

## Supplementary Information

### Submicron-Thick Single Anion-Conducting Polymer Electrolytes

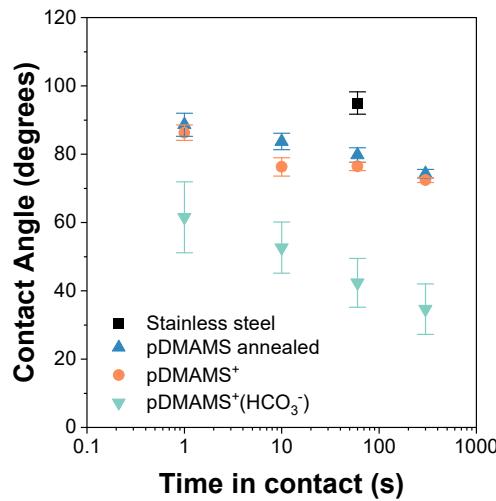
Hunter O. Ford,<sup>1</sup> Brian L. Chaloux,<sup>2</sup> Youngchan Kim,<sup>3</sup> Jeffrey W. Long,<sup>2</sup> Debra R. Rolison,<sup>2\*</sup> and Megan B. Sassin<sup>2\*</sup>

\*Corresponding Authors: E-mail: [megan.sassin@nrl.navy.mil](mailto:megan.sassin@nrl.navy.mil); [debra.r.rolison.civ@us.navy.mil](mailto:debra.r.rolison.civ@us.navy.mil)

<sup>1</sup> NRL–NRC Postdoctoral Associate in the Chemistry Division, U.S. Naval Research Laboratory, Washington, DC 20375, USA

<sup>2</sup> Chemistry Division, U.S. Naval Research Laboratory, Washington, DC 20375, USA

<sup>3</sup> Materials Science & Technology Division, U.S. Naval Research Laboratory, Washington, DC 20375, USA



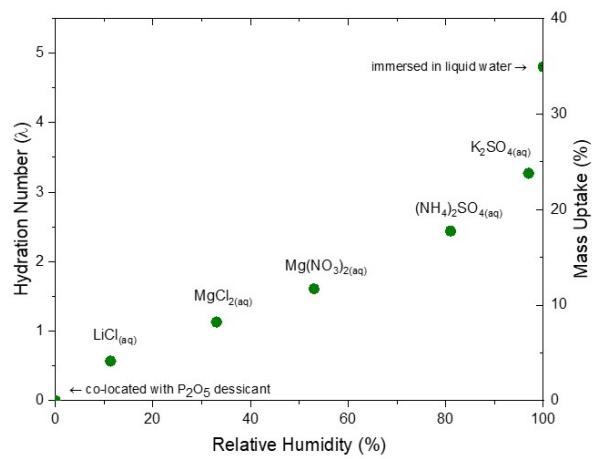
**Fig. S1** Contact-angle measurements for various iCVD-derived pDMAMS films.

**Table S1** Fit parameters from dry Nyquist equivalent-circuit models.

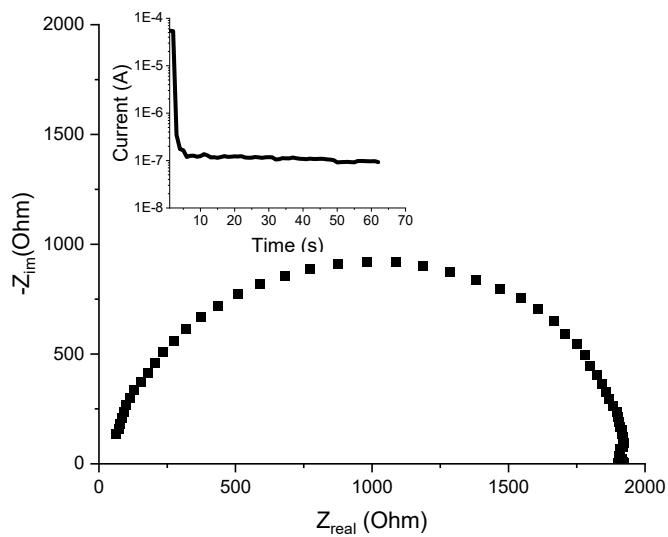
Sample ID	R1	R2	Q <sub>1</sub>	Q <sub>1_a</sub>
Br_1	67	1128	1.0×10 <sup>-10</sup>	1
Br_2	52	1291	1.2×10 <sup>-10</sup>	1
Br_3	103	1070	0.7×10 <sup>-10</sup>	1
Br_4	22	1414	2.8×10 <sup>-10</sup>	1
Br_5	38	885	1.4×10 <sup>-10</sup>	1
<b>Average</b>	<b>56</b>	<b>1158</b>	<b>1.4×10<sup>-10</sup></b>	<b>1</b>
HCO <sub>3</sub> _1	62	97	4.5×10 <sup>-9</sup>	0.9
HCO <sub>3</sub> _2	93	11	5.3×10 <sup>-10</sup>	1
HCO <sub>3</sub> _3	87	56	1.2×10 <sup>-9</sup>	1
HCO <sub>3</sub> _4	119	78	2.2×10 <sup>-8</sup>	0.8
HCO <sub>3</sub> _5	58	145	6.7×10 <sup>-10</sup>	1
HCO <sub>3</sub> _6	115	48	1.1×10 <sup>-7</sup>	0.75
<b>Average</b>	<b>84</b>	<b>77</b>	<b>5.8×10<sup>-9</sup></b>	<b>0.94</b>
OH_1	76	56	1.7×10 <sup>-8</sup>	0.85
OH_2	76	37	2.5×10 <sup>-8</sup>	0.85
OH_3	67	66	1.7×10 <sup>-8</sup>	0.85
OH_4	75	42	1.1×10 <sup>-8</sup>	0.9
OH_5	74	56	7.6×10 <sup>-9</sup>	0.9
OH_6	58	89	6.0×10 <sup>-9</sup>	0.9
<b>Average</b>	<b>74</b>	<b>51</b>	<b>1.6×10<sup>-8</sup></b>	<b>0.87</b>

