

Supplementary Information

Solar CO₂ Reduction using a Molecular Re(I) Catalyst Grafted on SiO₂ via Amide and Alkyl Amine Linkages

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Supplementary Figures

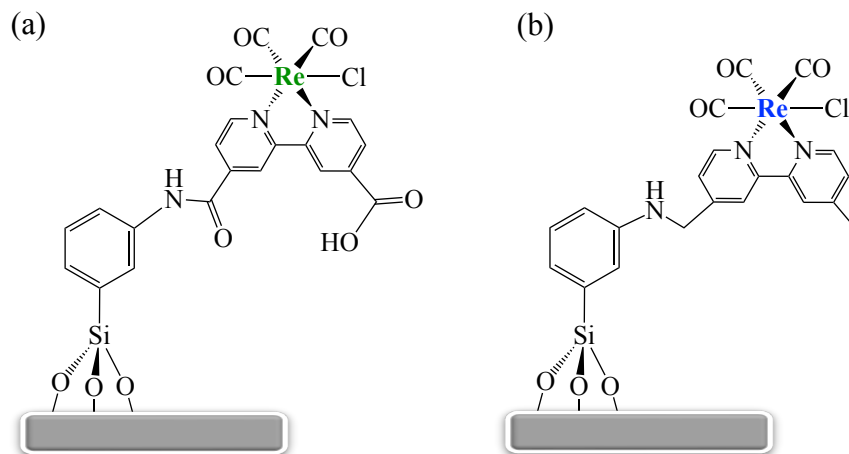


Figure S1. Structures of (a) Re-3-SiO₂ featuring an amide linkage, and (b) Re-4-SiO₂ featuring an alkyl amine linkage.

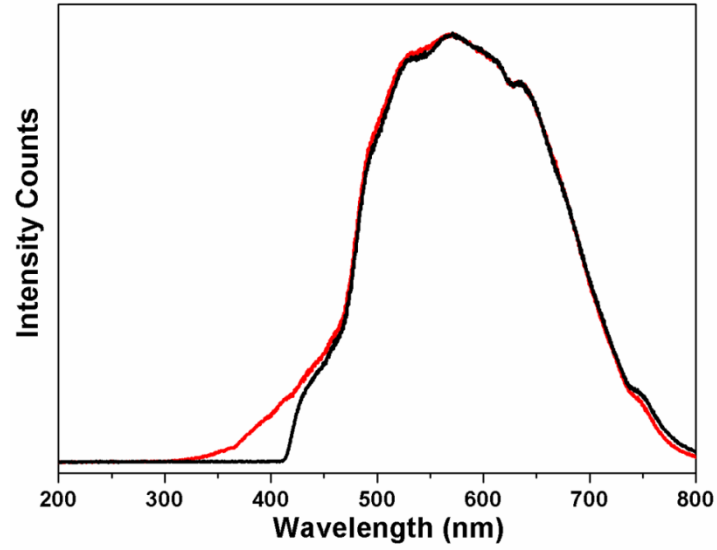


Figure S2. Output spectra of the halogen lamp used in this study without (red trace) and with (black trace) a 420-nm long-pass optical filter.

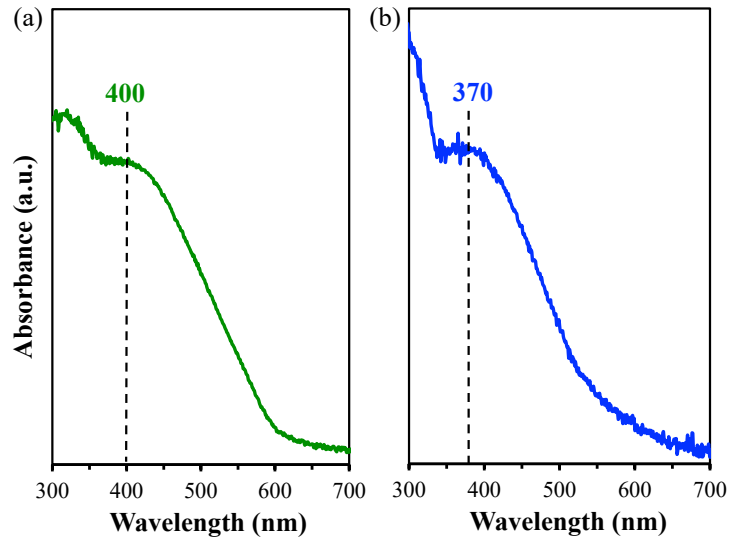


Figure S3. Diffuse reflectance UV-vis spectra of (a) Re-3-SiO₂ and (b) Re-4-SiO₂. Barium sulfate was used as the background.

