

Electronic Supplementary Information (ESI)

A dinuclear Cd^{II} cluster-based stable luminescent metal-organic framework for the consecutive and visual detection of H₂PO₄⁻ and OCN⁻

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Materials and instrumentations

All chemical reagents were obtained from commercial sources and used without further purification. Notably, H₂BTDB was purchased by Jilin Chinese Academy of Sciences -Yanshen Technology Co., Ltd. Thermogravimetric analysis (TGA) was performed under a N₂ flow at a heating rate of 10 °C min⁻¹ from 25 to 1000 °C on a NETZSCH STA2500 thermal analyzer. The powder X-ray diffraction (PXRD) patterns were recorded by Rigaku MiniFlex 600. The simulated PXRD pattern of single-crystal data was obtained using the Mercury (Hg) software, which is freely available on the Internet at <http://www.iucr.org>. IR spectra in the range of 4000-400 cm⁻¹ were collected with KBr particles on a Bruker Alpha FT-IR spectrometer. X-ray photoelectron spectrum (XPS) was obtained with an Axis Ultra DLD spectrometer. The fluorescence measurements were carried out on an F4600 (Hitachi) fluorescence spectrophotometer. The UV-vis absorption spectra for all samples were collected on a UV-2550 (SHIMADZU) spectrophotometer.

Crystal Description

X-ray single-crystal diffraction data of **JXUST-47** were recorded on a Bruker D8 QUEST with ω -scan pattern and Mo-K α radiation ($\lambda = 0.71073 \text{ \AA}$). The structure is solved by using direct methods with SHELXTL software and refined using full-matrix least-squares.^{S1} The anisotropic thermal parameters on non-hydrogen atoms are refined by F^2 . This structure is based on weak and highly inconsistent data. Fig. S1 shows that there exist electrons or unresolved atoms in **JXUST-47**. As shown in Fig. S2, **JXUST-47** there exist numerous outliers. The R merge rises above 15% at about 1.4 (Fig. S3). As shown in Fig. S4, the I/σ drops below 3 at $2\theta = 41$. The ORTEP drawing of **JXUST-47** is shown in Fig. S5. During the collection of crystal data, scan width is 0.50 and exposure time is 10 s. The exposure time of 10 s may not be optimal, and the reason for the weak data is merely the limited crystal quality and the short exposure time.

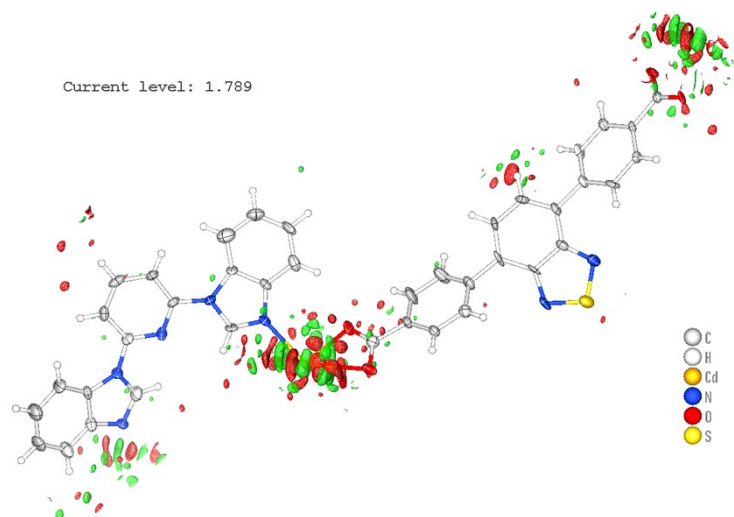


Fig. S1 The residual density map of JXUST-47.

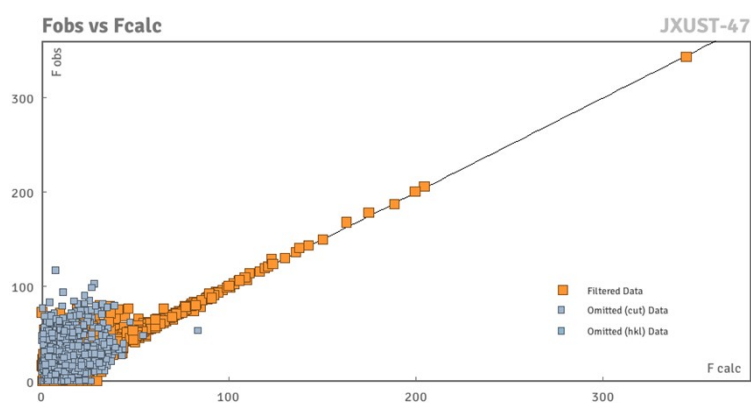


Fig. S2 The F_{obs} vs F_{calc} plot of JXUST-47.

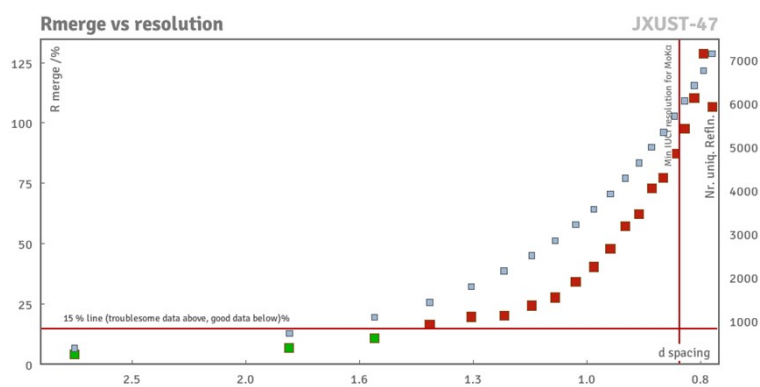


Fig. S3 The plot of R merge vs resolution of JXUST-47.