**Table S2** Fit parameters from water-swelled Nyquist equivalent-circuit models.

Sample ID	R1	R2	Q <sub>1</sub>	Q <sub>1_a</sub>	Q <sub>2</sub>	Q <sub>2_a</sub>
Br_1	24	23	7.3×10 <sup>-9</sup>	1	5.4×10 <sup>-5</sup>	0.68
Br_2	15	68	4.7×10 <sup>-9</sup>	1	4.2×10 <sup>-5</sup>	0.70
Br_3	21	34	6.8×10 <sup>-9</sup>	1	6.2×10 <sup>-5</sup>	0.70
Br_4	29	23	8.7×10 <sup>-9</sup>	1	2.0×10 <sup>-4</sup>	0.59
<b>average</b>	<b>22</b>	<b>37</b>	<b>6.9×10<sup>-9</sup></b>	<b>1</b>	<b>9.2×10<sup>-5</sup></b>	<b>0.67</b>
HCO <sub>3</sub> _1	20	7	5.4×10 <sup>-7</sup>	0.95	9.2×10 <sup>-5</sup>	0.67
HCO <sub>3</sub> _2	19	11	1.8×10 <sup>-7</sup>	1	1.0×10 <sup>-4</sup>	0.66
HCO <sub>3</sub> _3	30	11	2.3×10 <sup>-7</sup>	0.97	1.0×10 <sup>-4</sup>	0.67
<b>average</b>	<b>23</b>	<b>10</b>	<b>3.2×10<sup>-7</sup></b>	<b>0.97</b>	<b>9.8×10<sup>-5</sup></b>	<b>0.67</b>
OH_1	19	26	7.3×10 <sup>-9</sup>	1	8.3×10 <sup>-5</sup>	0.72
OH_2	0	17	1.1×10 <sup>-5</sup>	0.45	1.9×10 <sup>-4</sup>	0.63
OH_3	21	14	6.5×10 <sup>-6</sup>	0.8	4.9×10 <sup>-5</sup>	0.71
OH_4	16	8	2.2×10 <sup>-6</sup>	0.91	5.5×10 <sup>-5</sup>	0.71
OH_5	13	9	5.5×10 <sup>-6</sup>	0.85	5.2×10 <sup>-5</sup>	0.74
<b>average</b>	<b>14</b>	<b>15</b>	<b>5.0×10<sup>-6</sup></b>	<b>0.8</b>	<b>8.6×10<sup>-5</sup></b>	<b>0.70</b>



**Fig. S2** Water uptake in hydration number ( $\lambda$ ) and percent mass (of the dry membrane mass) for pDMAMS<sup>+</sup>(Br<sup>-</sup>) equilibrated over  $P_2O_5$  (0% RH), various salt solutions (11–97% RH), and in liquid deionized water (100% RH).



**Fig. S3** EIS spectra of pDMAMS<sup>+</sup>(Br<sup>-</sup>) showing adequate electrode-electrolyte contact. Inset: current response of potentiostatic hold at 0.1 V.