

Supporting Information

Closing the Loop: Polyimine Thermosets from Furfural Derived Bioresources

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1. Synthesis, characterization and stability of OBMF

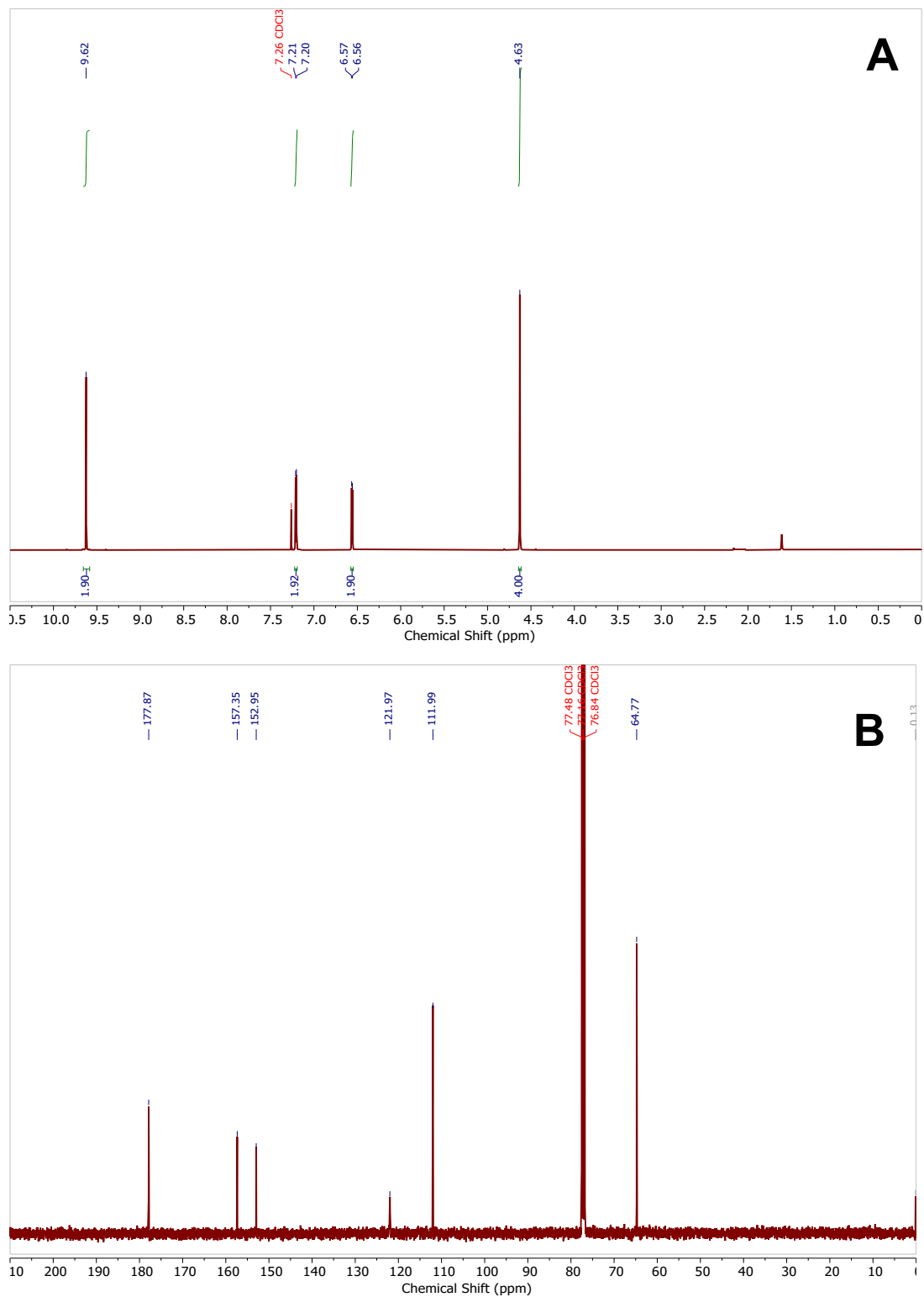


Figure S1. ^1H NMR (A) and ^{13}C NMR (B) spectra of OBMF recorded in CDCl_3 .