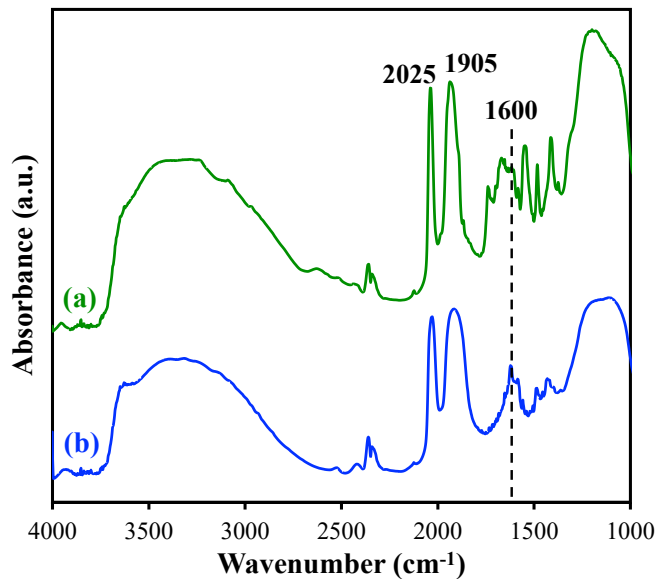


Figure S4. DRIFTS spectra of (a) Re-3-SiO₂ and (b) Re-4-SiO₂. Potassium bromide was used as the background.

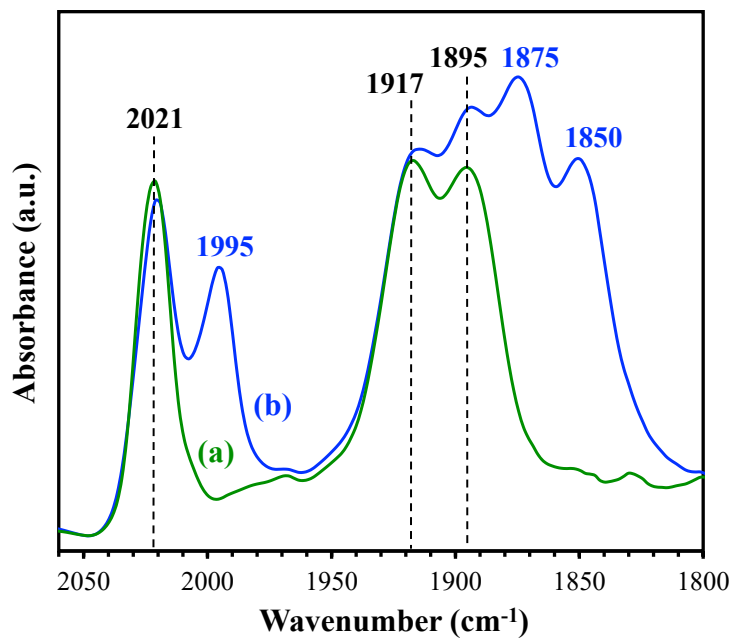


Figure S5. DRIFTS spectra of (a) Re-3-SiO₂ and (b) Re-4-SiO₂ in the presence of TEOA and CO₂.

