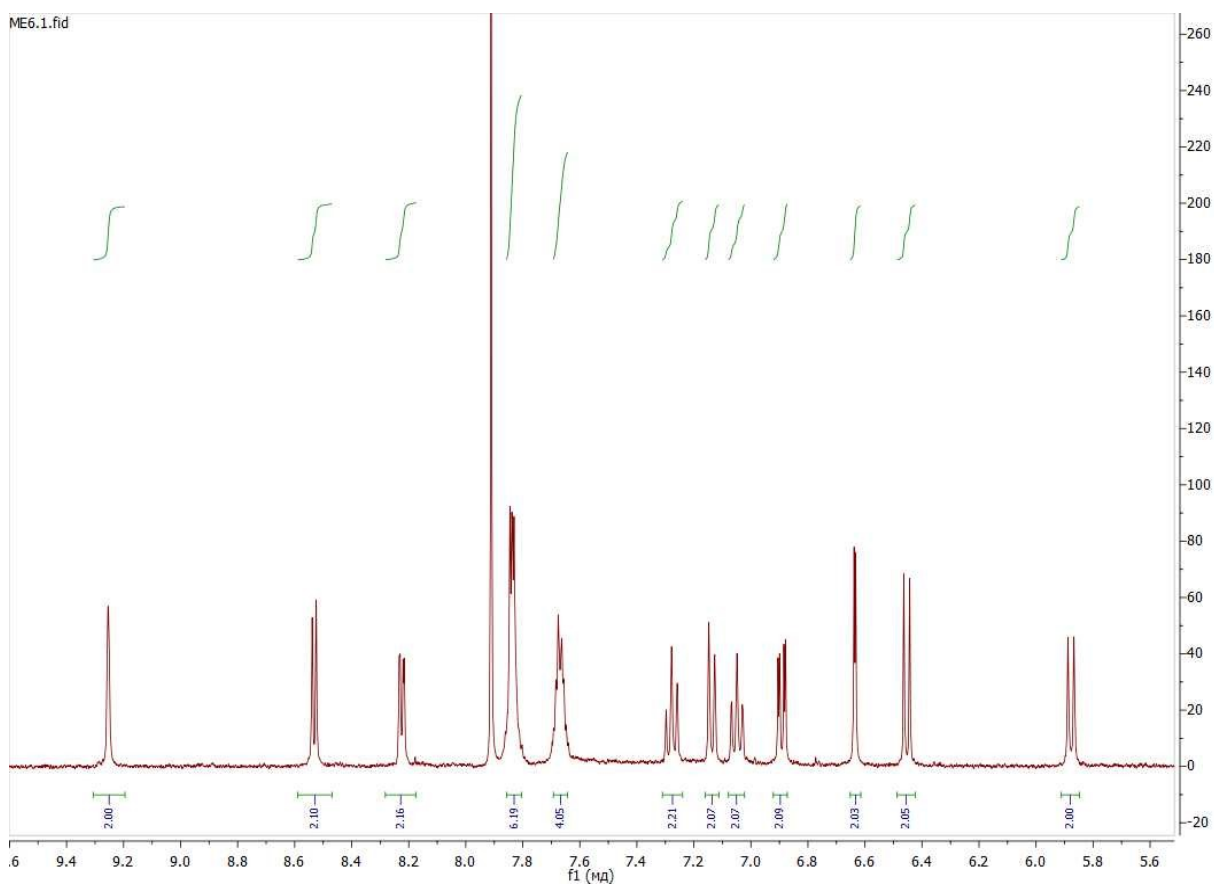


Figure S27. Aromatic region of ^1H NMR spectrum of **10** (400 MHz, 298K, $\text{CDCl}_3:\text{CD}_3\text{OD}$ 6:1).



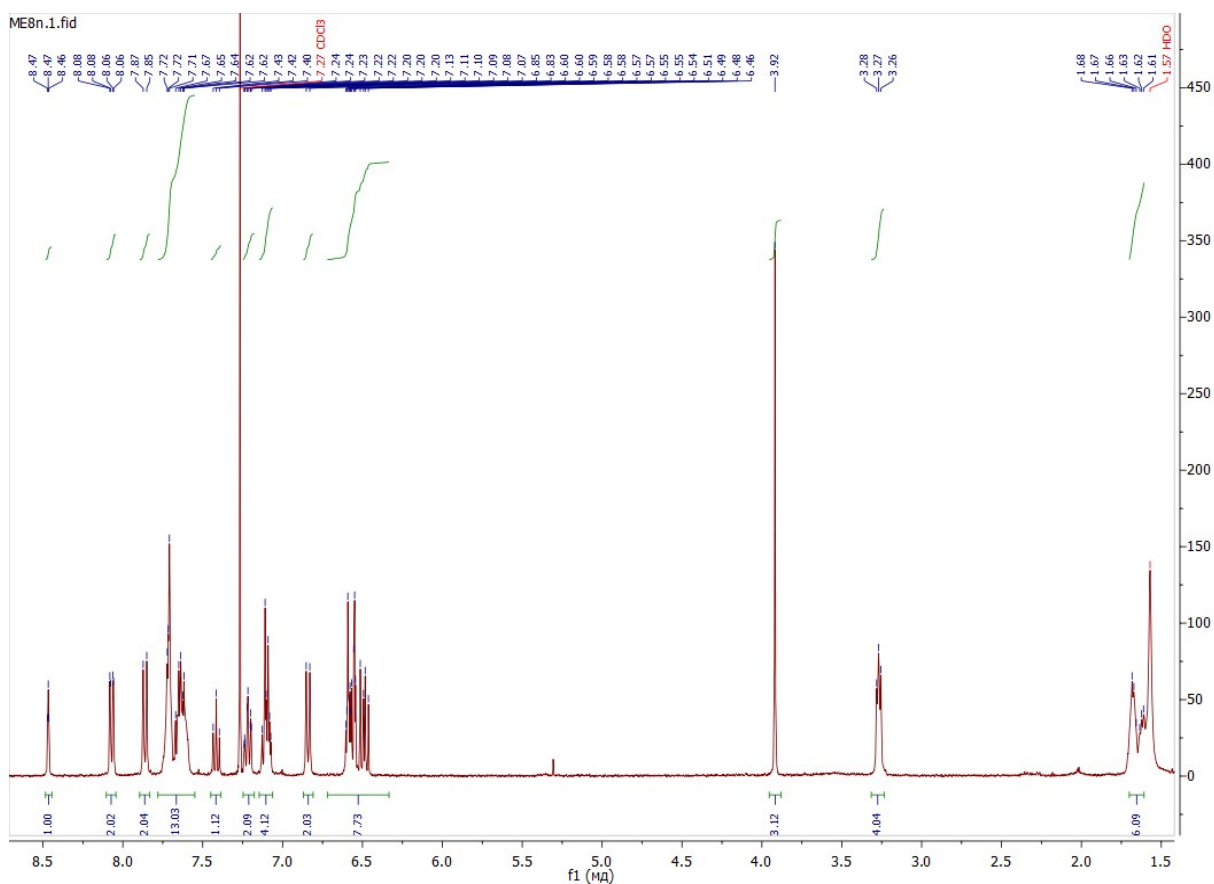


Figure S30. ^1H NMR spectrum of **13** (400 MHz, 298K, CDCl_3).

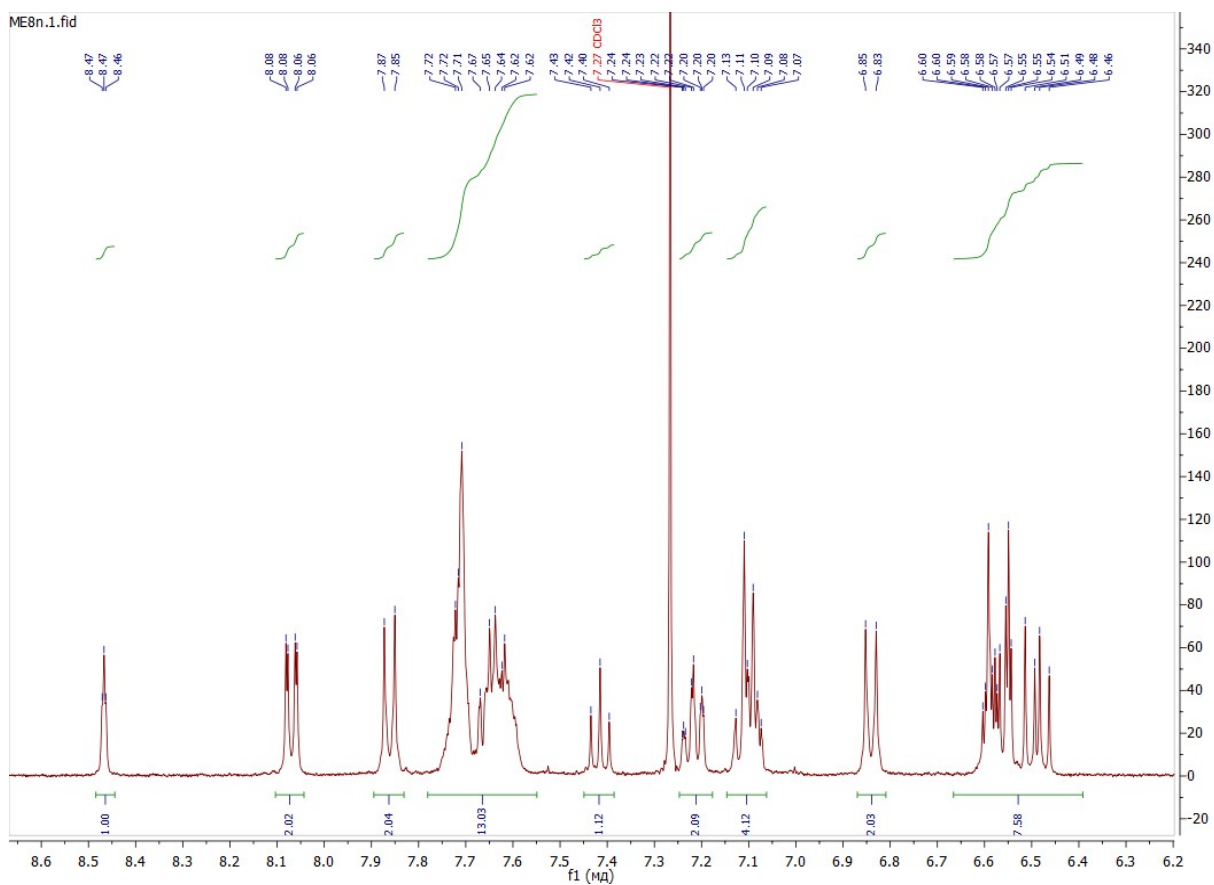


Figure S31. Aromatic region of ^1H NMR spectrum of **13** (400 MHz, 298K, CDCl_3)

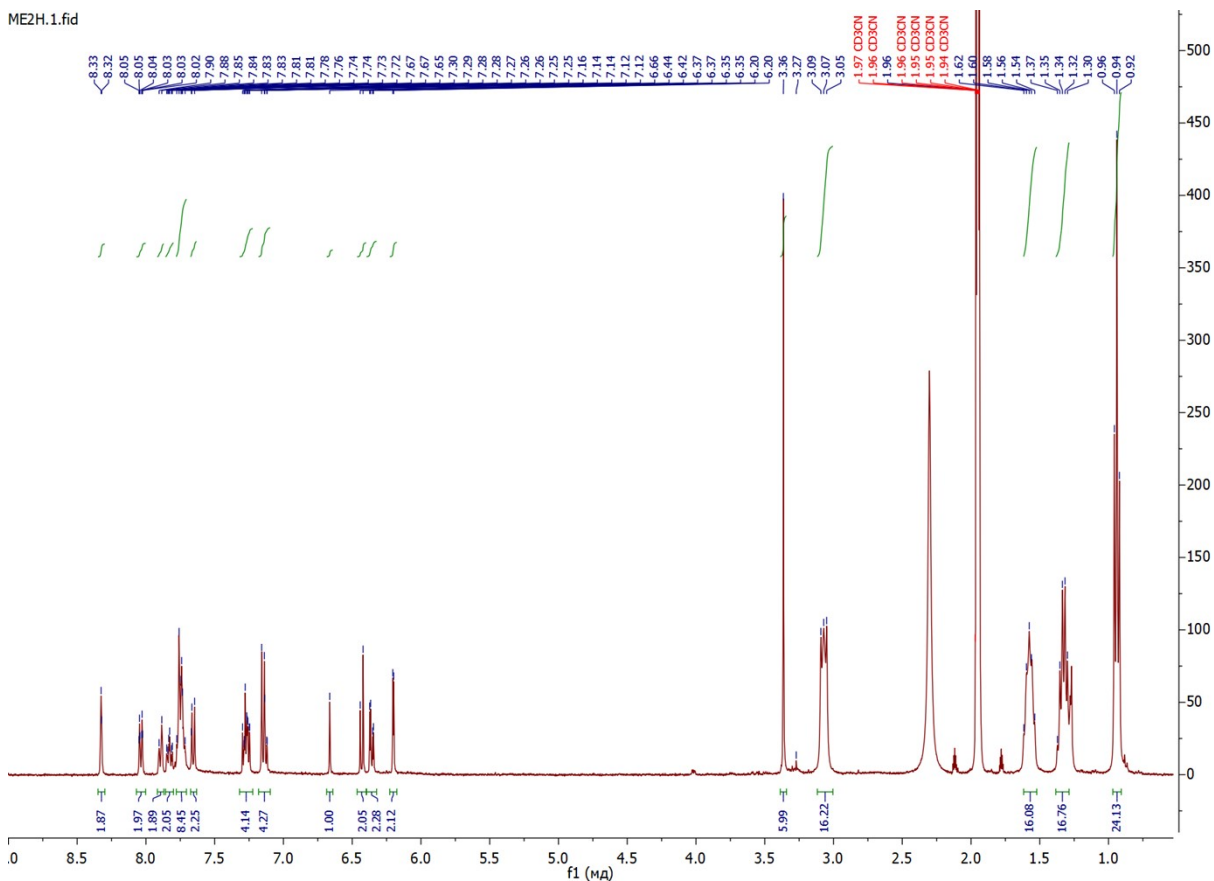


Figure S32. ¹H NMR spectrum of hydrolysed complex 2(TBA)⁺(**7H**)²⁻ (400 MHz, 298K, CD₃CN).

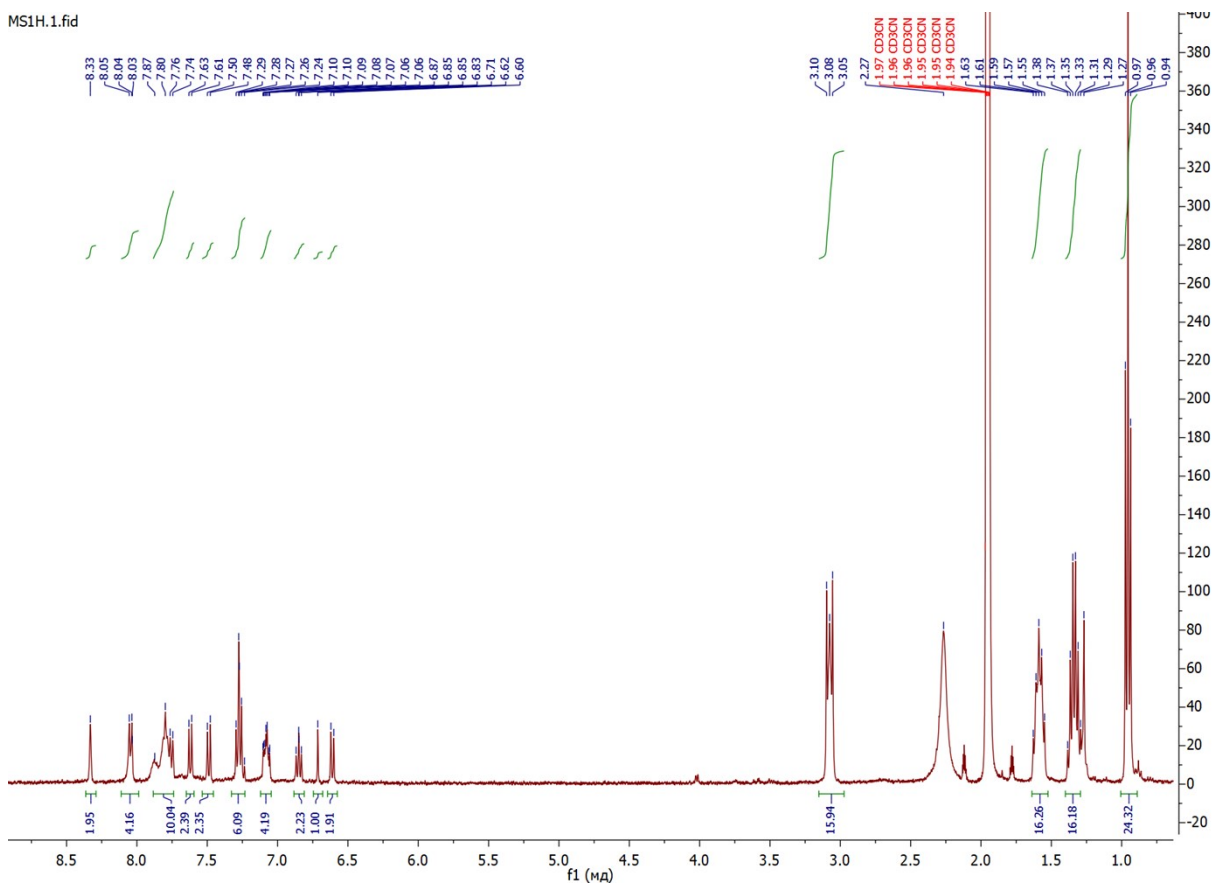


Figure S33. ¹H NMR spectrum of hydrolysed complex 2(TBA)⁺(**9H**)²⁻ (400 MHz, 298K, CD₃CN).

Table S1. High resolution mass-spectrum data (m/z) of complexes **1** – **13**.

Complex	[Ir(C [^] N) ₂ (L)] ⁺	[Ir(C [^] N) ₂] ⁺
1	1054.1986	767.1591
2	1086.1394	799.0993
3	1078.2371	791.1996
4	1170.2816	883.2417
5	1154.1941	867.1523
6	1218.3963	931.3566
7	1130.2886	832.2258
8	1138.1861	840.1249
9	1182.2107	884.1487
10	1035.2476	
11	1043.1473	797.0998
12	1087.1719	843.1219
12	1164.2613	799.0987

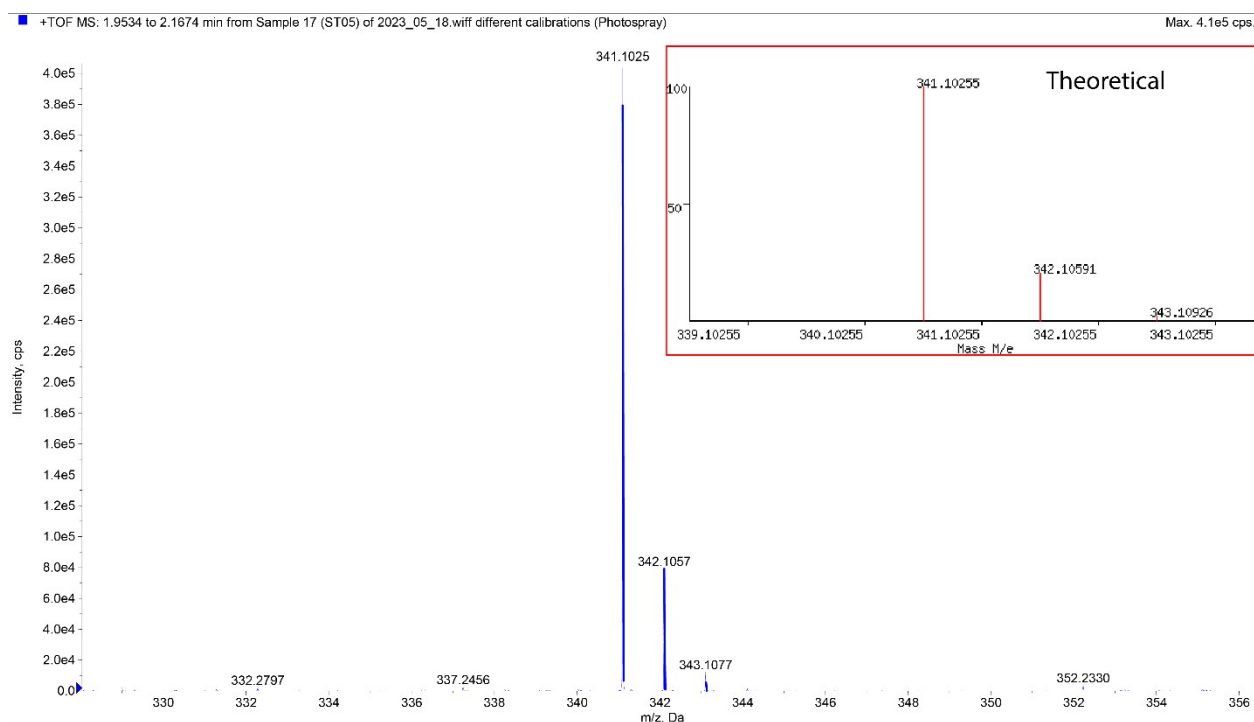


Figure S34. High resolution mass spectrum of L₂.