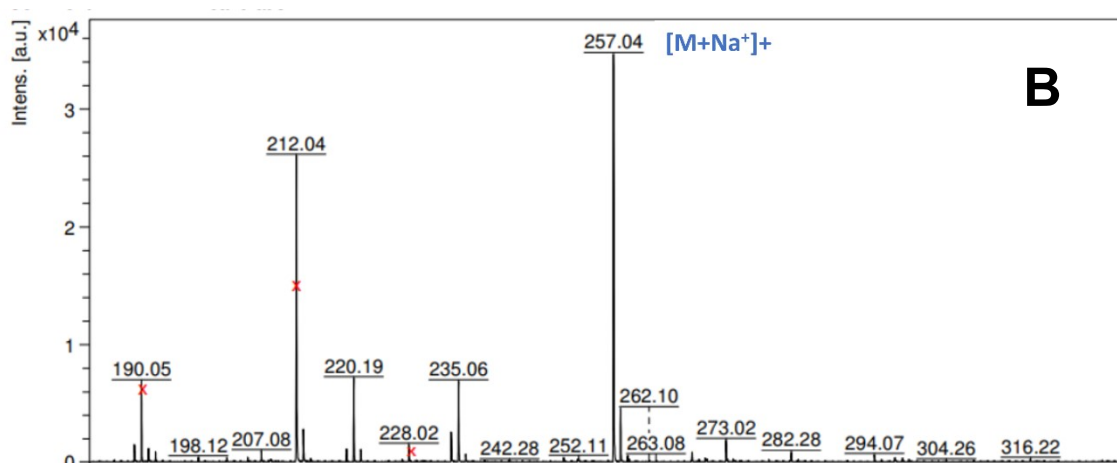
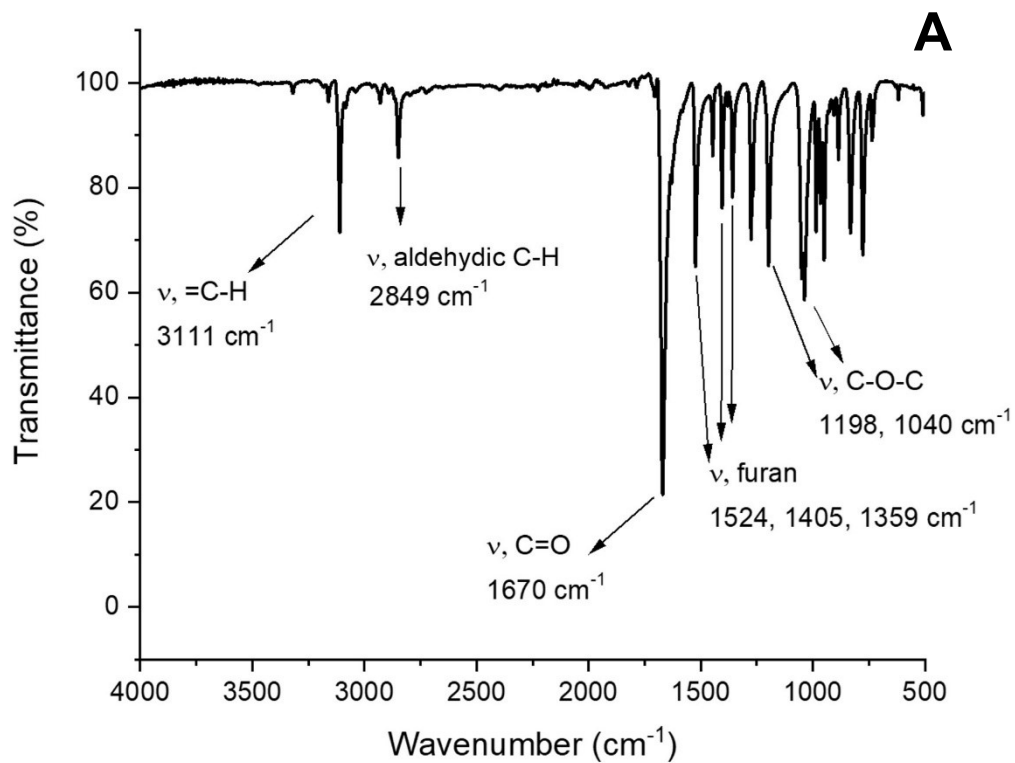


Figure S2. FTIR (A) and MALDI-TOF (B) spectra of OBMF. α -Cyano-4-hydroxycinnamic acid (CHCA) was used as matrix in MALDI measurement. The peaks assigned with x corresponds to the matrix peaks.