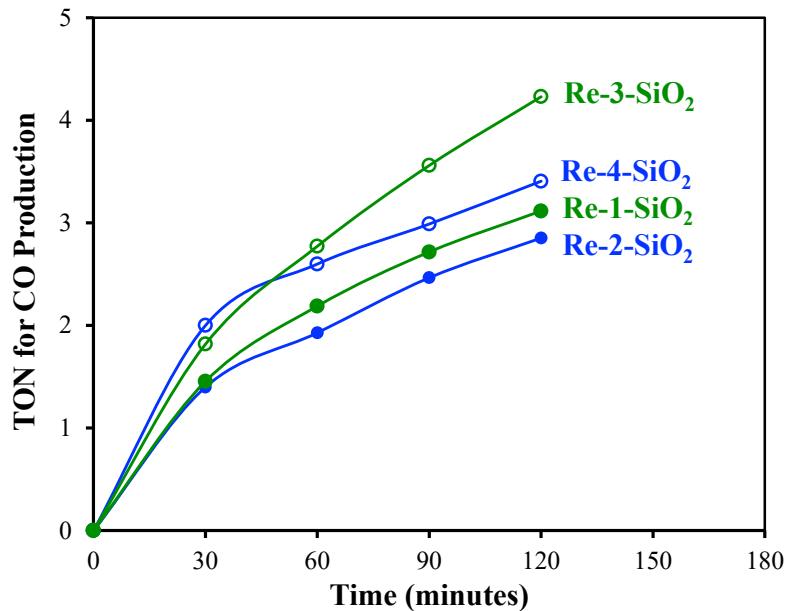


Figure S6. CO production in CO₂ reduction using different heterogenized Re(I) catalysts under simulated solar irradiation (provided by a Xe lamp and an AM1.5 optical filter) in the absence of the Ru(II) photosensitizer. Light intensity was 100 mW/cm².

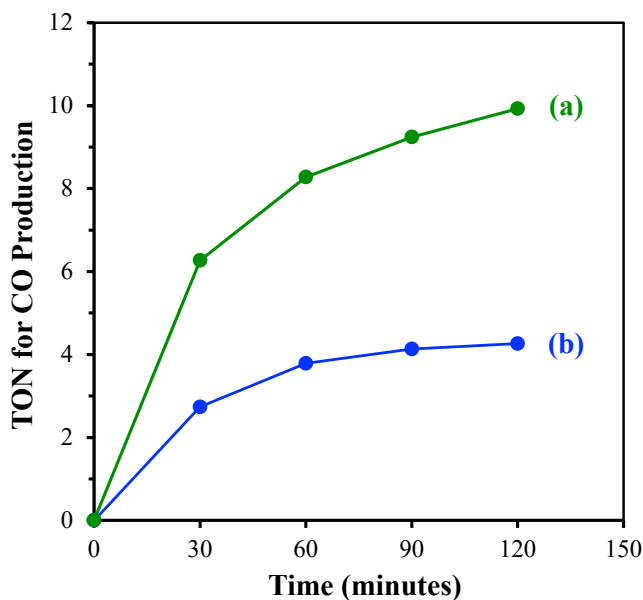


Figure S7. CO production in CO₂ reduction using (a) Re-3-SiO₂ and (b) Re-4-SiO₂ under visible light using a molecular Ru(II) photosensitizer. Light intensity was 100 mW/cm². The loadings of Re were determined to be 2.7 and 4.2 μmol per 10 mg of Re-3-SiO₂ and Re-4-SiO₂, respectively.

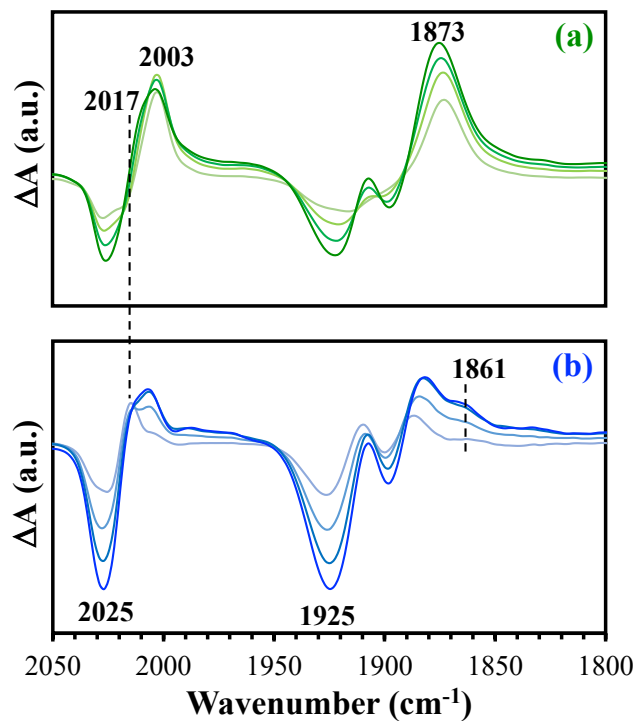


Figure S8. Difference DRIFTS spectra of (a) Re-3-SiO₂ and (b) Re-4-SiO₂ upon light irradiation in the presence of TEOA and CO₂. The spectra were obtained by subtracting spectra collected before light irradiation ($t = 0$) from corresponding spectra collected after light irradiation for different times (2, 10, 30, and 60 min).