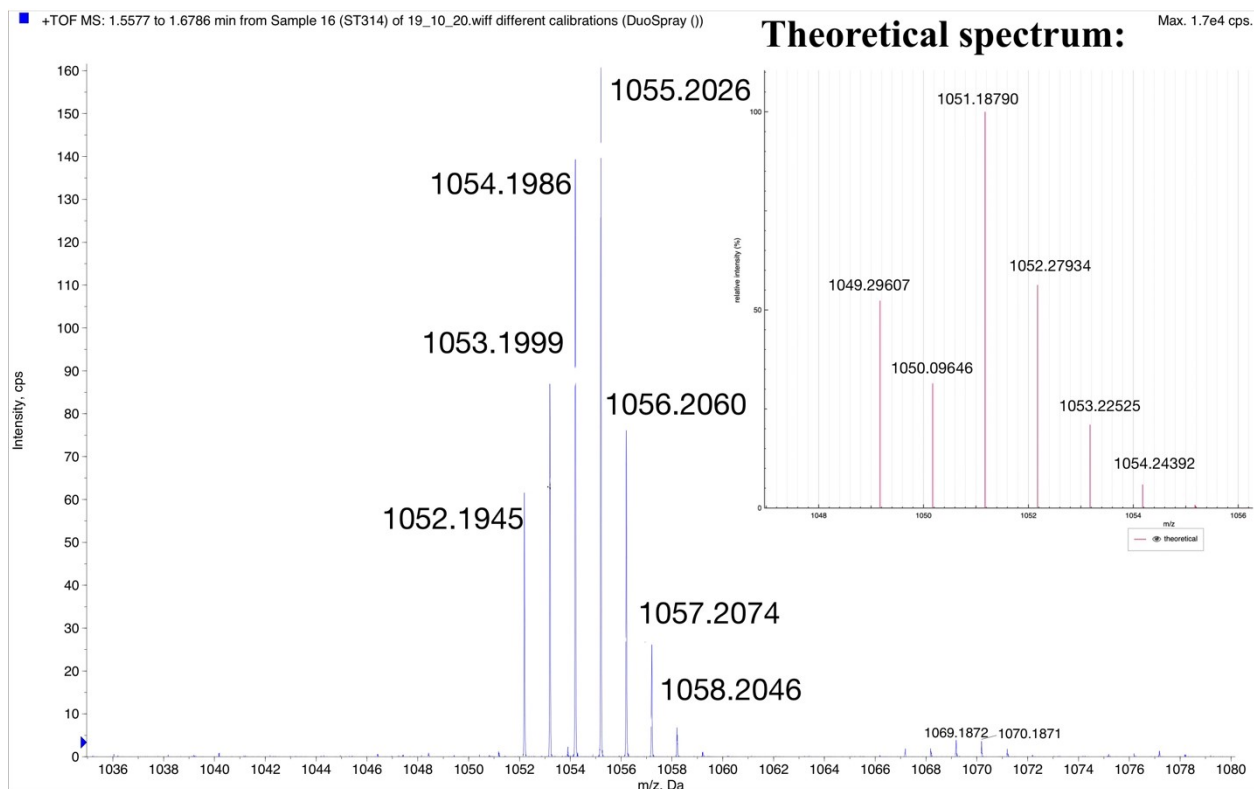


Figure S35. High resolution mass spectrum of **1**.

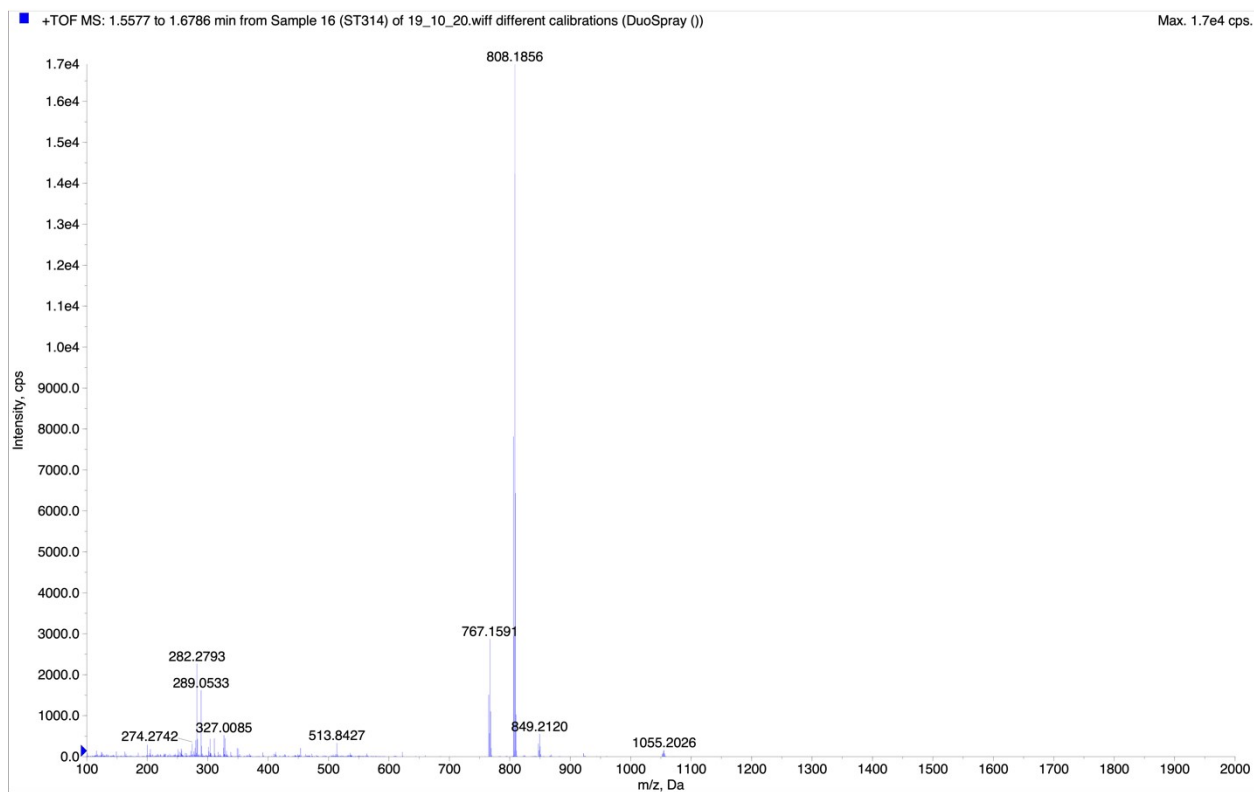


Figure S36. High resolution mass spectrum of **1**.

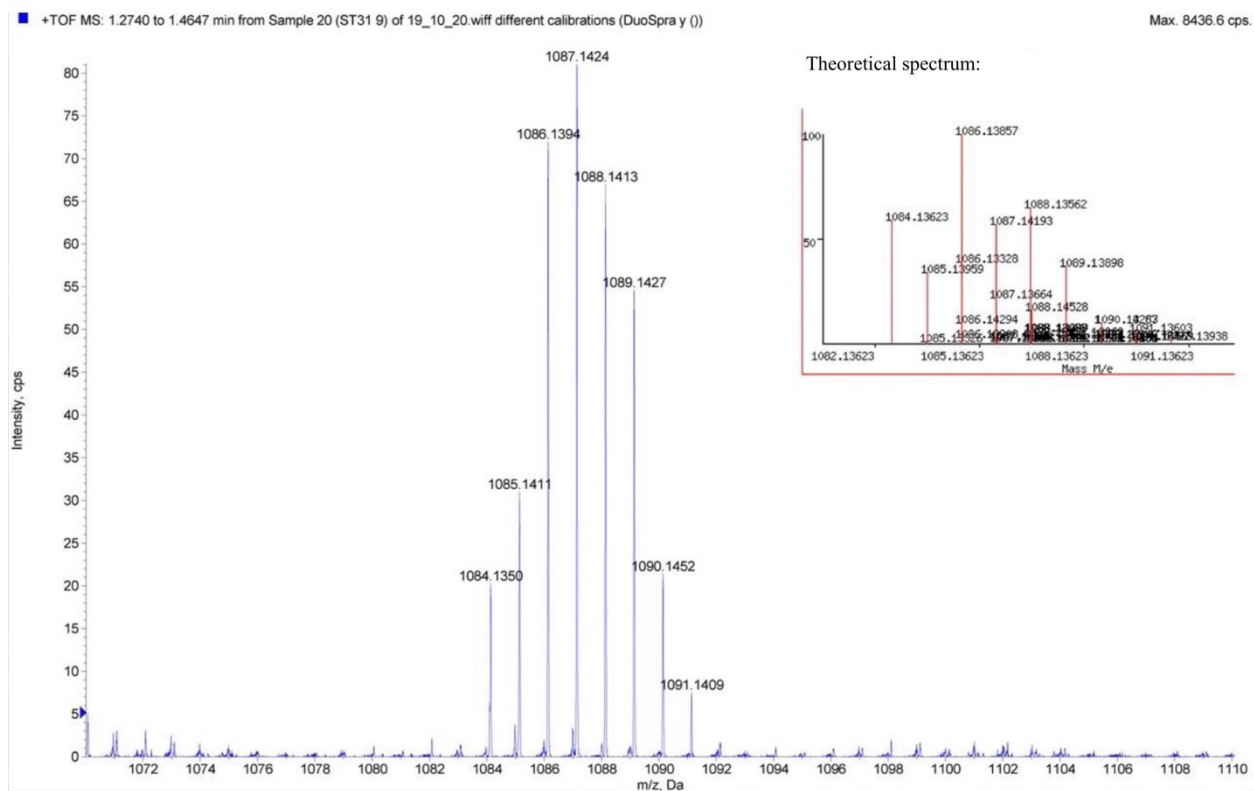


Figure S37. High resolution mass spectrum of 2.

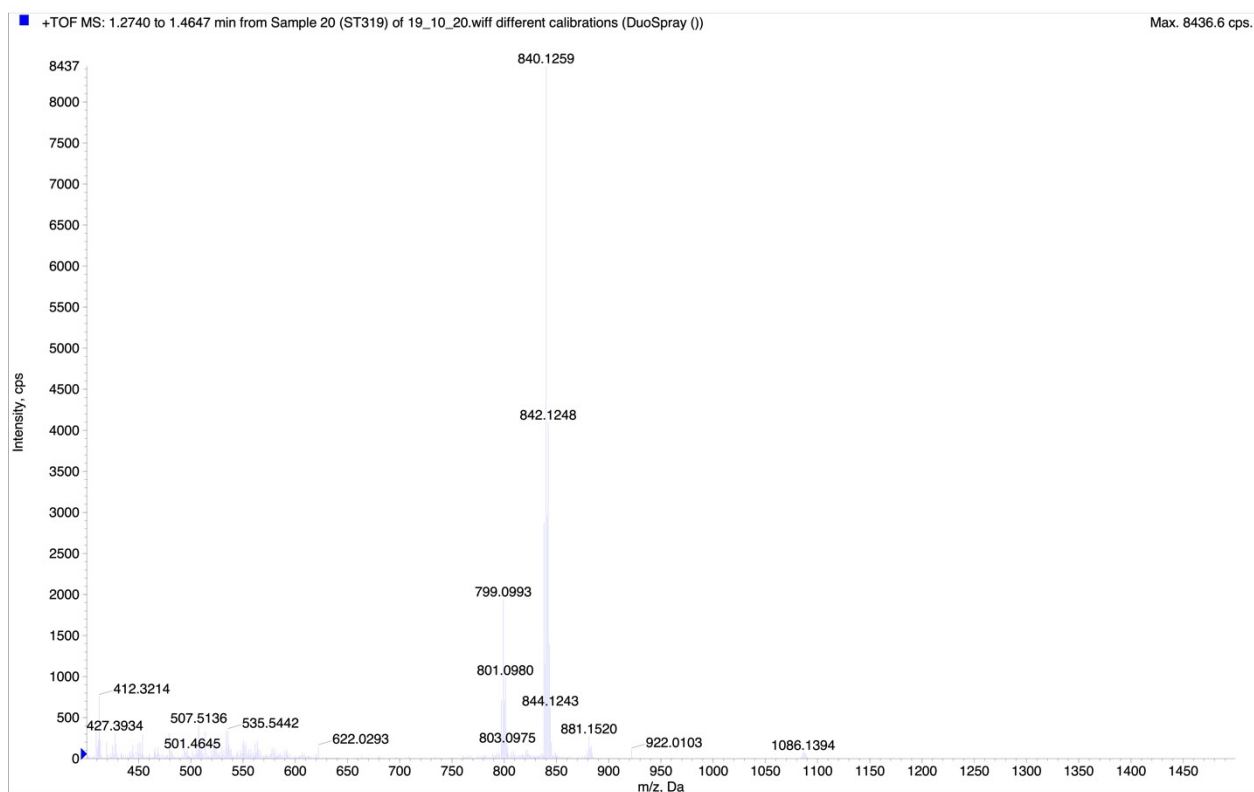


Figure S38. High resolution mass spectrum of 2.

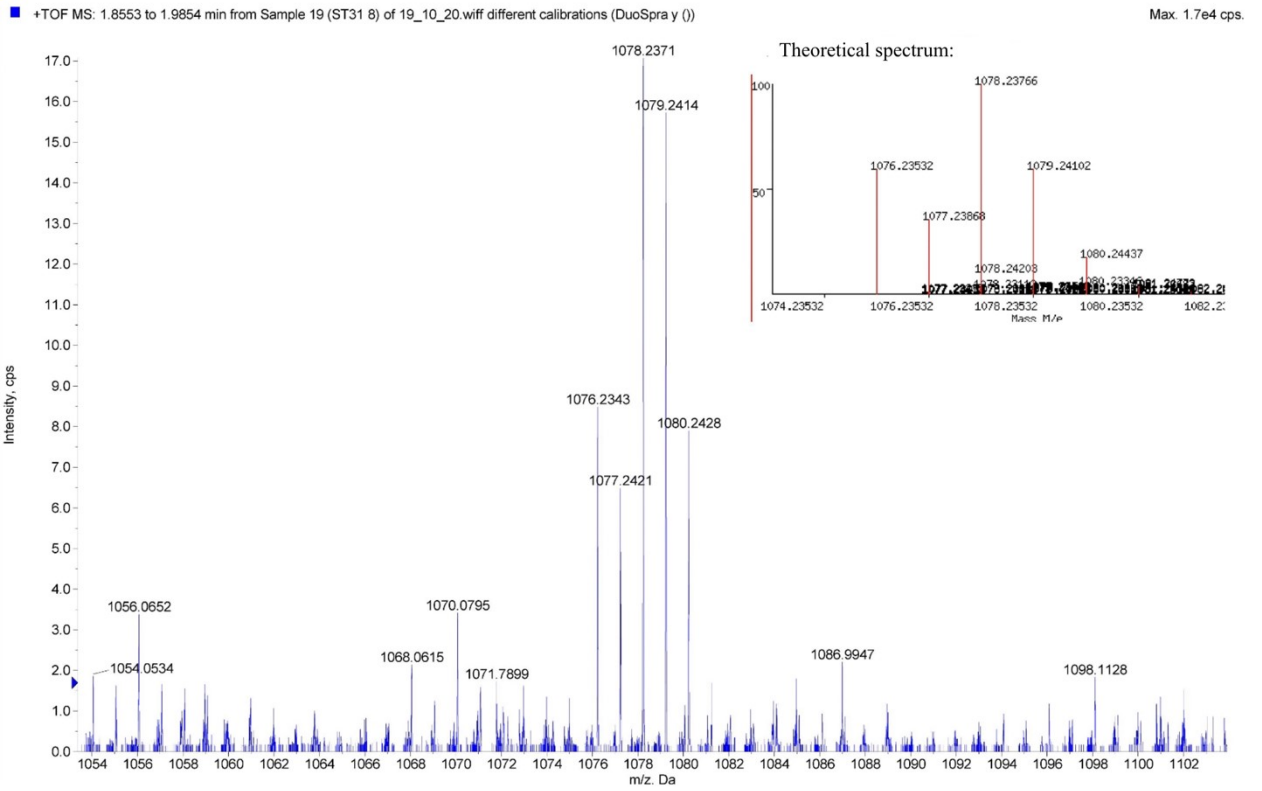


Figure S39. High resolution mass spectrum of 3.

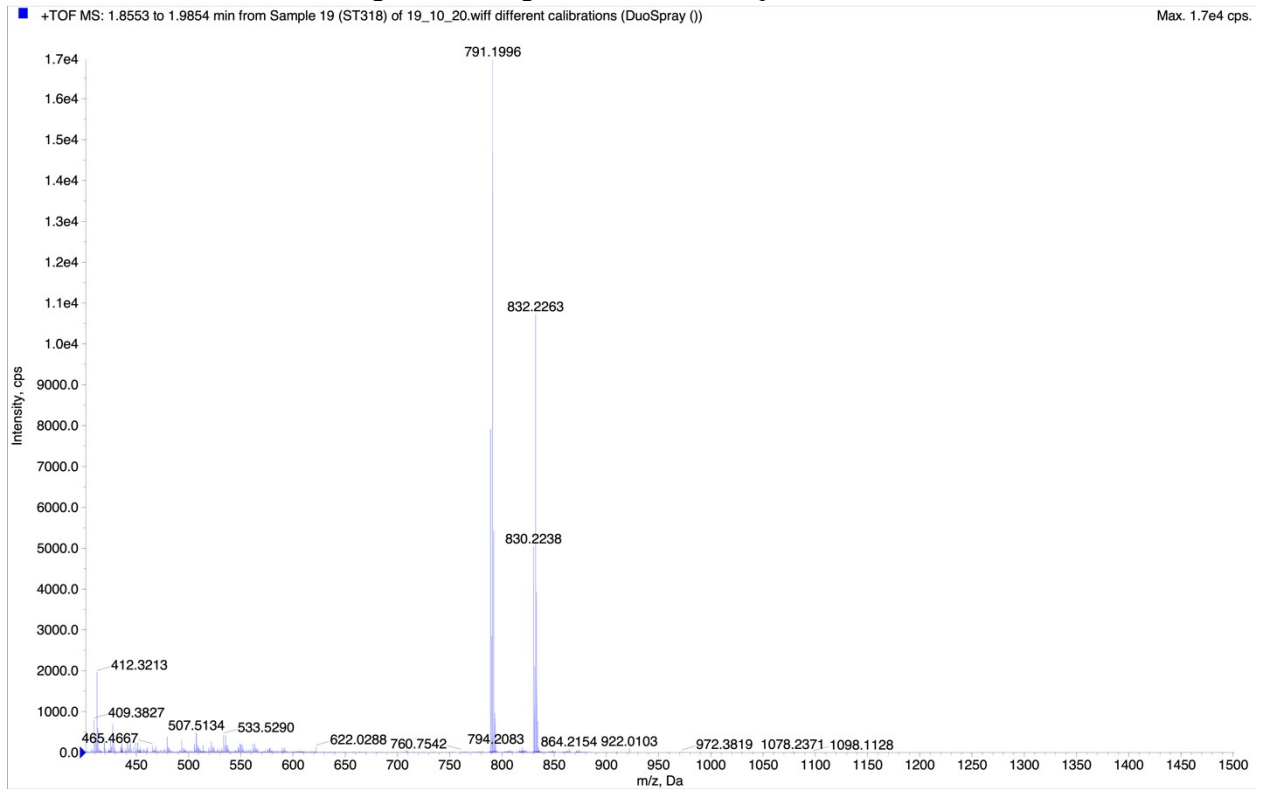


Figure S40. High resolution mass spectrum of 3.

