

Electronic Supplementary Information (ESI)

Terpyridine Isomerism as a Tool for Tuning of the Red-to-NIR Emissive Properties in Heteronuclear Gold(I)-Thallium(I) Complexes

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1. Characterization of the complexes

1. ^1H NMR signal assignement complexes.

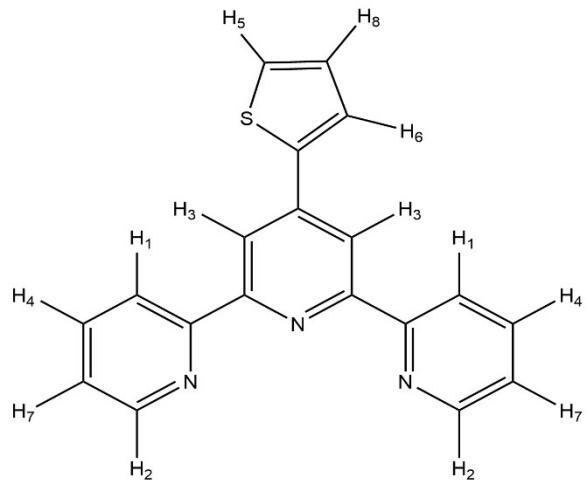


Figure S1: ^1H NMR assignment of ligand **L₁**.

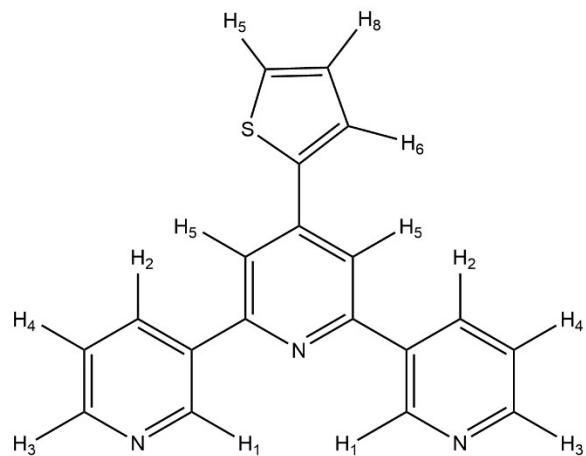


Figure S2: ^1H NMR assignment of ligand **L₂**.

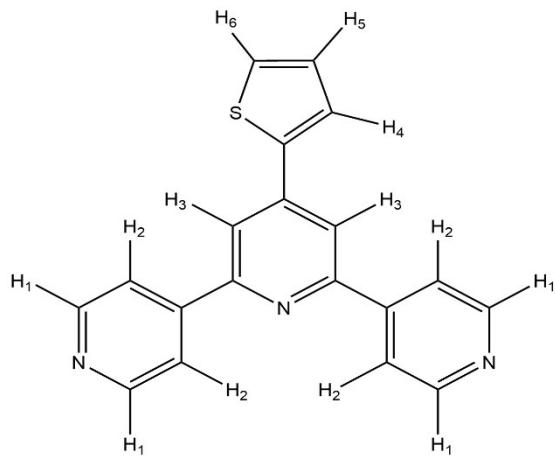
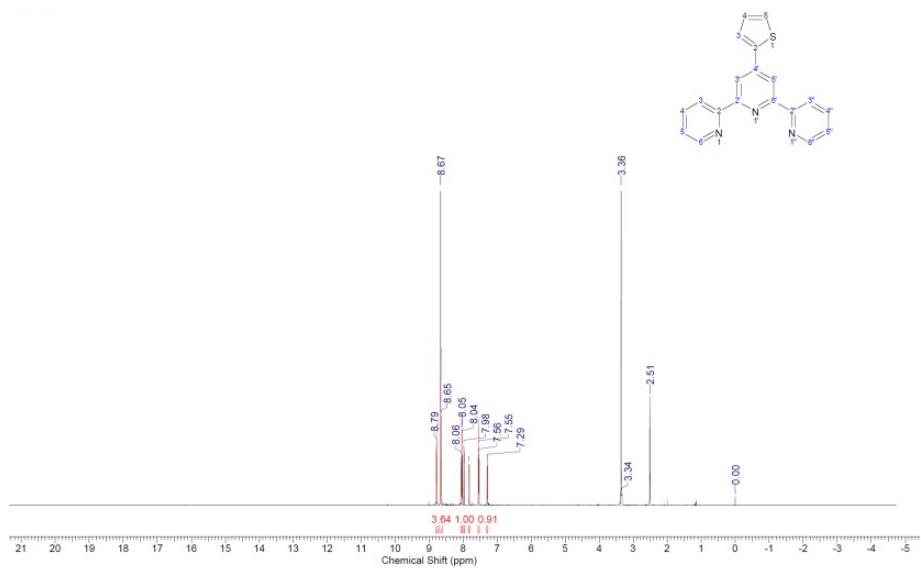


Figure S3: ¹H NMR assignment of ligand L₃.

2. ¹H NMR of L₁, L₂ and L₃ (600 MHz).



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gure S4 : ¹H NMR spectrum of L₁ in $[D_6]$ -DMSO.

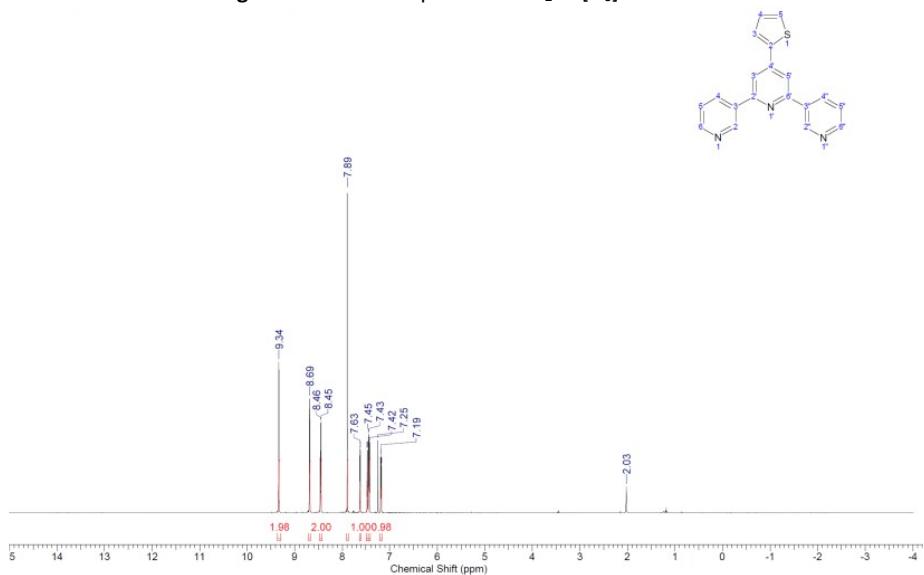


Figure S5 : ^1H NMR spectrum of L_2 in CDCl_3 .