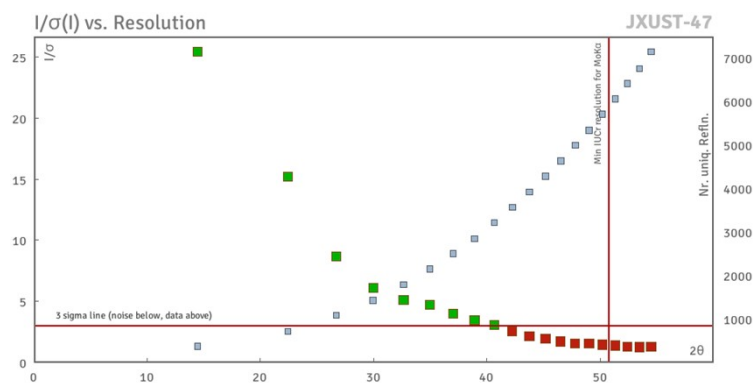


Fig. S4 The plot of $I/\sigma(I)$ vs resolution of JXUST-47.

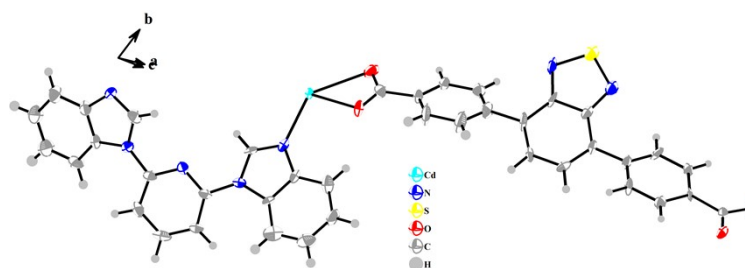


Fig. S5 The ORTEP drawing of JXUST-47.

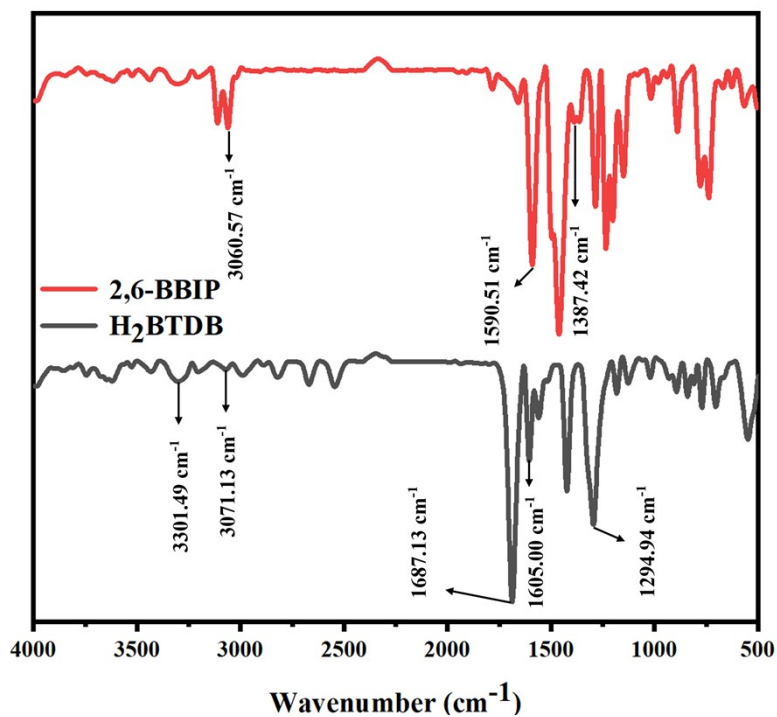


Fig. S6 The IR spectra of 2,6-BBIP and H₂BTDB.

