

Electronic Supplementary Information (ESI)

Ozonolysis–oxidation-driven top-down strategy for target preparation of ultrathin 2D metal–organic framework monolayers

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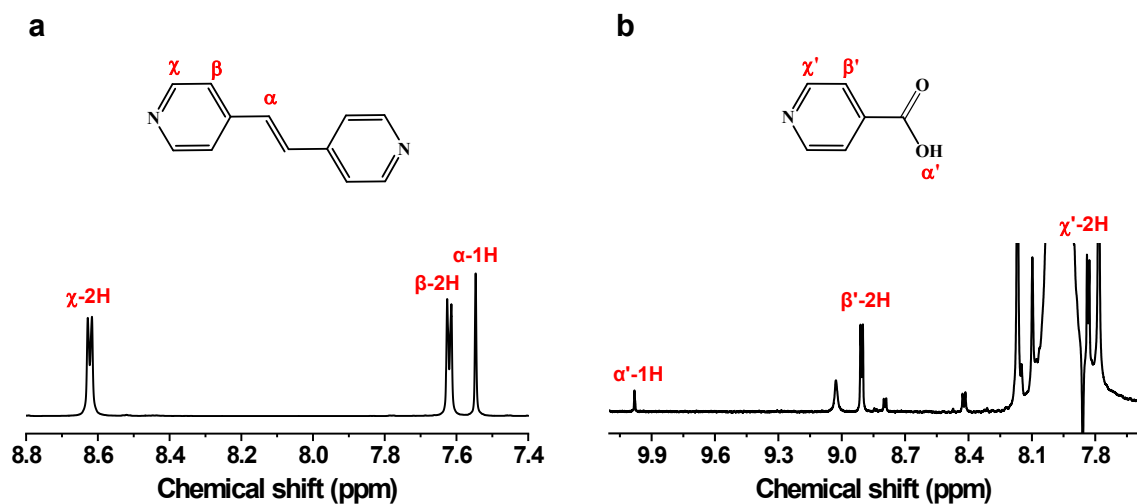


Fig. S1 (a) The ¹H NMR spectra of the *trans*-1, 2-bis(4pyridyl)ethylene; (b) The ¹H NMR spectra of the *trans*-1, 2-bis(4pyridyl)ethylene after ozone treatment in DMF at 0 °C.

The ¹H nuclear magnetic resonance (NMR) spectroscopy is employed to characterize the structure of the *trans*-1, 2-bis(4pyridyl)ethylene before and after ozone treatment. It can be seen from Fig. S2b that the chemical shifts at $\delta = 10.12$ ppm, 8.91 ppm, 8.90 ppm, 7.84 ppm and 7.83 ppm, which are consistent with the theoretical ¹H NMR spectra of the isonicotinic acid. These results confirm the breakage of the olefin group in the *trans*-1, 2-bis(4pyridyl)ethylene.

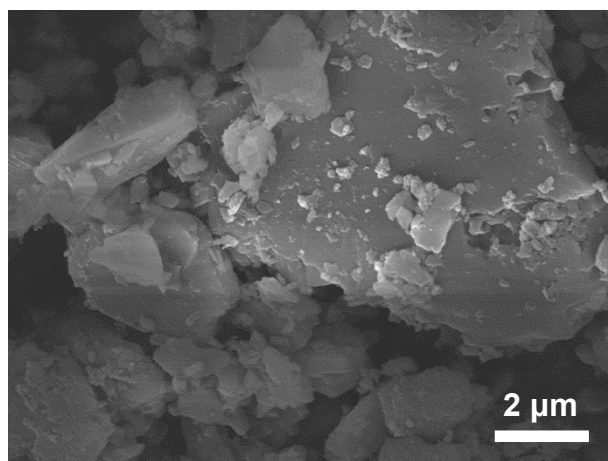


Fig. S2 SEM image of the 3D pillar-layered Co-MOF precursor.

