

## Supporting Information

### **Bimetallic sulfide/N-doped carbon composites derived from Prussian blue analogues/cellulose nanofibers (PBA/CNFs) film toward enhanced oxygen evolution reaction**

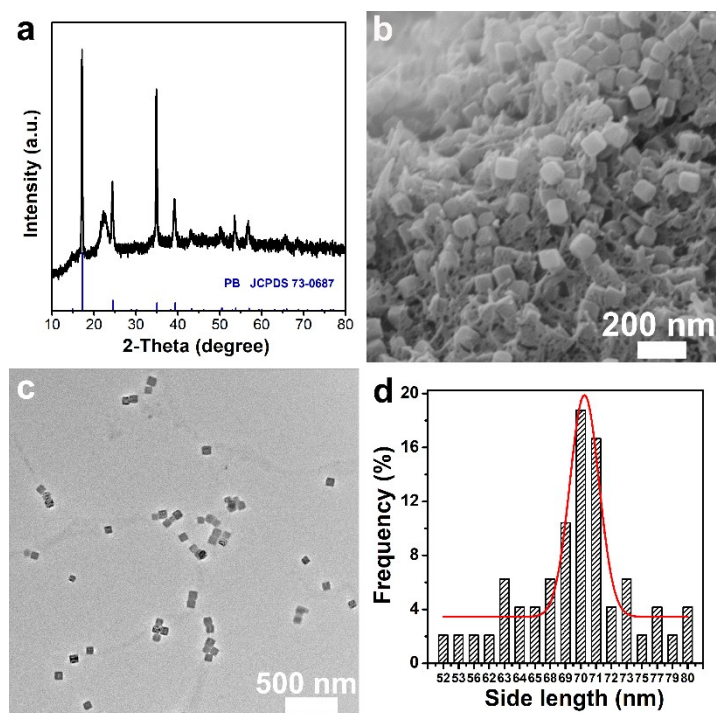
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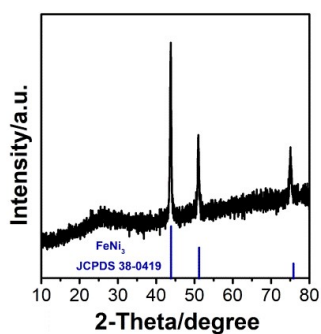
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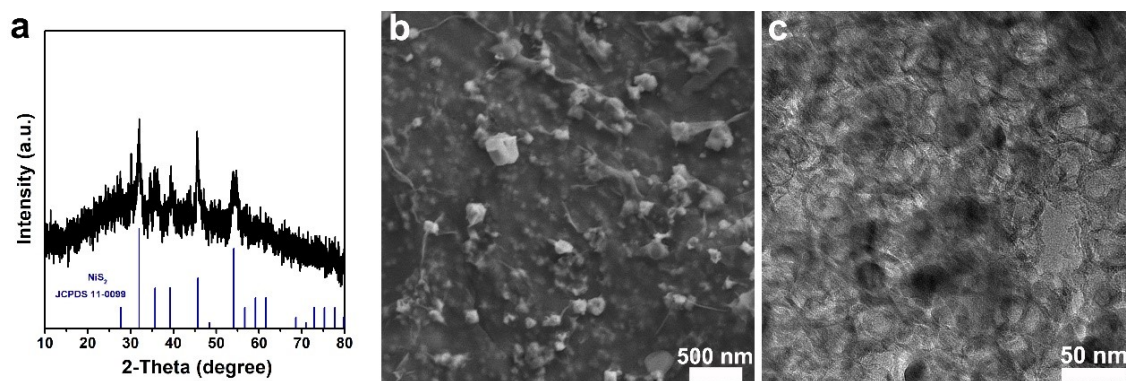
**Fig. S1.** a) XRD result, b) SEM and c) TEM image of NiFe-PBA/CNFs film, d) histograms of size distribution of NiFe-PBA NPs in NiFe-PBA/CNFs film.



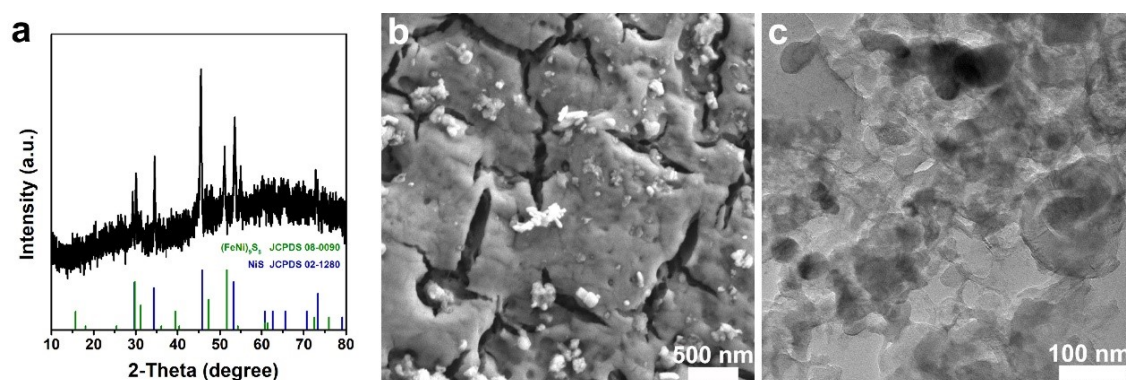
**Fig. S2.** XRD result of NiFe-alloy/NC composite.