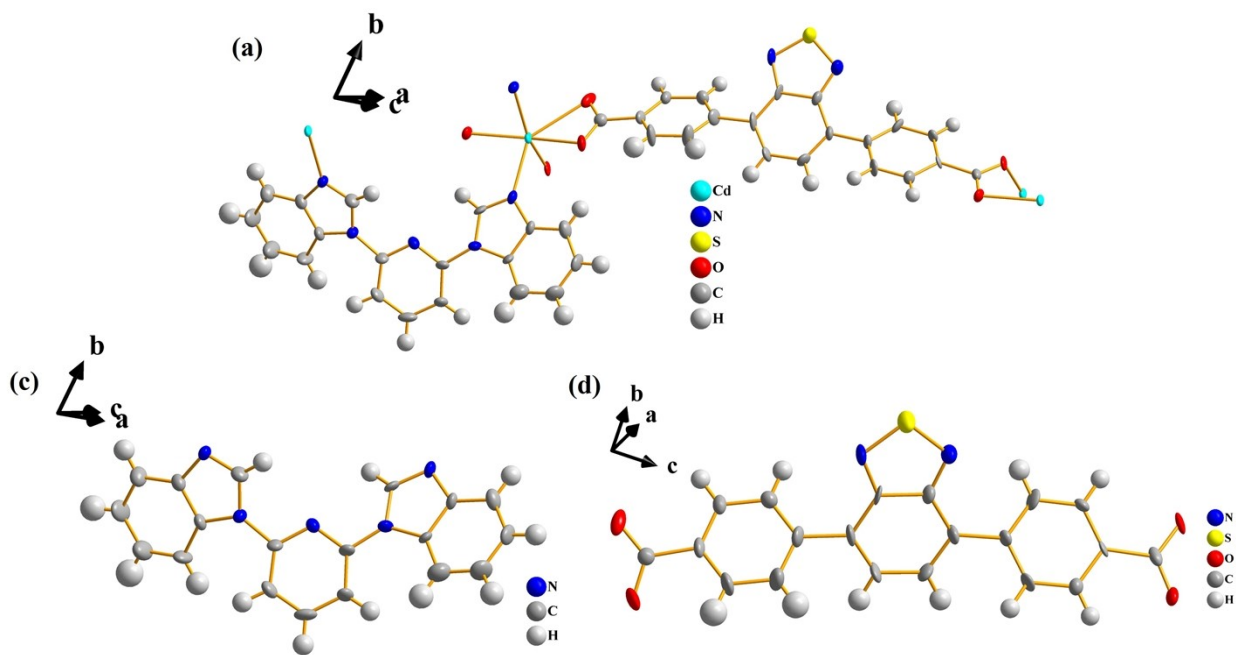


Fig. S7 The ellipsoid diagrams of (a) JXUST-47, (b) 2,6-BBIP and (c) H₂BTDB.

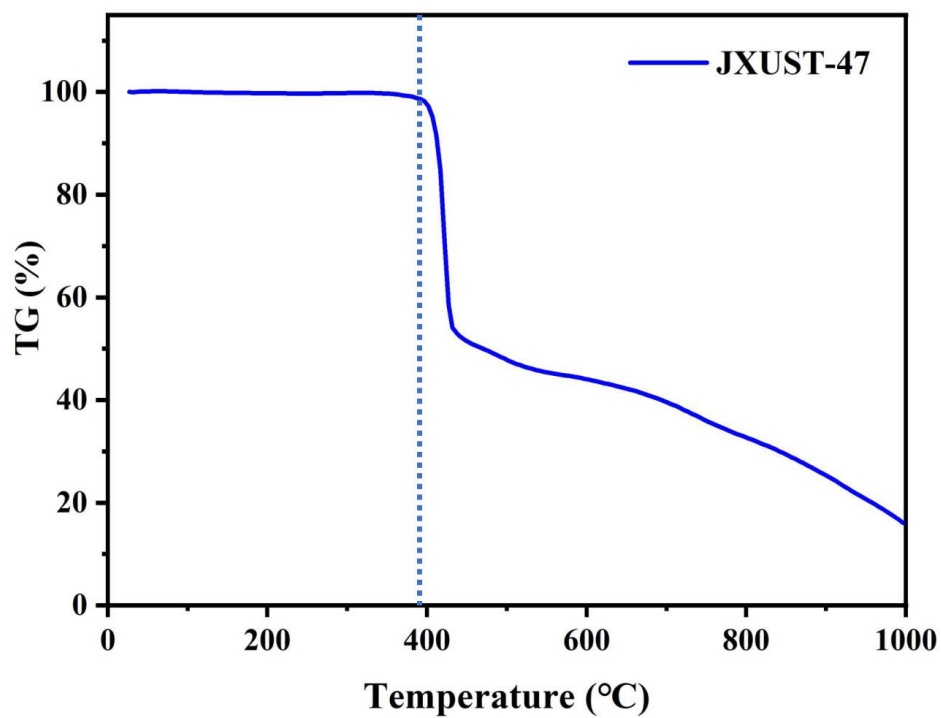


Fig. S8 The TGA curve of JXUST-47.

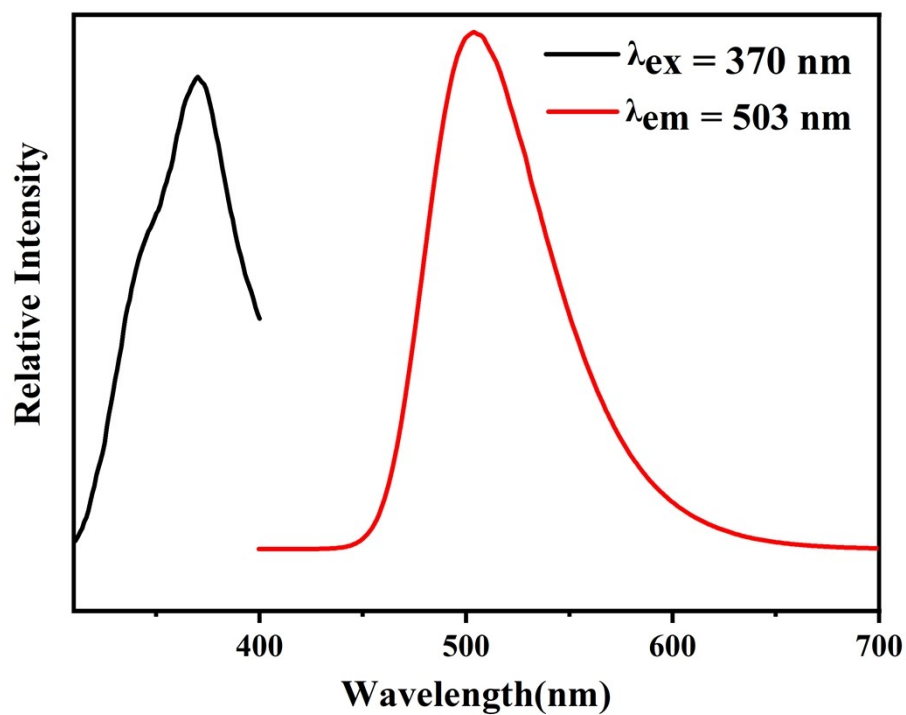


Fig. S9 The solid-state excitation and emission spectra of JXUST-47.