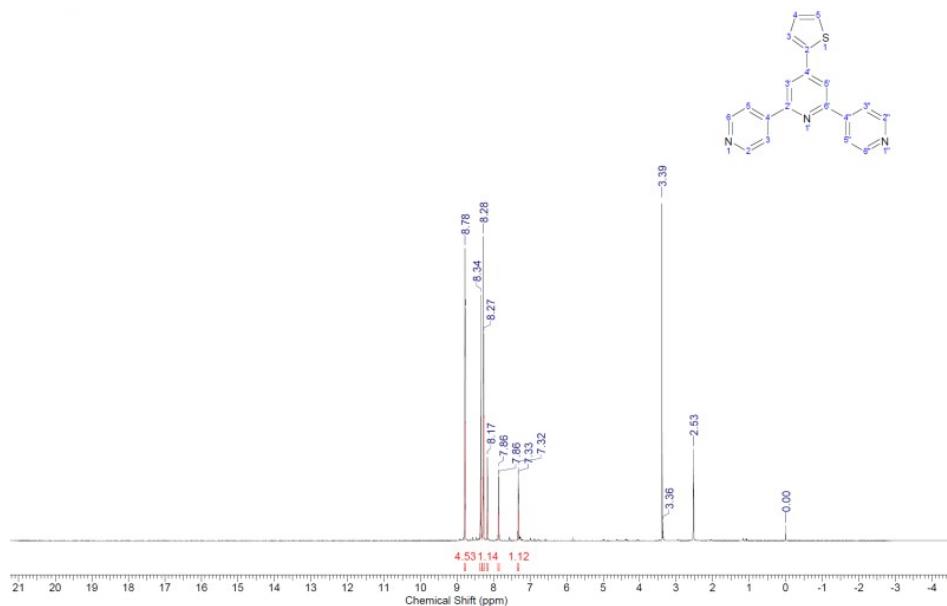


Figure S6 : ^1H NMR spectrum of L_3 in $[\text{D}_6]\text{-DMSO}$.

3. ^1H NMR spectra of complexes 1, 2 and 3 (300 MHz, 298K)

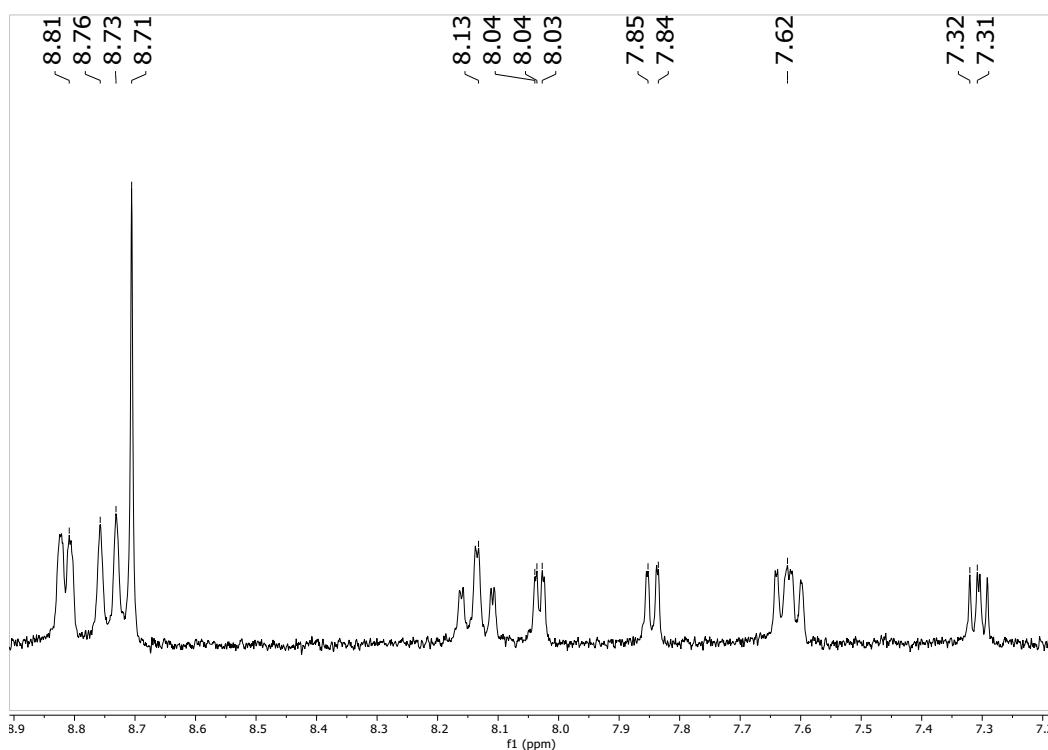


Figure S7: ^1H NMR spectrum of complex 1 in $[\text{D}_6]\text{-DMSO}$.

