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Electronic Supplementary Material (SZI) for Material Advances

Supporting Information

Reassembly of wood to plastic- and paper-like films via ultra-mild dissolution in formic

acid

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Table S1: Comparison of density, maximal stress and strain, Young's, storage (E`), and elastic (E``) moduli, and loss tangent (tan δ) of films from Eucalyptus, Japanese cedar sapwood and heartwood under dissolving temperatures of 30 °C to 50 °C. Table 1 with standard daviations. Five replicates were measured.

Wood	Temperature	Density	Stress	Strain	Young`s modulus	E`	Ε``	tand
powder	(°C)	(g/cm³)	(MPa)	(%)	(MPa)	(Tg /℃)	
	30.0	0.37	3.09	1.16	334.75	230	1/12	nd
	50.0	±0.004	±0.20	±0.11	±31.26	250	145	n.u.
Eucohyptuc	40.0	0.56	31.15	2.73	1745.84	220	242	nd
Eucalyptus	40.0	±0.003	±4.75	±0.62	±105.48	239	243	n.u.
	50.0	0.66	60.98	3.21	3096.73	110	175	177
	50.0	±0.004	±1.92	±0.58	±346.81	110	175	1//
	20.0	0.54	2.17	1.19	209.93	245	265	n d
lananaca	50.0	±0.002	±0.48	±0.06	±5.81	245	205	n.u.
Japanese	40.0	0.52	10.61	1.94	685.74	251	ام م	
(convood)	40.0	±0.003	±1.35	±0.15	±47.95	251	n.d.	n.a.
(Sapwood)	50.0	0.65	5.49	1.35	475.71	261	110	nd
	50.0	±0.004	±0.17	±0.07	±46.92	201	110	n.u.
	20.0	0.23	0.59	1.10	nd	nd	nd	nd
lananoso	30.0	±0.02	±0.16	±0.10	n.u.	n.u.	n.u.	n.u.
Japanese	40.0	0.50	6.10	1.68	453.45	260	124	nd
(hoortwood)	40.0	±0.008	±0.66	±0.13	±48.53	207	124	n.u.
(neartwood)	50.0	0.64	7.15	1.41	614.81	276		
	50.0	±0.004	±1.62	±0.23	±141.63	270	114	n.d.

Table S2: Comparison of density, maximal stress and strain, Young's, storage (E`), and elastic (E``) moduli, and loss tangent (tan δ) of the films from Japanese beech, wheat bran, and sugarcane bagasse under dissolving temperatures at 50 °C. Table 2 with standard daviations. Five replicates were measured.

Wood powder	Density	Stress	Strain	Young`s modulus	E,	E	tand
	(g/cm³)	(MPa)	(%)	(MPa)		(Tg /°C)	
Jananasa baash	0.90	44.0	2.6	2395.8	1.1.1	206	240
Japanese beech	±0.005	±11.91	±0.22	±222.51	141	200	240
M/haat hyan	0.44	28.3	3.34	1276.8	100	120	171
wheat bran	±0.002	±2.04	±0.25	±59.88	109	130	1/1
Sugarcane	0.82	60.4	3.27	3592.6	177	165	177
bagasse	±0.001	±3.21	±0.20	±102.20	123	122	1//



*The film was not peeled off from the cellophane

b

Background double lines: 0.6 mm (black line), 0.5 mm (interval), 0.6 mm (black line)



Figure S1: Biomass films and their transmittance spectra prepared from the sawdust of Eucalyptus and Japanese cedar at different dissolving temperatures. a) Biomass films prepared by dissolution of sawdust from Eucalyptus and sapwood and heartwood of Japanese cedar with three different particle sizes (1.05–355, 355–500, and 500–1000 μ m) at 30 °C–90 °C. The

films were categorized into (1) lump-containing sheet from limited solubilization at 30 $^{\circ}$ C (yellow box); (2) transparent or opaque films prepared at 40 $^{\circ}$ C-60 $^{\circ}$ C (red box); and (3) black and fragile films prepared at 70 $^{\circ}$ C-90 $^{\circ}$ C (green box). **b**, Transmittance spectra between 220 and 800 nm of the films. Transmittance is summarized in Table S3.



