## Aluminum ion intercalation mesoporous multilayer carbocatalyst promotes the conversion of glucose to 5-hydroxymethylfurfural

Mingfu Li <sup>a, b, c\*</sup>, Liqun Jiang <sup>a</sup>, Sufei Feng <sup>c, d</sup>, Junsheng Huang <sup>a, b</sup>, Pingjun Zhang <sup>a, b\*</sup>

Jian Zhang <sup>c, d\*</sup>

<sup>a</sup> Institute of Biological and Medical Engineering, Guangdong Academy of Sciences, Guangzhou, Guangdong 510316, China.

<sup>b</sup> Guangdong Province Engineering Research Center for Green Technology of Sugar Industry, Guangzhou, Guangdong 510316, China.

<sup>c</sup> Ningbo Institute of Materials Technology & Engineering, Chinese Academy of Sciences, Ningbo, Zhejiang 315201, China.

<sup>d</sup> University of Chinese Academy of Sciences, Beijing 100049, China.

\*Corresponding Author E-mail: limingfuzd51@126.com



Fig. S1 SEM of different carbon materials (a) SC (b) PSCA; (c) Al-PSCA1; (d) Al-PSCA3.



Fig. S2 XRD analysis of Al-PSCA carbocatalysts.

Samples	Chemical composition (at%)					Raman
	С	0	S	Cl	Al	$= \frac{\text{analysis}}{(I_D/I_G)}$
SC	73.52	26.48	0.64	0	0	0.29
PSCA	76.34	19.92	3.74	0	0	0.37
Al-PSCA1	77.06	19.54	1.37	0.46	1.57	0.32
Al-PSCA2	74.55	20.38	2.97	0.36	1.74	0.33
Al-PSCA3	76.25	17.44	4.15	0.34	1.82	0.35

 Table S1 XPS analysis and Raman analysis results of carbon materials



Fig. S3 Py-FTIR analysis of Al-PSCA catalysts.



**Fig. S4** Recyclability of catalysts. (a) Catalysts recycle, (b) XPS analysis, SEM analysis of Al-PSCA2 (c) and recycled Al-PSCA2 (d). Reaction condition: 20 mL THF-H<sub>2</sub>O ( $V_{THF}/V_{H2O}$  =4/1), 150 °C 2 h, 0.5 mM glucose, 90 mg catalysts.



**Fig. S5** Experimental (points) versus calculated kinetic parameter. (a) glucose, (b) levoglucosan, (c) fructose Reaction condition: 20 mL THF-H<sub>2</sub>O ( $V_{THF}/V_{H2O}$  =4/1), 140°C, 0.5 mM substrates, 90 mg catalysts.