## Supporting Information

# Role of Ancillary Ligands in Selectivity Towards Acceptorless Dehydrogenation versus Dehydrogenative Coupling of Alcohols and Amines Catalyzed by Cationic Ruthenium(II)-CNC Pincer Complexes

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### **Table of Contents**

S. No.	Contents	Page No.
1.	General Information	<b>S</b> 3
2.	Crystallographic parameters, selected bond lengths, and bond angles of complexes <b>4b</b> and <b>6b</b>	S4-S5
3.	<sup>1</sup> H NMR spectra of <b>CNC<sup>iPr</sup>·2HBr</b> and <b>CNC<sup>Cy</sup>·2HCl</b>	<b>S6</b>
4.	<sup>1</sup> H, <sup>31</sup> P, <sup>13</sup> C NMR, IR spectrum, & HRMS data of complexes <b>1b</b> , <b>1c</b> , and <b>2c</b>	S7-S15
5.	<sup>1</sup> H, <sup>31</sup> P, <sup>13</sup> C NMR and HRMS data of complexes <b>3b</b> , and <b>3c</b>	S16-S22
6.	<sup>1</sup> H, <sup>31</sup> P, <sup>13</sup> C NMR and HRMS data of complexes <b>4b</b> , and <b>4c</b>	S23-S28
7.	<sup>1</sup> H, <sup>31</sup> P, <sup>13</sup> C NMR and HRMS data of complexes <b>6b</b> , and <b>6c</b>	S29-S34
8.	HRMS data of complex <b>5b</b>	S35
9.	Plausible mechanism for transfer hydrogenation reaction	S36
10.	<sup>1</sup> H NMR of the catalytic reaction mixture	S37
11.	Mass data of catalytic reaction mixture	S38
12.	Mass data of catalytic reaction mixture of benzaldehyde with catalyst <b>4b</b>	<b>S</b> 39
13.	Plausible mechanism for the formation of other intermediates in catalytic reaction	S40
14.	Determination of % GC yield by gas chromatography	S41
15.	GC-MS spectra for entries <b>5-13</b> of table <b>1</b>	S42-S46
16.	GC-MS spectra for entries 6-14 of table 2	S46-S50
17.	GC-MS spectra for entries <b>1-15</b> of table <b>3</b>	S51-S61
18.	GC-MS spectra for entries <b>1-17</b> of table <b>4</b> and <b>1-5</b> of table <b>5</b>	S62-S74
19.	<sup>1</sup> H and <sup>13</sup> C NMR spectra of cyclohexanol and benzaldehyde	S75-S76
20.	DFT optimized structures of <b>[Ru-H]</b> <sup>L</sup>	S77
21.	Cartesian coordinates of DFT optimized structures	S78-S91
22.	References	S92

#### **General Information**

**Catalysis experiments:** Since TH and AAD reactions have been studied with the *N*-Me analogs,<sup>1–3</sup> no substrate scope was investigated again with the new compounds but only a confirmation of the observed trends in terms of the trans effect of ancillary ligands was checked. The catalyst loading may seem high; however, the focus of this manuscript is the effect of ancillary ligands on catalytic performance with a comparison of the four ligands (CO, COD, PPh<sub>3</sub>, DMSO) and the unexpected reversal in catalytic activity during the ADC reactions. The CNC pincer ligand platform provides a unique ligand framework with no metal-ligand cooperativity, which allows this comparison between a set of ancillary ligands. New complexes with two bulkier *N*-substituents have allowed us to observe the effect of ancillary ligands and confirmation of these effects in a larger set of compounds as well as characterization of an important Ru-H intermediate with CO ligand positioned trans to the product.