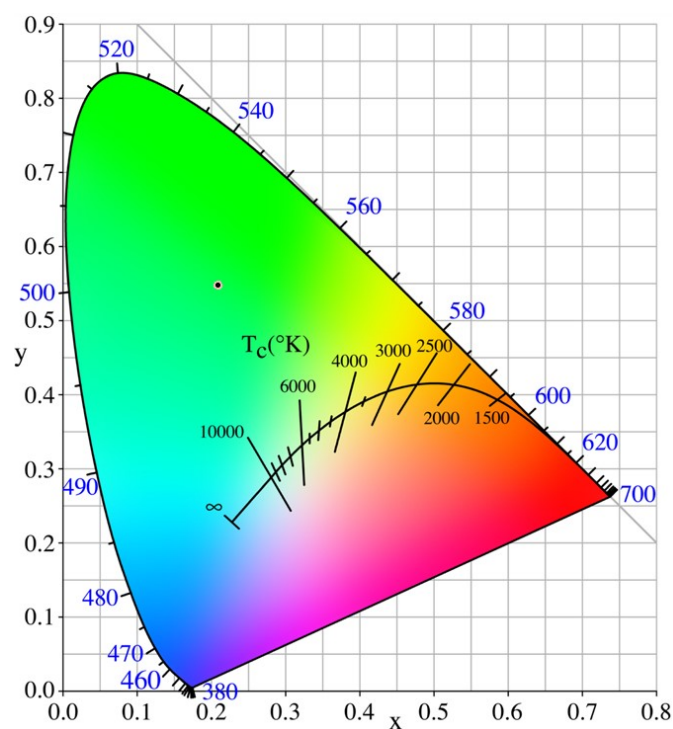


Fig. S10 CIE chromaticity diagram displaying the color coordinate of JXUST-47.

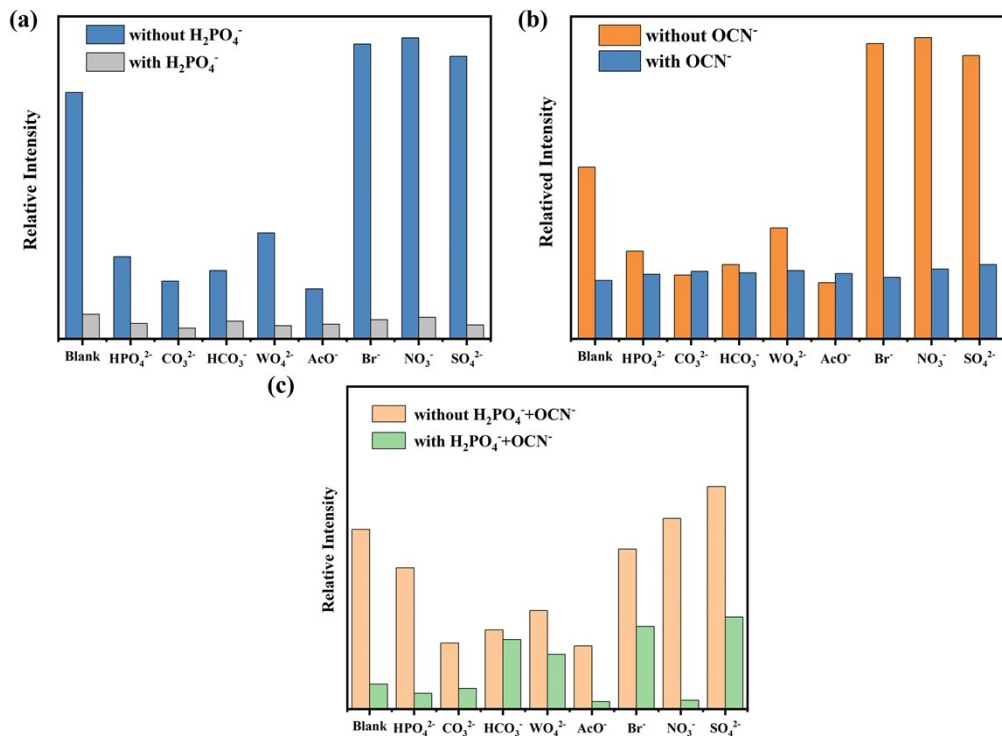


Fig. S11 Competitive experiments of **JXUST-47** in sensing (a) H_2PO_4^- , (b) OCN^- , and (c) H_2PO_4^- and OCN^- with the other interfering substance.