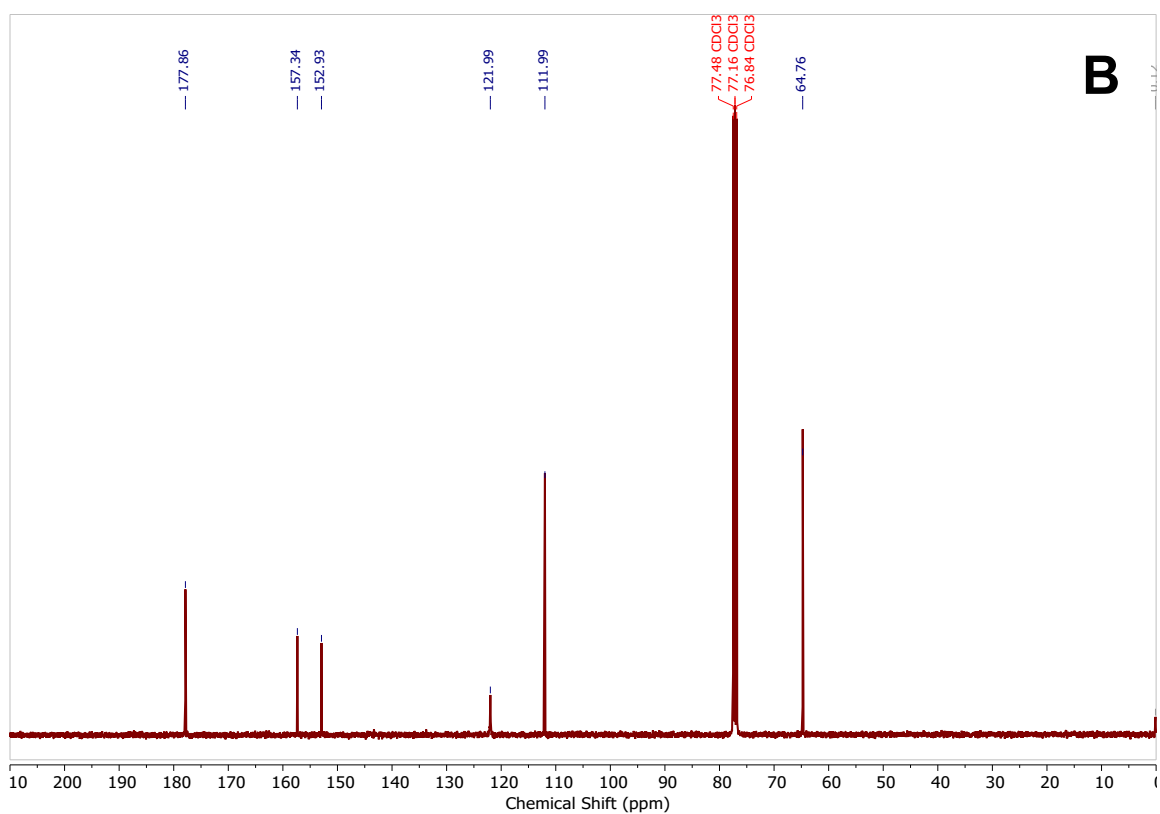
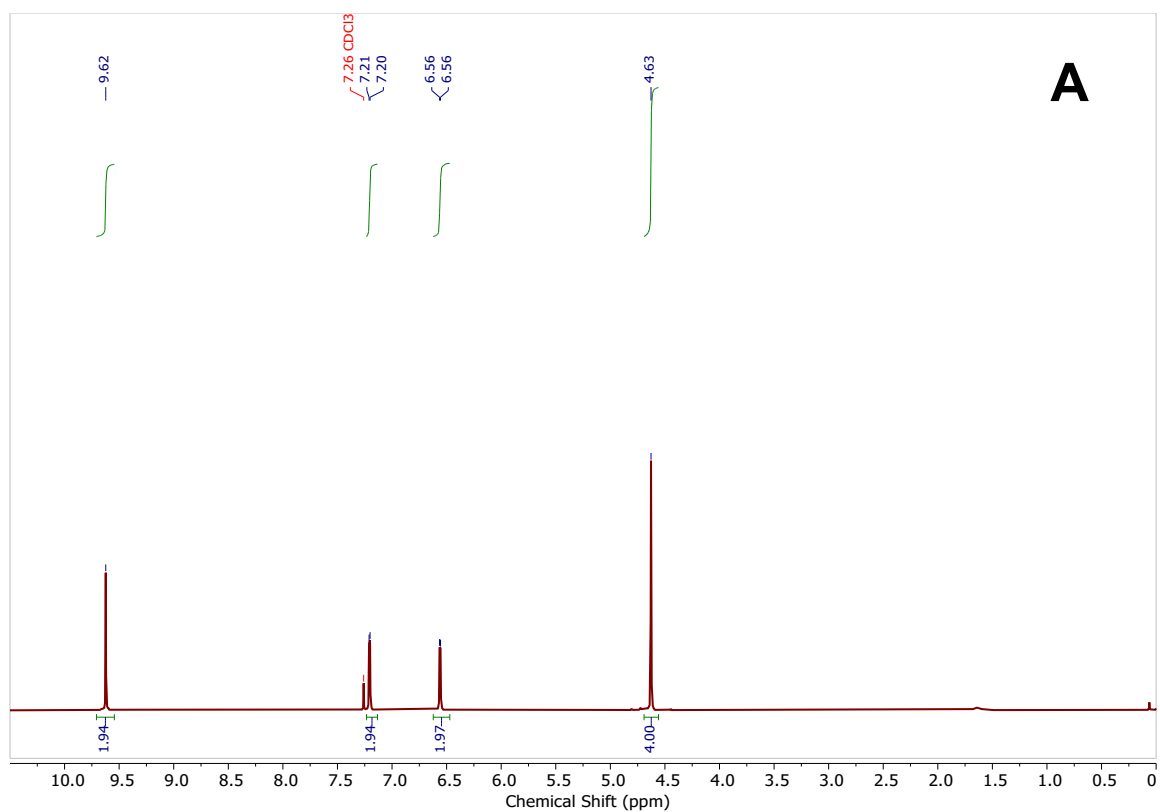


Figure S3. ¹H NMR (A) and ¹³C NMR (B) spectra of OBMF recorded in CDCl₃, after keeping the solids 1-year under ambient conditions.

Table S1. Isolated yields of OBMF at different concentrations of HMF.

