Supplementary Information

Surface and bulk defect formation during hydrothermal synthesis of LiCoPO₄ crystals and their electrochemical implications

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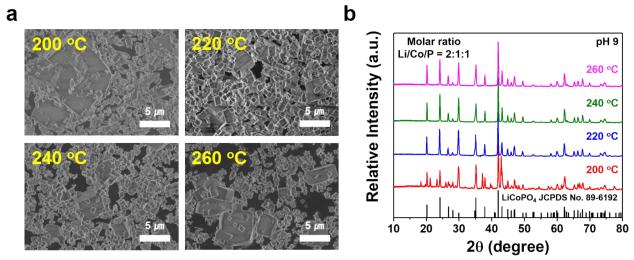


Figure S1. Morphology and phase characterizations of hydrothermally synthesized LCP (HT-LCP) samples for 6 hours at different temperatures via (a) SEM and (b) XRD. (The concentration of $CoSO_4$ and H_3PO_4 is fixed at 0.4 M).

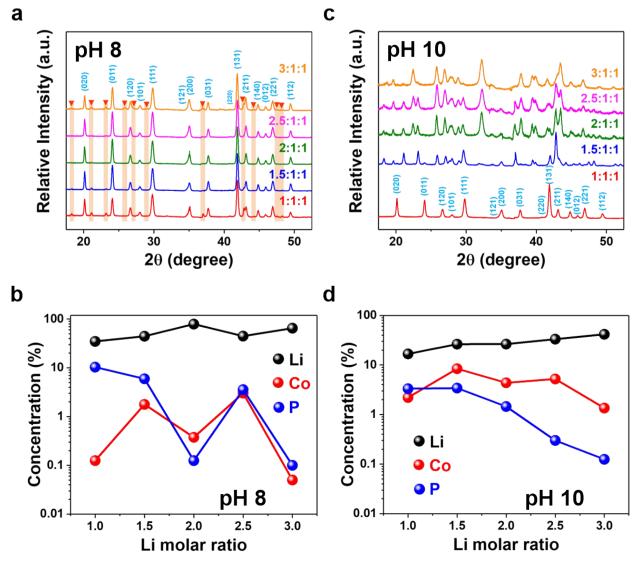


Figure S2. Phase and elemental comparison of HT-LCP samples from the precursor solutions with different Li/Co/P molar ratio at (a-b) pH 8 and (c-d) 10 (The concentration of $CoSO_4$ and H_3PO_4 is fixed at 0.4 M).

Table S1. pH value variations before and after HT-synthesis.

Li/Co/P Molar ratio	рН 8		рН 9		pH 10	
	Before	After	Before	After	Before	After
1:1:1	8	6.1	9	9.17	10	10
1.5:1:1	8	5.04	9	9.1	10	10.14
2:1:1	8	4.97	9	9.07	10	10.16
2.5:1:1	8	6.26	9	9.38	10	9.96
3:1:1	8	8.75	9	9.36	10	10

(The concentration of CoSO₄ and H₃PO₄ is fixed at 0.4 M)

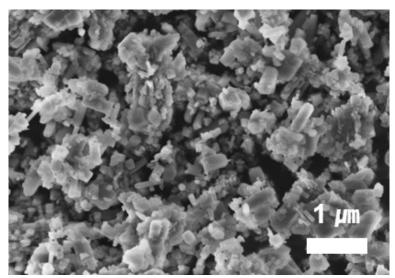


Figure S3. Morphology of LCP nanocrystals from the precursor solutions with the stoichiometric ratio 1:1:1 (Li:Co:P) at pH 10.

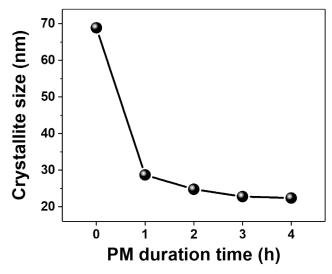


Figure S4. Crystallite size calculated based on Rietveld refinement with the increasing PM duration time.

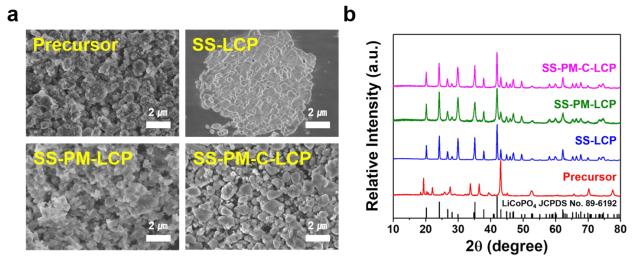


Figure S5. Morphology and phase characterizations of LCP samples obtained by solid-state reaction (SS-LCP) and post-synthesis treatments (SS-PM-LCP and SS-PM-C-LCP) via (a) SEM and (b) XRD.