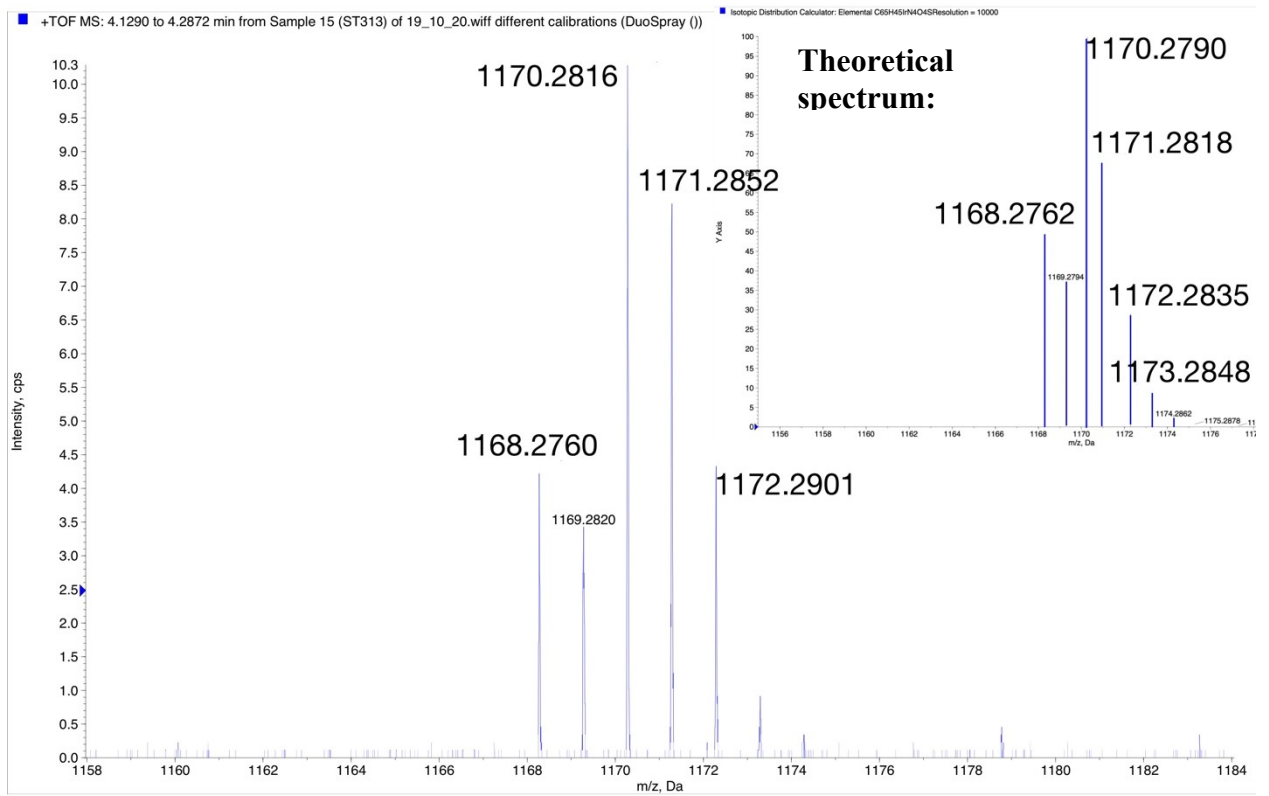


Figure S41. High resolution mass spectrum of 4.

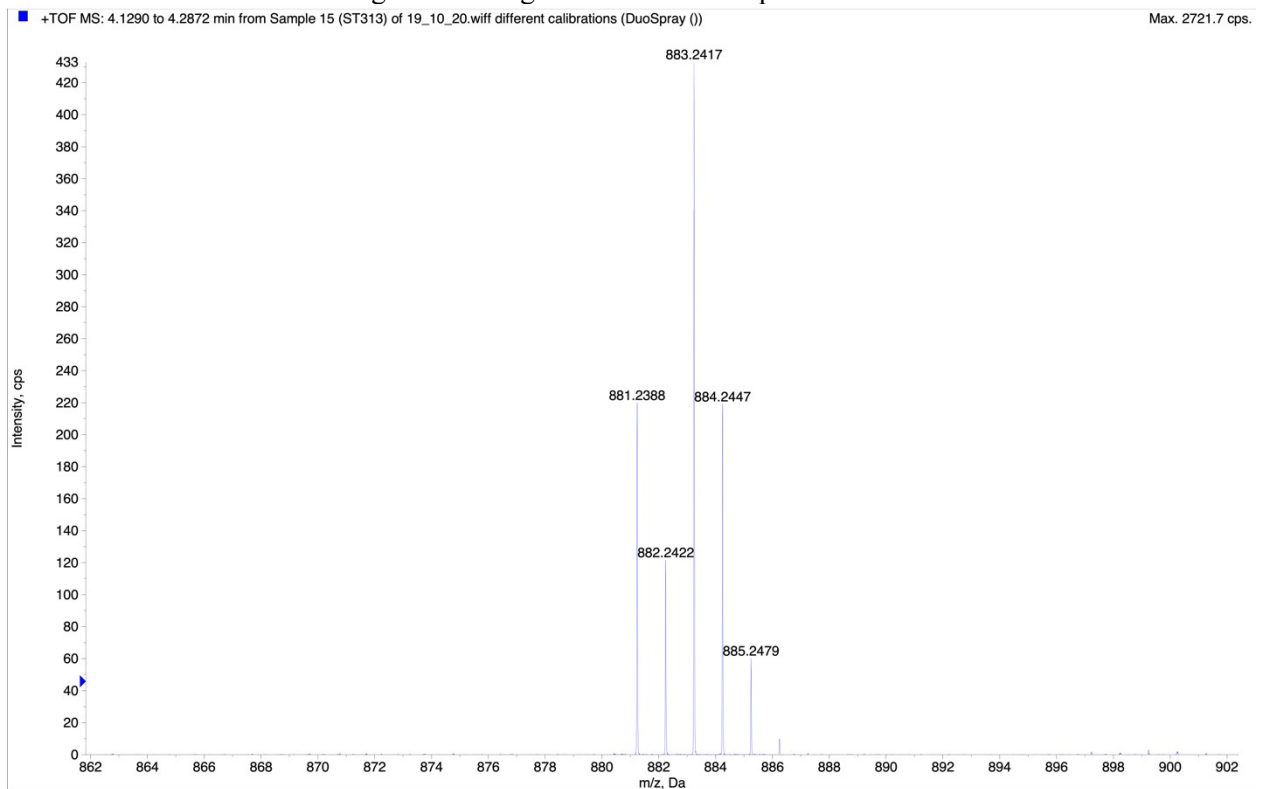


Figure S42. High resolution mass spectrum of 4.

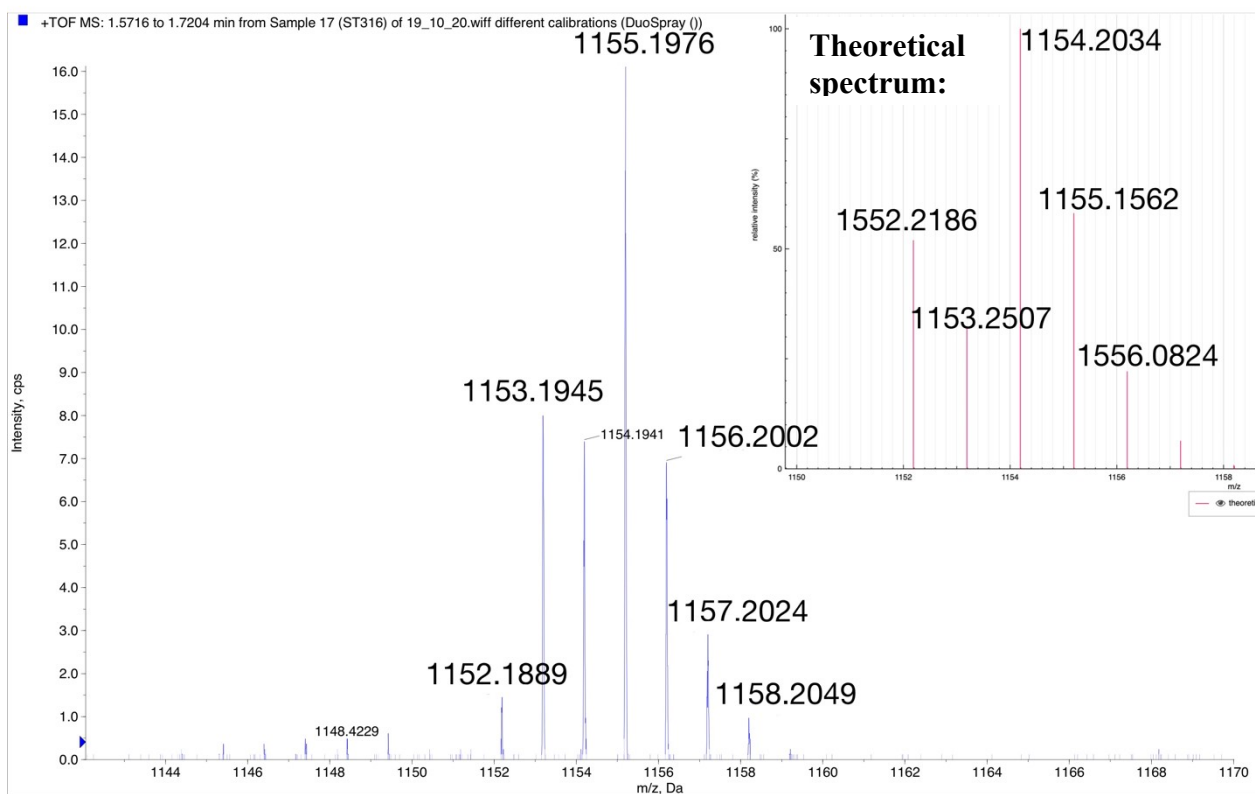


Figure S43. High resolution mass spectrum of 5.

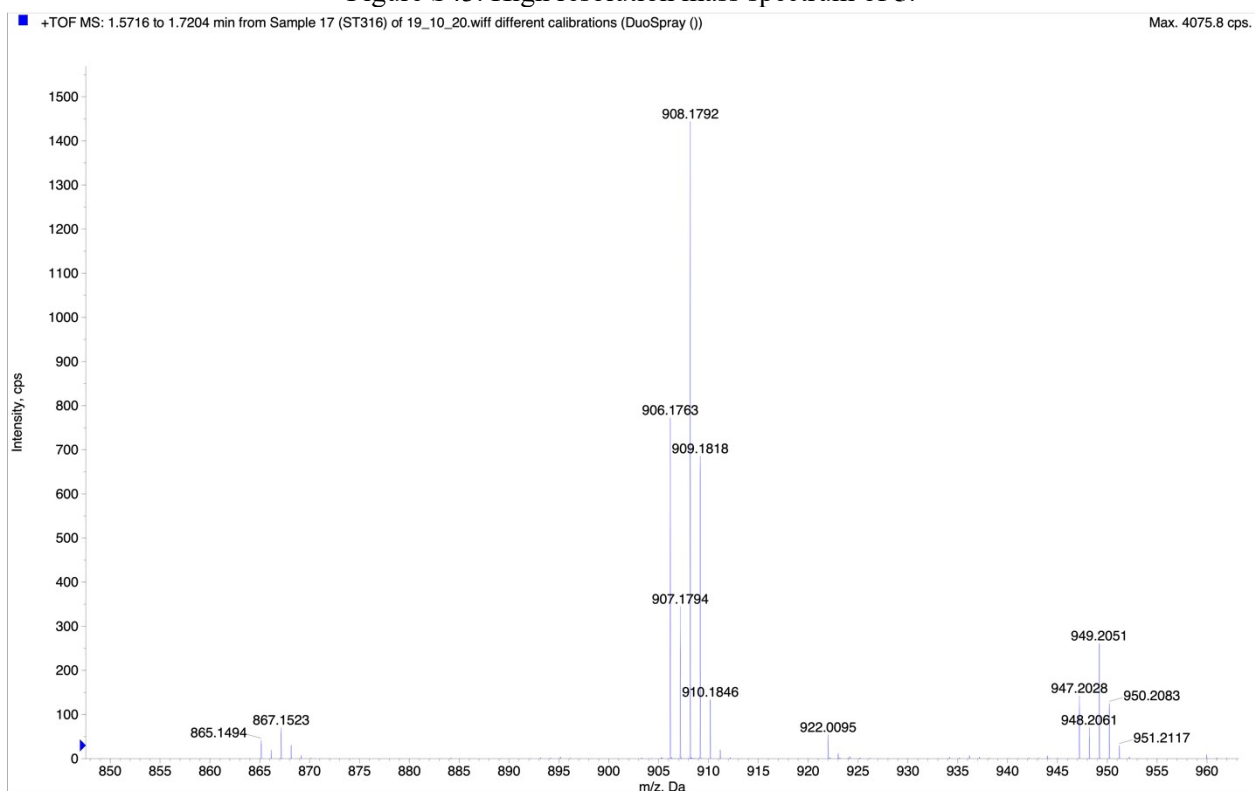


Figure S44. High resolution mass spectrum of 5.

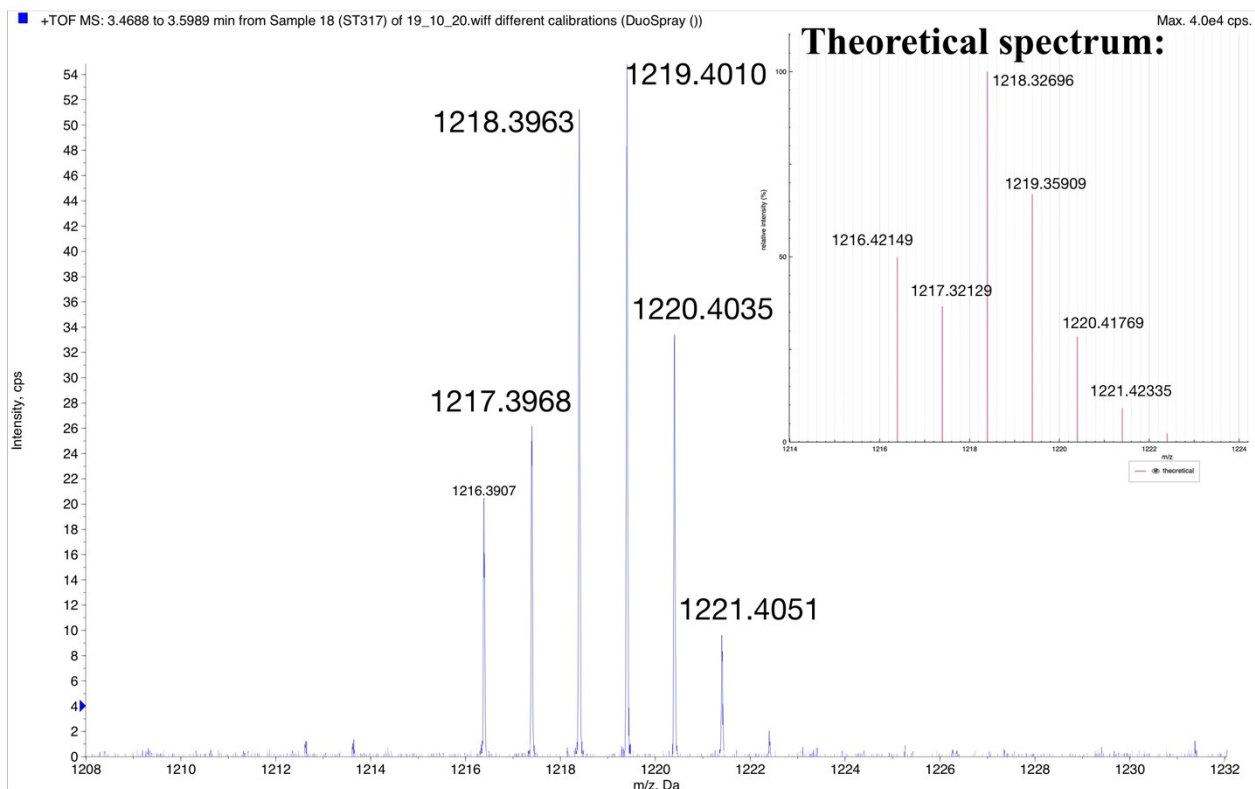


Figure S45. High resolution mass spectrum of 6.

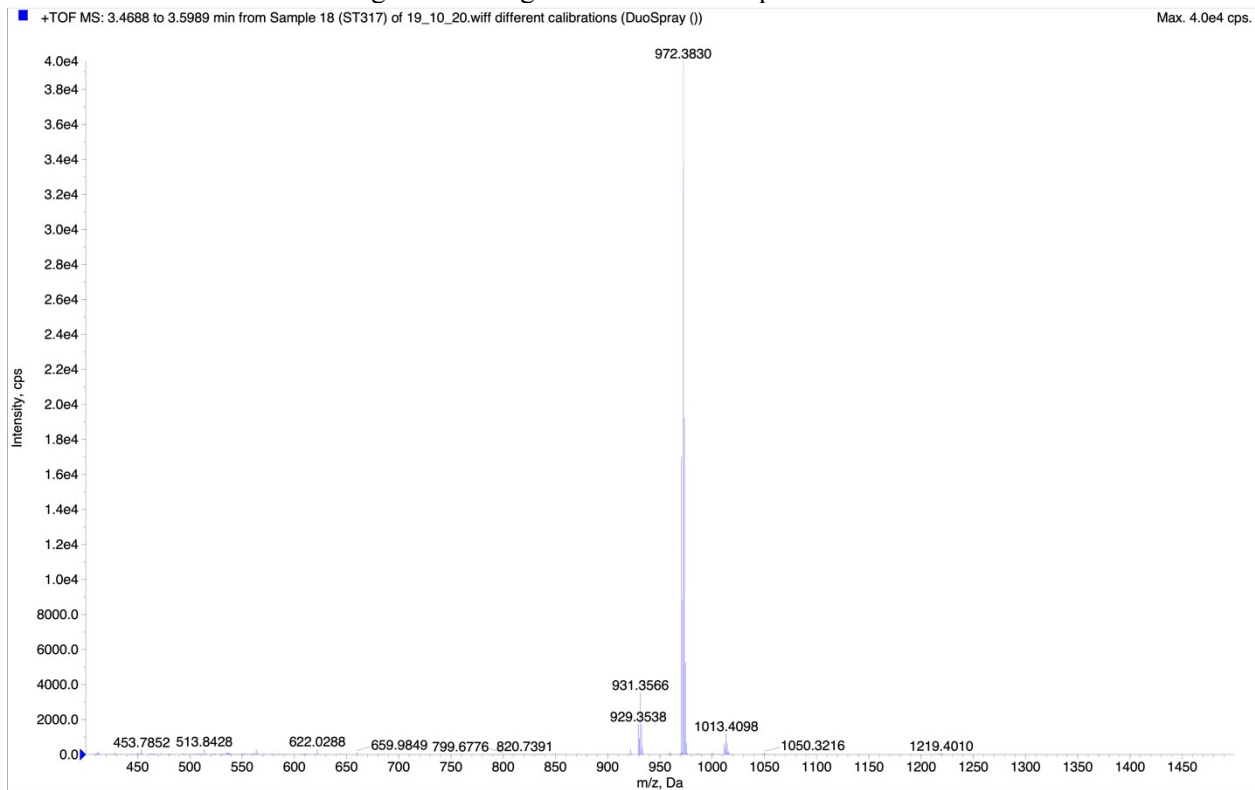


Figure S46. High resolution mass spectrum of 6.