	Concentration of HMF (mol/L)	Isolated Yield (%)
Entry 1	0.4	64
Entry 2	0.2	67
Entry 3	0.1	79

2. Synthesis and structural characterization of the polyimine networks

Table S2. Recipes for the preparation of polyimine networks.

	OBMF	Diamine	Triamine
P(Flm)-TREN	6 mmol	0	TREN 4 mmol
CFRP(Flm)-TREN	7.8 mmol	0	TREN 5.2 mmol
P(Flm)-PD	6 mmol	PD 3 mmol	TREN 2 mmol
P(Flm)-MD	6 mmol	MD 3 mmol	TREN 2 mmol
P(Flm)-FD	6 mmol	FD 3 mmol	TREN 2 mmol
P(Flm)-Pri	6 mmol	Pri 3 mmol	TREN 2 mmol

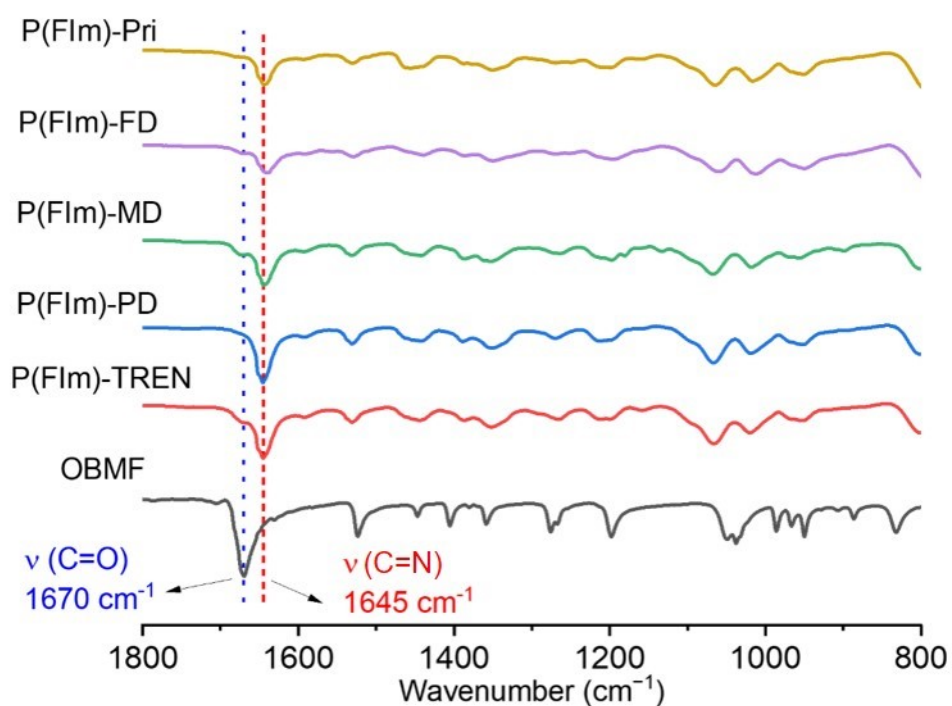


Figure S4. Characterization of the polyimine networks by FTIR.

3. Stability of polyimine networks in organic solvents and water

Table S3. Swelling measurements of **P(Fim)-TREN** in different solvents at room temperature.

Solvents	MeOH	EtOH	Acetone	IPA	THF	n-hexane	H ₂ O
Swelling ratio (%)	82.6	49.2	16.2	0.8	22.5	0	37.6
Gel content (%)	88.1	95.9	99.8	99.2	99.7	98.3	98.3