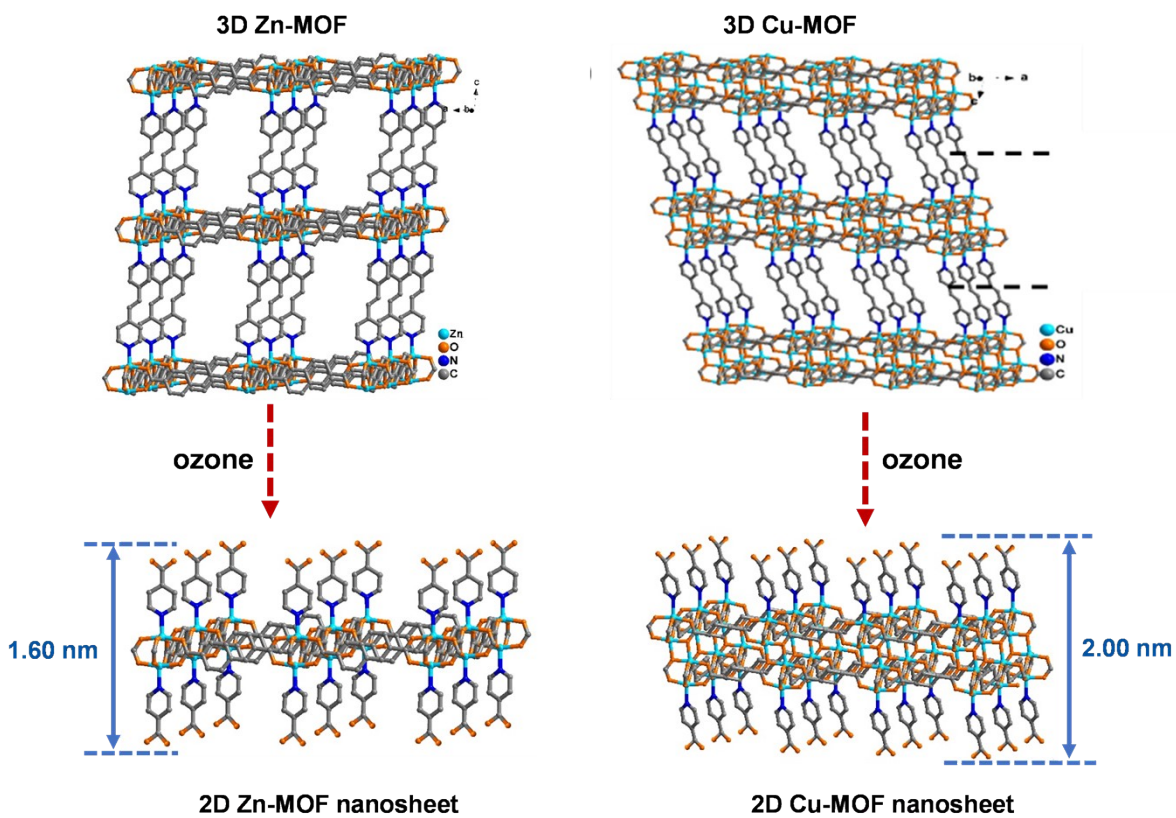


Fig. S3 The 3D pillar-layered structure of the Zn-MOF and Cu-MOF and the corresponding structure of the 2D Zn-MOF nanosheet and 2D Zn-MOF nanosheet.