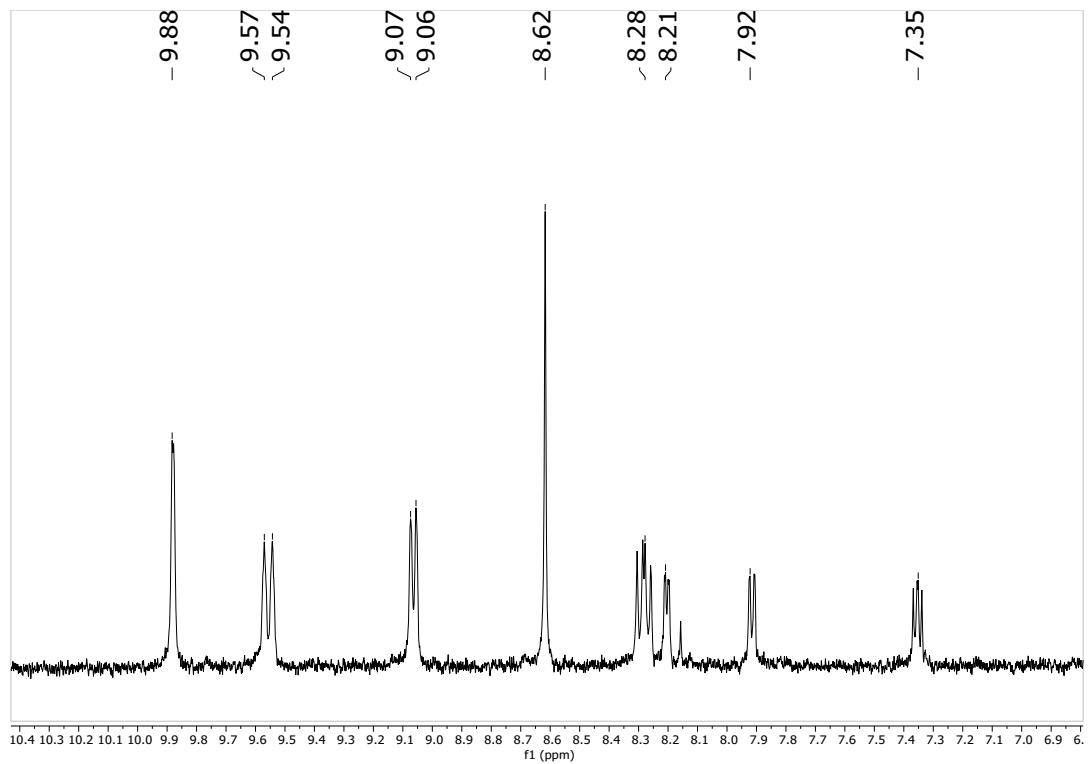


Figure S8: ¹H NMR spectrum of complex **2** in $[D_6]$ -DMSO.

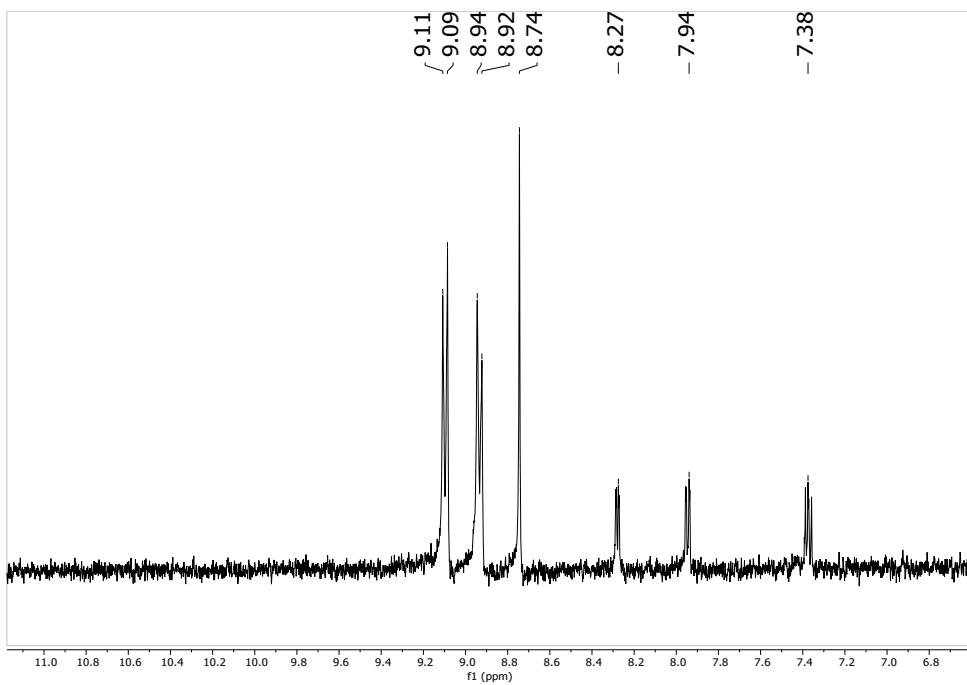


Figure S9: ¹H NMR spectrum of complex **3** in $[D_6]$ -DMSO.

4. Single crystal analysis of compounds 1-3.

Table S1 : Data collection and structure refinement details for 2-4

	1	2	3
Chemical Formula	$C_{31} H_{13} Au Cl_{10} N_3 S Tl$	$C_{43} H_{13} Au_2 Cl_{20} N_3 S Tl_2$	$C_{31} H_{13} Au Cl_{10} N_3 S Tl$
Crystal habit	Dark-Green plate	Red plate	Orange prism
Crystal size/mm	0.166x0.037x0.017	0.121x0.106x0.016	0.121x0.121x0.046
Crystal system	Triclinic	Monoclinic	Triclinic
Space group	P-1	C 2/c	P -1
$a/\text{\AA}$	10.5516(8)	44.5975(15)	11.2372(4)
$b/\text{\AA}$	10.9165(9)	10.7247(3)	12.8751(4)
$c/\text{\AA}$	15.3597(13)	30.8509(10)	13.5564(6)
$\alpha/^\circ$	89.270(3)°	90	76.6270(10)
$\beta/^\circ$	86.508(3)°.	124.1240(10)	85.9050(10)
$\gamma/^\circ$	88.089(3)°.	90	69.1070(10)°.
$V/\text{\AA}^3$	1764.9(2)	12215.2(7)	1782.56(12)
Z	2	8	2
$D_c/\text{g cm}^{-3}$	2.287	2.300	2.264
M	1215.34	2115.30	1215.34
F(000)	1132	7744	1129
T/°C	26	22	26
2θmax/°	53	56	56
$\mu(\text{Mo-K}\alpha)/\text{mm}^{-1}$	9.557	10.993	9.464
No. refl. Measured	38612	137038	41470
No. unique refl.	7481	14581	8531
R_{int}	0.0834	0.0871	0.0315
$R[F > 2\sigma(F)][a]$	0.0589	0.0696	0.0277
wR[F 2 , all refl.][b]	0.1548	0.2312	0.0545
No. of refl. Used [$F > 2\sigma(F)$]	7481	14581	8531
No. of parameters	423	613	519
No. of restraints	20	103	134
S [c]	1.021	1.020	1.071
Max. residual electron density/e·Å ⁻³	1.486	2.219	0.66

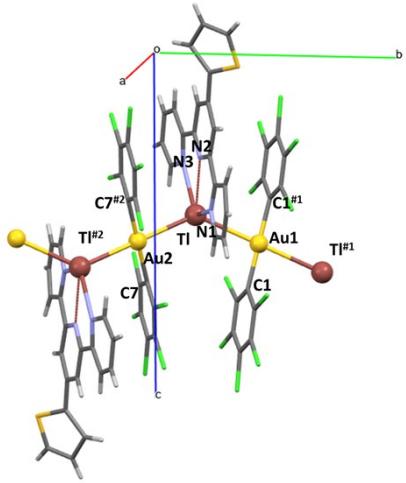


Figure S10: Partial view of the polymeric chain in the crystal structure of **1** formed via $\text{Au}\cdots\text{Te}$ interactions with the labelling scheme for the atom positions and crystallographic axes (red *a*, green *b* and blue *c*). Hydrogen atoms have been omitted for clarity. #1: $-x+1, -y+1, -z+1$; #2: $-x+1, -y, -z+1$.