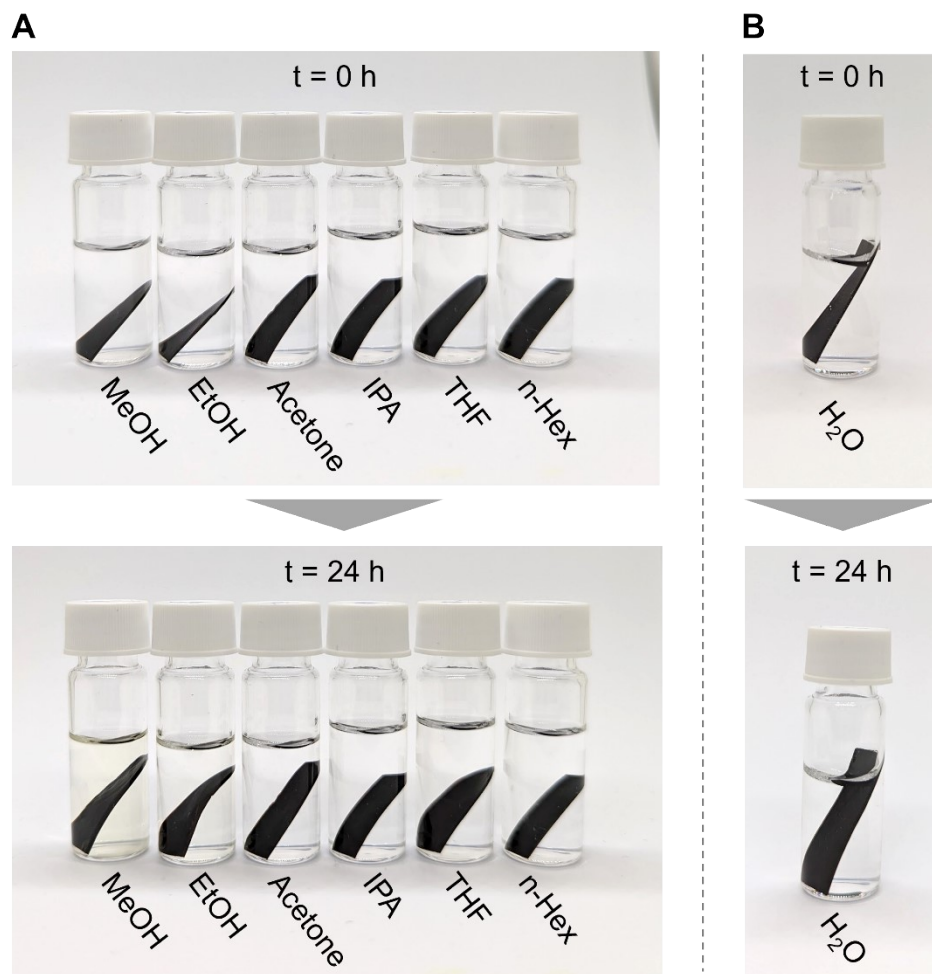


Figure S5. Photographs depicting the solvent resistance of **P(Fim)-TREN** in various organic solvents (A) and water (B) at different time intervals.

Table S4. Swelling measurements of **P(FIm)-FD** in different solvents at room temperature.

Solvents	MeOH	EtOH	Acetone	IPA	THF	n-hexane	H ₂ O
Swelling ratio (%)	19.2	0.6	0.17	3.7	0.5	0.6	18.4
Gel content (%)	95.2	99.6	99.3	99.4	99.8	99.6	97.3

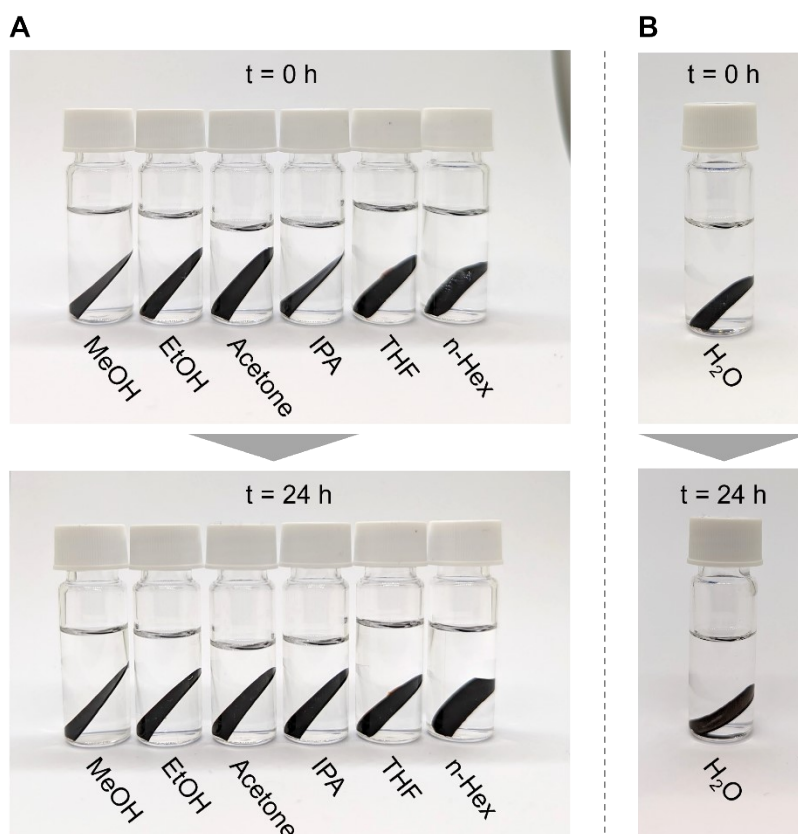


Figure S6. Photographs depicting the solvent resistance of **P(FIm)-FD** in various organic solvents (A) and water (B) at different time intervals.