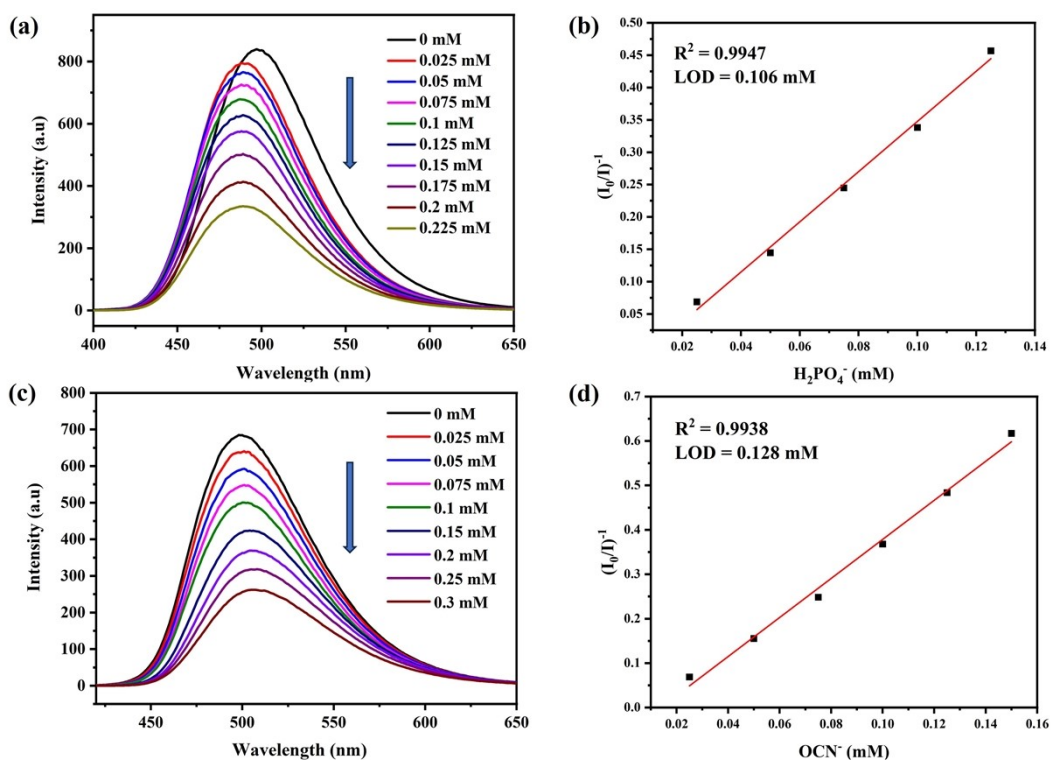


Fig. S12 Emission spectra of **JXUST-47** dispersed in DMF suspension with different concentrations for (a) H_2PO_4^- and (c) OCN^- ions, and the linear relationship in a low concentration range between the fluorescence intensity of **JXUST-47** and the concentration of (b) H_2PO_4^- and (d) OCN^- ions.

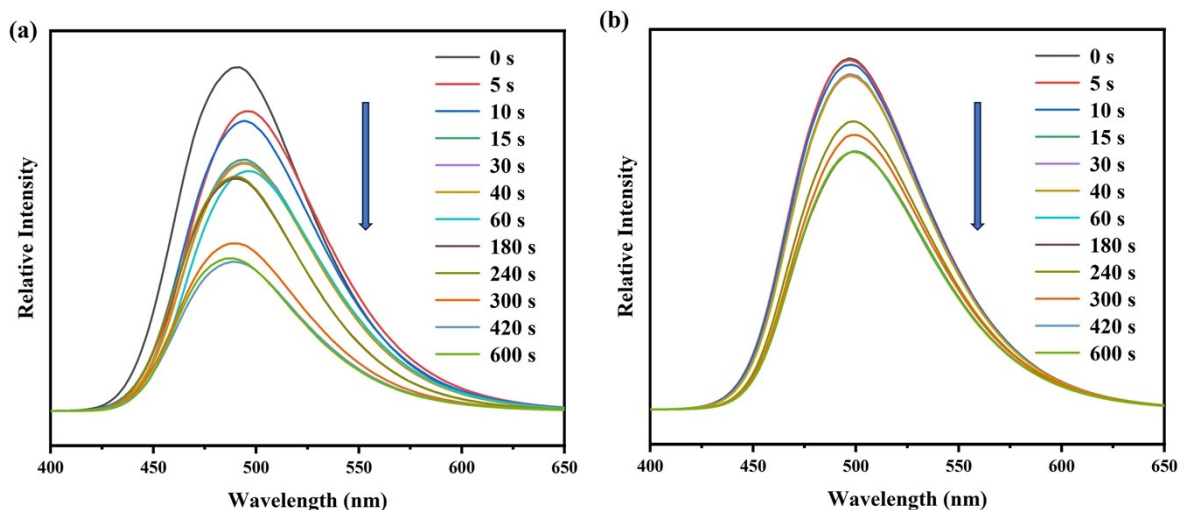


Fig. S13 Time-dependent emission spectra of the suspension after adding 0.2 M (a) H_2PO_4^- and (b) OCN^- .