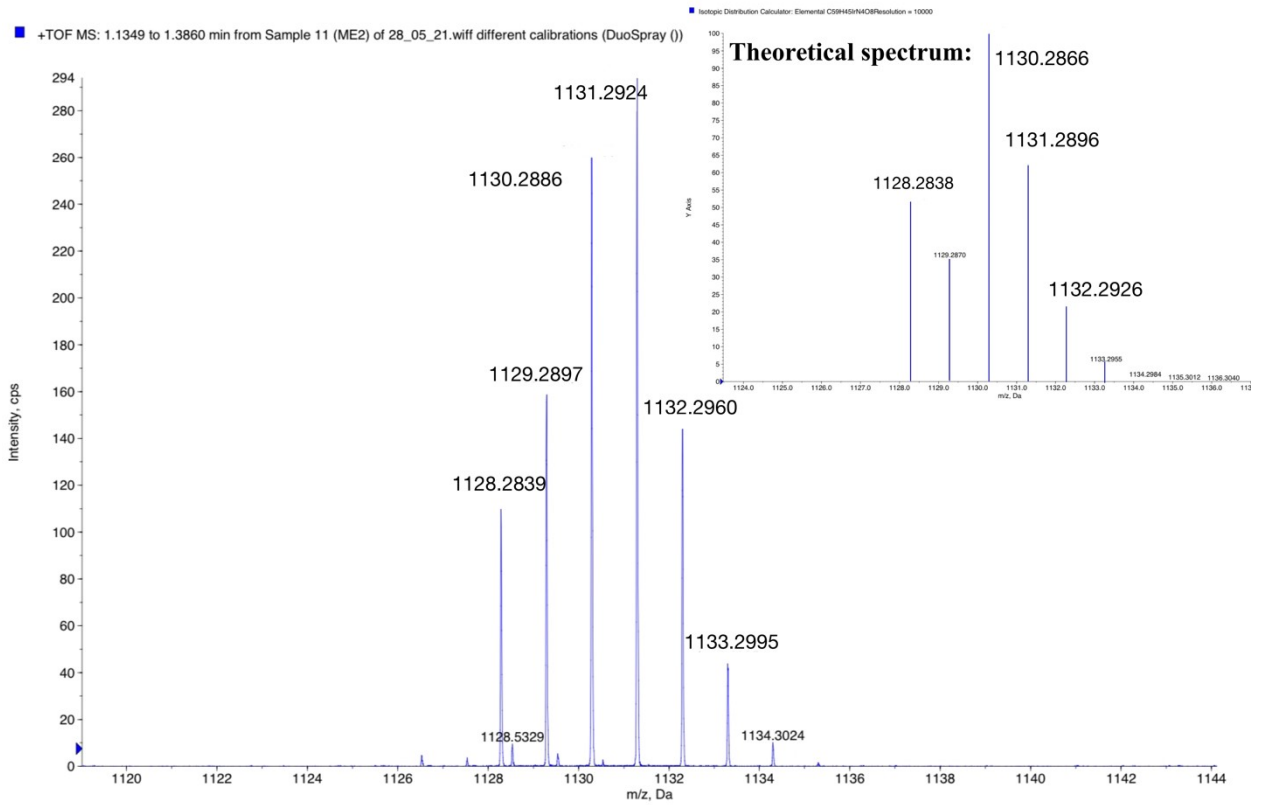


Figure S47. High resolution mass spectrum of 7.

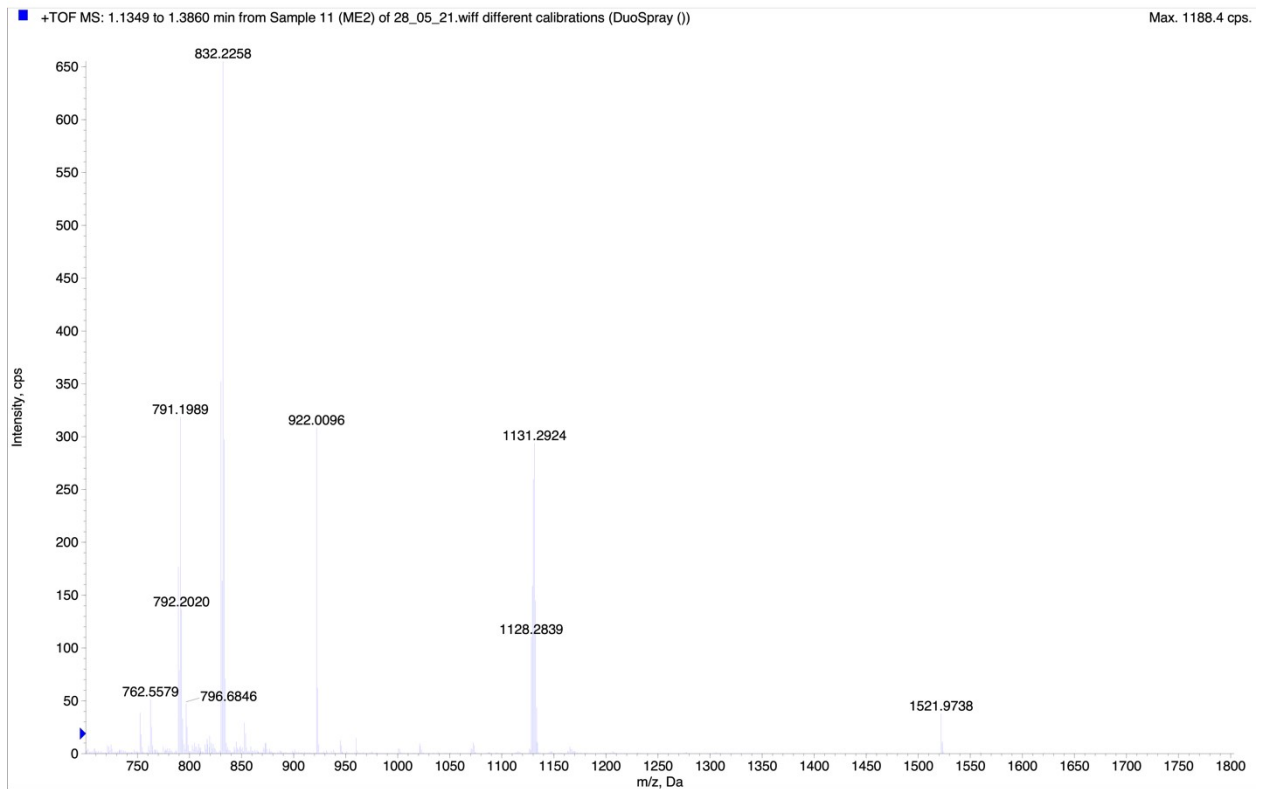


Figure S48. High resolution mass spectrum of 7.

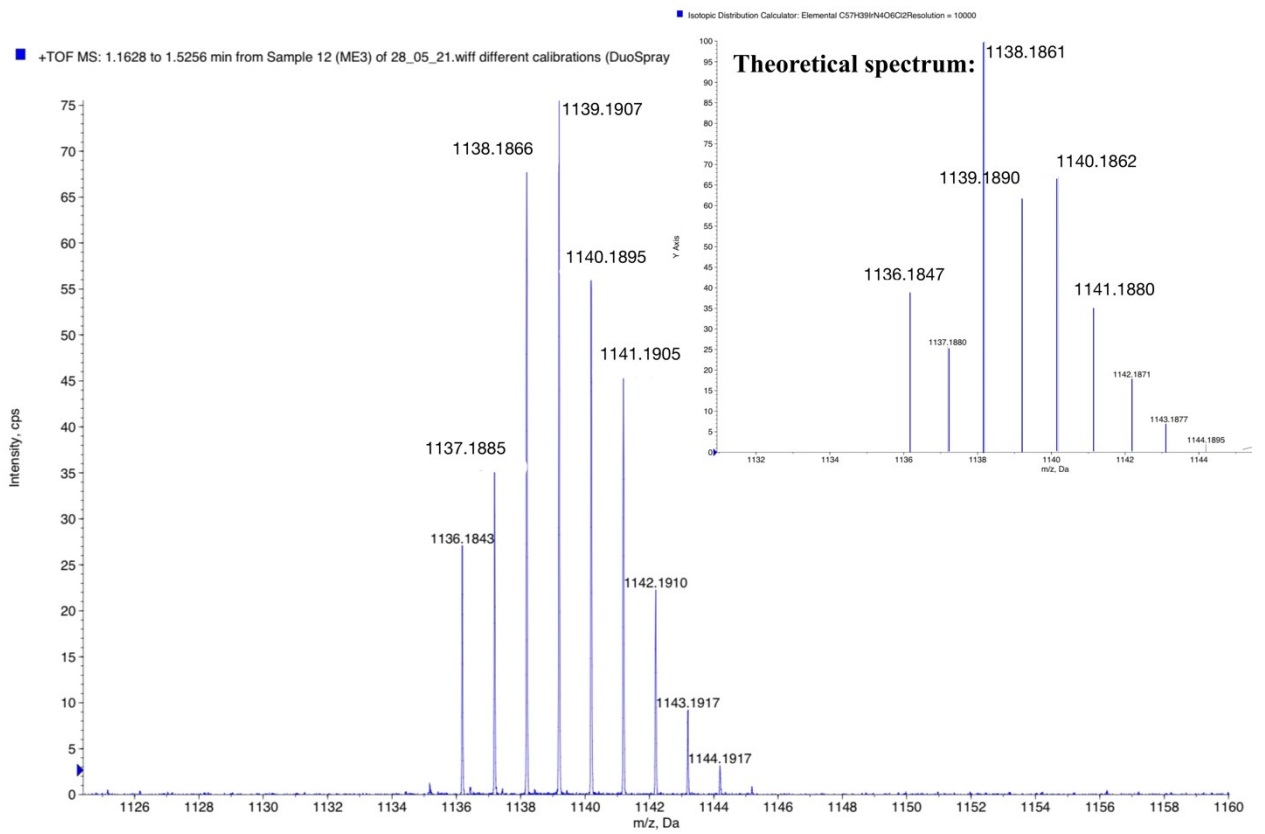


Figure S49. High resolution mass spectrum of **8**.

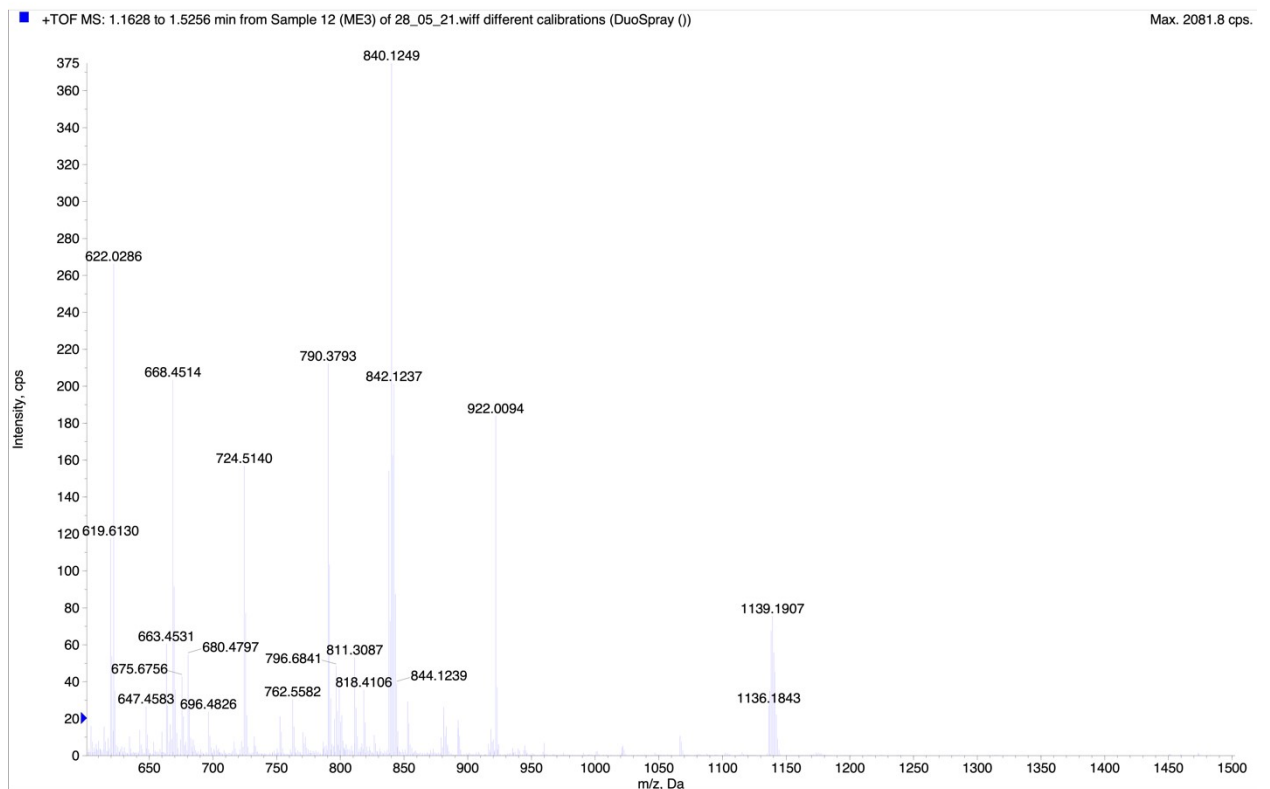


Figure S50. High resolution mass spectrum of **8**.

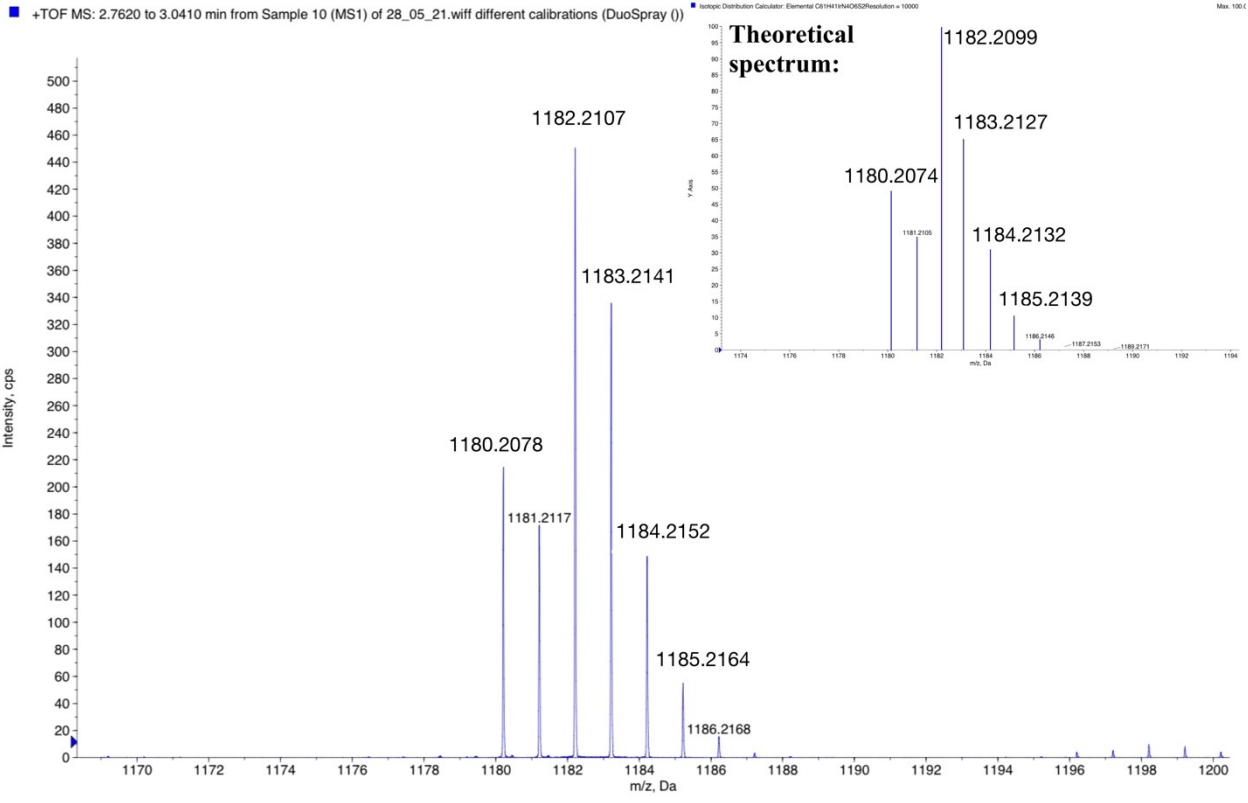


Figure S51. High resolution mass spectrum of **9**.

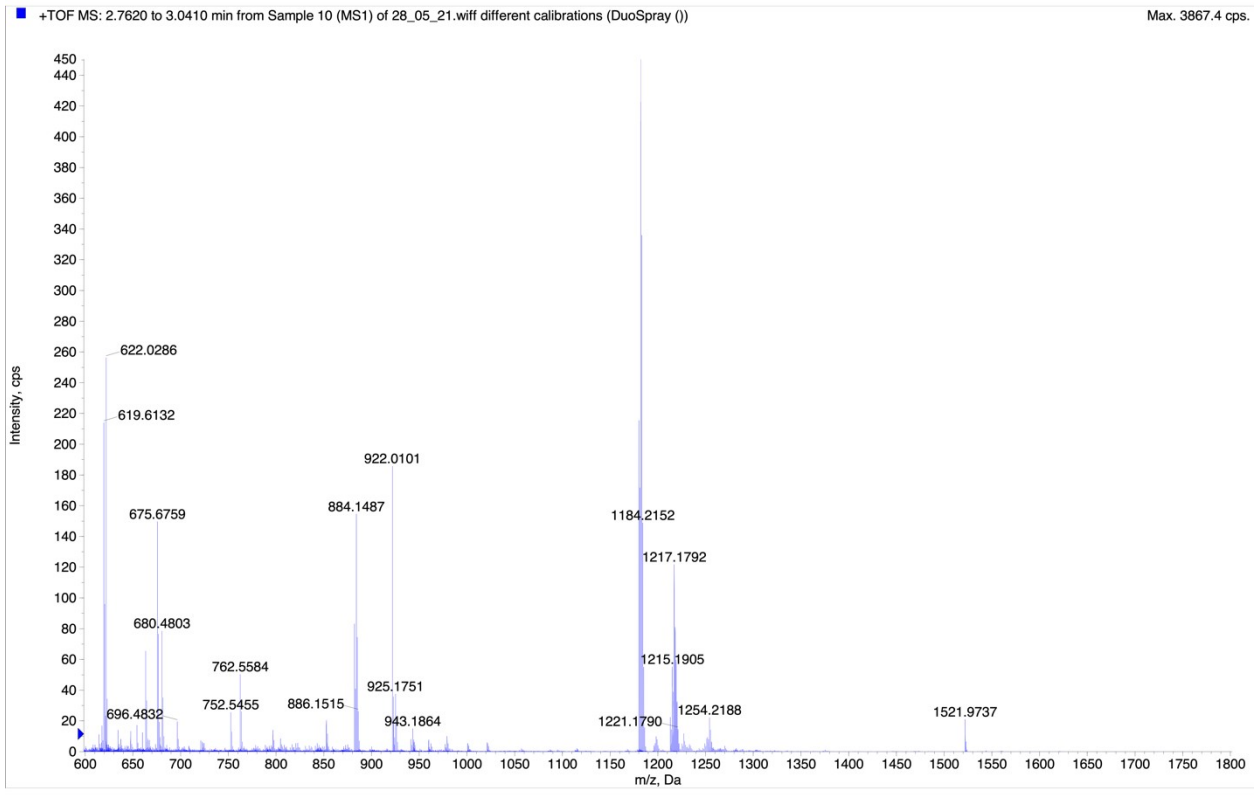


Figure S52. High resolution mass spectrum of **9**.