Figure S2: Correlation between transmittance and whiteness of the biomass films from Eucalyptus and Japanese cedar prepared by dissolution at different temperatures. The films were prepared from different sizes of sawdust, $1.05-355 \mu m$ (a, d), $355-500 \mu m$ (b, e), and $500-1000 \mu m$ (c, f), from Eucalyptus (yellow), Japanese cedar sapwood (pale green) and heartwood (blue) at dissolving temperatures of $30 \degree C-90 \degree C$. (a, b, c) All transmittance degrees were measured at 800 nm (Table S3). (d, e, f) Whiteness levels of the films. Whiteness is summarized in Table S4. Five points on the film were measured and calculated for the averages.

Table S3: Transmittance of the films formed from Eucalyptus and sapwood and heartwood of Japanese cedar by dissolution at 30 $^{\circ}C$ –90 $^{\circ}C$. Five points on the film were measured and calculated for the averages.

Temperature	Eucalyptus			Japanes	Japanese cedar sap (T%)			Japanese cedar heart		
(°C)		(T%)						(T%)		
		size (mm)			size (mm)			size (mm)		
	1.05-	355-	500-	1.05-	355-	500-	1.05-	355-	500-	
	355	500	1000	355	500	1000	355	500	1000	
30	2.82	0.36	0.46	3.54	0.19	0.39	0.87	*	*	
	± 1.07	± 0.02	± 0.03	± 0.66	± 0.01	± 0.05	± 0.30			
40	0.15	0.15	0.24	0.21	0.13	0.14	0.14	0.13	0.20	
	± 0.00	± 0.00	± 0.02	± 0.01	± 0.00	± 0.00	± 0.01	± 0.01	± 0.02	
50	8.14	7.53	9.05	0.38	0.45	0.57	0.34	0.22	0.26	
	± 0.00	± 0.00	± 0.02	± 0.11	± 0.04	± 0.19	± 0.01	± 0.00	± 0.02	
60	9.90	8.25 ±	9.47	0.26	0.18	0.29	0.23	0.21	0.20	
	± 0.08	0.11	± 0.78	± 0.00	± 0.03	± 0.01	± 0.00	± 0.02	± 0.01	
70	6.48	6.67	8.16	1.40	2.13	0.30	*	*	*	
	± 0.26	± 0.19	± 0.27	± 0.05	± 0.11	± 0.03				
80	6.94	6.23	6.96	4.66	2.92	0.82	*	*	*	
	± 0.20	± 0.53	± 0.19	± 0.04	± 0.10	± 0.10				
90	3.42	4.51	3.85	*	*	*	*	*	*	
	± 0.31	± 0.45	± 0.08							

*Not detected

Table S4. Whiteness of the films formed from Eucalyptus and sapwood and heartwood of
Japanese cedar by dissolving at 30 $^\circ\!\mathrm{C}$ –90 $^\circ\!\mathrm{C}.$ Five points on the film were measured and
calculated for the averages.