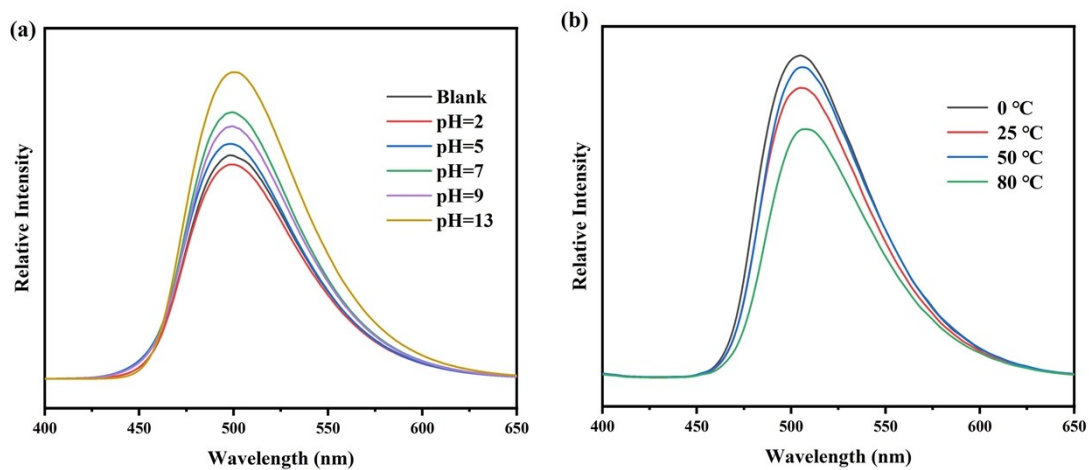


Fig. S14 The emission spectra of **JXUST-47** (a) in DMF aqueous solution with different pH values and (b) at different temperatures.

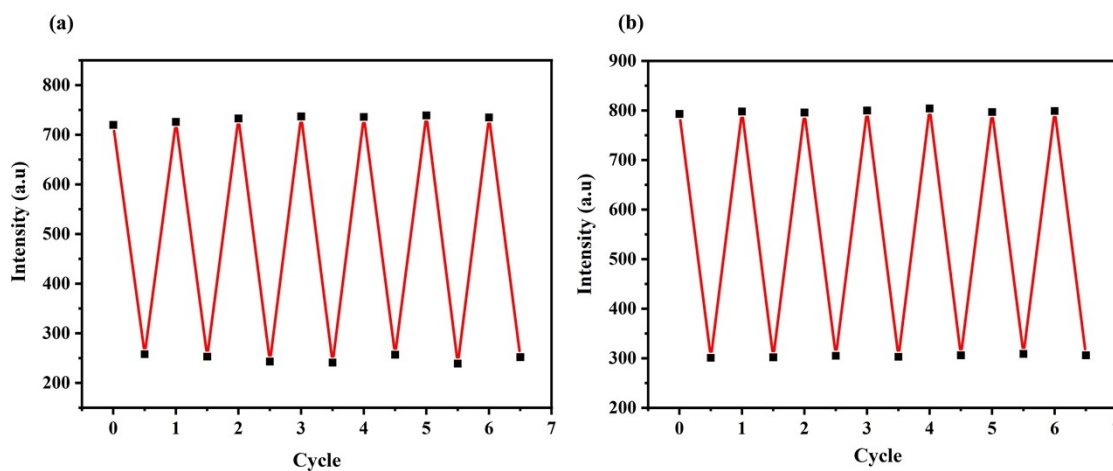


Fig. S15 Relative luminescent intensity of **JXUST-47** after seven cycles for (a) H_2PO_4^- and (b) OCN^- .

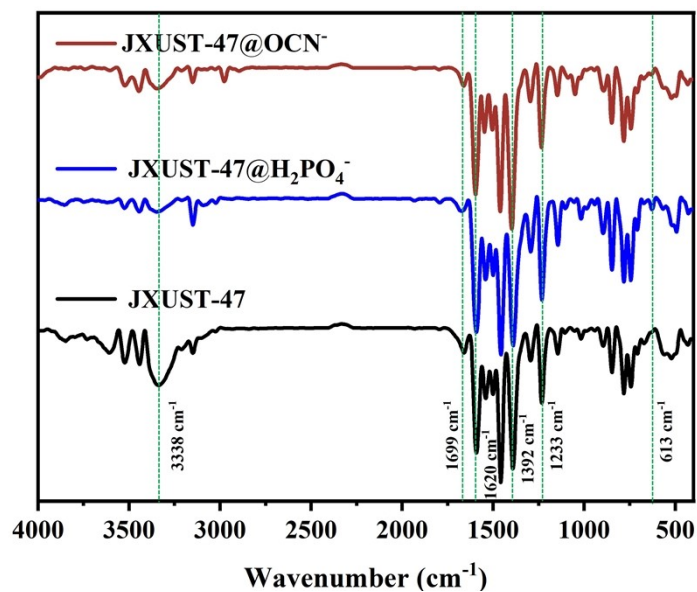


Fig. S16 The IR spectra of as-synthesized JXUST-47 and JXUST-47 after sensing H₂PO₄⁻/OCN⁻.