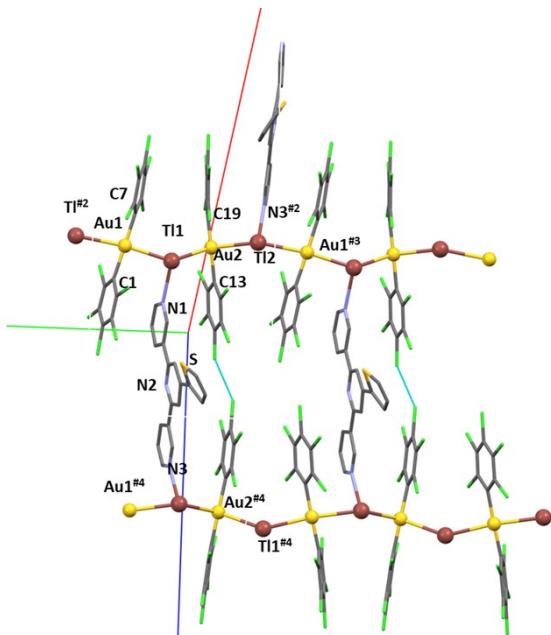


Figure S11: Partial view of the 2D network in the crystal structure of **2** formed via $\text{Au}\cdots\text{Te}$ interactions and bridging L_2 ligands with the labelling scheme for the atom positions and crystallographic axes (red *a*, green *b* and blue *c*). Hydrogen atoms have been omitted for clarity. #1: $x, y+1, z$; #2: $x, -y, z-1/2$; #3: $x, y-1, z$; #4: $x, -y, z+1/2$

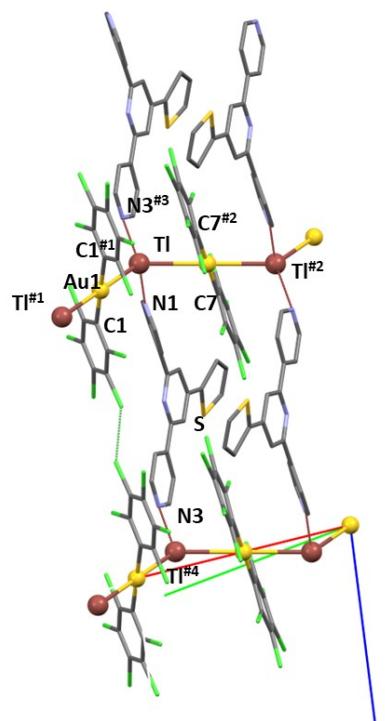


Figure S12: Partial view of the 2D network in the crystal structure of **3** formed via $\text{Au}\cdots\text{Tl}$ interactions and bridging L_3 ligands with the labelling scheme for the atom positions and crystallographic axes (red a , green b and blue c). Hydrogen atoms have been omitted for clarity. #1: $-x+2, -y, -z$; #2: $-x+1, -y, -z$; #3: $x, y, z-1$; #4: $-x+2, -y, -z+1$.

5. IR spectra

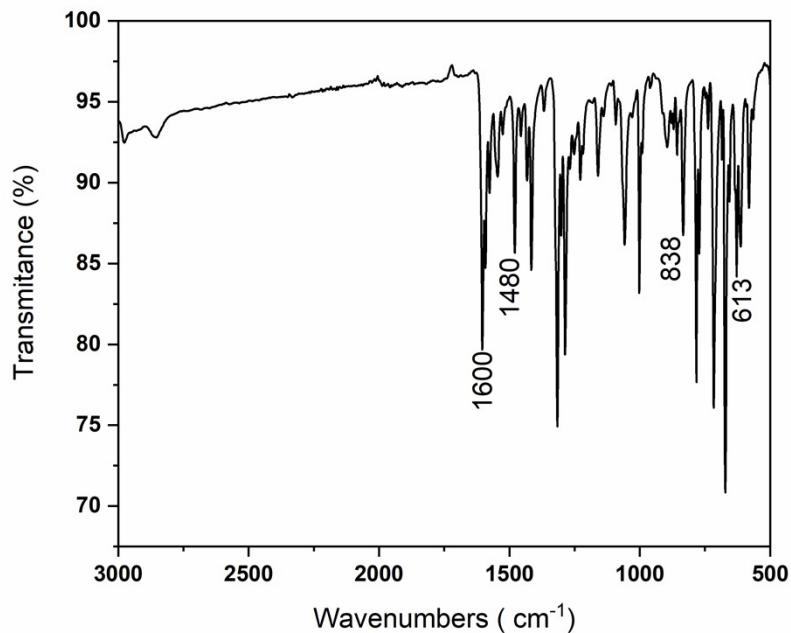


Figure S13: FT-IR spectrum of complex **1**.

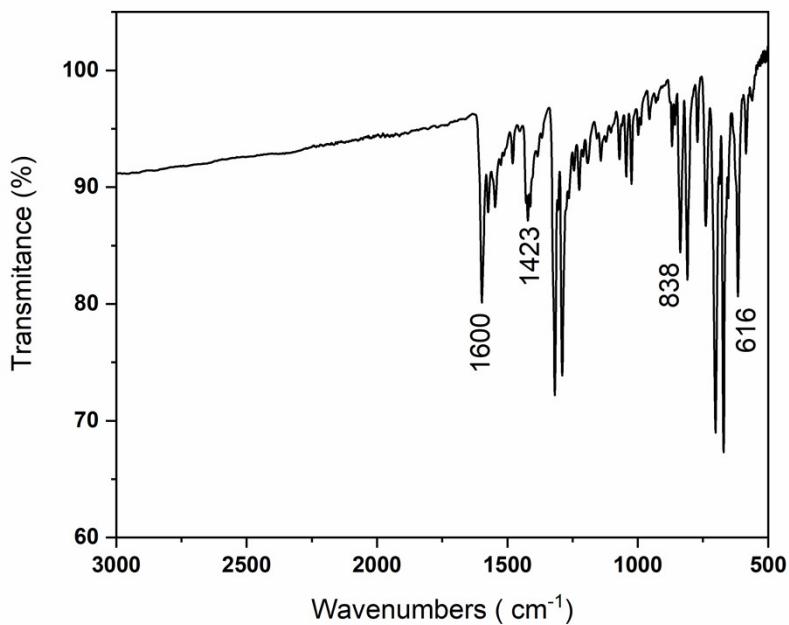


Figure S14: FT-IR spectrum of complex **2**.

