

Supporting information

Uptake of m-xylene and VOCs emission by mineral photocatalytic paints of indoor air building interest

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Table S1: Surface emission fluxes (molecules cm⁻² s⁻¹) of selected VOCs under UV irradiation for reference paint. The standard deviation (s.d) was calculated using twenty experimental data points.

Compounds	0 hour	s.d 0 hour	500 hours	s.d 500 hours	1000 hours	s.d 1000 hours
Formaldehyde	9.49·10 ⁹	9.70·10 ⁸	1.04·10 ¹⁰	7.76·10 ⁸	1.09·10 ¹⁰	8.31·10 ⁸
Methanol	4.26·10 ¹⁰	1.40·10 ⁹	2.48·10 ¹⁰	7.44·10 ⁸	2.53·10 ¹⁰	1.21·10 ⁹
Acetaldehyde	4.92·10 ⁹	3.02·10 ⁸	6.52·10 ⁹	2.60·10 ⁸	8.63·10 ⁹	2.93·10 ⁸
Formic Acid	1.04·10 ¹⁰	3.51·10 ⁸	1.50·10 ¹⁰	5.12·10 ⁸	1.49·10 ¹⁰	8.14·10 ⁸
Acrolein	6.82·10 ⁹	3.97·10 ⁸	6.14·10 ⁹	4.05·10 ⁸	6.07·10 ⁹	6.88·10 ⁸
Acetone + Propanal	1.04·10 ⁹	6.86·10 ⁷	2.34·10 ⁹	1.11·10 ⁸	2.99·10 ⁹	1.56·10 ⁸
Acetic Acid	4.05·10 ¹⁰	1.64·10 ⁹	4.58·10 ¹⁰	1.30·10 ⁹	2.42·10 ¹⁰	6.59·10 ⁸
MVK	4.14·10 ⁸	8.23·10 ⁷	5.90·10 ⁸	7.07·10 ⁷	4.65·10 ⁸	6.36·10 ⁷
Acrylic Acid	9.41·10 ⁸	1.07·10 ⁸	1.65·10 ⁹	1.58·10 ⁸	1.85·10 ⁹	2.45·10 ⁸
Propionic Acid	7.49·10 ⁹	5.45·10 ⁸	1.89·10 ⁹	1.69·10 ⁸	1.82·10 ⁹	1.39·10 ⁸
Benzene	5.40·10 ⁸	1.11·10 ⁸	7.71·10 ⁸	8.94·10 ⁷	4.65·10 ⁸	1.15·10 ⁸
Pentanal + Vinyl Acetate	1.73·10 ⁸	3.15·10 ⁷	3.00·10 ⁸	5.62·10 ⁷	3.12·10 ⁸	5.97·10 ⁷
Butanoic Acid	7.96·10 ⁸	1.02·10 ⁸	9.73·10 ⁸	9.80·10 ⁷	5.02·10 ⁸	6.11·10 ⁷
TMB + Ethyltoluene	2.77·10 ⁸	7.77·10 ⁷	4.94·10 ⁸	9.80·10 ⁷	5.35·10 ⁸	7.79·10 ⁷
Benzoic Acid	6.43·10 ⁷	3.54·10 ⁷	9.84·10 ⁷	3.73·10 ⁷	1.29·10 ⁸	2.89·10 ⁷
Octanal	8.53·10 ⁷	3.17·10 ⁷	1.80·10 ⁸	3.96·10 ⁷	2.78·10 ⁸	6.09·10 ⁷

Table S2: Surface emission fluxes (molecules cm⁻² s⁻¹) of selected VOCs under UV irradiation for paint nanoTiO₂. The standard deviation (s.d) was calculated from twenty points of the experiment.