	4b	6b
Empirical formula	$C_{53}H_{55}Br_{0.17}Cl_{0.85}N_5O_{1.5}P_2Ru$	$C_{21}H_{39}Cl_2N_5O_5RuS_2$
т/к	293(2)	293(2)
Crystal System	Monoclinic	Monoclinic
Space Group	<i>P</i> 2 <sub>1</sub> /n	<i>P</i> 2 <sub>1</sub> /n
a/ Å	12.3216(2)	11.3804(2)
b/Å	26.5103(11)	11.0157(2)
c/Å	14.8679(3)	24.3528(4)
<b>α/</b> °	90	90
<b>β/</b> °	92.768(2)	96.679(2)
γ <b>/</b> °	90	90
V/ų	4850.9(2)	3032.22(9)
Z	4	4
$\rho_{calc}g/cm^3$	1.360	1.484
λ/Å (Cu-Kα)	1.54184	1.54184
<b>Reflections Collected</b>	21244	13368
Data/restr./param.	8823/0/578	5531/0/342
R (int)	0.0843	0.0897
Final R indices [I>2 $\sigma$ (I)]	R1 = 0.0719, wR2 = 0.1823	R1 = 0.0866, wR2 = 0.2388
R indices (all data)	R1 = 0.0918, wR2 = 0.2003	R1 = 0.0909, wR2 = 0.2488
GOF on F2	1.058	1.039

 Table S1. Crystal data and structure refinement parameters for 4b and 6b.

Complex	Bond lengths (Å)	Bond angles (°)
4b	Ru1-N1, 2.058(4)	P2-Ru1-P1, 165.04(5)
	Ru1-C1, 2.022(5)	N1-Ru1-P1, 100.75(13)
	Ru1-C11, 2.049(6)	N1-Ru1-P2, 94.21(13)
	Ru1-P1, 2.3462(14)	C1-Ru1-P1, 91.58(16)
	Ru1-P2, 2.3303(15)	C1-Ru1-P2, 91.81(16)
		C1-Ru1-N1, 77.5(2)
		C1-Ru1-C11, 154.5(2)
		C11-Ru1-P1, 94.09(16)
		C11-Ru1-P2, 89.10(16)
		C11-Ru1-N1, 77.0(2)
6b	Ru1-N1, 1.995(5)	N1-Ru1-C1, 78.0(2)
	Ru1-C1, 2.062(6)	N1-Ru1-C11, 77.7(2)
	Ru1-C11, 2.061(6)	N1-Ru1-S1, 89.69(14)
	Ru1-S1, 2.2954(14)	N1-Ru1-S2, 89.35(14)
	Ru1-S2, 2.3138(14)	N1-Ru1-Cl1, 177.23(14)
	Ru1-Cl1, 2.4296(13)	C1-Ru1-S1, 88.29(16)
		C1-Ru1-S2, 87.69(16)
		C11-Ru1-Cl1, 101.60(15)
		C11-Ru1-C1, 155.7(2)
		C11-Ru1-S1, 91.74(15)
		C11-Ru1-S2, 91.87(15)
		C1-Ru1-Cl1, 102.67(16)
		S1-Ru1-Cl1, 87.65(5)
		S1-Ru1-S2, 175.98(6)
		S2-Ru1-Cl1, 93.36(5)

 Table S2.
 Selected bond lengths and bond angles of complex 4b and 6b.













100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)

Figure S4. <sup>31</sup>P NMR spectrum of Complex 1b.







Figure S6. IR spectra of Complex 1b.







100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)

Figure S9. <sup>31</sup>P NMR spectrum of Complex 1c.



Figure S11. IR spectrum of Complex 1c.



Figure S12. HRMS spectrum of complex 1c.



100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)

Figure S14. <sup>31</sup>P NMR spectrum of Complex 2c.



Figure S16. IR spectra of Complex 2c.









100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)

Figure S19. <sup>31</sup>P NMR spectrum of Complex 3b.



Figure S20. <sup>13</sup>C NMR spectrum of Complex 3b.













100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)





Figure S24. <sup>13</sup>C NMR spectrum of Complex 3c.



Figure S25. HRMS spectrogram of Complex 3c.



Figure S26. HRMS spectrogram of dicationic complex [3c-PF<sub>6</sub>-Cl]<sup>2+</sup>.



100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)

Figure S28. <sup>31</sup>P NMR spectrum of Complex 4b.