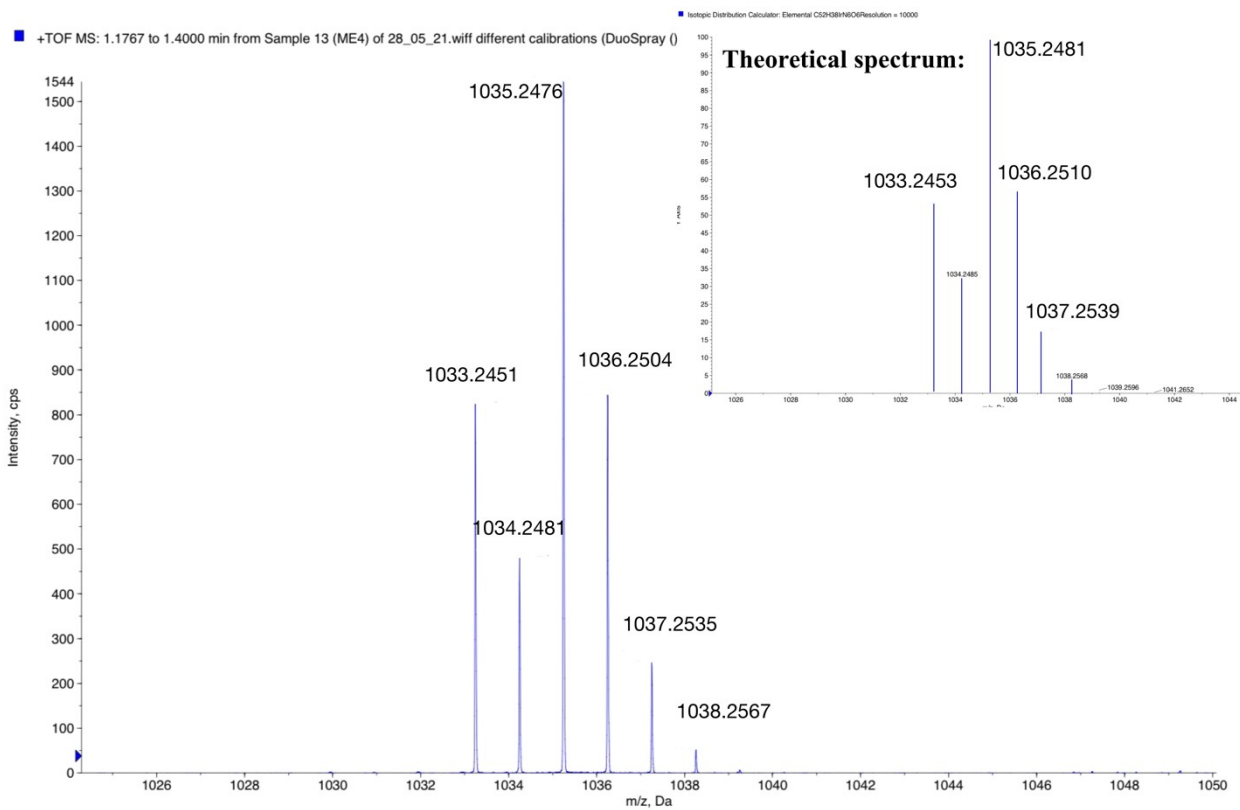


Figure S53. High resolution mass spectrum of **10**.

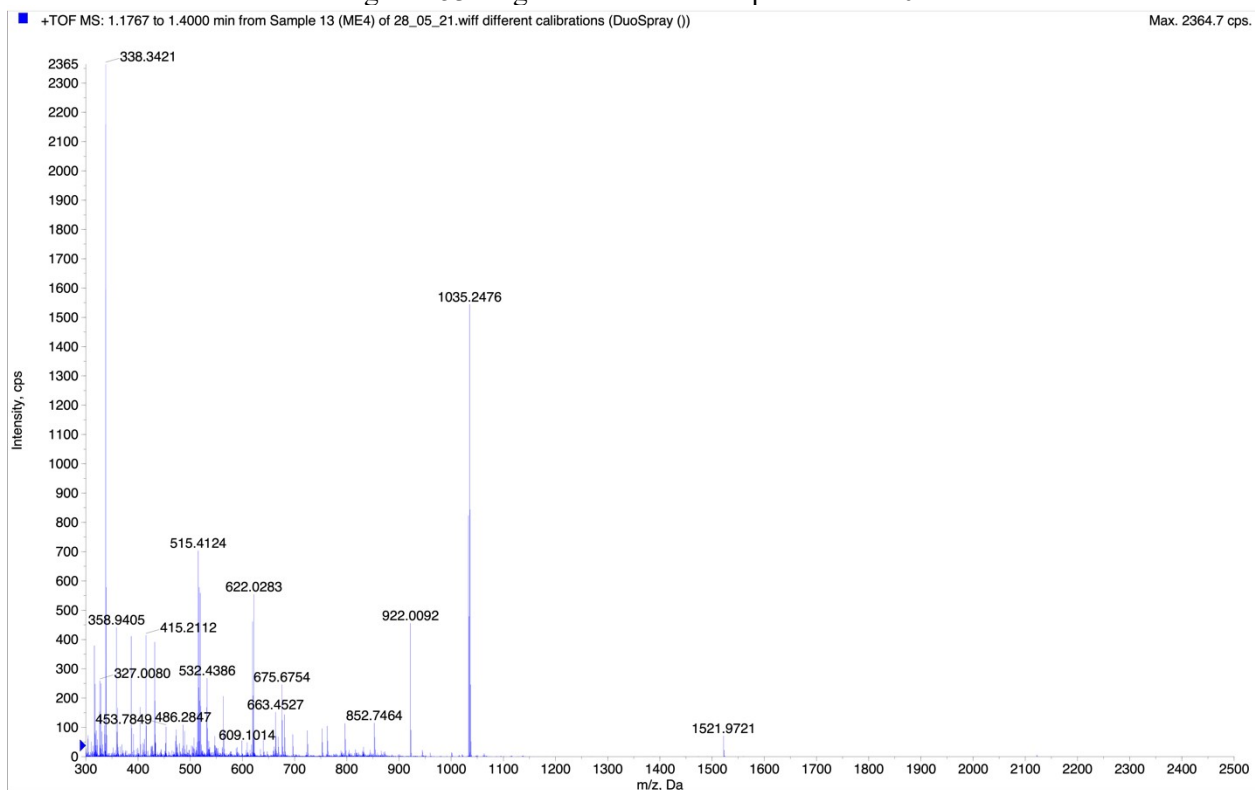


Figure S54. High resolution mass spectrum of **10**.

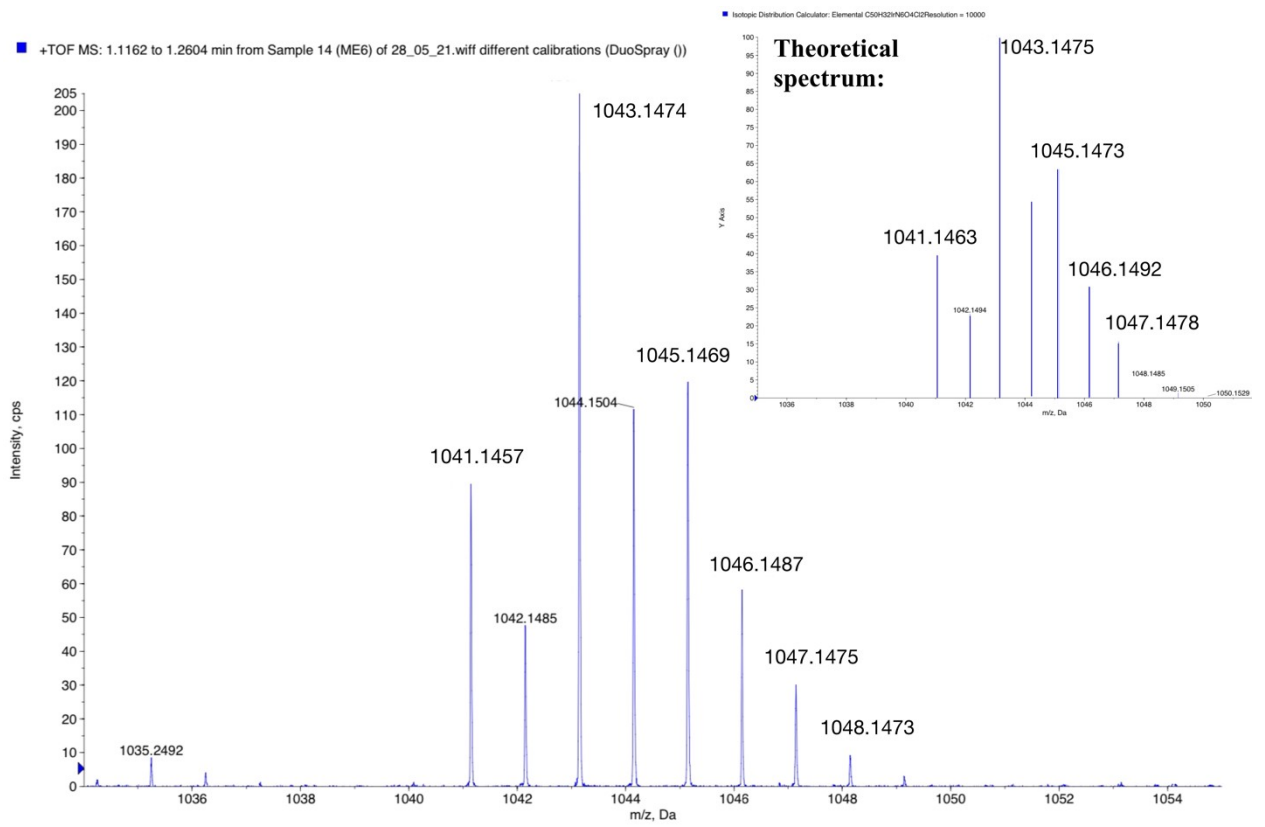


Figure S55. High resolution mass spectrum of **11**.

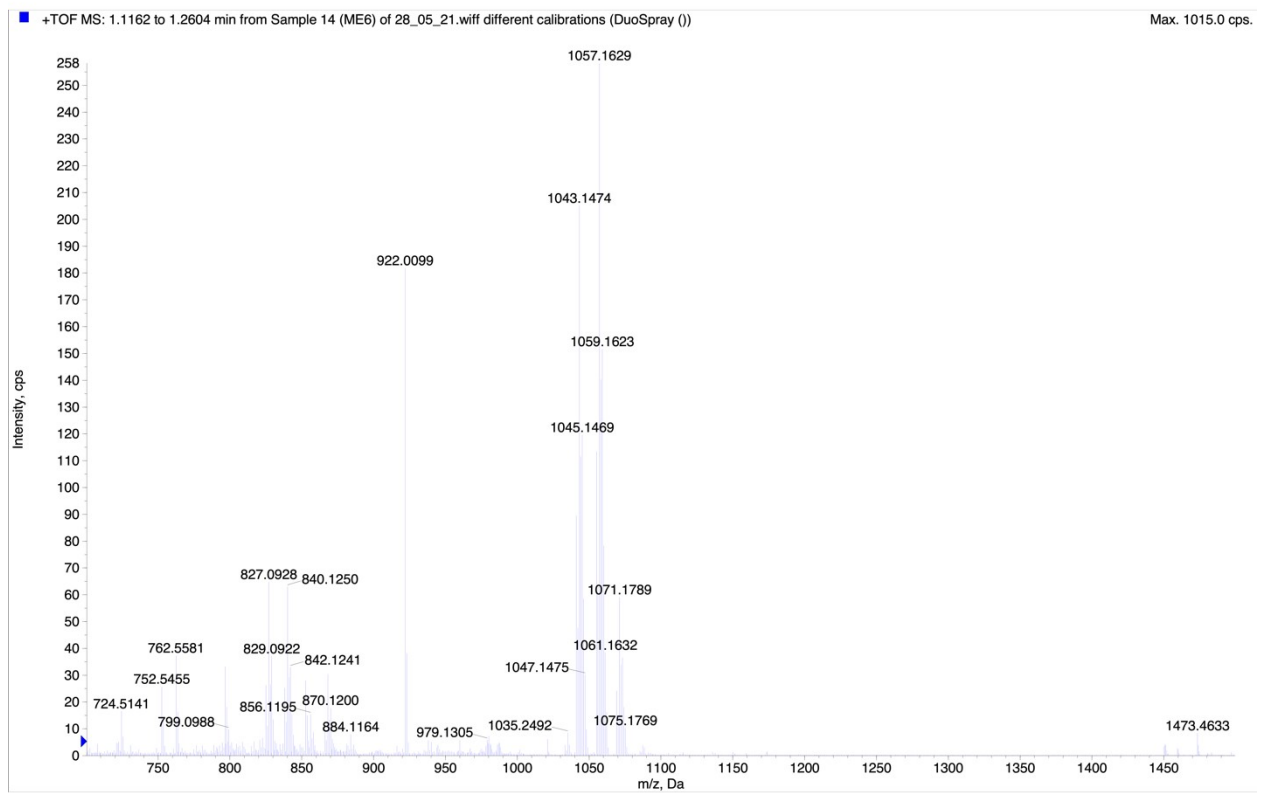


Figure S56. High resolution mass spectrum of **11**.

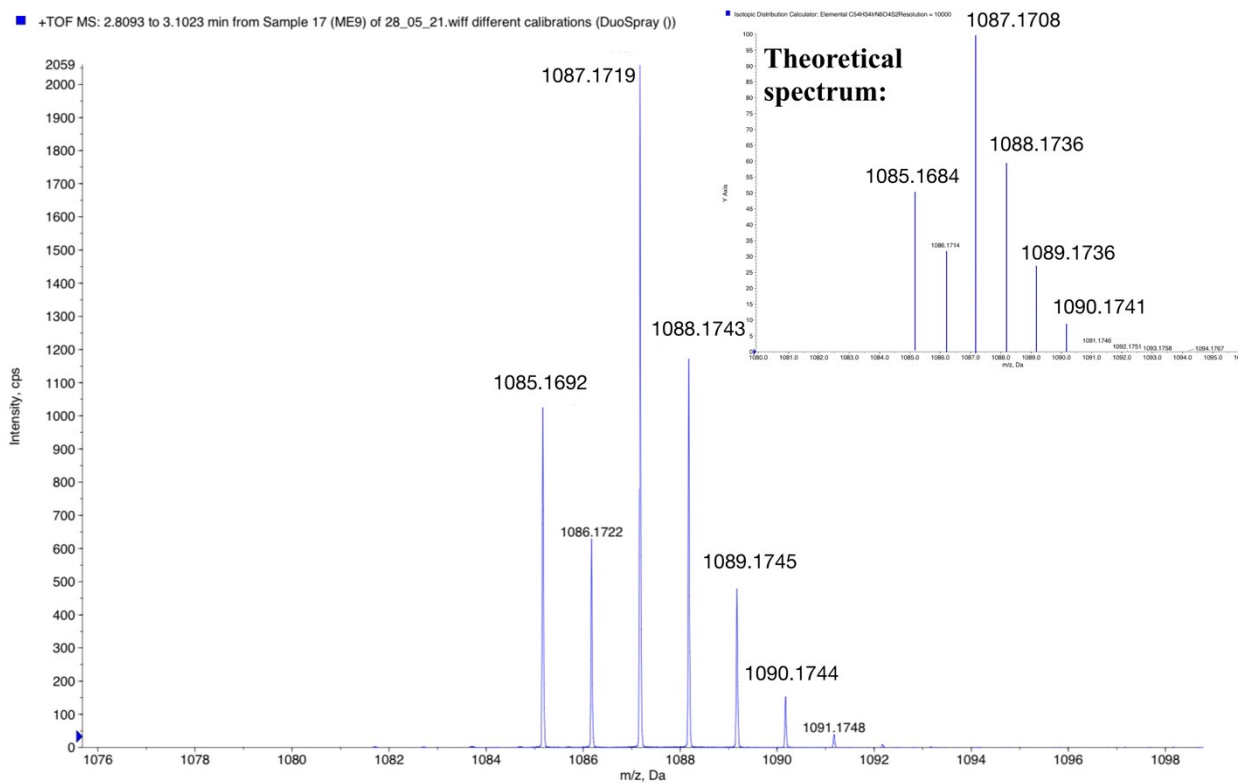


Figure S57. High resolution mass spectrum of **12**.

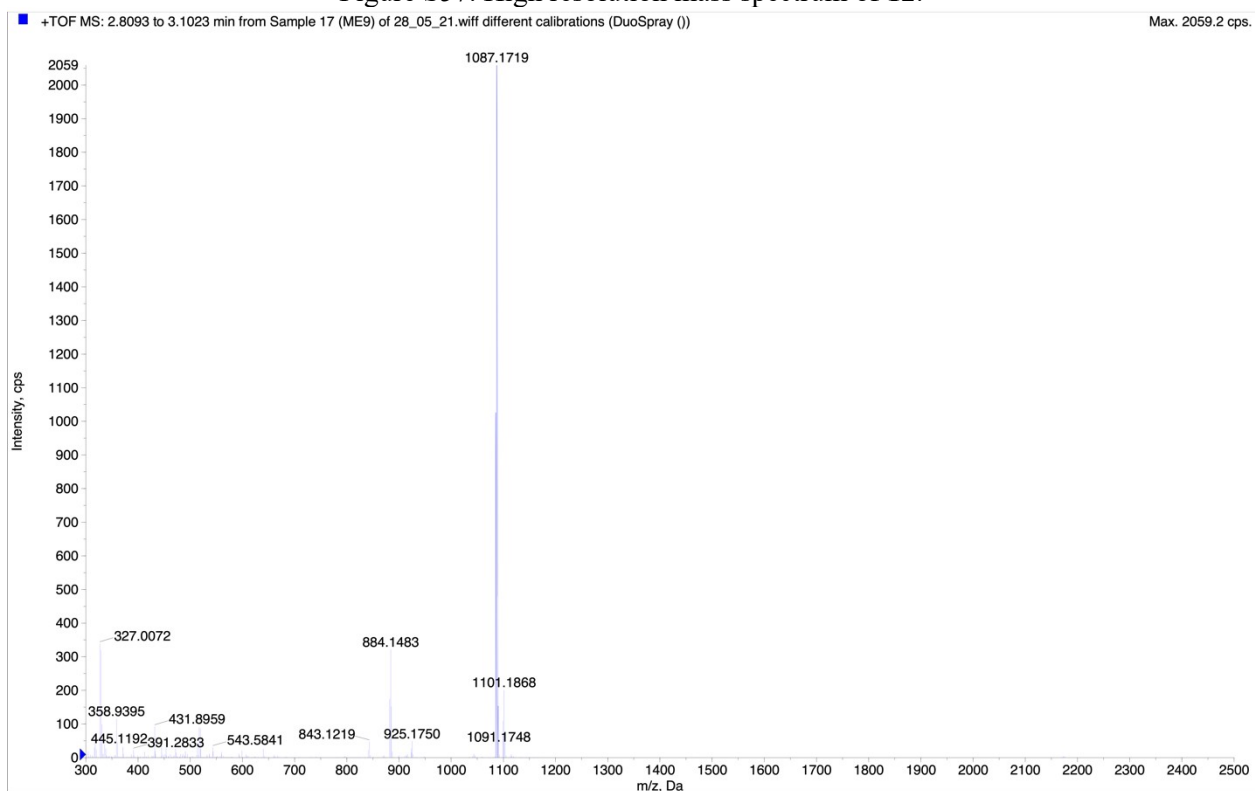


Figure S58. High resolution mass spectrum of **12**.