Compounds	0 hour	s.d 0 hour	500 hours	s.d 500 hours	1000 hours	s.d 1000 hours
Formaldehyde	1.15·10 ¹⁰	8.05·10 ⁸	5.64·10 ¹⁰	2.62·10 ⁹	5.29·10 ¹⁰	2.01·10 ⁹
Methanol	4.19·10 ¹⁰	1.65·10 ⁹	4.11·10 ¹⁰	1.36·10 ⁹	3.51·10 ¹⁰	1.43·10 ⁹
Acetaldehyde	6.86·10 ⁹	3.24·10 ⁸	4.52·10 ¹⁰	1.09·10 ⁹	5.28·10 ¹⁰	1.30·10 ⁹
Formic Acid	1.03·10 ¹⁰	5.27·10 ⁸	2.05·10 ¹⁰	7.30·10 ⁸	1.95·10 ¹⁰	7.46·10 ⁸
Acrolein	6.56·10 ⁹	5.37·10 ⁸	1.05·10 ¹⁰	3.88·10 ⁸	8.82·10 ⁹	4.30·10 ⁸
Acetone + Propanal	1.34·10 ⁹	1.10·10 ⁸	6.62·10 ⁹	2.63·10 ⁸	4.57·10 ⁹	2.15·10 ⁸
Acetic Acid	5.07·10 ¹⁰	3.70·10 ⁹	7.54·10 ¹⁰	1.74·10 ⁹	3.99·10 ¹⁰	1.06·10 ⁹
MVK	4.44·10 ⁸	9.71·10 ⁷	1.47·10 ⁹	1.19·10 ⁸	1.16·10 ⁹	1.87·10 ⁸
Acrylic Acid	1.13·10 ⁹	1.53·10 ⁸	5.81·10 ⁹	4.10·10 ⁸	3.37·10 ⁹	2.07·10 ⁸
Propionic Acid	9.07·10 ⁹	6.09·10 ⁸	9.19·10 ⁹	5.94·10 ⁸	9.01·10 ⁹	7.65·10 ⁸
Benzene	7.40·10 ⁸	1.33·10 ⁸	1.28·10 ⁹	1.67·10 ⁸	8.93·10 ⁸	1.68·10 ⁸
Pentanal + Vinyl Acetate	1.87·10 ⁸	4.50·10 ⁷	1.41·10 ⁹	1.11·10 ⁸	8.91·10 ⁸	1.11·10 ⁸
Butanoic Acid	8.42·10 ⁸	1.22·10 ⁸	1.90·10 ⁹	1.73·10 ⁸	7.38·10 ⁸	8.07·10 ⁷
TMB + Ethyltoluene	2.87·10 ⁸	7.84·10 ⁷	8.09·10 ⁸	1.48·10 ⁸	6.30·10 ⁸	1.45·10 ⁸
Benzoic Acid	4.64·10 ⁷	1.37·10 ⁷	1.73·10 ⁸	4.79·10 ⁷	1.42·10 ⁸	4.73·10 ⁷
Octanal	1.19·10 ⁸	4.39·10 ⁷	9.44·10 ⁸	1.63·10 ⁸	1.30·10 ⁹	1.69·10 ⁸

Table S3: Surface emission fluxes (molecules cm⁻² s⁻¹) of selected VOCs under UV irradiation for paint PEG 3350. The standard deviation (s.d) was calculated using twenty experimental data points.