Figure S29. <sup>13</sup>C NMR spectrum of Complex 4b.



Figure S30. HRMS spectrogram of Complex 4b.



Figure S32. <sup>31</sup>P NMR spectrum of Complex 4c.



Figure S33. <sup>13</sup>C NMR spectrum of Complex 4c.



Figure S34. HRMS spectrogram of Complex 4c.





100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)

Figure S36. <sup>31</sup>P NMR spectrum of Complex 6b.



Figure S37. <sup>13</sup>C NMR spectrum of Complex 6b.



Figure S38. HRMS spectrum of Complex 6b.





100 90 80 70 60 50 40 30 20 10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -200 f1 (ppm)





Figure S41. <sup>13</sup>C NMR spectrum of Complex 6c.







Figure S43. HRMS spectrogram of Complex 5b.

#### Mechanism for transfer hydrogenation reaction

A plausible mechanism for transfer hydrogenation is shown, with complex **1c** as the catalyst precursor and **2c** as the Ru-hydride intermediate. The detailed mechanistic investigation for TH reaction with the *N*-methyl complex was previously reported by our group.<sup>1</sup> The ruthenium alkoxide species **A** is produced, when complex **1c** is treated with NaO<sup>*i*</sup>Pr. The Ru-H intermediate **2c'** is formed from **A** via  $\beta$ -H elimination by releasing one molecule of acetone, or by dissociation of a PPh<sub>3</sub> ligand if starting from **2c**. The addition of cyclohexanone to the intermediate **2c'** produces another ruthenium alkoxide intermediate **B**, which releases the hydrogenated product upon protonation from <sup>*i*</sup>PrOH resulting in the formation of **A** again.



**Figure S44.** Plausible mechanism for transfer hydrogenation reaction by complex **1c** with key intermediates **2c**.


**Figure S45.** <sup>1</sup>H NMR experiment in DMSO-d<sub>6</sub> to observe the generation of ruthenium hydride intermediate from complex **3b** under catalytic reaction conditions.



Figure S46. LCMS spectrogram of catalytic reaction mixture (L = PPh<sub>3</sub>).

## **Display Report**

#### Analysis Info Acquisition Date 22-06-2022 16:24:20 Analysis Name C:\Users\Rahul Kumar Singh\Desktop\NMR and Mass\Mass data\m chem aks-rks-c-110-3h\_RA3\_01\_18656.d Method 8. LCMS tune wide MeOH.m Operator IIT Indore m chem aks-rks-c-110-3h Sample Name Instrument micrOTOF-Q 228888.10348 Comment Acquisition Parameter lon Polarity Set Capillary Set End Plate Offset Positive 4500 V -500 V 2.0 Bar 250 °C 7.0 l/min Source Type ESI Set Nebulizer Not active Set Dry Heater Set Dry Gas Focus Scan Begin 50 m/z Scan End 3000 m/z Set Collision Cell RF 650.0 Vpp Set Divert Valve Waste Intens. x10<sup>5</sup> 6 4 2 0 0.5 1.0 1.5 2.0 2.5 Time [min] BPC +All MS Intens. m chem aks-rks-c-110-3h\_RA3\_01\_18656.d: +MS, 0.1min #3 x10<sup>6</sup> 7+ 1+ 0.8 922.2317 L 0.6 Έ E 4b 0.4 1+ 780.1677 0.2 579.1338 0.0 400 600 800 1000 1200 1400 1600 m/z Intens. m chem aks-rks-c-110-3h\_RA3\_01\_18656.d: +MS, 0.1min #3 ×10<sup>5</sup> 1+ 780.1677 2 1+ 1+ 777.1662 778.1686 779.1688 1+ 782.1698 781.1705 1+ 1 783.1707 1+ 1+ 774,1688 776.1684 784.1725 785.1903 0 C42H41N5O2RuP, M, 780.2036 1 +780.2047 2000 1+ 1+ 1+ 779.2057 1+ 782.2056 1+ 778.2051 781.2073 1+ 1000 777.2054 1+ 1+ 1+ 774.2068 775.2100 783.2082 1+ 784.2110 A А 0 774 778 780 782 784 776 786 m/z

Figure S47. LCMS spectrogram of catalytic reaction mixture with benzaldehyde and catalyst **4b** formed intermediate  $I (L = PPh_3)$ .