4. Thermal analysis of polyimine networks

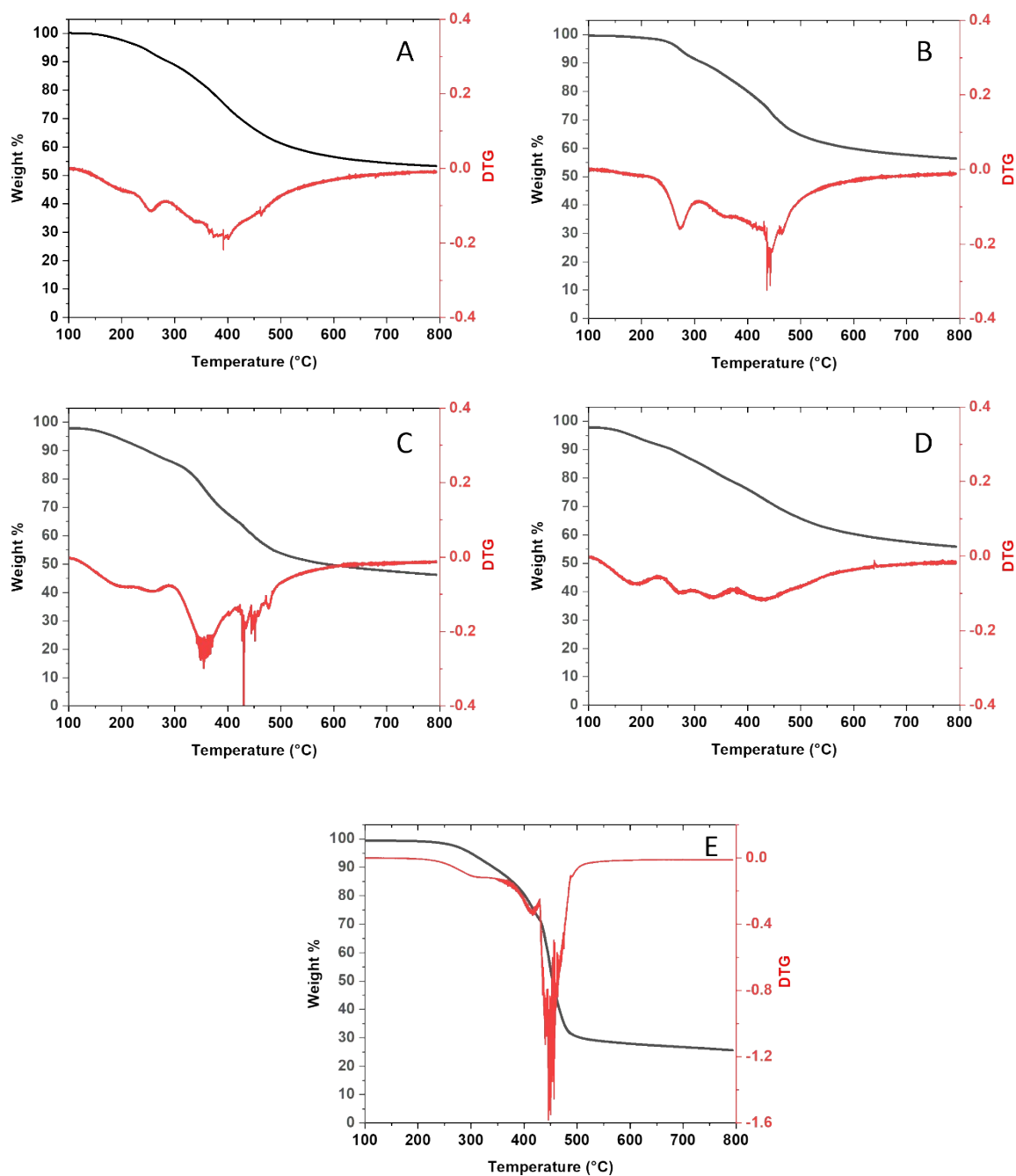


Figure S7. Thermogravimetric analysis (TGA) of the polyimines, performed under N₂: **P(FIm)-TREN** (A), **P(FIm)-PD** (B), **P(FIm)-pMD** (C), **P(FIm)-FD** (D), **P(FIm)-Pri** (E).