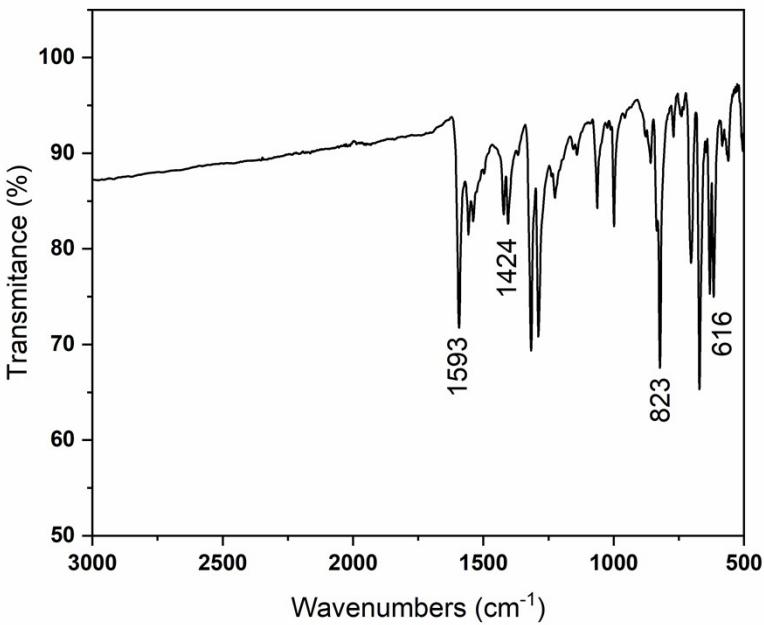


Figure S15: FT-IR spectrum of complex **3**.

2. Optical properties

1. UV-Vis absorption spectra in solution

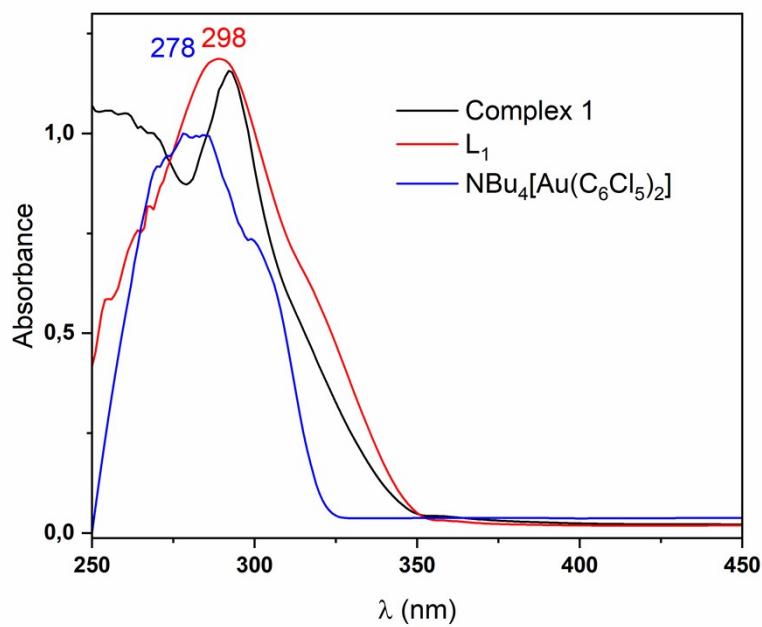


Figure S16: UV-Vis absorption spectra in DMSO solution of complex **1** ($2.96 \cdot 10^{-5} M$), L₁ ($1.9 \cdot 10^{-5} M$) and [NBu₄][Au(C₆Cl₅)₂] ($3.39 \cdot 10^{-5} M$).

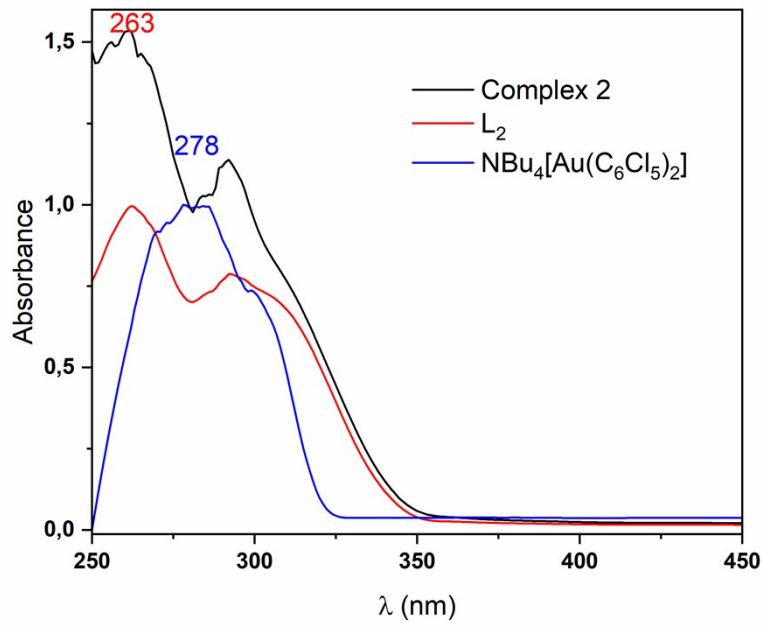


Figure S17: UV-Vis absorption spectra in DMSO solution of complex **2** ($4.28 \cdot 10^{-5} M$), L₂ ($2.71 \cdot 10^{-5} M$) and [NBu₄][Au(C₆Cl₅)₂] ($3.39 \cdot 10^{-5} M$).

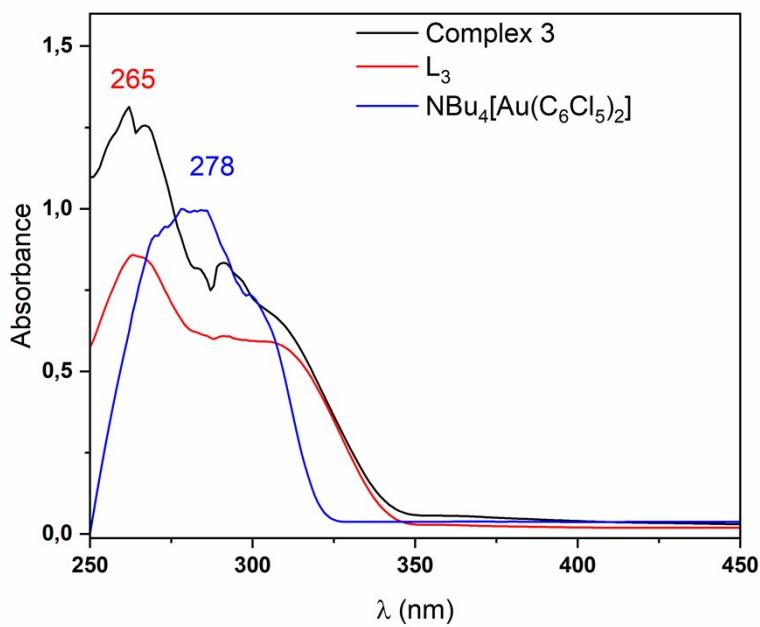


Figure S18: UV-Vis absorption spectra in DMSO solution of complex **3** ($3.75 \cdot 10^{-5}$ M), L_3 ($2.03 \cdot 10^{-5}$ M) and $[NBu_4][Au(C_6Cl_5)_2]$ ($3.39 \cdot 10^{-5}$ M).