Formation of other intermediates **I**, **II**, and **III** appeared during the catalytic reactions (Figure S46). Intermediate **I** also appeared in LCMS during the reaction of benzaldehyde with Ru hydride catalyst **4b** (Figure S47). This information suggests that the *ortho* C-H activation of benzaldehyde takes place and generated intermediate **D**, which further reacted with moisture (while recording mass data) and appeared as intermediate **I** (Figure S47).



**Figure S48.** Plausible mechanism for the formation of intermediates I and II in catalytic reaction mixture by complex **3b**.



**Figure S49.** Plausible mechanism for the formation of intermediate **III** in catalytic reaction mixture by complex **3b**.

#### Determination of % GC yield by gas chromatography

GC Samples were analysed in Shimadzu QP2010 Ultra gas chromatograph. Yields of the product were determined using *n*-decane as an internal standard. Samples were prepared by filtering the reaction mixture through a celite pad with chloroform and further dilution with methanol solution. The additional peaks in some GC-MS traces are of PPh<sub>3</sub> and sometimes OPPh<sub>3</sub> due to aerial oxidation in the GC sample. The poor signal separation, only in case of transfer hydrogenation of cyclohexanone, is due to very close retention time of cyclohexanone and cyclohexanol. For uniformity, we have followed the automated integration by the GC-MS software of our instrument. The reactants and products relative response factors (RF) were calculated using *n*-decane as the internal standard. *n*-Decane was added to the reaction mixture prior to start of catalysis. The following equations are used to calculate the % GC yields.<sup>4</sup>

Response factors were calculated using the following equation:

$$RF = \frac{Area \text{ percentage of internal standard} \times Moles \text{ of analyte}}{Area \text{ percentage of analyte } \times Moles \text{ of internal standard}}$$

Moles of remaining reactants and products were calculated using the following equation:

$$Moles of analyte = \frac{RF \times Moles of internal standard \times Area percentage of analyte}{Area percentage of internal standard}$$

The products of catalysis experiments (TH and AAD) are characterized by <sup>1</sup>H and <sup>13</sup>C NMR as well as GC-MS. ADC catalysis experiments are analyzed with GC-MS only as the imine products are prone to hydrolysis during column chromatography.



#### GC-MS spectra of Transfer hydrogenation of cyclohexanone products for table 1.

Figure S51. GC-MS spectrum for entry 6 of table 1.

m/z







Figure S53. GC-MS spectrum for entry 8 of table 1.



Figure S54. GC-MS spectrum for entry 9 of table 1.



Figure S55. GC-MS spectrum for entry 10 of table 1.





Figure S57. GC-MS spectrum for entry 12 of table 1.



Figure S58. GC-MS spectrum for entry 13 of table 1.





Figure S59. GC-MS spectrum for entry 6 of table 2.



Figure S61. GC-MS spectrum for entry 8 of table 2.



Figure S63. GC-MS spectrum for entry 10 of table 2.





Figure S65. GC-MS spectrum for entry 12 of table 2.



Figure S67. GC-MS spectrum for entry 14 of table 2.

# GC-MS spectra of Acceptorless dehydrogenative coupling of aniline and benzyl alcohol products for table 3.



Figure S68. GC-MS spectrum for entry 1 of table 3.



Figure S69. GC-MS spectrum for entry 2 of table 3.







Figure S70. GC-MS spectrum for entry 3 of table 3.



Figure S71. GC-MS spectrum for entry 4 of table 3.













Figure S73. GC-MS spectrum for entry 6 of table 3.





Figure S75. GC-MS spectrum for entry 8 of table 3.





Figure S76. GC-MS spectrum for entry 9 of table 3.