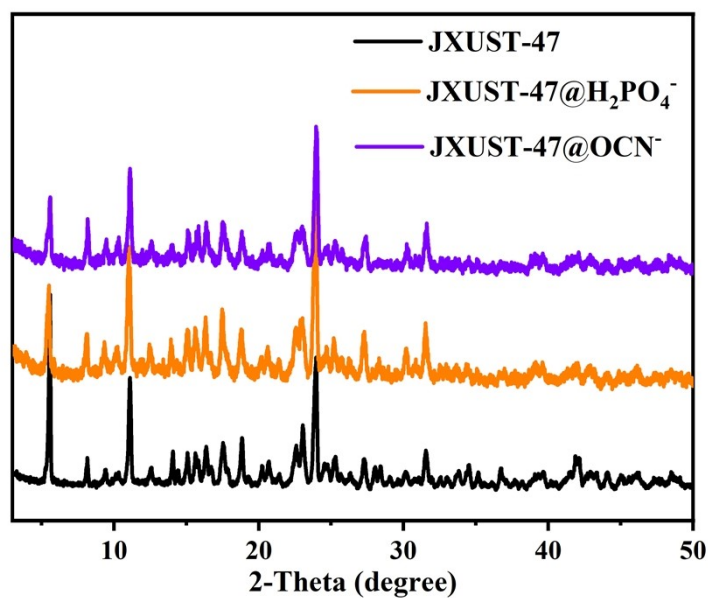


Fig. S17 The PXRD patterns of as-synthesized JXUST-47 and JXUST-47 after sensing H₂PO₄⁻/OCN⁻.

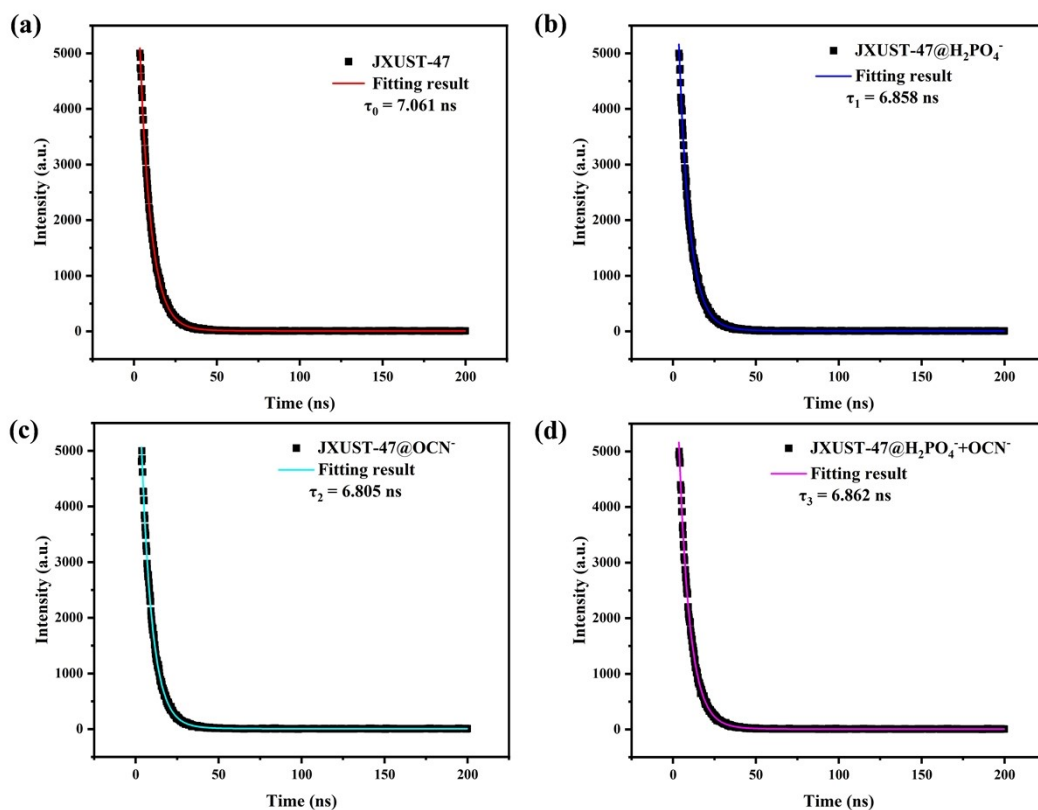


Fig. S18 The luminescence decay curves of (a) JXUST-47, (b) JXUST-47@ H₂PO₄⁻, (c) JXUST-47@ OCN⁻ and (d) JXUST-47@ H₂PO₄⁻+OCN⁻ at room temperature.

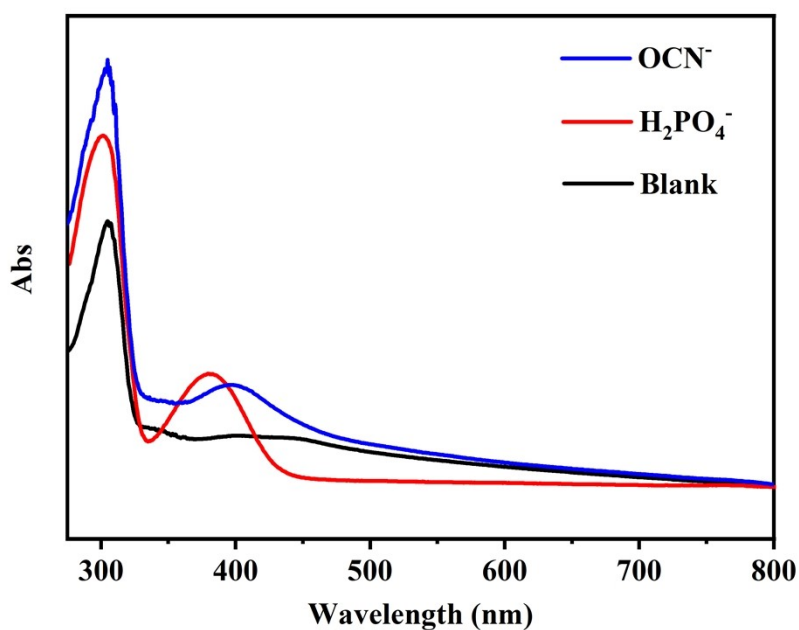


Fig. S19 The UV-vis absorption spectra of JXUST-47 dispersed in DMF solution and JXUST-47 dispersed in DMF solution after adding H₂PO₄⁻ and OCN⁻.