2. Experimental UV-vis solid state absorption and TD-DFT singlet-singlet and singlet-triplet for complexes.

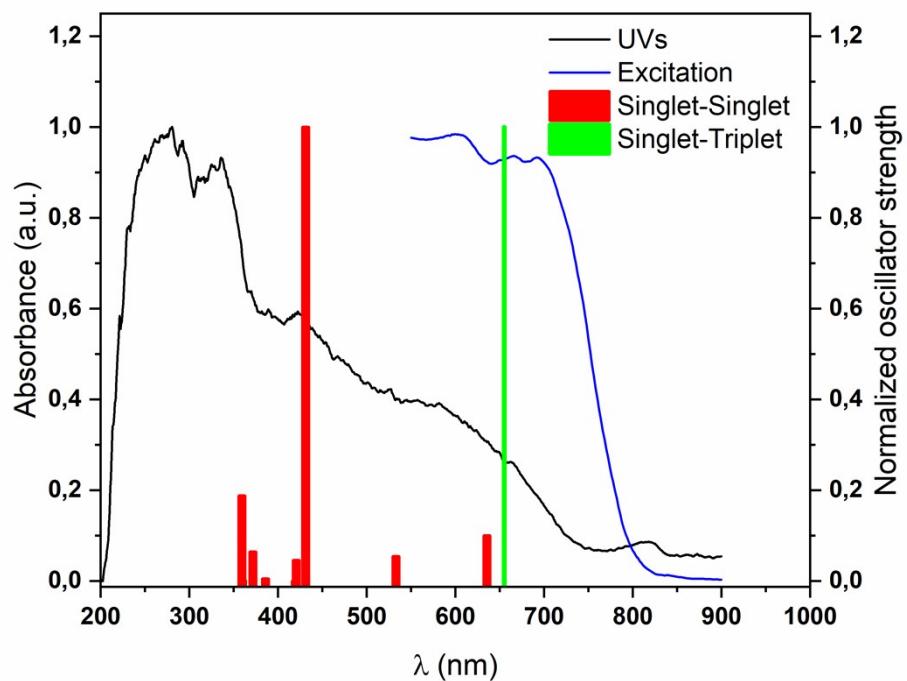


Figure S19: Experimental UV-vis solid state absorption spectrum (black line) , excitation (blue) ,TD-DFT singlet-singlet excitations (red bars) and singlet-triplet (green bars) for complex **1**. The green bar only represents the energy

of the lowest singlet-Triplet transitions since the oscillator strength cannot be calculated.

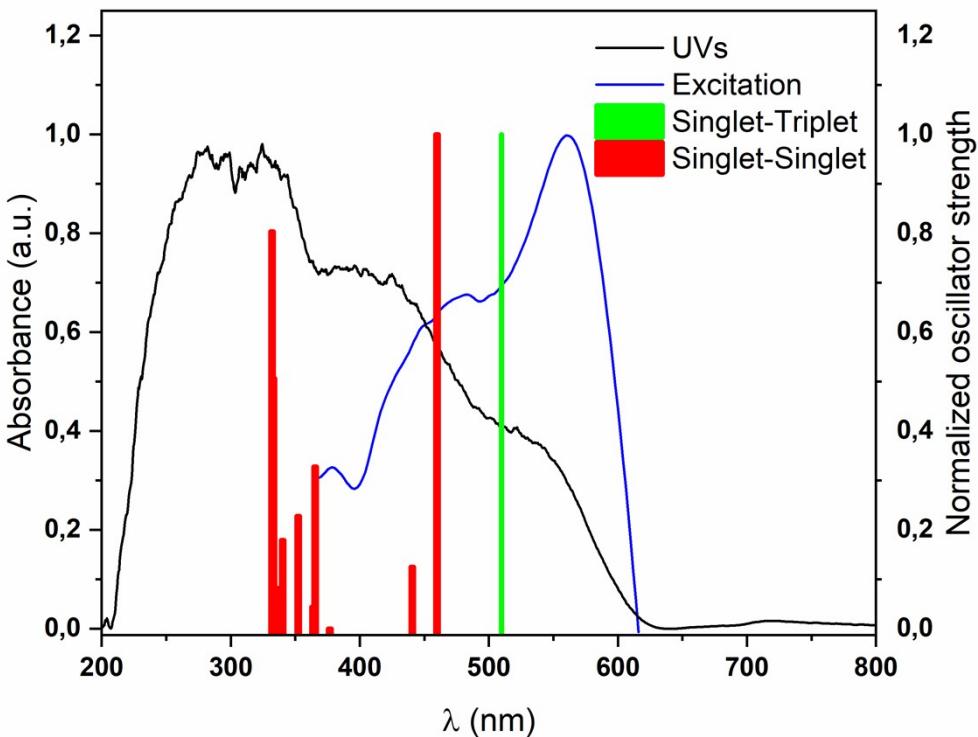


Figure S20: Experimental UV-vis solid state absorption spectrum (black line), excitation (blue), TD-DFT singlet-singlet excitations (red bars) and singlet-triplet (green bars) for complex **2**. The green bar only represents the energy of the lowest singlet-Triplet transitions since the oscillator strength cannot be calculated.