Figure S78. GC-MS spectrum for entry 11 of table 3.



Figure S79. GC-MS spectrum for entry 12 of table 3.





Figure S80. GC-MS spectrum for entry 13 of table 3.









Figure S82. GC-MS spectrum for entry 15 of table 3.



### GC-MS spectra of Acceptorless dehydrogenative coupling products for table 4.

Figure S83. GC-MS spectrum for entry 1 of table 4.





Figure S84. GC-MS spectrum for entry 2 of table 4.















Figure S87. GC-MS spectrum for entry 5 of table 4.





Figure S88. GC-MS spectrum for entry 6 of table 4.



Figure S89. GC-MS spectrum for entry 7 of table 4.







Figure S91. GC-MS spectrum for entry 9 of table 4.







Figure S93. GC-MS spectrum for entry 11 of table 4.







Figure S95. GC-MS spectrum for entry 13 of table 4.







Figure S97. GC-MS spectrum for entry 15 of table 4.







Figure S99. GC-MS spectrum for entry 17 of table 4.



#### GC-MS spectra of Acceptorless dehydrogenative coupling products for table 5.

Figure S100. GC-MS spectrum for entry 1 of table 5.








Figure S102. GC-MS spectrum for entry 3 of table 5.







Figure S103. GC-MS spectrum for entry 4 of table 5.

Figure S104. GC-MS spectrum for entry 5 of table 5.

NMR spectra of product after transfer hydrogenation reaction.



**Figure S105.** <sup>1</sup>H NMR spectrum of cyclohexanol.



Figure S106. <sup>13</sup>C NMR spectrum of cyclohexanol.

## NMR spectra of product after acceptorless alcohol dehydrogenation.



Figure S108. <sup>13</sup>C NMR spectrum of benzaldehyde.



**Figure S109.** DFT optimized structures of **[Ru-H]**<sup>L</sup> after aldehyde dissociation from **B**<sup>L</sup> (**L** = CO, DMSO and PPh<sub>3</sub>). For **L** = PPh<sub>3</sub>, two structures with hydride position w.r.t. PPh<sub>3</sub> ligand are calculated. The **[Ru-transH]**<sup>PPh<sub>3</sub></sup> (model for **4b**'', confirmed in <sup>1</sup>H NMR) is found -1.7 kcal/mol lower than **[Ru-cisH]**<sup>PPh<sub>3</sub></sup> (model for **4b**') possibly due to an agostic interaction between Ru and a phenyl ring of PPh<sub>3</sub> ligand.

# Cartesian coordinates of DFT optimized structures

## B-CO

Ru	1 -1.15372206112549	0.97723485532526	0.00700472651732
Н	-1.78428764605351	0.32165240741799	1.35010717571905
Ν	-0.21217954645072	1.84843454058005	-1.64123392543552
Ν	-0.03021446760500	3.31188843659006	1.96175694434061
Ν	0.56362526489258	3.32163371694917	-0.10756312207798
Ν	-2.49509898029893	-1.28604603735579	-1.86986927086123
0	-3.89121220348032	2.16189375233839	0.15371659196936
Ν	-1.12334350955022	0.11165802602076	-2.76893005709475
С	-0.18578290352311	2.60583946222437	0.82792899217961
С	-1.70092811271409	-0.25016081616334	-1.55009739532875
С	-0.27692718489713	1.22661727259435	-2.82029458032239
С	0.59673694388663	2.89002929456211	-1.43978248158780
С	-2.82440389722069	1.72886008559574	0.07168840573374
С	0.78579098379245	4.42482186452312	1.77022743003579
Н	1.02515830339403	5.10933295050600	2.56983741952701
С	0.44214143577264	1.68437504032524	-3.91614508176723
Н	0.39304157759952	1.19502207487260	-4.88133534059150
С	1.35626614650844	3.42144170078990	-2.47351955156384
Н	2.01441970403031	4.26902009200090	-2.32599841204036
С	-1.56056477190500	-0.71861476612200	-3.79502086463825
Н	-1.24481931367083	-0.61668859486426	-4.82095273670268
С	-2.41840896463264	-1.59437518967307	-3.22484889646511
Н	-2.98220928152978	-2.40291810344679	-3.66494426525815
С	1.16105760502979	4.43446790636434	0.47246429318946
Н	1.78362743499761	5.12827711158688	-0.06930449307091
С	1.25192799393140	2.80134270151571	-3.71741707551693
Н	1.83113404344705	3.18791531060387	-4.54964464506400
С	-0.64749217571081	2.95478819624116	3.23605344685193