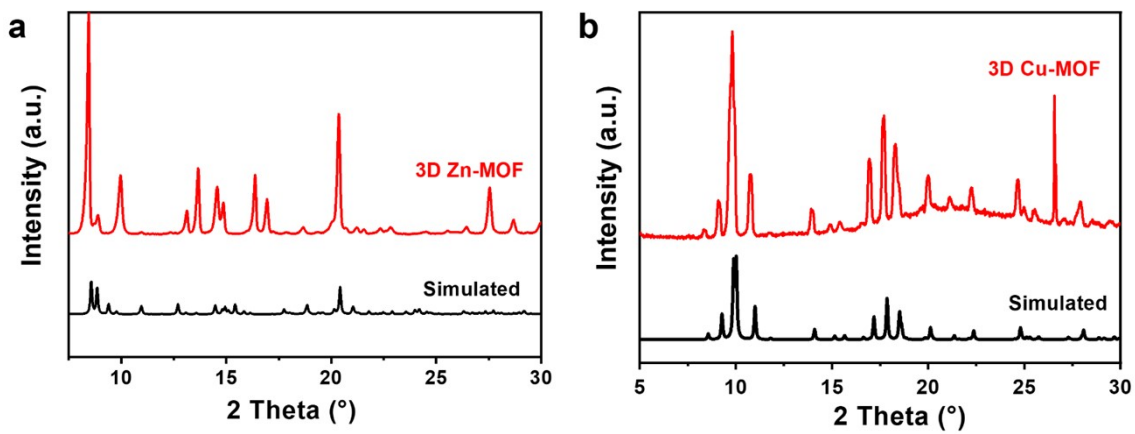


Fig. S4 The simulated XRD patterns of the 3D Zn-MOF and 3D Cu-MOF

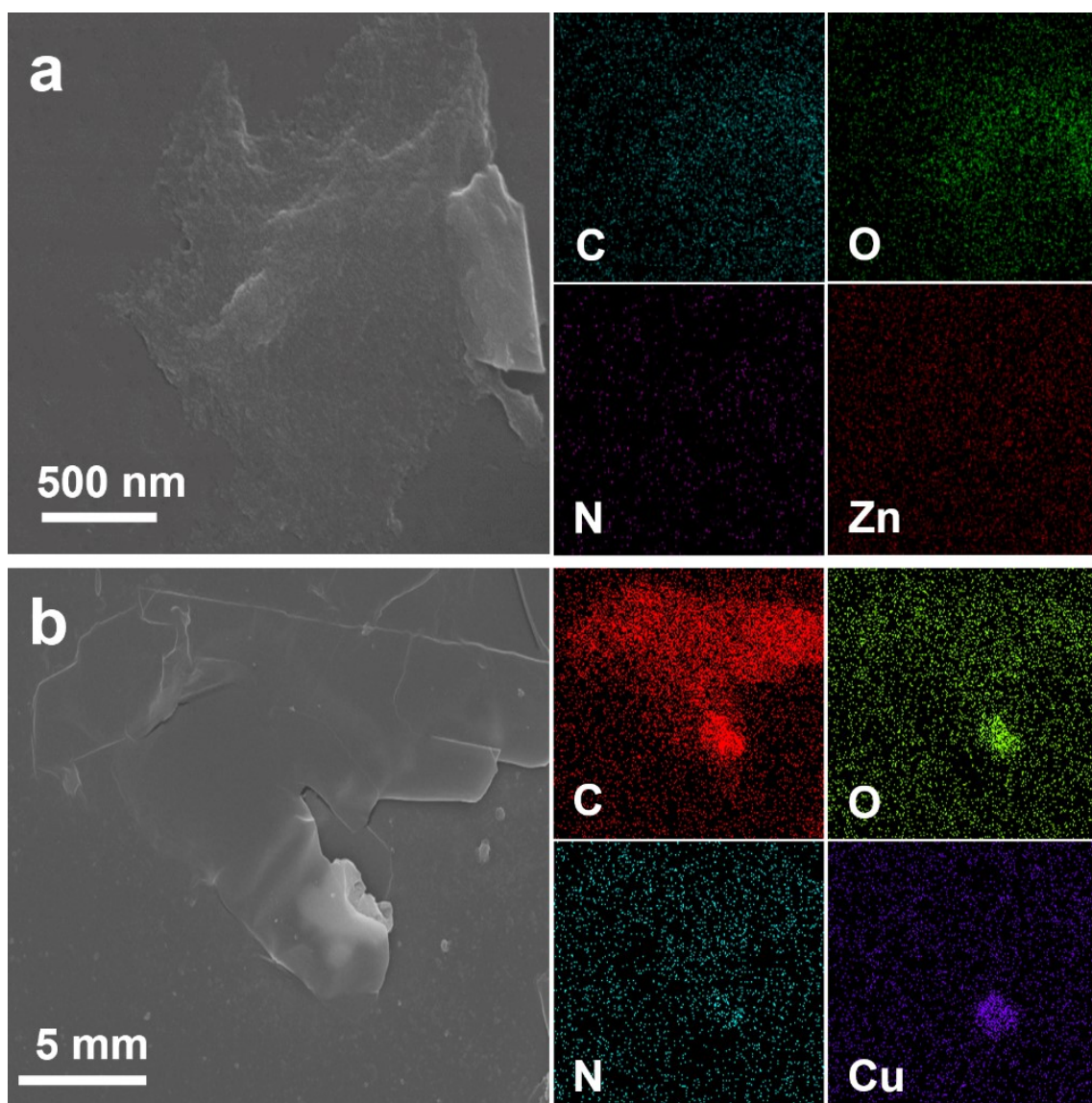


Fig. S5 (a) SEM images of 2D Zn-MOF and corresponding elemental maps; b) SEM images of 2D Cu-MOF and corresponding elemental maps.