Table S1. Crystal data and structure refinements for **JXUST-47**.

Compound	JXUST-47
Formula	C ₃₉ H ₂₃ N ₇ O ₄ SCd
<i>M</i> _r	798.10
Temperature (K)	293(2)
Crystal system	triclinic
Space group	<i>P</i> $\bar{1}$
<i>a</i> (Å)	9.4237(11)
<i>b</i> (Å)	11.6946(13)
<i>c</i> (Å)	16.1030(17)
α (°)	82.624(5)
β (°)	83.013(5)
γ (°)	68.067(4)
<i>V</i> (Å ³)	1627.3(3)
<i>Z</i>	2
<i>F</i> (000)	804.0
<i>D</i> _{calc} (g cm ⁻³)	1.629
μ (mm ⁻¹)	0.792
Reflections collected/unique	19017/5584
<i>R</i> _{int}	0.1686
<i>R</i> ₁ ^a / <i>wR</i> ₂ ^b [<i>I</i> > 2σ(<i>I</i>)]	0.1032/0.1275
<i>R</i> ₁ ^a / <i>wR</i> ₂ ^b (all data)	0.1708/0.1453
GOF on <i>F</i> ²	1.074

$${}^aR_1 = \Sigma(|F_0| - |F_C|)/\Sigma|F_0|; {}^b wR_2 = [\Sigma w(|F_0|^2 - |F_C|^2)^2 / (\Sigma w|F_0|^2)^2]^{1/2}.$$

Table S2. Selected bond lengths (Å) and angles (°) for **JXUST-47**^a.

Cd1—O1	2.211(6)	Cd1—N3	2.325(7)
Cd1—O2	2.643(6)	Cd1—O4 ⁱⁱ	2.388(6)
Cd1—O3 ⁱ	2.268(5)	Cd1—N7 ⁱⁱⁱ	2.483(7)
O3 ⁱ —Cd1—O2	139.3(2)	O3 ⁱ —Cd1—N7 ⁱⁱⁱ	78.8(2)
O3 ⁱ —Cd1—N3	81.6(2)	N3—Cd1—O2	138.3(2)
O3 ⁱ —Cd1—O4 ⁱⁱ	102.9(2)	N3—Cd1—O4 ⁱⁱ	82.1(2)
O1—Cd1—O2	53.6(2)	N3—Cd1—N7 ⁱⁱⁱ	117.7(3)
O1—Cd1—O3 ⁱ	159.3(2)	O4 ⁱⁱ —Cd1—O2	80.4(2)
O1—Cd1—N3	90.8(2)	O4 ⁱⁱ —Cd1—N7 ⁱⁱⁱ	160.0(2)
O1—Cd1—O4 ⁱⁱ	95.1(2)	N7 ⁱⁱⁱ —Cd1—O2	85.6(2)
O1—Cd1—N7 ⁱⁱⁱ	88.0(2)		

^aSymmetry codes: (i) *x*-1, *y*, *z*-1; (ii) -*x*+2, -*y*+1, -*z*+1; (iii) -*x*, -*y*+1, -*z*; (iv) *x*+1, *y*, *z*+1.

Table S3. Comparison of others work with **JXUST-47** for the fluorescence detection of H_2PO_4^- .

MOFs	Sensing medium	LOD	Interferences	Bio-imaging	Reference
UiO-66-NH ₂	H ₂ O	0.73 μM	I ⁻ , Br ⁻ , Cl ⁻ , ClO ₄ ⁻ , AcO ⁻ , SO ₄ ²⁻ , HSO ₄ ⁻ , P ₂ O ₇ ⁴⁻ , SCN ⁻ , AsO ₄ ³⁻ , IO ₃ ⁻ , AsO ₃ ³⁻ , SO ₄ ²⁻ , F ⁻ ,	–	S2
F-MOF	CH ₃ OH–H ₂ O	3.903 μM	Br ⁻ , NO ₃ ⁻ , CH ₃ COO ⁻ , PO ₄ ³⁻ , HF ₂ ⁻ , AsO ₂ ⁻ , CN ⁻ , HPO ₄ ²⁻	WI-38 cell	S3
Zn-DMBI	CH ₃ OH	1.3 μM	F ⁻ , SO ₄ ²⁻ , ClO ₄ ⁻ , NO ₃ ⁻ , I ⁻ , TfO ⁻ , BF ₄ ⁻ , AcO ⁻ , Br ⁻ , OH ⁻ , HCO ₃ ⁻ , HSO ₄ ⁻	–	S4
JXUST-44	DMF	0.121 μM	F ⁻ , Br ⁻ , Cl ⁻ , I ⁻ , NO ₃ ⁻ , SO ₄ ²⁻ , WO ₄ ²⁻ , HPO ₄ ²⁻ , PO ₄ ³⁻	–	S5
JXUST-47	DMF	0.106 mM	HPO ₄ ²⁻ , HCO ₃ ⁻ , CO ₃ ²⁻ , WO ₄ ²⁻ , Br ⁻ , SO ₄ ²⁻ , AcO ⁻ , NO ₃ ⁻	–	This work

References

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