Н	-1.30538710195471	3.76205995131466	3.57079404872831
н	0.12521631156922	2.77567982559966	3.98953570816342
н	-1.23206667100007	2.04700487870003	3.07924089470719
С	-3.29664276257894	-2.02182673489568	-0.89633223871741
н	-2.95065652671720	-3.05845982386968	-0.83866647021412
Н	-4.35106763889112	-1.99441549231753	-1.18591101971391
Н	-3.16460449099827	-1.53798435162578	0.07283832695381
0	0.82568733297474	-0.04950251184711	0.07540292483926
С	0.27495392674291	-0.81877002496880	0.90758112552435
Н	0.26490317552739	-0.54717412671205	1.97728065856173
С	-0.10798930056635	-2.19919578301298	0.58966975555584
С	-0.71987683974946	-2.98577266008129	1.57140513485848
С	0.18546385114629	-2.74933681893820	-0.66427503489637
С	-1.05314072131629	-4.30563988443533	1.29812261805362
Н	-0.93621004155599	-2.55520997010817	2.54662279103895
С	-0.14300368701721	-4.06936131447295	-0.93200241320763
Н	0.68593904220513	-2.13240804402438	-1.40541450082428
С	-0.76568326506539	-4.84741238640444	0.04663713229069
Н	-1.52723273127814	-4.91697066963036	2.05971599820646
Н	0.09846192131609	-4.50475630713559	-1.89727417896498
Н	-1.01419619970673	-5.88369104303687	-0.16345449161990

#### B-PPh₃

Rι	1.16084743422634	0.92809232195030	-0.03677624538470
Н	-1.98847592726597	0.14287679153210	1.10970224634811
Ν	-0.23512045079430	2.11443673310386	-1.46456402375284
Ν	-0.18352304508486	2.85750130993059	2.35101593401675
Ν	0.40135440851084	3.33149383998587	0.33049447760864
Ν	-2.17414266515597	-1.14790763499487	-2.34153335412350
Ν	-0.94993589946208	0.53121461221387	-2.90787361074217
С	-0.33024385134561	2.40484708755813	1.08810711107207

С	-1.53873515490173	-0.08846042826215	-1.79272981769814
С	-0.23502822567491	1.71900150539039	-2.73830904382887
С	0.47213168115913	3.17127678717402	-1.05727361477771
С	0.58308249371551	4.01782087548625	2.40223040363039
Н	0.80358303507384	4.52383876165026	3.32998339824991
С	0.43235571478147	2.44500342177671	-3.71724029502764
Н	0.43055190614618	2.14361479120267	-4.75752908042255
С	1.17256187908087	3.96036691079615	-1.96005840224792
Н	1.74405340131964	4.82485713551346	-1.64485743998253
С	-1.23967558426897	-0.14986675846658	-4.08191900378698
Н	-0.88201845430501	0.15408580610390	-5.05249136281090
С	-2.00621193288469	-1.20291339020269	-3.72159520149477
Н	-2.43641107496365	-1.98933093119175	-4.32325843576940
С	0.95408573183888	4.31887880959168	1.13751245559507
Н	1.56173242140530	5.12691172960999	0.76254728821430
С	1.12524267126285	3.57947619994629	-3.30043885026306
Н	1.65876510453583	4.17213120190827	-4.03635011944869
С	-0.84139476487069	2.25425784958783	3.50370098674235
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Н	0.05417313383386	-4.35144716365108	-2.02969339562076

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