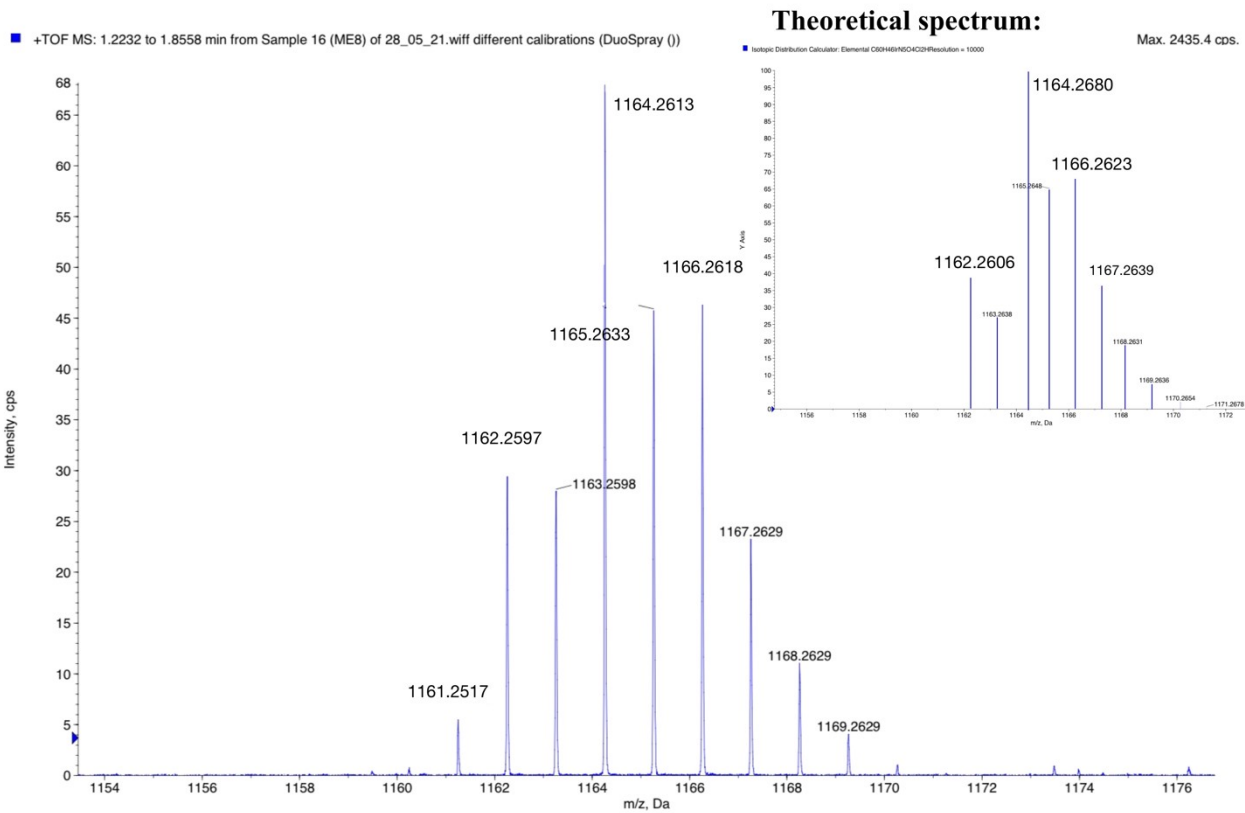


Figure S59. High resolution mass spectrum of **13**.

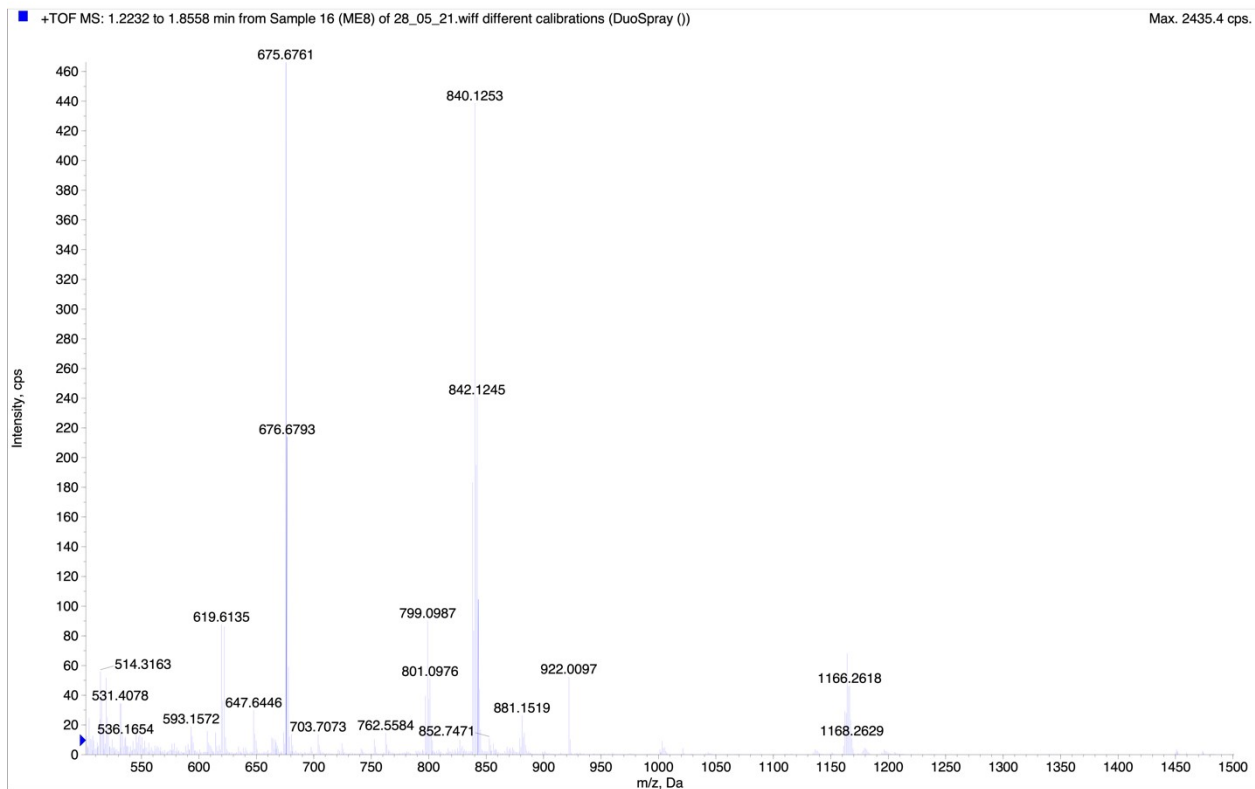
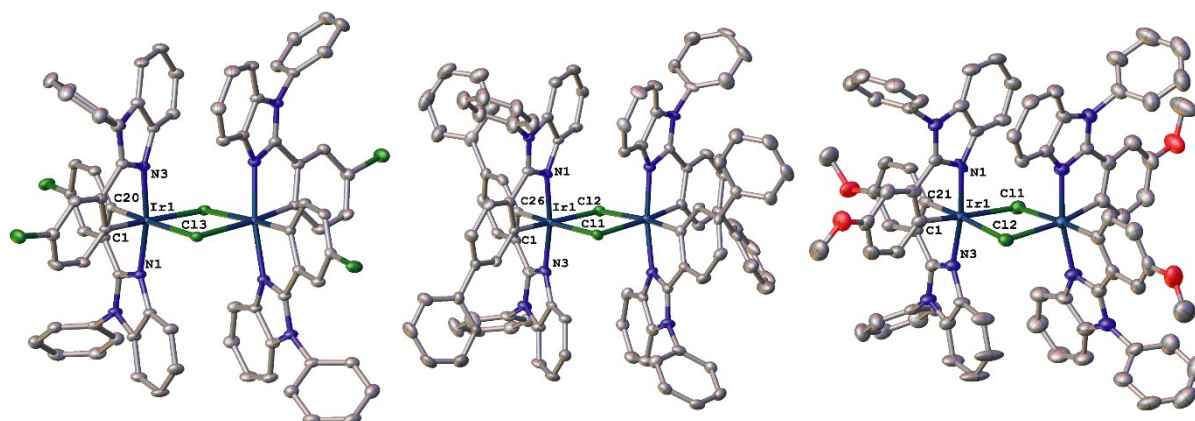


Figure S60. High resolution mass spectrum of **13**.

2. X-ray data.



$[\text{Ir}(\text{3cbi})_2\text{Cl}]_2$

$[\text{Ir}(\text{dpbi})_2\text{Cl}]_2$

$[\text{Ir}(\text{3mbi})_2\text{Cl}]_2$

Figure S61. Molecular structures of the chloride dimers. Displacement ellipsoids are shown at 50% probability level. Minor components of disordered groups and hydrogen atoms are not shown.

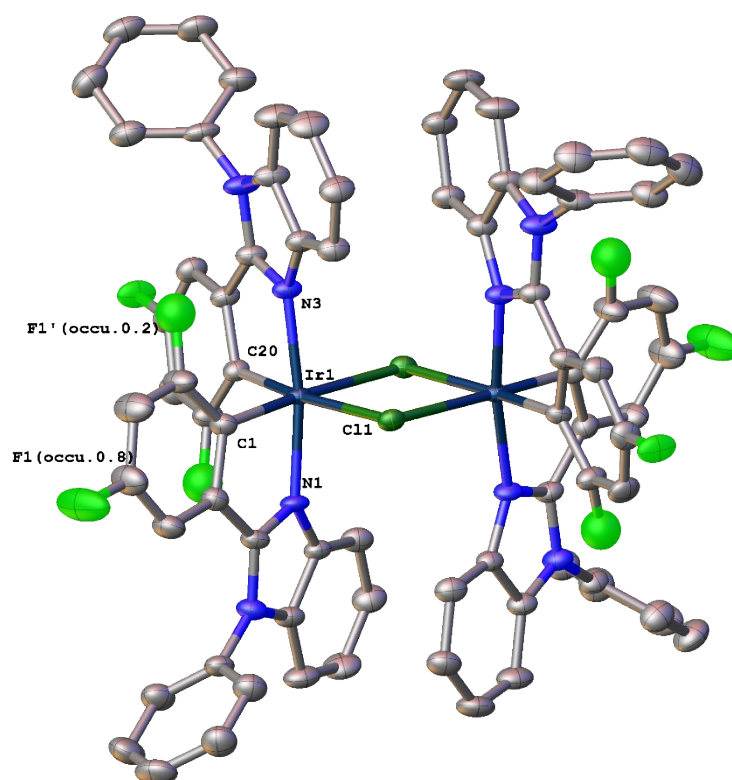


Figure S62. Molecular structure of the $[\text{Ir}(\text{3fbi})_2\text{Cl}]_2$. Displacement ellipsoids are shown at 50% probability level. The hydrogen atoms are not shown. Both fluorine components are shown with their relative occupancies.

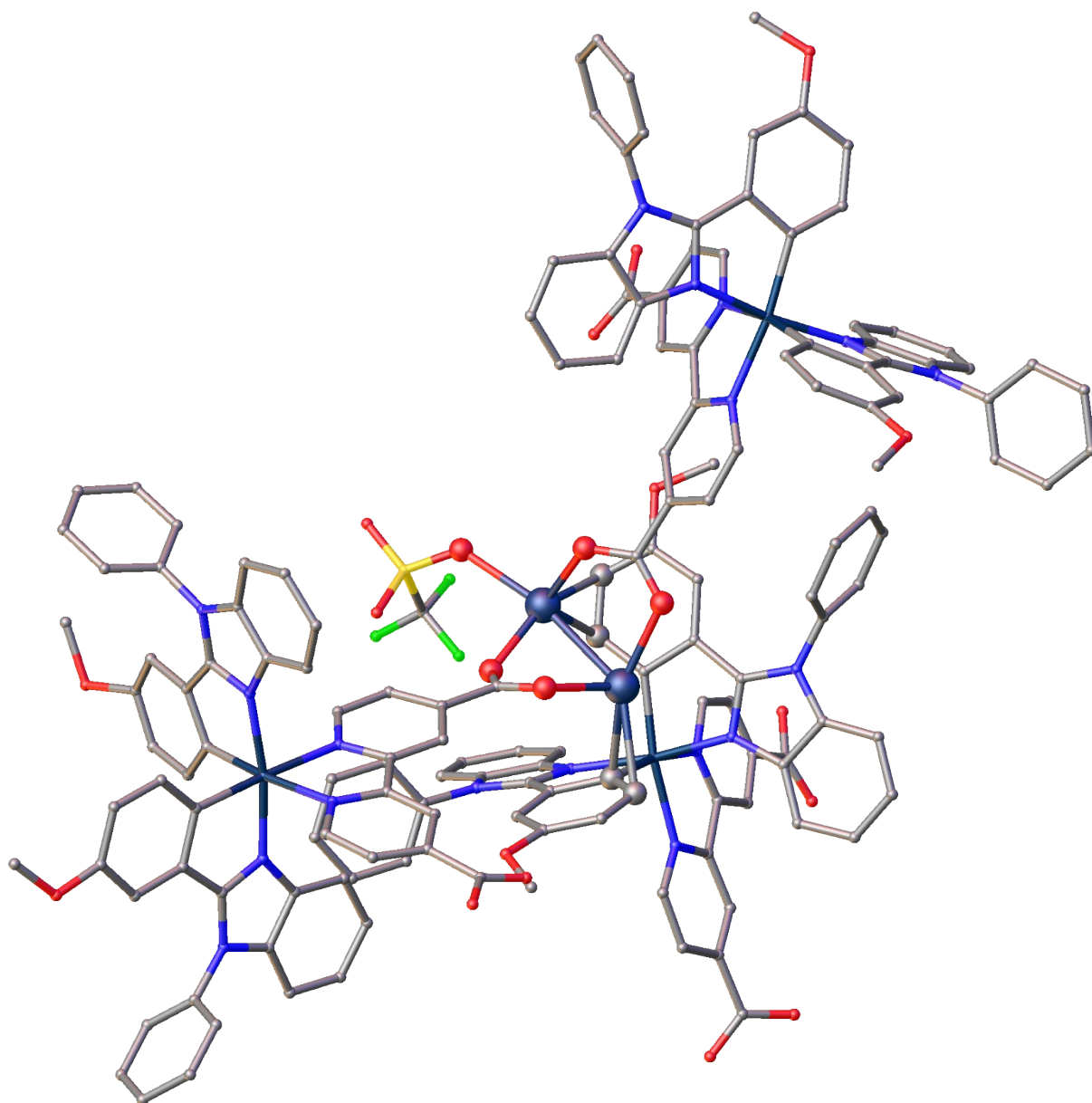


Figure S63. Coordination environment of the silver ions in the Ir₂Ag₂ structure.

Table S2. Details of the X-ray crystal data collection and structure refinement for dimeric compounds

	[Ir(3cbi) ₂ Cl] ₂	[Ir(dpbi) ₂ Cl] ₂	[Ir(3mbi) ₂ Cl] ₂	[Ir(3fbi) ₂ Cl] ₂
Empirical formula	C ₈₀ H ₅₆ Cl ₁₄ Ir ₂ N ₈ + 4CH ₂ Cl ₂	C ₁₀₀ H ₆₈ Cl ₂ Ir ₂ N ₈ +solvent	C ₈₁ H ₆₄ Cl ₂ Ir ₂ N ₈ O ₅ +1 CH ₃ OH	C ₈₀ H ₅₂ Cl ₁₄ F ₄ Ir ₂ N ₈ + 4 CHCl ₃
M _w	2010.03	1989.79	1684.70	2081.99
Temperature (K)	100	100	296.15	150
Size (mm)	0.30 x 0.27 x 0.04	0.4 × 0.38 × 0.08	0.20 × 0.12 × 0.07	0.15 × 0.15 × 0.10
Cryst. system	monoclinic	triclinic	monoclinic	monoclinic
Space group	C2/c	P-1	P2 ₁ /n	C2/c
<i>a</i> (Å)	21.0487(9)	13.0955(3)	20.4452(6)	20.4091(8)
<i>b</i> (Å)	15.2627(6)	19.1698(6)	15.2416(4)	15.0372(5)
<i>c</i> (Å)	23.7169(10)	19.3990(5)	23.5774(7)	25.0183(11)
<i>α</i> (°)	90.00	103.9670(10)	90.00	90.00
<i>β</i> (°)	98.824(2)	93.2770(10)	108.1450(10)	99.107
<i>γ</i> (°)	90.00	101.9530(10)	90.00	90.00
<i>V</i> (Å ³)	7529.1(5)	4593.9(2)	6981.8(3)	7581.2(5)
<i>Z</i>	4	2	4	4
<i>ρ</i> _{calcd} (g·cm ⁻³)	1.773	1.438	1.603	1.824
Abs coeff (mm ⁻¹)	4.080	3.107	3.944	4.063
<i>F</i> (000)	3936.0	1975.0	3336.0	4064.0
<i>θ</i> range (deg)	4.58 < <i>θ</i> < 61.1	3.662 < <i>θ</i> < 52	3.232 < <i>θ</i> < 50.1	3.898 < <i>θ</i> < 57.398
no. of collected/unique rflns	70680/11449	30399/17642	52970/ 12349	40903/ 9781
Completeness to <i>θ</i> (%)	99.2	97.7	99.9	100
no. of data/restraints/params	11449/0/469	17642/14/1031	12349/1/903	9781/6/501
Goodness of fit on <i>F</i> ²	1.028	1.022	1.026	1.041
Final <i>R</i> indices (<i>I</i> > 2σ(<i>I</i>))	R ₁ = 0.0308, wR ₂ = 0.0645	R ₁ = 0.0416, wR ₂ = 0.0951	R ₁ = 0.0372, wR ₂ = 0.0917	R ₁ = 0.0406, wR ₂ = 0.1014
<i>R</i> indices (all data)	R ₁ = 0.0424, wR ₂ = 0.0681	R ₁ = 0.0607, wR ₂ = 0.1013	R ₁ = 0.0536, wR ₂ = 0.1000	R ₁ = 0.0536, wR ₂ = 0.1092
Largest diff peak/hole (e/Å ³)	1.89/-1.41	1.87/-1.59	1.80/-1.48	1.62/-1.73

Table S3. Details of the X-ray crystal data collection and structure refinement for complexes 1, 2, 5, 7-10, 12, 13

	1	2	5	7	8	9	10	12	13
Empirical formula	$C_{53}H_{35}F_2IrN_4O_4S$ + solvent	$C_{53}H_{35}Cl_2IrN_4O_4S$	$C_{55}H_{35}F_6IrN_4O_4S$ + $CH_2Cl_2/$ CH_3OH	$C_{59}H_{45}N_4O_8Ir$ + solvent	$C_{57}H_{39}Cl_2IrN_4O_6$ + solvent	$C_{61}H_{41}IrN_4O_6S_2$ + solvent	$[C_{52}H_{37}IrN_6O_6]OTf$	$[C_{54}H_{36}IrN_6O_4S_2]OTf$	$C_{60}H_{46}Cl_2IrN_5O_4$
M_w	1054.11	1087.01	1212.61	1130.19	1139.02	1182.30	1109.11	866.52	1164.12
Temperature (K)	100.00	100.00	100.00	200.00	150.00	100.15	150.00	150.00	100.00
Size (mm)	$0.14 \times 0.12 \times$ 0.005	$0.1 \times 0.1 \times 0.004$	$0.22 \times 0.20 \times$ 0.11	$0.42 \times 0.28 \times$ 0.20	$0.35 \times 0.30 \times$ 0.30	$0.2 \times 0.03 \times$ 0.003	$0.25 \times 0.22 \times 0.05$	$0.30 \times 0.30 \times 0.07$	$0.45 \times 0.40 \times 0.25$
Cryst. system	monoclinic	monoclinic	triclinic	orthorhombic	orthorhombic	triclinic	triclinic	triclinic	triclinic
Space group	$P2_1/n$	$P2_1/n$	P-1	$Pnc2$	$Pbcn$	P-1	P-1	P-1	P-1
a (Å)	11.3870(9)	13.2139(5)	11.6380(3)	29.2455(14)	13.7388(5)	12.7102(7)	15.1677(14)	20.2850(8)	12.8762(11)
b (Å)	22.3107(18)	24.0162(16)	12.0786(4)	13.7868(7)	12.7766(5)	14.9577(8)	17.1131(17)	21.1480(8)	12.9389(11)
c (Å)	18.4099(14)	13.8114(8)	18.5157(5)	12.7980(6)	28.3069(9)	15.6155(8)	20.8414(18)	28.2673(11)	15.9754(15)
α (°)	90.00	90.00	91.2080(10)	90.00	90.00	115.4936(19)	69.915(3)	80.1990(10)	80.087(3)
β (°)	102.871(3)	90.525(2)	96.2940(10)	90.00	90.00	104.484(2)	80.152(3)	71.5230(10)	89.178(3)
γ (°)	90.00	90.00	100.4940(10)	90.00	90.00	100.722(2)	72.408(4)	81.6440(10)	69.118(3)
V (Å ³)	4559.6(6)	4382.8(4)	2541.56(13)	5160.2(4)	4968.9(3)	2443.6(2)	4829.4(8)	11279.1(8)	2446.4(4)
Z	4	4	2	4	4	2	4	8	2
ρ_{cald} (g·cm ⁻³)	1.536	1.647	1.585	1.455	1.523	1.607	1.525	1.531	1.580
Abs coeff (mm ⁻¹)	3.033	3.270	2.794	2.647	2.850	2.878	2.852	2.525	2.894
$F(000)$	2096.0	2160.0	1204.0	2272.0	2272.0	1184.0	2212.0	5210.0	1168.0