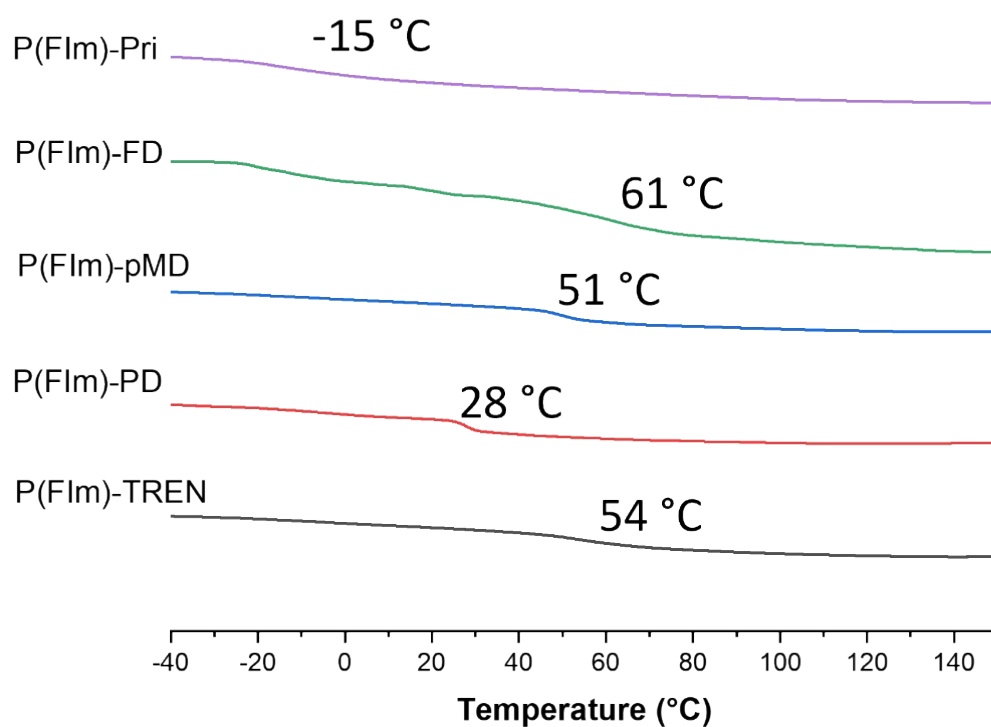


Figure S8. Differential scanning calorimetry (DSC) of the polyimines, performed under N₂. The heating rate of P(Flm)-FD and P(Flm)-TREN was 20 °C/min, whereas that of the others was 10 °C/min.

5. Thermomechanical analysis of polyimine networks

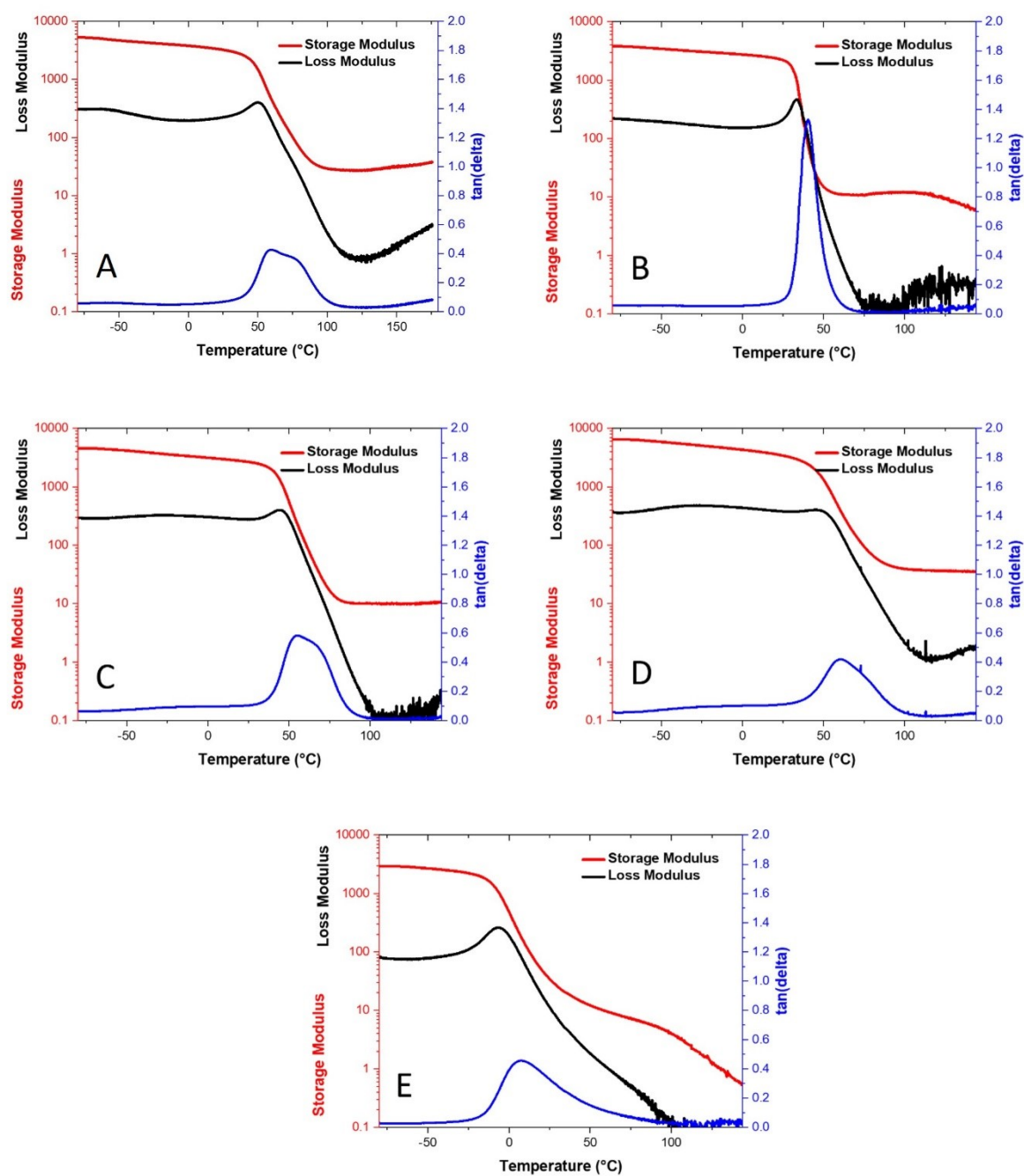
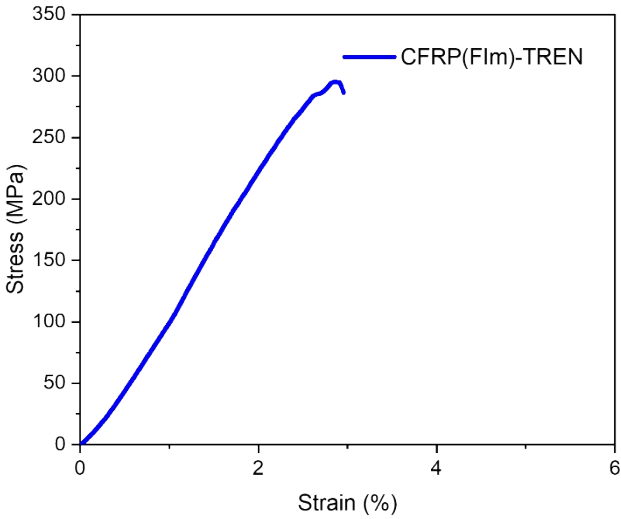