**Table S1** Elemental composition of Ni and Fe determined by ICP-OES

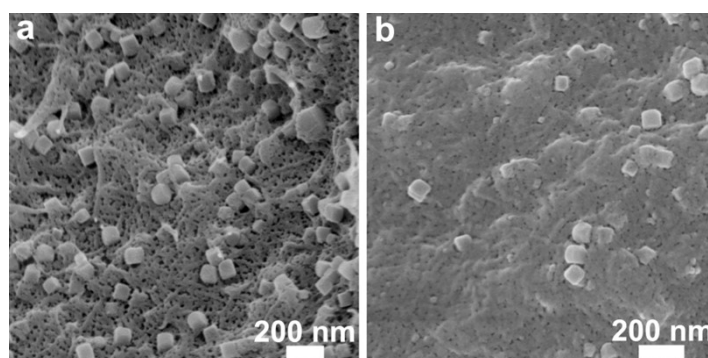
Samples	Ni content (mg/L)	Fe content (mg/L)	Ni/Fe molar ratio
$\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DF}$	9.069	6.092	1.416
$\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DA}$	13.90	9.009	1.468
$\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}/\text{NC NPs}$	22.72	15.36	1.407



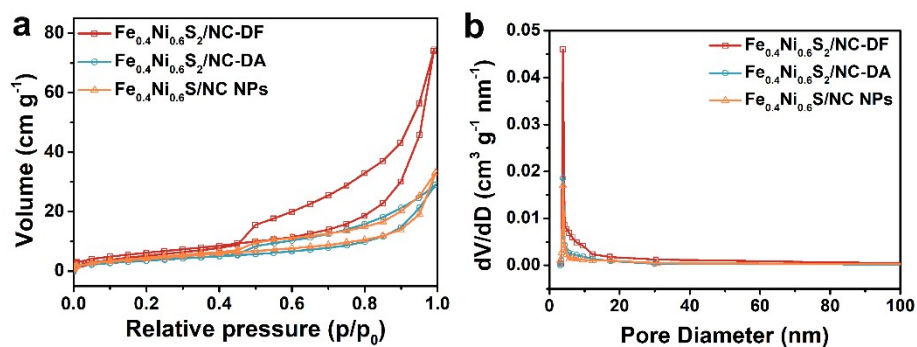
**Fig. S3.** a) XRD result, b) SEM and c) TEM image of  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DA}$ .



**Fig. S4.** a) XRD result, b) SEM and c) TEM image of  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}/\text{NC}$  NPs.



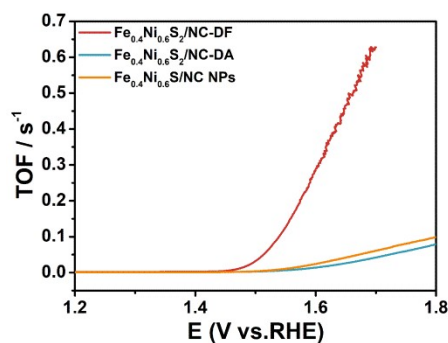
**Fig. S5.** SEM image of a) NiFe-PBA/CNFs film and b) NiFe-PBA/CNFs aggregates.



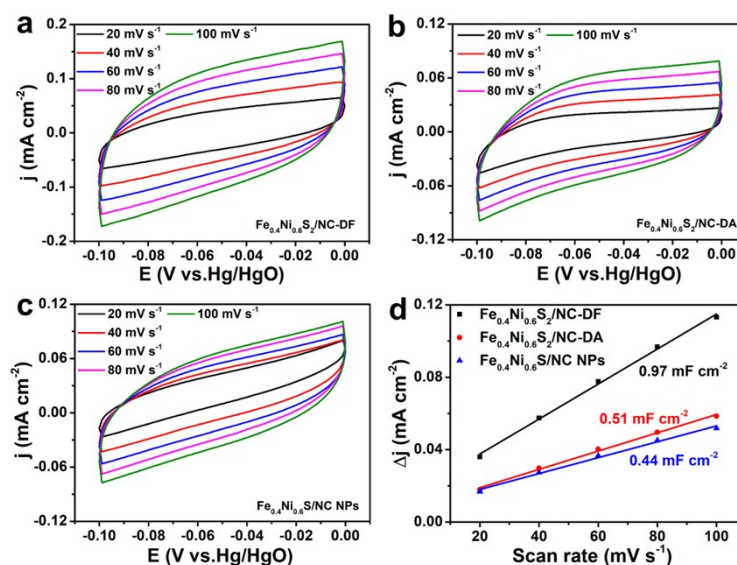
**Fig. S6.** a)  $\text{N}_2$  adsorption-desorption isotherm and b) the pore size distribution curve of  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DF}$ ,  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DA}$  and  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}/\text{NC}$  NPs.