θ range (deg)	2.912 < θ < 52.81	3.402 < θ < 51.406	3.99 < θ < 52.774	3.266 < θ < 50.096	4.132 < θ < 52.79	4.66 < θ < 53.996	3.312 < θ < 50.098	2.676 < θ < 55.998	3.856 < θ < 52.762
no. of collected/unique rflns	47812/9340	8346/8297	53682/10384	82138/9157	49989/5089	21603/10623	40635/16932	163620/54411	39476/9973
Completeness to θ (%)	99.9	99.4	99.8	100	99.9	99.9	99.9	99.9	99.6
no. of data/restraints/params	9340/2/590	8297/257/592	10384/61/584	9157/200/643	5089/0/318	10623/193/755	16932/39/1280	54411/62/2900	9973/1/614
Goodness of fit on F^2	1.080	1.141	1.154	1.086	1.063	1.014	1.035	1.020	1.050
Final R indices ($I > 2\sigma(I)$)	$R_1 = 0.0469$, $wR_2 = 0.1074$	$R_1 = 0.0649$, $wR_2 = 0.1562$	$R_1 = 0.0450$, $wR_2 = 0.1164$	$R_1 = 0.0308$, $wR_2 = 0.0747$	$R_1 = 0.0239$, $wR_2 = 0.0522$	$R_1 = 0.0441$, $wR_2 = 0.0761$	$R_1 = 0.0556$, $wR_2 = 0.1134$	$R_1 = 0.0373$, $wR_2 = 0.0723$	$R_1 = 0.0392$, $wR_2 = 0.0824$
R indices (all data)	$R_1 = 0.0659$, $wR_2 = 0.1161$	$R_1 = 0.0797$, $wR_2 = 0.1636$	$R_1 = 0.0477$, $wR_2 = 0.1179$	$R_1 = 0.0521$, $wR_2 = 0.0881$	$R_1 = 0.0341$, $wR_2 = 0.0561$	$R_1 = 0.0659$, $wR_2 = 0.0837$	$R_1 = 0.0962$, $wR_2 = 0.1281$	$R_1 = 0.0558$, $wR_2 = 0.0780$	$R_1 = 0.0529$, $wR_2 = 0.0883$
Largest diff peak/hole ($e/\text{\AA}^3$)	2.23/-1.69	3.05/-2.27	2.16/-1.42	1.69/-0.82	0.89/-1.16	0.96/-1.26	1.93/-1.48	1.97/-0.74	0.80/-0.95

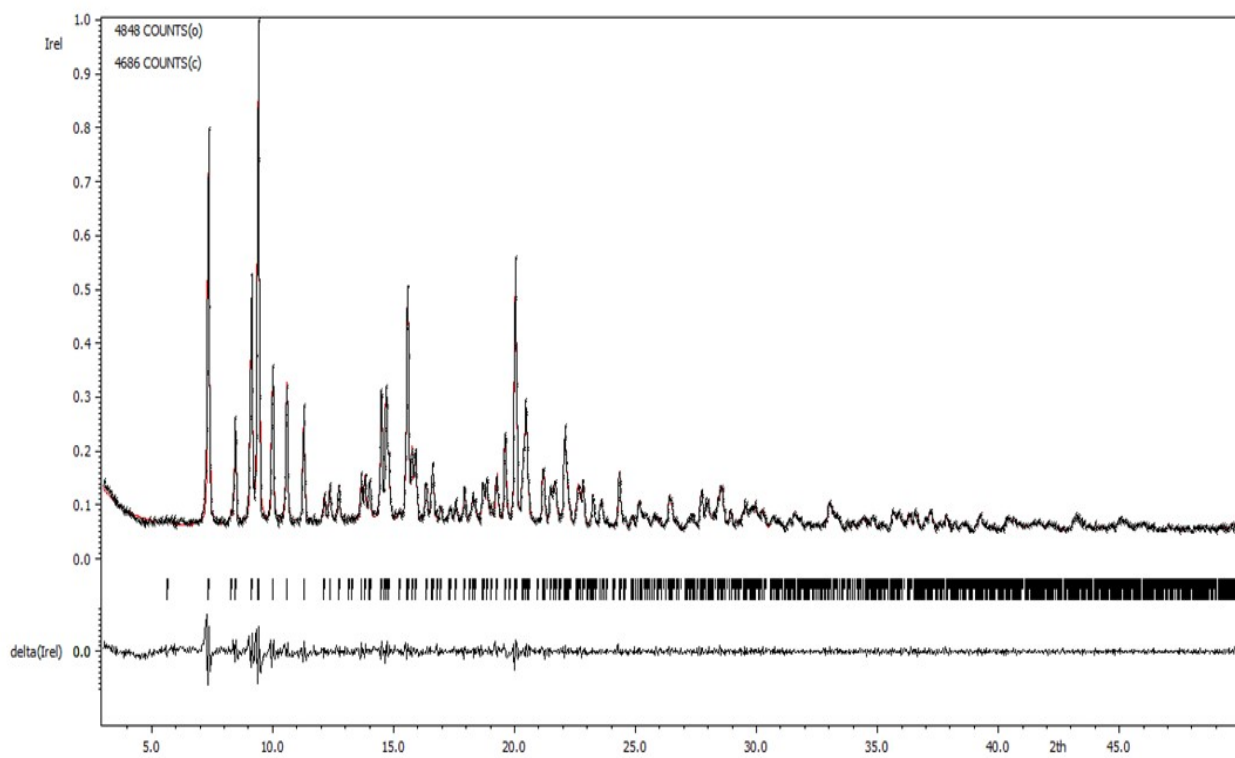


Figure S64. Powder XRD pattern of complex **13** (black line) with the simulated pattern (red line). The differential is presented in the bottom.

3. Redox and optical properties.

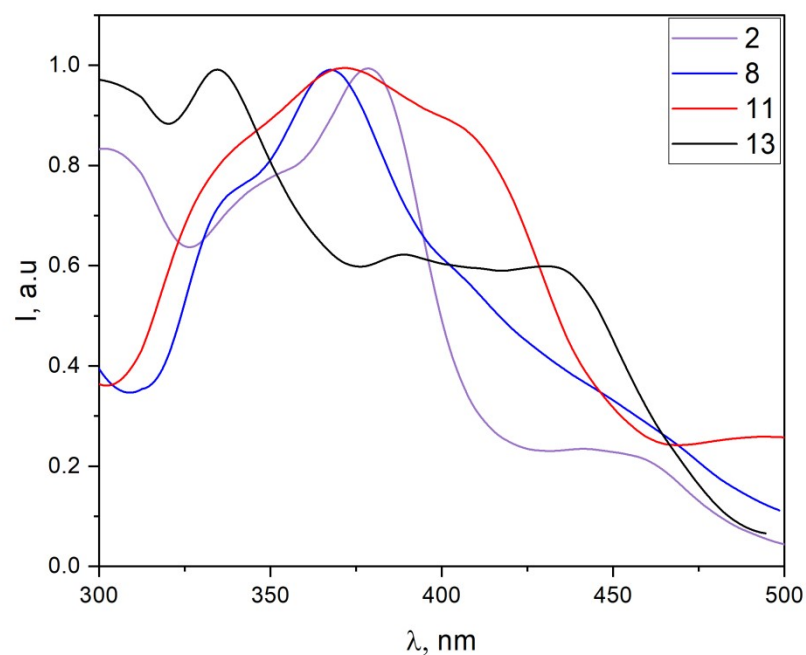


Figure S65. Normalised excitation spectra of complexes **2**, **8**, **11**, **13** with 3cbi as cyclometalated ligand measured under Ar atmosphere in CH_2Cl_2 at 25°C

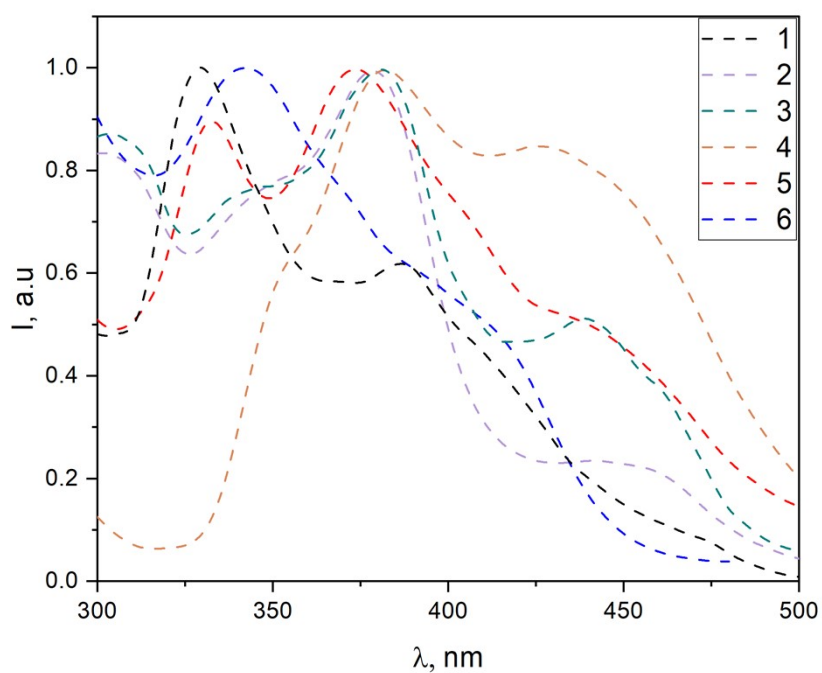


Figure S66. Normalised excitation spectra of complexes **1-6** with L_1 as ancillary ligand measured under Ar atmosphere in CH_2Cl_2 at 25°C

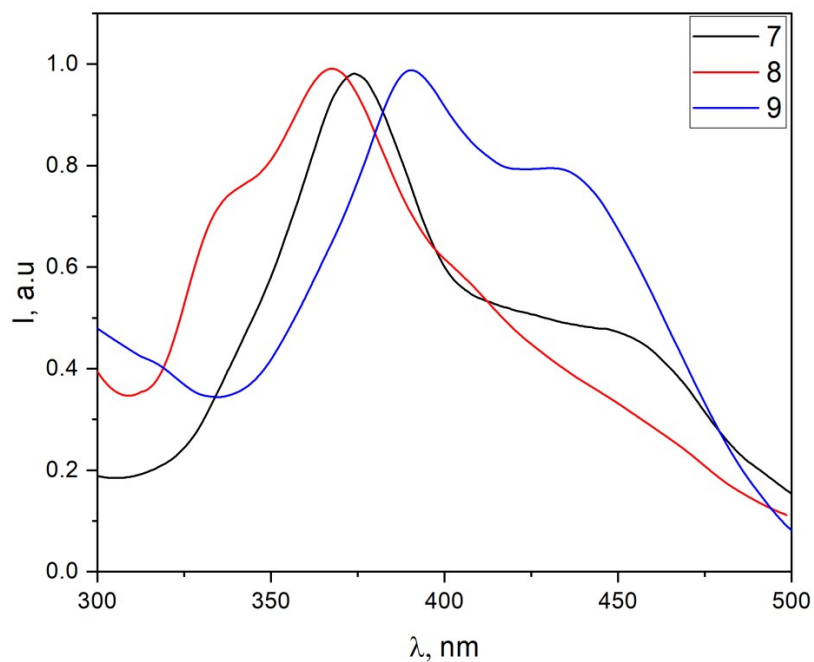


Figure S67. Normalised excitation spectra of complexes **7-9** with L_2 as ancillary ligand measured under Ar atmosphere in CH_2Cl_2 at $25^\circ C$

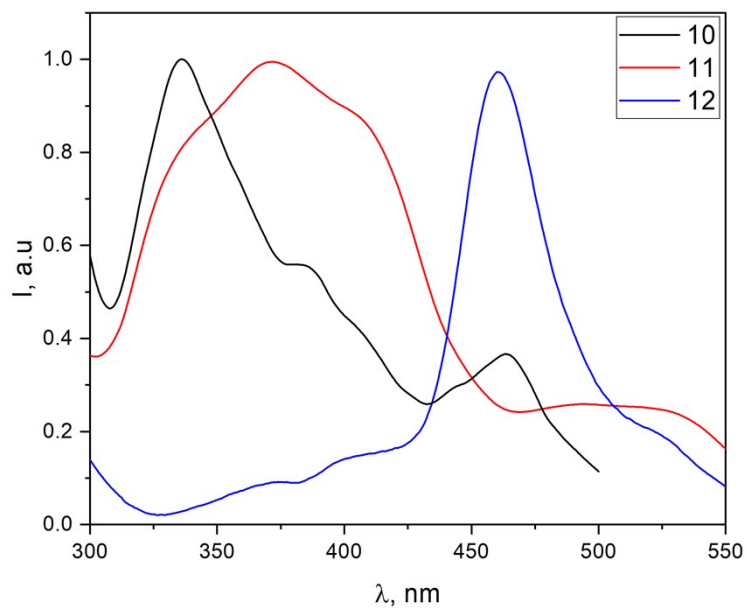


Figure S68. Normalised excitation spectra of complexes **10-12** with L_3 as ancillary ligand measured under Ar atmosphere in CH_2Cl_2 with 0,01M CF_3COOH at $25^\circ C$

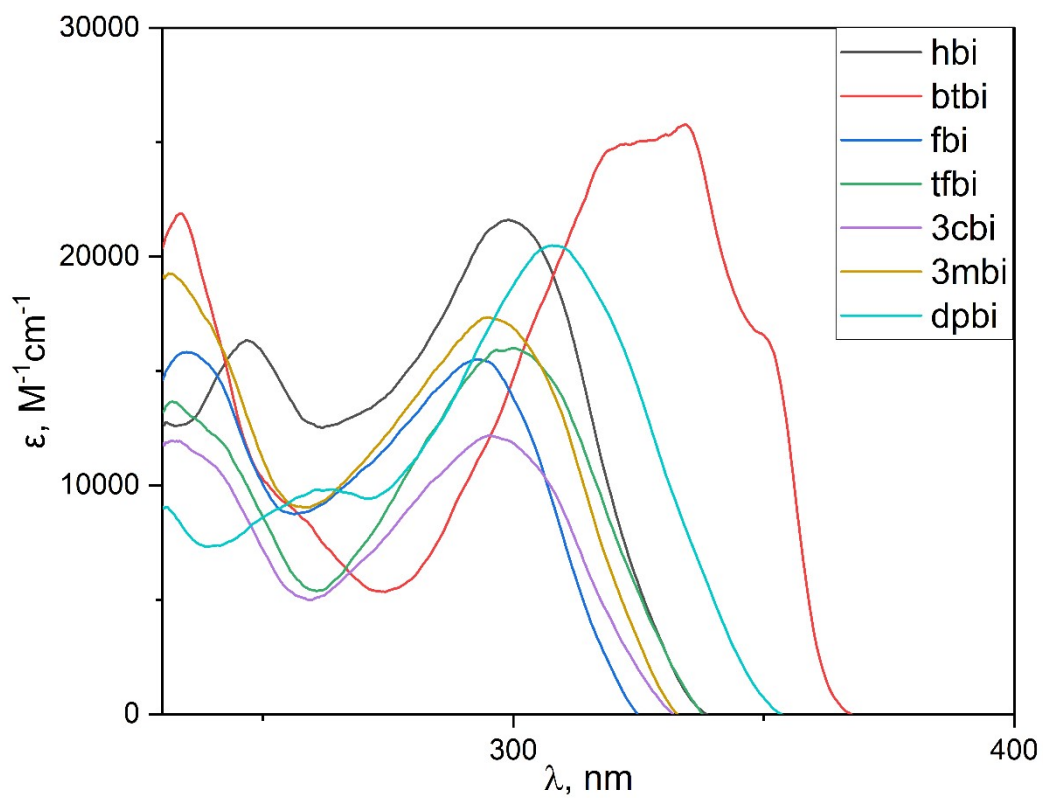


Figure S69. Absorption spectra of cyclometalating benzimidazole ligands (CH_2Cl_2).

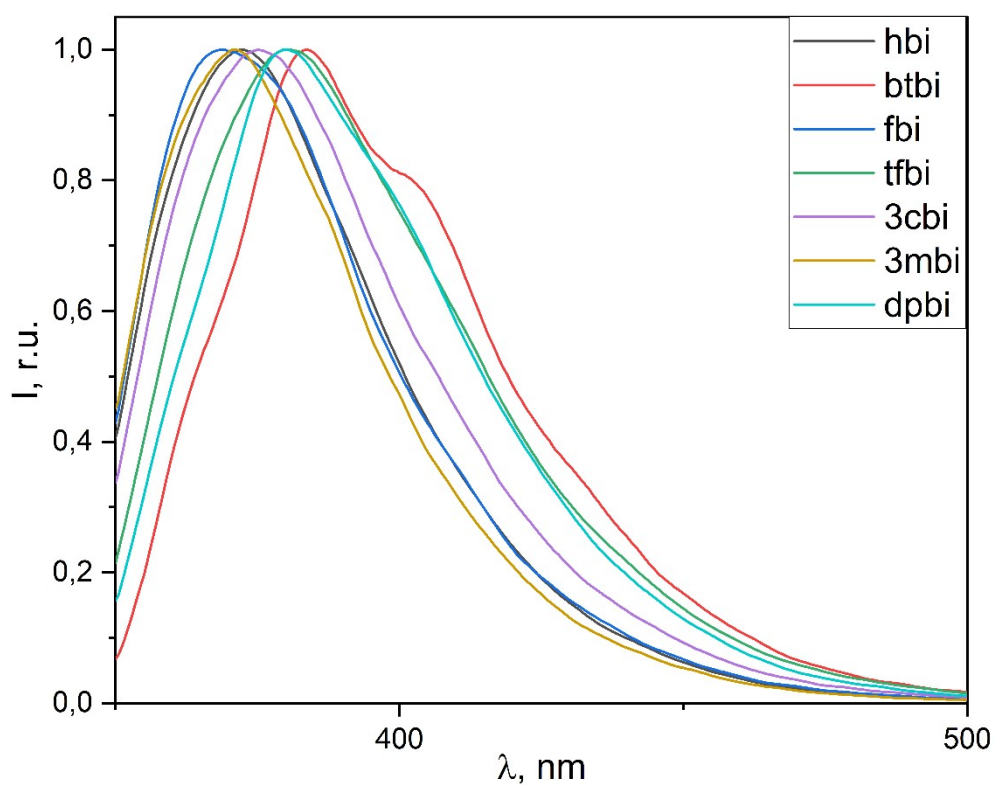


Figure S70. Normalised luminescence spectra of cyclometalating benzimidazole ligands (CH_2Cl_2).