Compounds	0 hour	s.d 0 hour	500 hours	s.d 500 hours	1000 hours	s.d 1000 hours
Formaldehyde	9.61·10 ⁹	8.08·10 ⁸	4.46·10 ¹⁰	1.73·10 ⁹	4.73·10 ¹⁰	2.44·10 ⁹
Methanol	4.37·10 ¹⁰	1.06·10 ⁹	2.99·10 ¹⁰	1.32·10 ⁹	3.73·10 ¹⁰	1.37·10 ⁹
Acetaldehyde	5.80·10 ⁹	3.05·10 ⁸	2.80·10 ¹⁰	9.20·10 ⁸	4.95·10 ¹⁰	1.94·10 ⁹
Formic Acid	1.08·10 ¹⁰	4.43·10 ⁸	1.41·10 ¹⁰	5.63·10 ⁸	1.83·10 ¹⁰	4.99·10 ⁸
Acrolein	6.49·10 ⁹	4.04·10 ⁸	6.90·10 ⁹	4.57·10 ⁸	9.33·10 ⁹	4.42·10 ⁸
Acetone + Propanal	1.25·10 ⁹	8.68·10 ⁷	5.78·10 ⁹	2.45·10 ⁸	3.17·10 ⁹	1.72·10 ⁸
Acetic Acid	5.93·10 ¹⁰	1.66·10 ⁹	5.50·10 ¹⁰	1.36·10 ⁹	4.45·10 ¹⁰	1.15·10 ⁹
MVK	4.95·10 ⁸	8.04·10 ⁷	1.08·10 ⁹	1.27·10 ⁸	1.34·10 ⁹	1.72·10 ⁸
Acrylic Acid	1.14·10 ⁹	1.71·10 ⁸	3.78·10 ⁹	2.80·10 ⁸	3.34·10 ⁹	2.87·10 ⁸
Propionic Acid	8.12·10 ⁹	3.31·10 ⁸	6.01·10 ⁹	3.70·10 ⁸	1.80·10 ¹⁰	1.53·10 ⁹
Benzene	8.26·10 ⁸	7.39·10 ⁷	1.02·10 ⁹	1.29·10 ⁸	1.00·10 ⁹	1.35·10 ⁸
Pentanal + Vinyl Acetate	1.80·10 ⁸	4.50·10 ⁷	8.16·10 ⁸	8.54·10 ⁷	7.54·10 ⁸	7.75·10 ⁷
Butanoic Acid	7.97·10 ⁸	9.51·10 ⁷	1.26·10 ⁹	1.29·10 ⁸	7.24·10 ⁸	1.22·10 ⁸
TMB + Ethyltoluene	2.95·10 ⁸	7.94·10 ⁷	6.68·10 ⁸	1.21·10 ⁸	5.60·10 ⁸	1.18·10 ⁸
Benzoic Acid	3.82·10 ⁷	1.95·10 ⁷	1.09·10 ⁸	2.61·10 ⁸	1.24·10 ⁸	3.86·10 ⁷
Octanal	1.11·10 ⁸	4.41·10 ⁷	6.64·10 ⁸	1.30·10 ⁸	1.23·10 ⁹	1.72·10 ⁸

Table S4: Surface emission fluxes (molecules cm⁻² s⁻¹) of selected VOCs under UV irradiation for paint CNC. The standard deviation (s.d) was calculated using twenty experimental data points.

Compounds	0 hour	s.d 0 hour	500 hours	s.d 500 hours	1000 hours	s.d 1000 hours
Formaldehyde	$8.81 \cdot 10^9$	$8.18 \cdot 10^8$	$1.09 \cdot 10^{10}$	$8.23 \cdot 10^8$	$1.18 \cdot 10^{10}$	$1.15 \cdot 10^9$
Methanol	$5.93 \cdot 10^{10}$	$2.65 \cdot 10^9$	$2.85 \cdot 10^{10}$	$1.27 \cdot 10^9$	$2.93 \cdot 10^{10}$	$9.64 \cdot 10^8$
Acetaldehyde	$5.57 \cdot 10^9$	$3.10 \cdot 10^8$	$6.42 \cdot 10^9$	$3.35 \cdot 10^8$	$1.11 \cdot 10^{10}$	$3.25 \cdot 10^8$
Formic Acid	$1.04 \cdot 10^{10}$	$6.40 \cdot 10^8$	$1.36 \cdot 10^{10}$	$5.30 \cdot 10^8$	$2.09 \cdot 10^{10}$	$7.11 \cdot 10^8$
Acrolein	$6.14 \cdot 10^9$	$3.35 \cdot 10^8$	$5.88 \cdot 10^9$	$3.85 \cdot 10^8$	$8.86 \cdot 10^9$	$3.56 \cdot 10^8$
Acetone + Propanal	$2.50 \cdot 10^9$	$2.53 \cdot 10^8$	$1.59 \cdot 10^9$	$9.57 \cdot 10^7$	$2.60 \cdot 10^9$	$1.29 \cdot 10^8$
Acetic Acid	$3.89 \cdot 10^{10}$	$2.06 \cdot 10^9$	$3.09 \cdot 10^{10}$	$9.45 \cdot 10^8$	$4.07 \cdot 10^{10}$	$1.03 \cdot 10^9$
MVK	$3.79 \cdot 10^8$	$7.50 \cdot 10^7$	$5.27 \cdot 10^8$	$7.48 \cdot 10^7$	$8.80 \cdot 10^8$	$9.72 \cdot 10^7$
Acrylic Acid	$1.17 \cdot 10^9$	$1.61 \cdot 10^8$	$1.56 \cdot 10^9$	$1.17 \cdot 10^8$	$3.56 \cdot 10^9$	$2.11 \cdot 10^8$
Propionic Acid	$5.18 \cdot 10^9$	$4.63 \cdot 10^8$	$1.20 \cdot 10^9$	$1.79 \cdot 10^8$	$2.50 \cdot 10^9$	$2.17 \cdot 10^8$
Benzene	$6.76 \cdot 10^8$	$1.65 \cdot 10^8$	$5.07 \cdot 10^8$	$1.21 \cdot 10^8$	$8.28 \cdot 10^8$	$1.10 \cdot 10^8$
Pentanal + Vinyl Acetate	$1.99 \cdot 10^8$	$5.18 \cdot 10^7$	$3.12 \cdot 10^8$	$4.59 \cdot 10^7$	$1.42 \cdot 10^9$	$1.26 \cdot 10^8$
Butanoic Acid	$6.23 \cdot 10^8$	$1.17 \cdot 10^8$	$8.15 \cdot 10^8$	$1.32 \cdot 10^8$	$1.65 \cdot 10^9$	$1.35 \cdot 10^8$
TMB + Ethyltoluene	$3.23 \cdot 10^8$	$1.11 \cdot 10^8$	$4.89 \cdot 10^8$	$1.06 \cdot 10^8$	$5.50 \cdot 10^8$	$8.74 \cdot 10^7$
Benzoic Acid	$4.24 \cdot 10^7$	$2.92 \cdot 10^7$	$7.07 \cdot 10^7$	$3.32 \cdot 10^7$	$1.75 \cdot 10^8$	$5.12 \cdot 10^7$
Octanal	$9.34 \cdot 10^7$	$3.76 \cdot 10^7$	$2.27 \cdot 10^8$	$7.18 \cdot 10^7$	$3.30 \cdot 10^8$	$8.61 \cdot 10^7$