Temperature		Eucalyptus			Japanese cedar sap (T%)			Japanese cedar heart		
(°C)		(T%)						(T%)		
		size (mm)			size (mm)			size (mm)		
	1.05-	355-	500-	1.05-	355-	500-	1.05-	355-	500-	
	355	500	1000	355	500	1000	355	500	1000	
30	32.84	40.62 ±	40.32 ±	31.44	42.66 ±	38.36 ±	*	19.42 ±	29.74 ±	
	± 0.95	0.94	0.45	± 1.17	0.88	0.56		0.34	0.42	
40	29.82	33.88 ±	32.86 ±	32.28	37.82 ±	34.02 ±	28.60 ±	27.40 ±	26.44 ±	
	± 0.50	1.08	1.21	± 0.38	0.20	0.31	0.26	0.37	0.36	
50	10.02	9.58	10.12 ±	29.72	25.76 ±	31.18 ±	23.46 ±	27.22 ±	25.94 ±	
	± 0.32	± 0.13	0.11	± 0.43	0.75	0.75	0.13	0.38	0.26	
60	9.00	9.22	8.90	23.72	28.42 ±	22.04 ±	22.36 ±	23.44 ±	23.20 ±	
	± 0.07	± 0.15	± 0.16	± 0.11	1.29	0.23	0.16	0.11	0.10	
70	5.94	6.26	6.66	12.28	12.36 ±	20.46 ±	13.96 ±	8.80	16.04 ±	
	± 0.17	± 0.23	± 0.09	± 0.25	0.21	0.32	0.26	± 0.12	0.22	
80	5.86	5.90	6.50	5.70	7.14	9.36	6.66	6.18	6.08	
	± 0.09	± 0.19	± 0.23	± 0.00	± 0.05	± 0.11	± 0.21	±0.11	± 0.04	
90	5.78	5.02	5.32	8.34	10.78 ±	6.40	5.72	6.94	7.96	
	±0.16	± 0.13	± 0.08	± 0.15	0.08	± 0.07	± 0.11	± 0.15	± 0.05	

*Not detected

Table S5. Comparison of transmittance and whiteness of the films formed for different species, tissues, and sources by dissolving at 30 $^{\circ}$ C–90 $^{\circ}$ C. Five points on the film were cmeasured and alculated for the averages.

Materials	Transmittance (T%)	Whiteness (%)
Japanese cedar (sapwood)	0.21 ± 0.01	36.84 ± 1.62
Japanese cedar (heartwood)	0.29 ± 0.06	27.00 ± 0.56
Rice fir	0.34 ± 0.01	40.14 ± 0.85
Japanese beech	6.70 ± 0.85	7.54 ± 0.39
Japanese cypress	5.88 ± 0.25	7.90 ± 0.35
Japanese red pine	8.02 ± 0.66	8.56 ± 0.18
Sugarcane baggase	5.98 ± 0.12	9.40 ± 0.12
Bomboo	6.11 ± 0.60	9.30 ± 0.37
Wheat bran	16.09 ± 0.88	8.96 ± 0.06



Figure S3: Comparison of soulubilization in different solvent and fractionation of biomass films by phase separation. Solubilazation of Eucaluptus (a) and Japanese cedar (b) biomass film in different solvent. Biomass film was stirred in each solvent for overnight at 40 $^{\circ}$ C. (c) Biomass films were prepared by dissolution of Eucalyptus and Japanese cedar heartwood at 50 $^{\circ}$ C. The films were redissolved in formic acid and separated by phase-partition with 2MeTHF and water. Each fraction was analyzed by NMR spectroscopy. (d) Recoveries of the three fractions from Eucalyptus and Japanese cedar films.



Figure S4: Aliphatic sidechain region of 2D ¹H-¹³C HSQC spectra from the Japanese cedar heartwood film formed at 50 °C. (a) ¹H-¹³C HSQC spectrum of the polysaccharide-rich aqueous fraction. The signals from cellulose, arabinoglucuronoxylan, galactoglucomannan, and formylated compounds are shown in red, green, yellow/red, and pink, respectively. (b) ¹H-¹³C

HSQC spectrum of the lignin-rich 2MeTHF fraction. The lignin linkages of β -O-4, β - β , β -5, and LCC are shown in blue, brown, purple, and yellow, respectively. The signal from A α in the β -O-4 lignin unit differed from that of the benzyl ether linkage in LCC. Signals from galactoglucomannan are shown in yellow as M1 to M6/M6^{\chi}. The signal assignment is shown in Tables S6, S7.