3. Computational studies

1. xyz for model 1a

Tl	10.60940000	4.40310000	0.91970000	C	7.74550000	2.87940000	2.29790000
Au	7.85850000	5.04140000	0.00000000	C	7.78980000	2.64250000	3.64130000
Au	13.27210000	5.04140000	0.00000000	C	12.13540000	5.96810000	-2.75750000
Cl	7.95370000	1.81210000	1.09240000	C	11.96220000	5.99880000	-4.04780000
Cl	11.54330000	7.38260000	-4.95110000	C	12.49270000	3.63020000	-3.97710000
Cl	6.93240000	6.90040000	2.57010000	C	12.33270000	4.90350000	-4.61350000
Cl	7.64240000	3.11550000	6.32810000	C	7.43880000	3.64740000	4.68420000
Cl	7.15970000	6.10890000	5.55560000	C	7.46620000	4.35720000	1.90900000
Cl	12.06860000	7.55230000	-1.89310000	C	7.33090000	5.26610000	2.93420000
Cl	11.95170000	4.68310000	-6.45180000	C	12.64980000	3.69500000	-2.70450000
Cl	7.94950000	0.97080000	4.09030000	Cl	12.99190000	2.22580000	-1.78000000
Cl	12.60890000	2.10930000	-4.86630000	Tl	5.10750000	5.67960000	-0.91970000
C	12.69640000	4.85280000	-1.96210000	Cl	7.76320000	8.27060000	-1.09240000
C	7.45310000	4.81340000	4.36600000	Cl	8.78450000	3.18230000	-2.57010000

Cl	8.07450000	6.96720000	-6.32810000	C	11.91210000	1.11870000	4.41900000
Cl	8.55720000	3.97380000	-5.55560000	H	12.20190000	1.15360000	5.32150000
Cl	7.76750000	9.11190000	-4.09030000	C	10.22900000	7.46920000	1.14900000
C	8.26380000	5.26930000	-4.36600000	H	10.32000000	7.19710000	0.23730000
C	7.97140000	7.20330000	-2.29790000	C	10.70970000	5.98660000	8.69670000
C	7.92710000	7.44020000	-3.64130000	H	10.08000000	6.63750000	8.42230000
C	8.27810000	6.43530000	-4.68420000	C	11.33670000	3.66020000	5.81550000
C	8.25070000	5.72560000	-1.90900000	H	11.59280000	2.89380000	6.31410000
C	8.38600000	4.81660000	-2.93420000	C	11.22910000	5.89130000	10.07540000
Cl	15.00100000	2.70020000	4.95110000	H	11.06470000	6.56610000	10.72570000
Cl	14.47560000	2.53050000	1.89310000	C	11.73790000	0.00540000	2.26250000
Cl	14.59260000	5.39960000	6.45180000	H	11.89080000	-0.76280000	1.72810000
Cl	13.93540000	7.97350000	4.86630000	C	11.85960000	4.90120000	10.34060000
C	13.84790000	5.22990000	1.96210000	H	12.14910000	4.66970000	11.2160000
C	14.40880000	4.11470000	2.75750000	C	12.02010000	-0.11530000	3.62360000
C	14.58200000	4.08390000	4.04780000	H	12.27420000	-0.94370000	4.01430000
C	14.05150000	6.45250000	3.97710000	C	11.23990000	5.06510000	7.90120000
C	14.21150000	5.17920000	4.61350000	C	11.21930000	3.62970000	4.45440000
C	13.89440000	6.38770000	2.70450000	C	10.31560000	6.99630000	3.46450000
Cl	13.55240000	7.85690000	1.78000000	S	3.52560000	6.14370000	-8.89460000
S	12.19130000	3.93900000	8.89460000	N	4.83040000	5.55960000	-3.65900000
N	10.88650000	4.52310000	3.65900000	N	4.71410000	7.94260000	-2.45700000
N	11.00280000	2.14010000	2.45700000	C	4.92090000	4.09500000	-5.67410000
C	10.79600000	5.98780000	5.67410000	H	5.06400000	3.25400000	-6.08520000
H	10.65290000	6.82870000	6.08520000	C	4.42330000	9.03510000	-1.69690000
C	11.29360000	1.04760000	1.69690000	H	4.54350000	9.02110000	-0.74930000
H	11.17340000	1.06160000	0.74930000	C	5.02100000	4.24130000	-4.33070000
C	10.69590000	5.84150000	4.33070000	C	4.61400000	5.23080000	-6.43410000
C	11.10290000	4.85200000	6.43410000	N	5.34780000	3.27310000	-2.12110000
N	10.36910000	6.80970000	2.12110000	C	5.86360000	1.09440000	-1.74990000
C	9.85340000	8.98840000	1.74990000	H	6.08640000	0.41550000	-1.11770000
H	9.63060000	9.66720000	1.11770000	C	4.37490000	7.83230000	-3.74730000
C	11.34200000	2.25050000	3.74730000	C	5.58770000	1.80520000	-3.97710000
C	10.12920000	8.27750000	3.97710000	H	5.52740000	1.63250000	-4.90120000
H	10.18950000	8.45020000	4.90120000	C	5.87040000	0.77710000	-3.05800000
C	9.84650000	9.30560000	3.05800000	H	6.06060000	-0.10270000	-3.34680000
H	9.65640000	10.18540000	3.34680000	C	3.80480000	8.96400000	-4.41900000

H	3.51500000	8.92920000	-5.32150000	C	3.97900000	10.07730000	-2.26250000
C	5.48790000	2.61360000	-1.14900000	H	3.82610000	10.84560000	-1.72810000
H	5.39690000	2.88560000	-0.23730000	C	3.85730000	5.18160000	-10.34060000
C	5.00720000	4.09610000	-8.69670000	H	3.56780000	5.41310000	-11.21600000
H	5.63690000	3.44520000	-8.42230000	C	3.69680000	10.19810000	-3.62360000
C	4.38020000	6.42250000	-5.81550000	H	3.44270000	11.02640000	-4.01430000
H	4.12410000	7.18890000	-6.31410000	C	4.47700000	5.01760000	-7.90120000
C	4.48780000	4.19140000	-10.07540000	C	4.49760000	6.45300000	-4.45440000
H	4.65220000	3.51660000	-10.72570000	C	5.40130000	3.08650000	-3.46450000

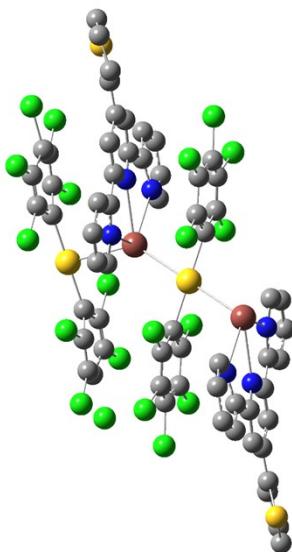


Figure S21: Model 1a. Color code: yellow (gold), grey (carbon), blue (nitrogen), orange (sulfur), brown (thallium) and green (chlorine).