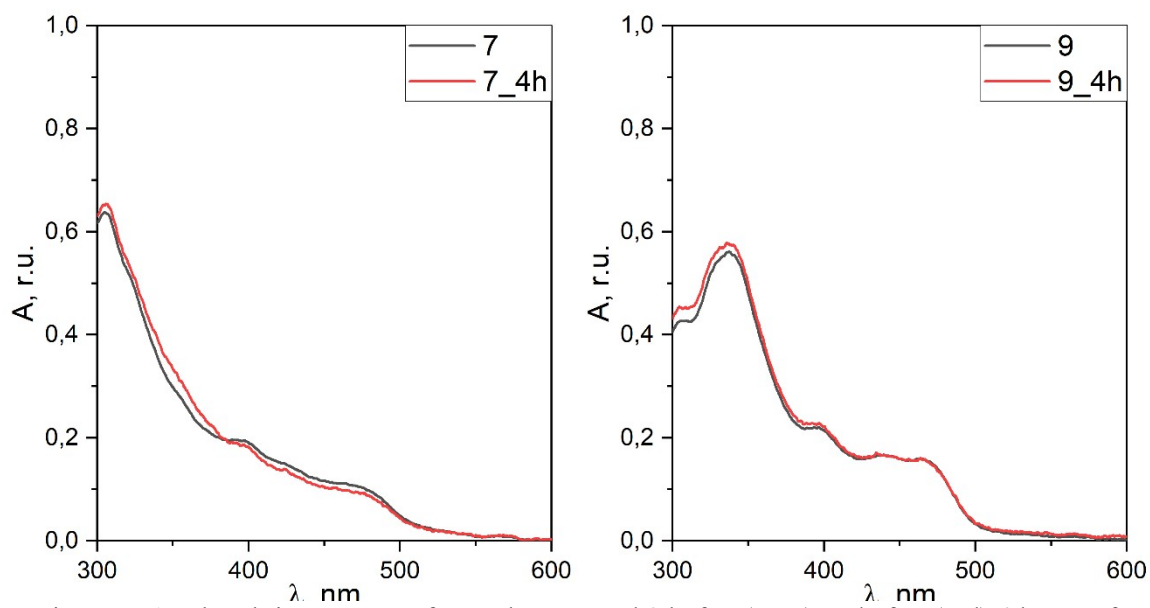


Figure S71. Absorbtion spectra of complexes 7 and 9 before(grey) and after (red) 4 hours of UV irradiation.

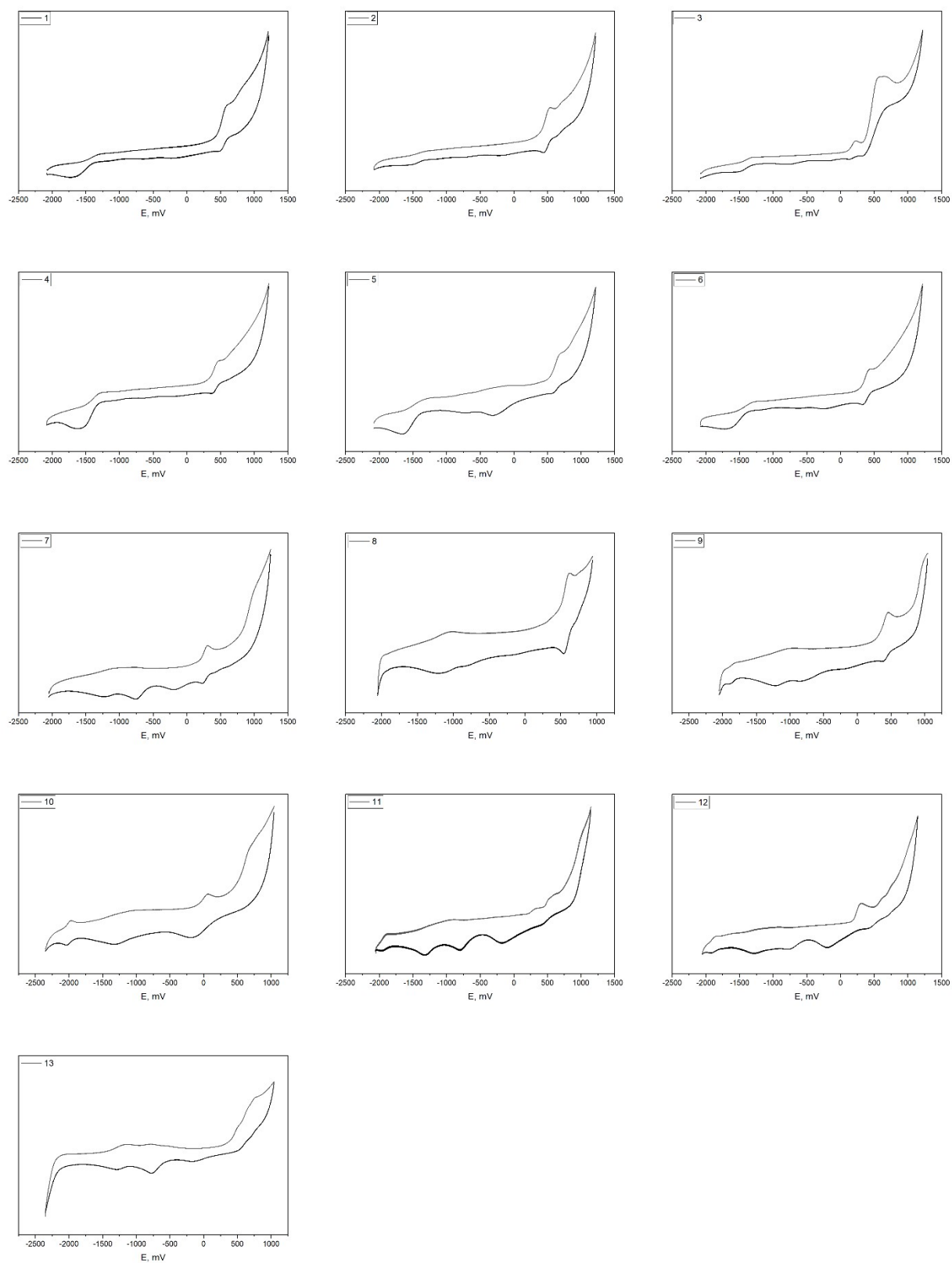


Figure S72. Cyclic voltammetry curves of complexes **1** – **13**. E is presented vs Fc^+/Fc couple.

4. Computational details

Table S4. Comparison of the bond lengths from the crystal structures and the calculations
(structure/calculation)

	1	2	5	7	8	9	13
Ir – C1	1.997 / 2.0173	1.986 / 2.0144	1.989 / 2.0188	1.971 / 2.0163	1.998 / 2.0115	2.001/2.0231	1.993 / 2.0182
Ir – C2	2.003 / 2.0109	1.990 / 2.0145	2.000 / 2.0120	1.971 / 2.0164	1.998 / 2.0115	1.997/2.0207	1.998 / 2.0140
Ir – N1	2.028 / 2.0810	2.042 / 2.0722	2.023 / 2.0714	2.040 / 2.0677	2.045 / 2.0630	2.028/2.0818	2.039 / 2.0802
Ir – N2	2.043 / 2.0619	2.046 / 2.0732	2.033 / 2.0660	2.040 / 2.0675	2.045 / 2.0632	2.054/2.0858	2.030 / 2.0738
Ir – O1	2.152 / 2.2175	2.140 / 2.1866	2.112 / 2.1824	2.133 / 2.1940	2.145 / 2.1745	2.108/2.1731	2.119 / 2.1925
Ir – O2	2.121 / 2.1732	2.152 / 2.1915	2.130 / 2.1889	2.133 / 2.1925	2.145 / 2.1745	2.154/2.1750	2.133 / 2.1774

Table S5. Comparison of the bond lengths from the crystal structures and the calculations
(structure/calculation)

*for the complexes **10** and **12** data for one on the molecules in the independent unit is shown.

	10*	12*
Ir – C1	2.001/2.0359	2.027 / 2.0496
Ir – C2	2.031/2.0390	2.042 / 2.0463
Ir – N1	2.043/2.0893	2.044 / 2.1042
Ir – N2	2.046/2.0874	2.065 / 2.0990
Ir – N3	2.113/2.2078	2.121 / 2.1980
Ir – N4	2.141/2.2095	2.102 / 2.1988

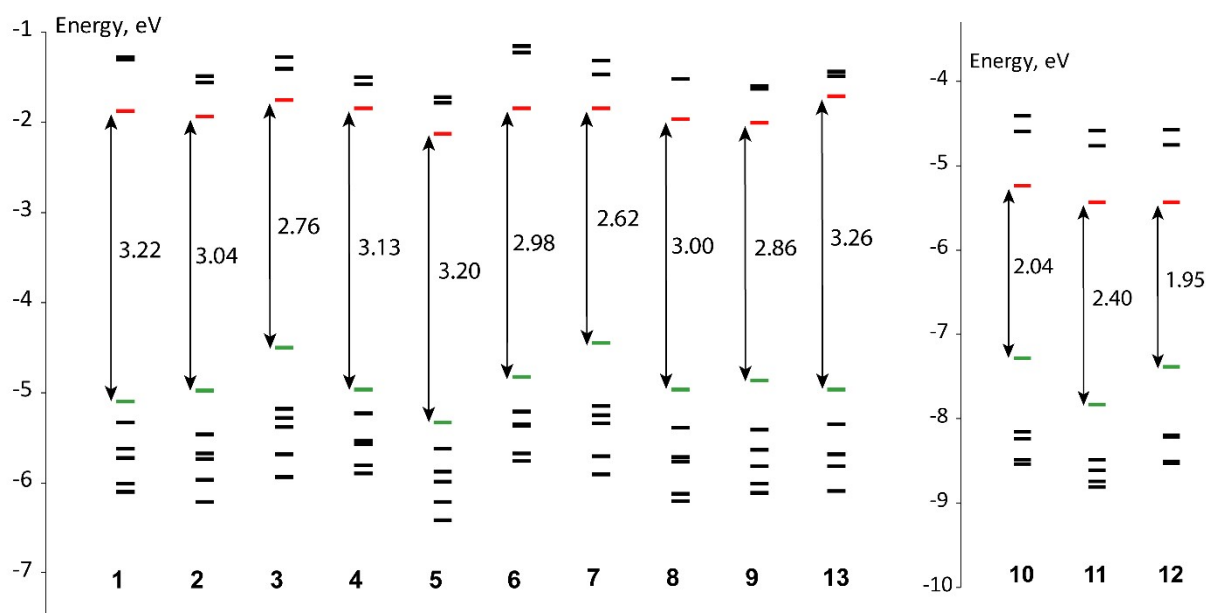


Figure S73. Energy of frontier molecular orbitals of complexes **1** – **9**, **13** (left) and **10** – **12** (right) and HOMO-LUMO gaps calculated by DFT (HOMO – highest occupied molecular orbital, LUMO – lowest unoccupied molecular orbital, DFT – Density Functional Theory).

Table S6. Composition of frontier molecular orbitals for **1 – 13**.

complex		1	2	3	4	5	6	7	8	9	10	11	12	13
HOMO-1	Ir	0,46	0,50	0,51	0,42	0,42	0,36	0,57	0,52	0,13	0,35	0,14	0,01	0,17
	C^N	0,22	0,19	0,23	0,27	0,14	0,51	0,32	0,23	0,84	0,51	0,86	0,79	0,04
	O^O	0,32	0,31	0,26	0,30	0,44	0,13	0,11	0,25	0,03	0,17	0,00	0,20	0,79
HOMO	Ir	0,46	0,42	0,39	0,45	0,46	0,40	0,40	0,42	0,11	0,33	0,31	0,17	0,44
	C^N	0,50	0,55	0,55	0,51	0,51	0,53	0,58	0,55	0,72	0,68	0,69	0,65	0,49
	O^O	0,04	0,03	0,06	0,04	0,03	0,07	0,02	0,03	0,17	0,03	0,00	0,18	0,07
LUMO	Ir	0,01	0,01	0,01	0,01	0,02	0,01	0,02	0,02	0,02	0,02	0,03	0,03	0,01
	C^N	0,01	0,11	0,02	0,02	0,04	0,46	0,15	0,01	0,90	0,63	0,09	0,63	0,06
	O^O	0,98	0,88	0,97	0,97	0,94	0,53	0,84	0,97	0,08	0,35	0,88	0,34	0,93
LUMO+1	Ir	0,02	0,01	0,00	0,02	0,03	0,00	0,00	0,00	0,54	0,02	0,01	0,01	0,02
	C^N	0,96	0,41	0,17	0,98	0,96	0,37	0,02	0,00	0,43	0,50	0,02	0,64	0,96
	O^O	0,02	0,58	0,82	0,00	0,00	0,63	0,98	1	0,03	0,48	0,97	0,36	0,03

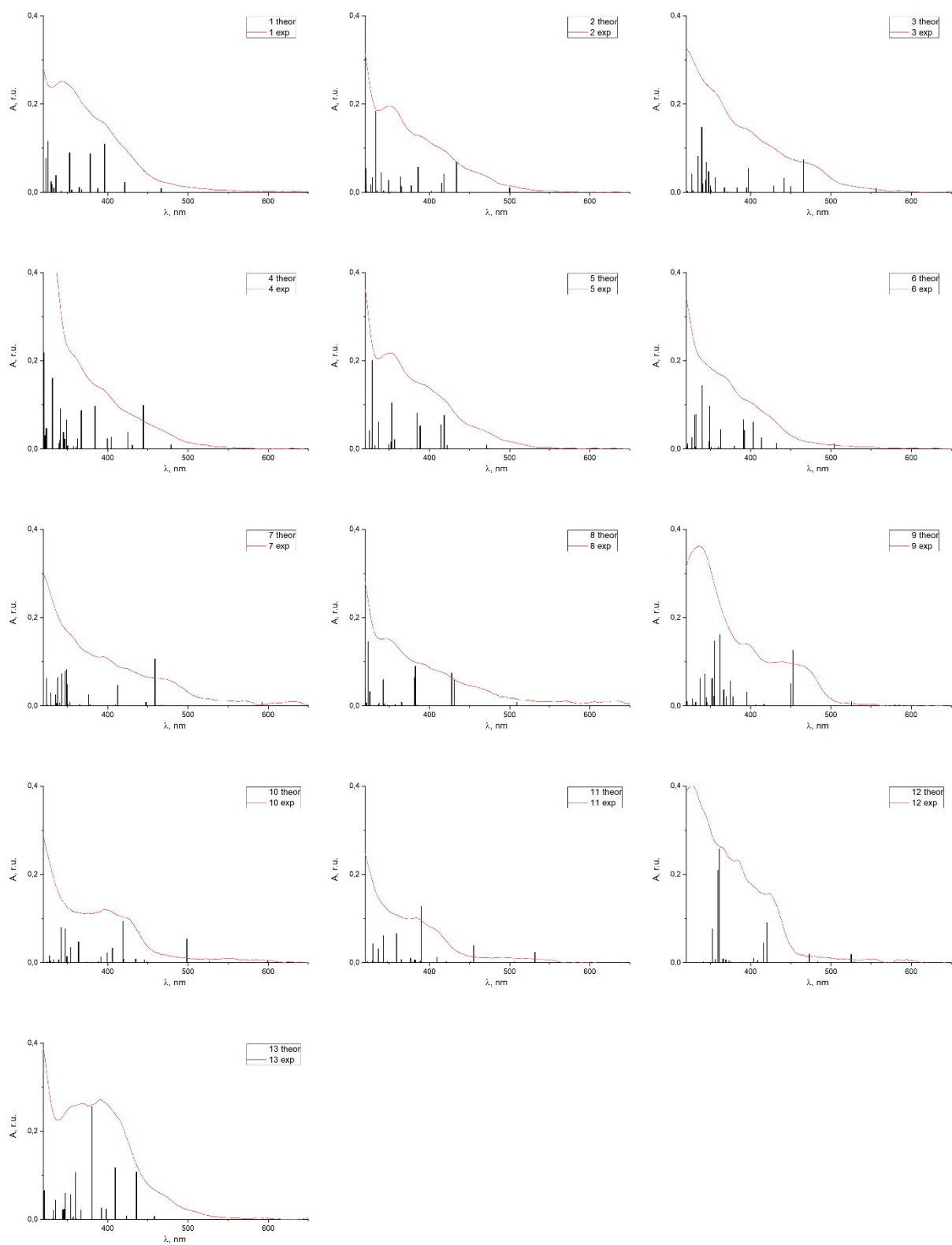


Figure S74. TDDFT (black) and experimental (red) electronic spectra of complexes **1** – **13**.

5. Photovoltaic measurements

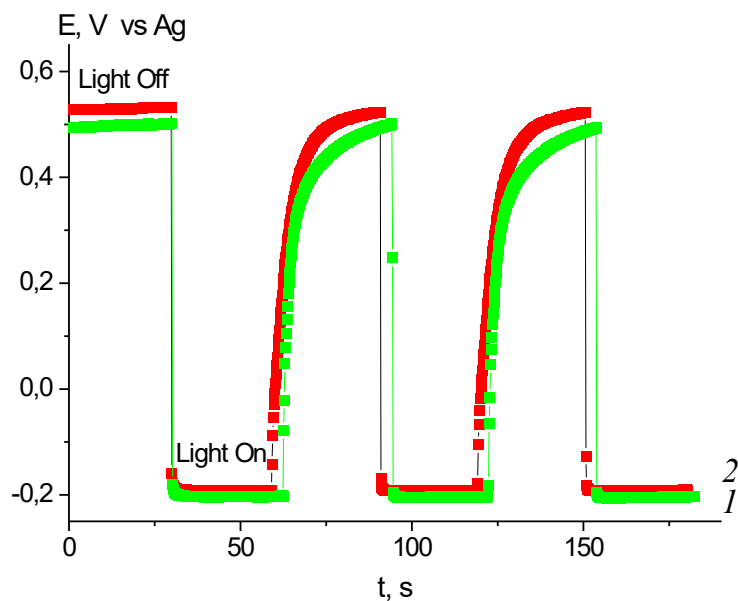


Figure S75. Time dependence of photopotential for TiO_2 photoanodes sensitized by complexes **7H** (1) and **9H** (2) in the dark and on exposure to AM 1.5 G simulated solar light (100 mW cm^{-2}).

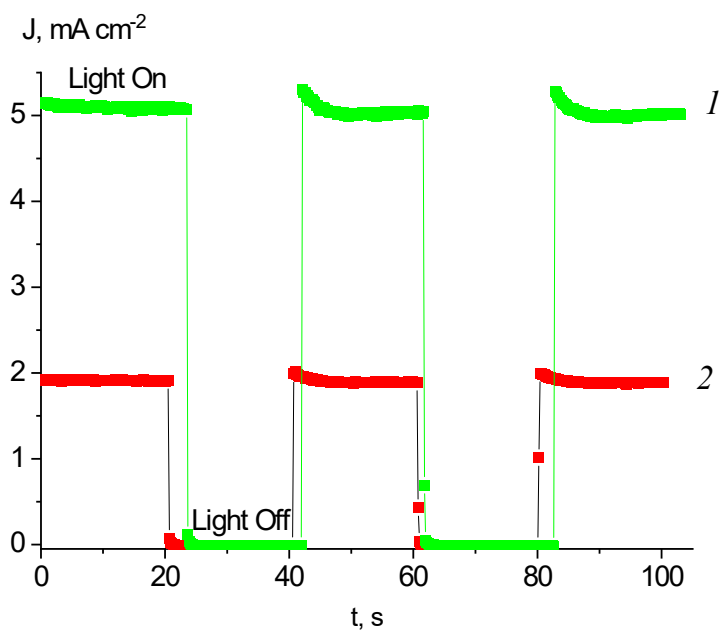


Figure S76. Time dependence of photocurrent at short-circuit for TiO_2 photoanodes sensitized by complexes **7H** (1) and **9H** (2) in the dark, and on exposure to AM 1.5 G simulated solar light (100 mW cm^{-2}).