а									
l ch de		Eucal	yptus		Japanese cedar				
Labers	30 °C	40 °C	50 °C	60 °C	30 °C	40 °C	50 °C	60 ℃	
C1-NR									
C1-I									
C1-R6*									
C1-Ra*1									
C2-NR									
C2-I									
C2-R6*									
C2-Rat*2									
C3-NR* ³									
C 3-I									
C3-RB									
C 3-Ra.									
C4-NR									
C4-I									
C4-RB									
C4-Ra									
C5-NR* ³									
C.5-I* ³									
C.5-RB*4									
C5-Ra									
C6-NR									
C6-I									
C6-R02									
C6-RB									
6 F C 6									

S	ignal inte	nsity

b

U U								
Labole+A2:136		Euca	yptus			Japane	se ceda	r
Labers+Az.130	30 °C	40 °C	50 °C	60 °C	30 °C	40 °C	50 ℃	60 °C
X1-NR								
X1-I								
X1-RB								
<u>X1-Rα*</u> 1								
X2-NR								
X2-I						_		
X2-RB								
X2-Ra*2								
X3-NR								
X3-l* ³								
X3-Rβ* ⁴								
X3-Ra								
X4-NR								
X4-I								
X4-RB								
X4-Ra								
X5-NR a								
X5-NR b								
X5-I a								
X5-I b								
X5-Roca								
X5-Rocb								
X5-Rβa								
X5-Rβb								
MGA1								
MGA2								
MGA3* ⁵								
MGA4								
MGA5								
MGAOME								
2-0-Ac-X1								
2,3-0-Ac-X1								
2-0-Ac-X2								
3-0-Ac-X3								
β-Ara1								
Ara2								
3-F-X1								
3-F-X3								





Figure S5: Comparison of signal intensity changes in cellulose, xylan, mannan, and lignin between Eucalyptus and Japanese cedar at 30 °C–60 °C dissolving temperatures. a) Relative signal intensities of cellulose and formylated glucose residues. C-NR, cellulose nonreducing end unit; C-I, cellulose internal unit; C-R α , α reducing end unit of cellulose; C-R β , β reducing

end unit of cellulose; 6-F-C, glucose residue formylated at 6 position in cellulose. (b) Relative signal intensities of xylan and formylated xylose residues. X-NR, xylan nonreducing end unit; X-I, xylan internal unit; X-R α , α reducing end unit of xylan; X-R β , β reducing end unit of xylan; MGA, 4-O-methyl- α -D-glucuronic acid residue; Ac-X, acetylated xylose residue in xylan; Ara, arabinose residue. (c) Relative signal intensities of glucomannan. M-NR, nonreducing end unit in glucomannan; M-I, internal unit of glucomannan; M-R α , α reducing end unit of glucomannan; M-R β , β reducing end unit of glucomannan; Ac-M, acetylated mannose residue in glucomannan; Gal-NR, galactose nonreducing end unit; Gal-I, galactose internal unit; Gal- $R\alpha$, galactose α reducing end unit; Gal- $R\beta$, galactose β reducing end unit. (d) Relative signal intensities of lignin. S, syringyl unit; S`, α -etherified syringyl unit; G, guaiacyl unit; G`, α etherified guaiacyl unit; A α , A β , A γ , α , β and γ positions in β -O-4 unit; α , γ -F-A, β -O-4 unit formylated at α and/or γ positions; B α , B β , α and β positions in β -5 unit; C α , C β , C γ , α , β and γ positions in β - β linkage. The chemical structures and positions are shown in Figure 7, and Scheme 1 and Figure S4. The signal assignment is tabulated in Tables S6, S7. The relative signal intensities were categorized into four groups: pale yellow, not detected; green, weak; orange, middle; pink, strong.

	HSC	QC		
Labels	δH (ppm)	δC (ppm)	Asingments	Compositions
C1-NR	4.20	103.27	C_1 -H ₁ of β -D-glucopyranoside at non-reducing end	
C1-I	4.30	103.03	C_1 -H ₁ of β -D-glucopyranoside in internal	
C1-Rβ*	4.49	97.10	C_1 -H ₁ of β -D-glucopyranoside at reducing end	
C1-Rα* ¹	4.91	92.36	C_1 -H ₁ of α -D-glucopyranoside at reducing end	
C2-NR	2.91	72.89	C_2 -H ₂ of β -D-glucopyranoside at non-reducing end	
C2-I	2.98	72.80	C_2 -H ₂ of β -D-glucopyranoside in internal	Collulaça
C2-R β*	2.97	74.64	C_2 -H ₂ of β -D-glucopyranoside at reducing end	Glucomannan Galactoglucomannan
C2-Rα*2	3.20	72.16	C_2 -H ₂ of α -D-glucopyranoside at reducing end	
C3-NR* ³	3.38	76.85	C_3 -H ₃ of β -D-glucopyranoside at non-reducing end	
C3-I	3 47	73 32	C_3 -H ₃ of β -D-glucopyranoside in internal	
C3-R β	5.47	13.32	C_3 -H ₃ of β -D-glucopyranoside at reducing end	
C3-Rα	3.65	71.54	C_3 -H ₃ of α -D-glucopyranoside at reducing end	
C4-NR	3.32	69.84	C₄-H₄ of β-D-glucopyranoside at non-reducing end	