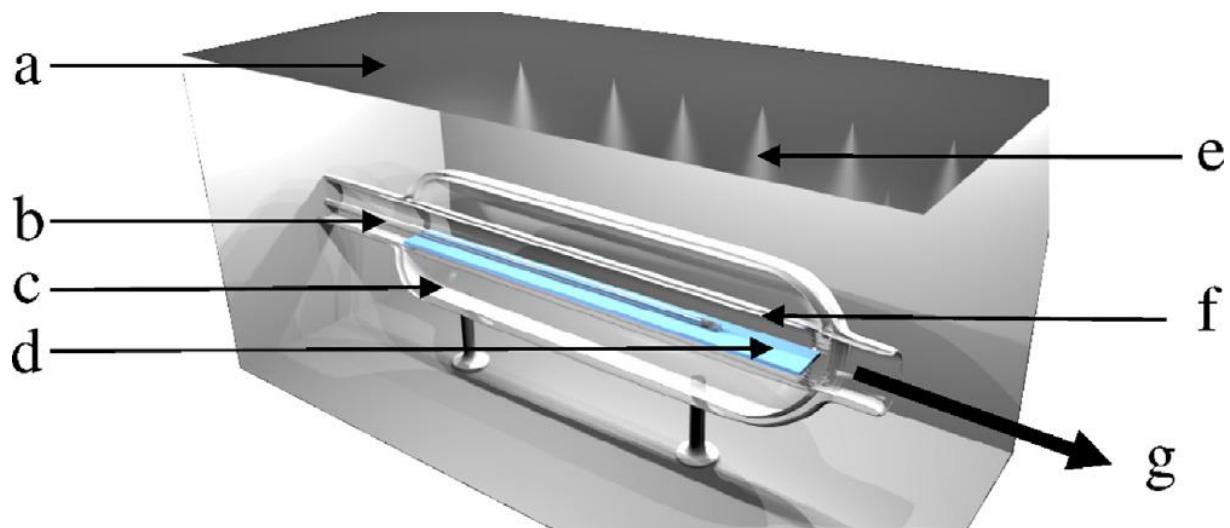


Fig. S1. 3D-sketch of (a) stainless steel box, (b) mobile injector, (c) double wall thermostated glass envelope, (d) glass plate covered with the paint, (e) UV lamps (f) the flow tube photo-reactor, and (g) exit of the flow gas to the PTR-MS analyzers.