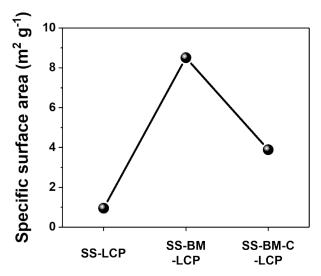
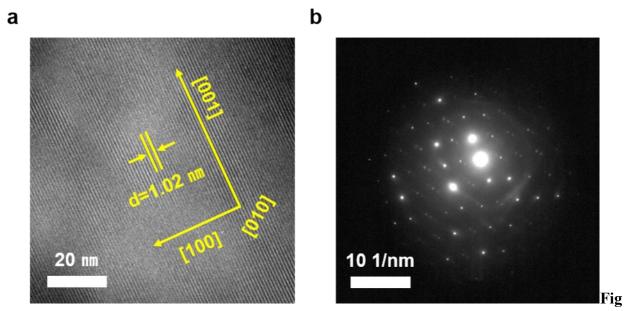


Figure S6. Specific surface area of SS-synthesis samples.



ure S7. (a) HRTEM and (b) SAED pattern of HT-LCP.

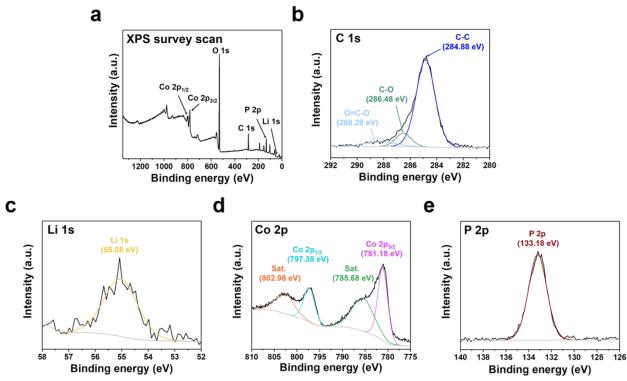


Figure S8. XPS spectra of HT-LCP: (a) Survey scan, (b) C 1s spectrum, (c) Li 1s spectrum, (d) Co 2p spectrum, and (e) P 2p spectrum.

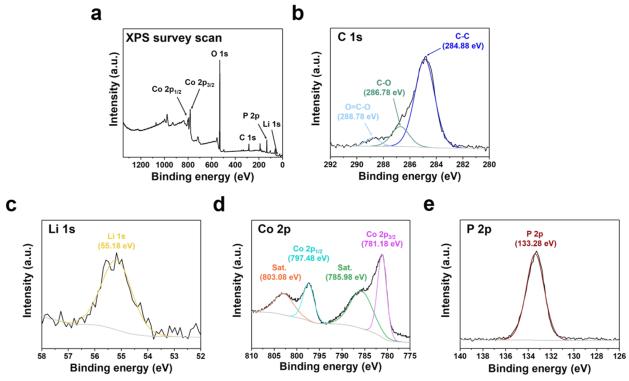


Figure S9. XPS spectra of SS-LCP: (a) Survey scan, (b) C 1s spectrum, (c) Li 1s spectrum, (d) Co 2p spectrum, and (e) P 2p spectrum.

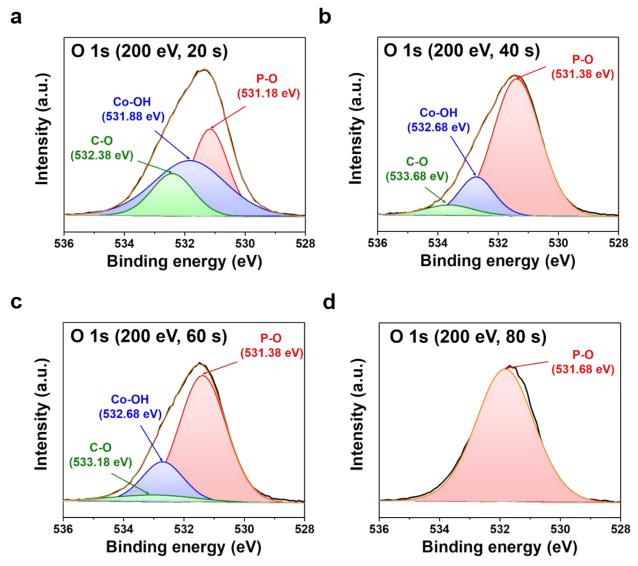


Figure S10. XPS depth profile (O 1s) of HT-LCP with 200 eV laser for (a) 20 seconds, (b) 40 seconds, (c) 60 seconds and (d) 80 seconds.

