

Supplementary Material

The effect and mechanism of a microbial agent used for corrosion control in circulating cooling water

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Supplementary figures

1. Fig. S1

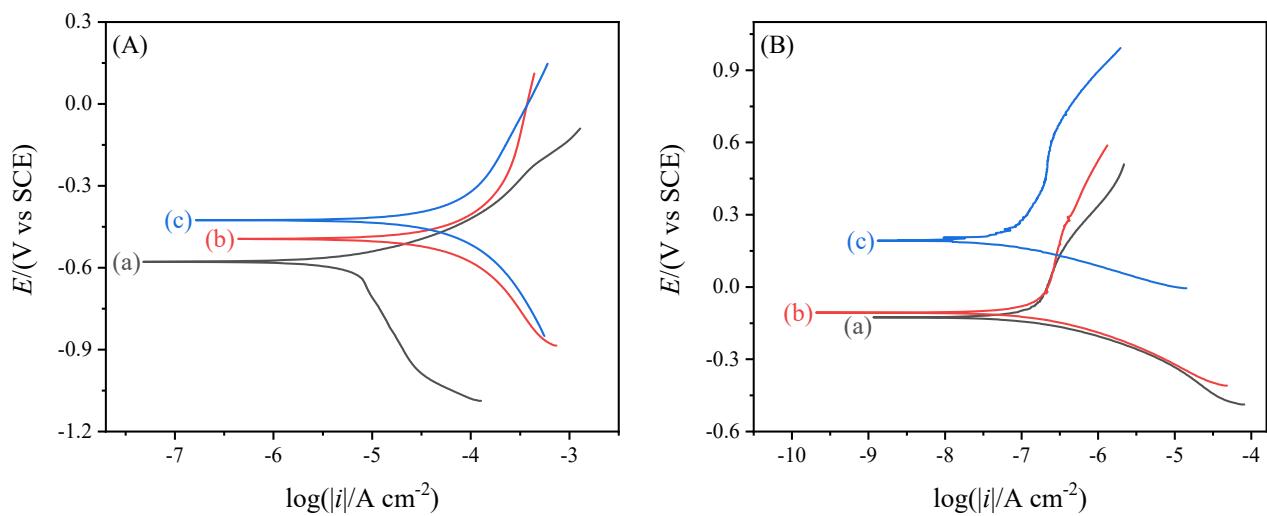


Fig. S1 Potentiodynamic polarization curves of different electrodes. (A) Q235B electrode; (B) 316L electrode; (a) new electrode at 0 d; (b) without microbial agent after 30 d; (c) with microbial agent after 30 d.

2. Fig. S2

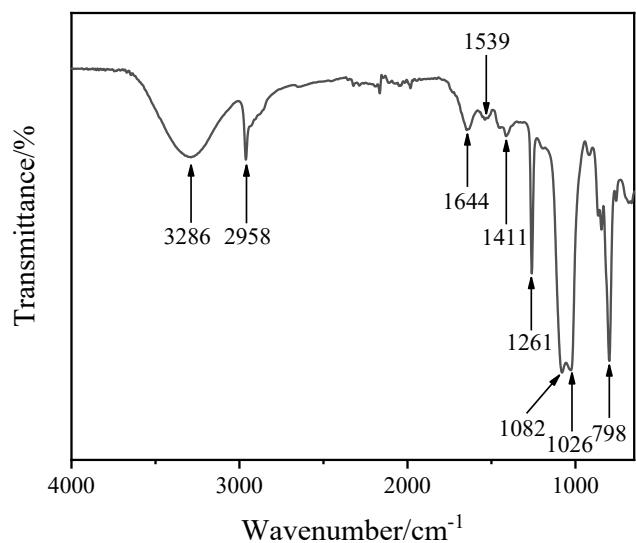


Fig. S2 FTIR spectrum of the biofilm on the coupon surface.

Supplementary tables

1. Table S1

Table S1 Water quality index of circulating cooling water.

Index	Value	Index	Value
pH/(25°C)	7.61-7.80	NH ₃ -N/(mg L ⁻¹)	0.17-0.28
Electric conductivity/(ms cm ⁻¹)	0.63-0.67	COD/(mg L ⁻¹)	12.36-19.30
Suspended solids/(mg L ⁻¹)	2.50-6.20	TP/(mg L ⁻¹)	0.33-0.51
Turbidity/(NTU)	0.21-1.47	Ca ²⁺ /(mg L ⁻¹)	47.34-65.33
Total hardness/(mmol L ⁻¹)	1.57-2.65	Cl ⁻ /(mg L ⁻¹)	44.75-60.73
Total alkalinity/(mmol L ⁻¹)	1.54-2.49	SO ₄ ²⁻ /(mg L ⁻¹)	42.16-64.29

2. Table S2

Table S2 Corrosion rate of the coupon under different conditions.

Type	Corrosion weight loss/(g)	Corrosion rate/(mm a ⁻¹)	Average corrosion rate/(mm a ⁻¹)
Q235B without microbial agent	0.5794	0.7319	0.7261 ± 0.0167
	0.5852	0.7392	
	0.5599	0.7072	
Q235B with microbial agent	0.5230	0.6606	0.6606 ± 0.0098
	0.5307	0.6704	
	0.5152	0.6508	
316L without microbial agent	0.0005	0.0006	0.0006 ± 0.0001
	0.0004	0.0005	
	0.0005	0.0006	
316L with microbial agent	0.0003	0.0004	0.0005 ± 0.0002
	0.0006	0.0007	
	0.0004	0.0005	

3. Table S3

Table S3 Surface roughness parameters of the coupon at different times.

Time/(d)	$S_a/(\mu\text{m})$	$S_q/(\mu\text{m})$
0	0.27	0.35
5	0.35	0.53
10	0.45	0.75
15	0.53	0.89
20	0.76	1.28
25	0.72	0.94
30	0.78	1.20