Figure S9. Dynamic mechanical analysis (DMA) of the polyimines: **P(FIm)-TREN** (A), **P(FIm)-PD** (B), **P(FIm)-pMD** (C), **P(FIm)-FD** (D), **P(FIm)-Pri** (E).

6. Mechanical analysis



	Maximum stress (MPa)	Strain at break (%)	Young's modulus (GPa)
CFRP(FIm)-TREN	295.1 ± 46.5	3.0 ± 0.8	10.7 ± 0.9

Figure S10. Tensile stress-strain curve of the polyimine based carbon-fiber composite, **CFRP(FIm)-TREN**.

7. Scanning electron microscopy

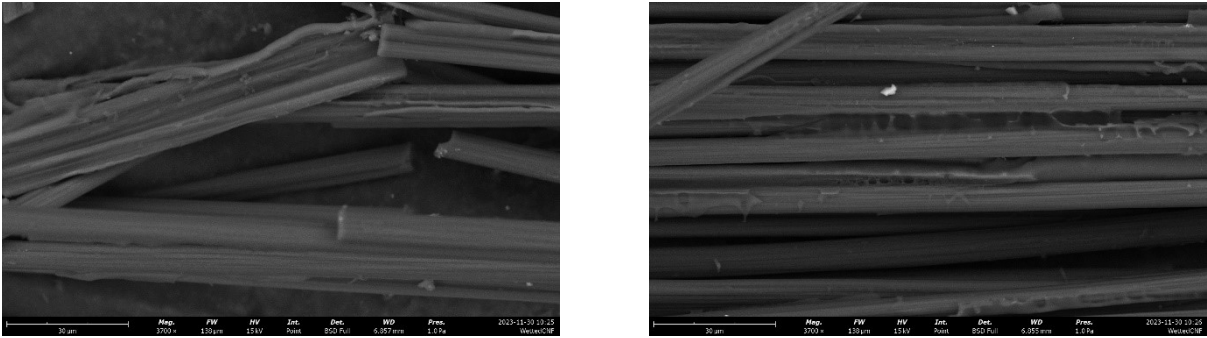


Figure S11. Investigation of the fractured **CFRP(FIm)-TREN** composite by scanning electron microscopy. Scale bar is 30 µm.

8. Chemical recycling of the P(FIm)-TREN network

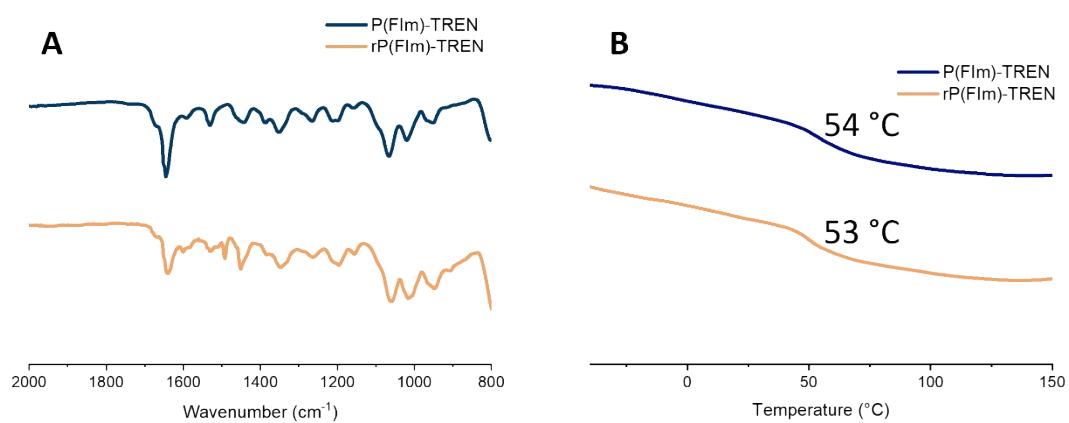


Figure S12. Comparison of the virgin **P(FIm)-TREN** and recycled **rP(FIm)-TREN**: FTIR (A), DSC (B).