Table S6: Assignment of 2D ¹H-¹³C HSQC NMR data for polysaccharides of Eucalyptus and Japanese cedar films. The names of lables are corresponding to those in Fugure S5.

C4-I			C₄-H₄ of β-D-glucopyranoside in internal			
C4-Rβ	3.41	79.14	C ₄ -H ₄ of β -D-glucopyranoside at reducing end			
C4-Rα			C ₄ -H ₄ of α -D-glucopyranoside at reducing end			
C5-NR* ³	3 38	76 85	C ₅ -H ₅ of β -D-glucopyranoside at non-reducing end			
C5-I* ³	0.00	10.00	C_5 -H $_5$ of β -D-glucopyranoside in internal			
C5-Rβ*4	3.28	74.28	C_5 -H $_5$ of β -D-glucopyranoside at reducing end			
C5- Rα	3.84	69.78	C ₅ -H ₅ of α -D-glucopyranoside at reducing end			
C6-NR	3.38	61.10	C_6 -H ₆ of β -D-glucopyranoside at non-reducing end			
C6-I	3.51	60.64	C_6 -H ₆ of β -D-glucopyranoside in internal			
C6-Ra	3 60	60 20	C_6 -H ₆ of α -D-glucopyranoside at reducing end			
C6-R β	5.05	00.23	C_6 -H ₆ of β -D-glucopyranoside at reducing end			
6-F-C6	4.04	62.48	C ₆ -H ₆ of formylated β-D- glucopyranoside at non-reducing end	Formylated cellulose		
X1-NR	4 29	101 02	C_1 -H ₁ of β -D-xylopyranoside at non-reducing end			
X1-I	4.20	101.05	C_1 -H ₁ of β -D-xylopyranoside in internal	Glucurono xylan Arabinoglucurono		
Χ1-Rβ	4.29	97.44	C_1 - H_1 of β -D-xylopyranoside at reducing end	Xylan		
X1-Rα* ¹	4.92	92.36	C_1 - H_1 of α -D-xylopyranoside at reducing end			

X2-NR	2 11	70 57	C_2 - H_2 of β -D-xylopyranoside at non- reducing end
X2-I	3.11	12.31	C_2 -H ₂ of β -D-xylopyranoside in internal
Χ2- Rβ	2.97	74.64	C_2 -H ₂ of β -D-xylopyranoside at reducing end
X2-Rα*²	3.20	72.16	C ₂ -H ₂ of α -D-xylopyranoside at reducing end
X3-NR	3.15	76.65	C_3 -H ₃ of β -D-xylopyranoside at non-reducing end
X3-I* ³	3 28	74 28	C_3 -H ₃ of β -D-xylopyranoside in internal
X3-Rβ* ⁴	3.20	74.20	C_3 -H ₃ of β -D-xylopyranoside at reducing end
X3-Rα	3.51	71.59	C ₃ -H ₃ of α -D-xylopyranoside at reducing end
X4-NR	3.49	69.57	C_4 -H ₄ of β -D-xylopyranoside at non-reducing end
X4-I			C₄-H₄ of β-D-xylopyranoside in internal
Χ4-Rβ	3.55	75.35	C₄-H₄ of β-D-xylopyranoside at reducing end
X4-Rα			C_4 -H ₄ of α -D-xylopyranoside at reducing end
X5-NR a	3.10	65.61	C_5 -H $_5$ of β -D-xylopyranoside at non-reducing end a
X5-NR b	3.74	65.73	C_5 -H $_5$ of β -D-xylopyranoside at non-reducing end b
X5-I a	3.35	63.04	C_5 -H $_5$ of β -D-xylopyranoside in internal a
X5-I b	3.92	63.15	C_5 -H $_5$ of β -D-xylopyranoside in internal b
X5-Rα a	3.52	58.12	C ₅ -H ₅ of α -D-xylopyranoside at reducing end a