**Table S2** Calculated values of OER electrocatalysts based on the fitted equivalent circuit in 1M KOH.

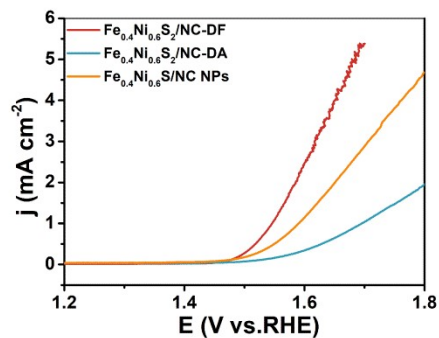
Samples	$R_s$ ( $\Omega$ )	$R_{ct}$ ( $\Omega$ )
$\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DF}$	9.68	18.62
$\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DA}$	3.27	271.4
$\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}/\text{NC NPs}$	7.33	63.86
$\text{RuO}_2$	11.78	42.93



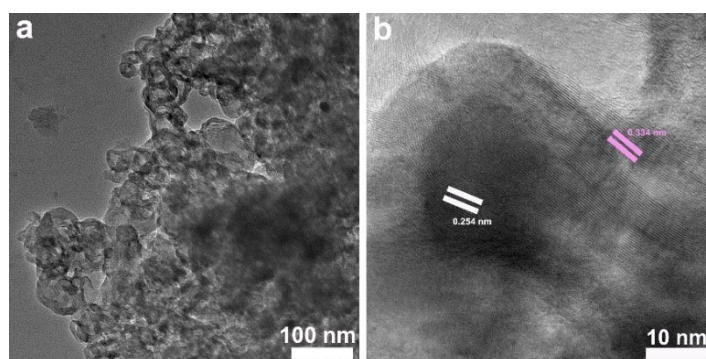
**Fig. S7.** TOF values of  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DF}$ ,  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DA}$  and  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}/\text{NC NPs}$  at different potentials.



**Fig. S8.** CV curves of a)  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DF}$ , b)  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DA}$ , and c)  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}/\text{NC NPs}$  in the potential range of -0.1-0 V (vs. Hg/HgO) at scanning rate of 20, 40, 60, 80, and 100  $\text{mV s}^{-1}$ , and d) linear fitting curve of  $\Delta j = 1/2(j_a - j_c)$  at different scanning speeds at -0.05 V (vs. Hg/HgO)



**Fig. S9.** The specific activity of  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DF}$ ,  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DA}$  and  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}/\text{NC NPs}$  normalized by ECSA.



**Fig. S10** a) TEM and b) HRTEM image of  $\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DF}$  after stability.

**Table S3** Comparison of OER performances over nickel sulfide-based electrocatalysts in 1M KOH electrolyte recently reported in the literatures. ( $\eta_{\text{OER}}$  are the overpotentials of OER at specific current density)

Catalysts	$\eta_{\text{OER}}$ (mV)	Tafel (mV $\text{dec}^{-1}$ )	Mass loading ( $\text{mg cm}^{-2}$ )	Substrate	Ref.
<b><math>\text{Fe}_{0.4}\text{Ni}_{0.6}\text{S}_2/\text{NC-DF}</math></b>	<b>282(10)</b>	<b>59.7</b>	<b>0.2</b>	<b>GC</b>	<b>This work</b>
Fe-NiS <sub>2</sub> /MoS <sub>2</sub>	280(10)	91.2	0.25	GC	1
Fe-NiS	270(10)	45.3	0.40	GC	2
NiS/NiS <sub>2</sub>	416(100)	156.5	2.4	Carbon cloth	3
FeNi <sub>2</sub> S <sub>4</sub>	273(10)	66	0.75	Nickel foam	4
MoS <sub>2</sub> /NiS <sub>2</sub>	278(10)	91.7	NA	Carbon cloth	5
NiS <sub>2</sub>	375(10)	138	-	Carbon cloth	6
N-CNTs@NiS <sub>2</sub> /Fe <sub>7</sub> S <sub>8</sub>	330(50)	51.49	0.22	GC	7
(Ni,Fe)S <sub>2</sub> /NiFe(3:1)-CNFs	263(10)	52	1.63	GC	8

N-doped carbon/NiS <sub>2</sub>	264(10)	51.3	2.4	Carbon cloth	9
FeS <sub>2</sub> /NiS <sub>2</sub>	280(50)	33	0.48	Carbon paper	10

## References

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