2. xyz for model 2a

Au	12.91460000	0.47210000	3.06930000	Cl	8.18870000	-0.12870000	7.63650000
Au	14.17810000	5.76410000	4.07020000	Cl	15.71260000	4.23090000	-1.20700000
Tl	14.41300000	-2.06080000	3.65180000	Cl	18.26990000	1.20970000	1.16280000
Tl	13.55200000	2.95960000	4.58610000	Cl	12.24630000	0.59410000	-0.16290000
Cl	16.21020000	1.00810000	3.43940000	Cl	17.35980000	0.98130000	-1.79110000
Cl	13.60580000	4.48610000	1.04170000	Cl	13.39880000	-0.25850000	6.38170000
Cl	11.15730000	-0.51590000	8.48050000	Cl	18.69940000	4.69530000	-0.60040000

Cl	9.73740000	0.73460000	2.57920000	C	11.47850000	6.64930000	5.09510000
Cl	14.33240000	0.69280000	-2.45690000	C	16.60600000	1.01990000	0.79940000
Cl	9.44630000	6.96030000	8.51120000	C	6.30680000	2.95040000	11.20660000
Cl	11.09190000	6.81020000	3.39440000	H	6.42100000	3.11720000	10.29800000
Cl	8.84040000	7.27460000	5.47300000	C	9.83040000	2.41200000	12.21280000
Cl	7.48540000	0.46010000	4.69080000	H	9.71730000	2.23030000	13.11890000
Cl	12.33490000	6.29110000	9.43770000	C	11.11080000	2.43670000	11.63310000
Cl	19.57640000	5.62830000	2.20150000	C	10.04890000	2.91500000	9.55420000
Cl	14.54600000	5.77850000	7.34970000	C	12.37240000	1.44350000	13.75280000
Cl	17.44400000	6.08950000	4.41570000	H	11.66700000	1.03480000	14.19460000
S	13.89140000	2.77770000	12.12340000	C	10.15690000	3.47270000	5.40660000
C	14.29670000	0.80440000	1.56810000	H	10.18190000	3.57400000	4.48320000
C	8.74850000	2.66620000	11.40070000	C	16.95370000	5.49100000	2.88590000
C	13.91650000	0.71860000	0.22730000	C	11.20370000	2.73800000	10.27180000
C	10.74840000	-0.21340000	6.82410000	H	12.03230000	2.81810000	9.86200000
C	11.45980000	0.16410000	4.49740000	C	12.96460000	6.22030000	6.81130000
C	16.21840000	0.94380000	-0.51080000	C	17.56460000	4.87970000	0.61550000
N	11.33590000	3.24530000	6.04260000	C	12.31670000	2.14070000	12.53210000
C	15.64120000	0.95240000	1.79540000	C	4.92910000	2.75410000	13.06840000
C	12.74510000	6.31150000	5.43730000	H	4.08060000	2.76470000	13.44300000
C	9.41710000	-0.03110000	6.45890000	C	11.26910000	3.06510000	7.33230000
C	9.13970000	0.23590000	5.13080000	H	12.05930000	2.86160000	7.77510000
C	16.20330000	4.67490000	0.39330000	C	8.96130000	3.42550000	7.40120000
N	5.91140000	2.52670000	13.84990000	H	8.16630000	3.52140000	7.87230000
C	7.35340000	2.69080000	11.95230000	C	11.96820000	6.43050000	7.74860000
C	15.24570000	4.80470000	1.39190000	C	8.94820000	3.56490000	5.99910000
C	10.10330000	0.33890000	4.16290000	H	8.16170000	3.71010000	5.52090000
C	11.73400000	-0.09330000	5.84340000	C	5.02560000	2.98040000	11.74800000
N	8.85170000	2.92680000	10.10330000	H	4.27410000	3.14790000	11.22400000
C	10.42780000	6.88950000	5.99400000	C	17.95150000	5.30870000	1.88730000
C	14.86100000	0.78400000	-0.75850000	C	14.61780000	2.09130000	13.51280000
C	15.59640000	5.26150000	2.68420000	H	15.51680000	2.16750000	13.73300000
C	7.11000000	2.50310000	13.28550000	C	10.11840000	3.14980000	8.09590000
H	7.84210000	2.34600000	13.83670000	C	13.76300000	1.49070000	14.18190000
C	10.70790000	6.76190000	7.32460000	H	14.01710000	1.06070000	14.96820000

N	14.56490000	-2.52670000	1.08030000	C	19.85720000	-2.73800000	-2.49770000
S	22.54480000	-2.77770000	-0.64610000	H	20.68580000	-2.81810000	-2.90760000
C	17.40200000	-2.66620000	-1.36890000	C	20.97010000	-2.14070000	-0.23750000
N	19.98940000	-3.24530000	-6.72700000	C	13.58260000	-2.75410000	0.29880000
C	16.00690000	-2.69080000	-0.81730000	H	12.73410000	-2.76470000	0.67340000
N	17.50520000	-2.92680000	-2.66630000	C	19.92260000	-3.06510000	-5.43730000
C	15.76350000	-2.50310000	0.51590000	H	20.71280000	-2.86160000	-4.99440000
H	16.49560000	-2.34600000	1.06710000	C	17.61480000	-3.42550000	-5.36830000
C	14.96020000	-2.95040000	-1.56300000	H	16.81980000	-3.52140000	-4.89730000
H	15.07450000	-3.11720000	-2.47160000	C	17.60160000	-3.56490000	-6.77040000
C	18.48390000	-2.41200000	-0.55680000	H	16.81520000	-3.71010000	-7.24870000
H	18.37080000	-2.23030000	0.34940000	C	13.67910000	-2.98040000	-1.02160000
C	19.76420000	-2.43670000	-1.13650000	H	12.92760000	-3.14790000	-1.54560000
C	18.70230000	-2.91500000	-3.21540000	C	23.27130000	-2.09130000	0.74320000
C	21.02590000	-1.44350000	0.98330000	H	24.17020000	-2.16750000	0.96350000
H	20.32040000	-1.03480000	1.42500000	C	18.77180000	-3.14980000	-4.67370000
C	18.81030000	-3.47270000	-7.36290000	C	22.41650000	-1.49070000	1.41230000
H	18.83540000	-3.57400000	-8.28640000	H	22.67060000	-1.06070000	2.19860000

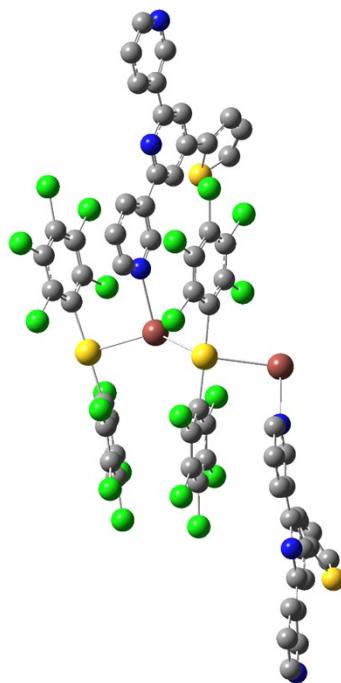


Figure S22: Model **2a**. Color code: yellow (gold), grey (carbon), blue (nitrogen), orange (sulfur), brown (thallium) and green (chlorine).