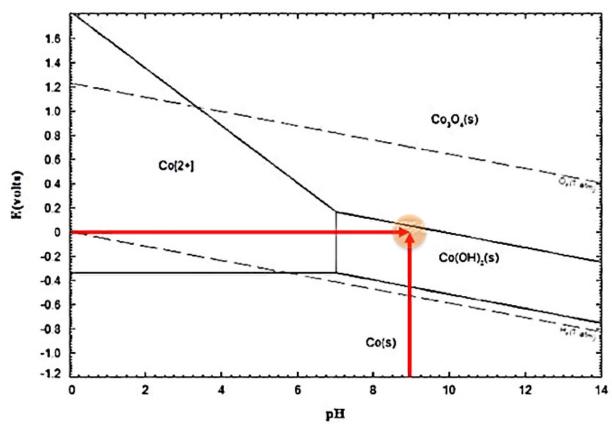


Figure S11. Pourbaix diagram of cobalt (Co-H₂O at 298.15 K, 0.01 M).⁸⁹

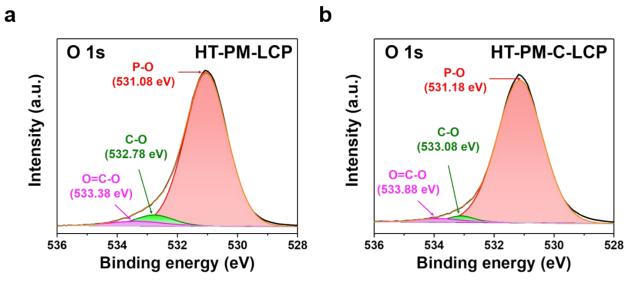
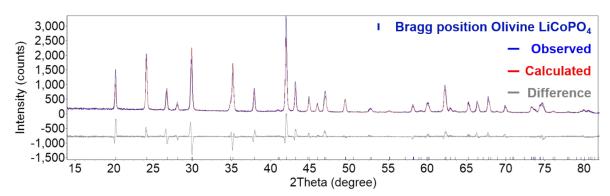
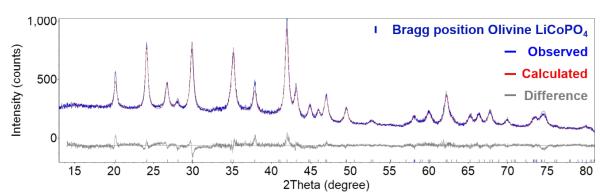


Figure S12. XPS O 1s spectra of (a) HT-PM-LCP, and (b) HT-PM-C-LCP.

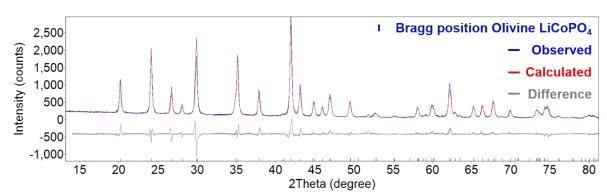
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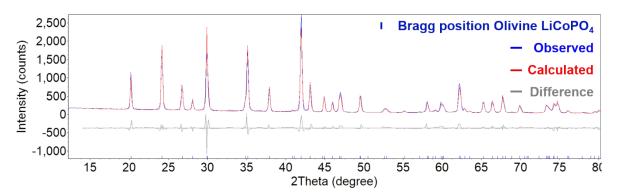
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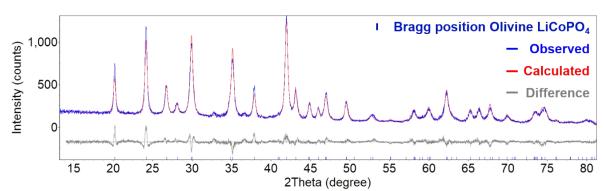
C











f

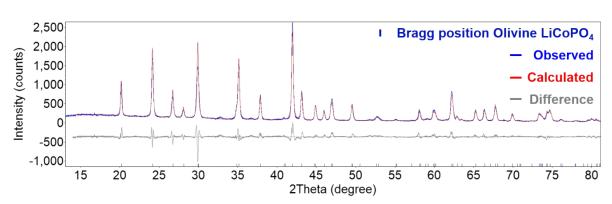


Figure S13. XRD Rietveld refinement of HT-LCP and SS-LCP samples: (a) HT-LCP, (b) HT-PM-LCP, (c) HT-PM-C-LCP, (d) SS-LCP, (e) SS-PM-LCP, and (f) SS-PM-C-LCP.