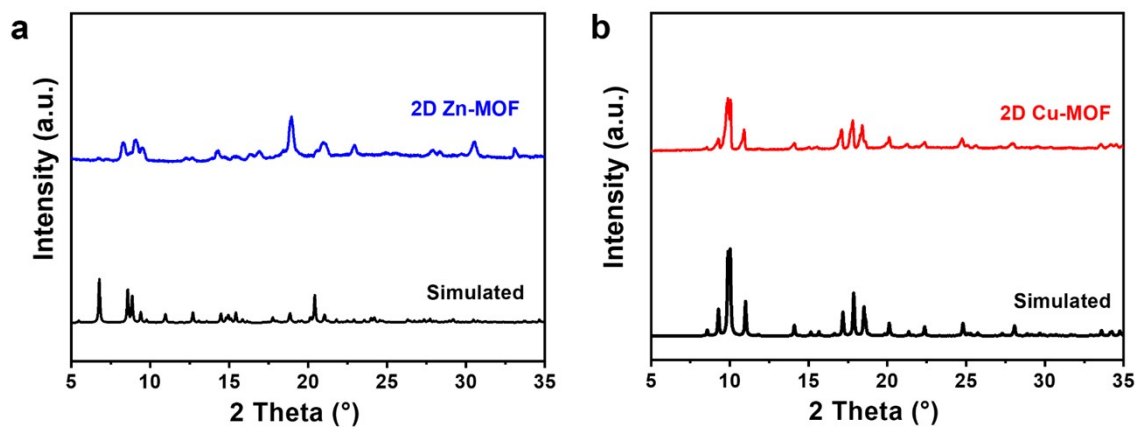


Fig. S6 The simulated XRD patterns of the 2D Zn-MOF and 2D Cu-MOF.

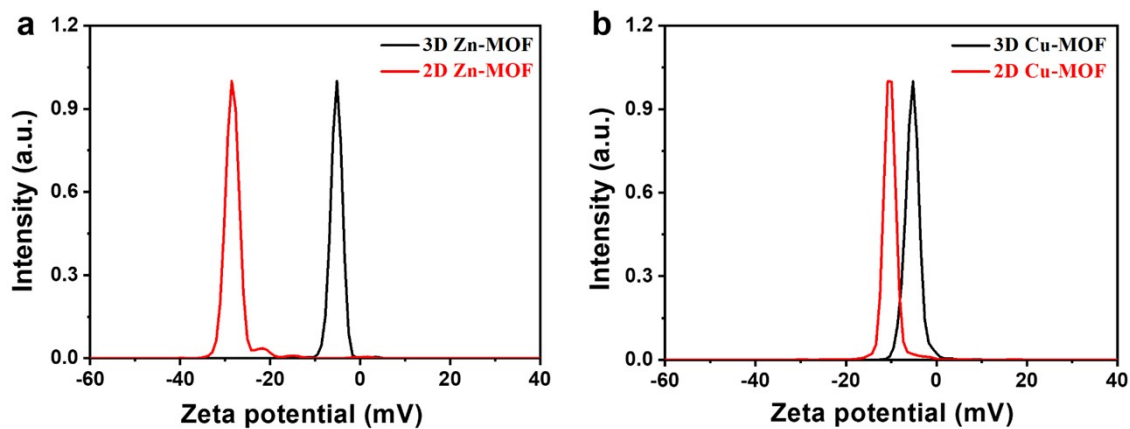


Fig. S7 Zeta potential of the (a) 3D Zn-MOF precursor, 2D Zn-MOF nanosheets and the (b) 3D Cu-MOF precursor, 2DCu-MOF nanosheet.