X5-Rα b	3.66	58.48	C₅-H₅ of α-D-xylopyranoside at reducing end b
X5-Rβ a	3.20	63.11	C ₅ -H ₅ of β -D-xylopyranoside at reducing end a
X5-Rβ b	3.42	62.91	C_5 -H $_5$ of β -D-xylopyranoside at reducing end b
MGA1	5.16	97.37	C ₁ -H ₁ of 4- <i>O</i> -methyl-α-D-gluconic acid pyranoside at non-reducing end
MGA2	3.73	73.06	C_2 -H ₂ of 4- <i>O</i> -methyl- α -D-gluconic acid pyranoside at non-reducing end
MGA3* ⁵	3.74	73.16	C_3 - H_3 of 4-O-methyl- α -D-gluconic acid pyranoside at non-reducing end
MGA4	3.15	81.59	C_4 -H ₄ of 4- <i>O</i> -methyl- α -D-gluconic acid pyranoside at non-reducing end
MGA5	4.49	70.16	C_5 -H ₅ of 4-O-methyl- α -D-gluconic acid pyranoside at non-reducing end
MGAOME	3.38	59.02	C _{OMe} -H _{OMe} of 4-O-methyl-α-D- gluconic acid pyranoside acid at non-reducing end
2- <i>0</i> -Ac-X1	4.52	99.31	C₁-H₁ of 2- <i>O</i> -acetyl-β-D- xylopyranoside
2,3- <i>0</i> -Ac-X1	4.62	101.06	C₁-H₁ of 2,3- <i>O</i> -acethyl-β-D- xylopyranoside
2- <i>0</i> -Ac-X2	4.56	73.28	C₂-H₂ of 2- <i>O</i> -acethyl-β-D- xylopyranoside
3- <i>0</i> -Ac-X3	4.87	74.76	C ₃ -H ₃ of 3- <i>O</i> -acethyl-β-D- xylopyranoside
R- A ra1	5.00	101.77	C_1 -H ₁ of β -D-arabinofranoside
	5.08	100.45	C_1 -H ₁ of β -D-arabinofranoside

Ara2	3.85	82.80	C_2 - H_2 of β -D-arabinofranoside	
Μ1-β	4.48	101.50	C_1 -H ₁ of β -D-mannopyranoside at reducing end	
Μ1-α	4.97	93.73	C_1 -H ₁ of α -D-mannopyranoside at reducing end	
M2	3.85	71.17	C_2 - H_2 of β -D-mannopyranoside	
M3* ⁵	3.74	72.53	C ₃ -H ₃ of β -D-mannopyranoside	
M4-NR	3.67	67.79	C_4 -H ₄ of β -D-mannopyranoside at non-reducing end	
M4-I			C₄-H₄ of β-D-mannopyranoside in internal	
Μ4- Rβ	3.76	76.55	C_4 -H ₄ of β -D-mannopyranoside at reducing end	Glucomannan
M4-Rα			C_4 -H ₄ of α -D-mannopyranoside at reducing end	Galactoglucomannan LCC
	3.38	76.85		
M5* ⁴	3.51	76.63	$\textbf{C}_{5}\text{-}\textbf{H}_{5}$ of $\beta\text{-}\textbf{D}\text{-}mannopyranoside}$	
	3.70	76.55		
M5`	3.69	69.26	LCC C5 -H5 of β -D-xylopyranoside	
M6 a			C_6 -H ₆ of β -D-mannopyranoside a	
M6 b	3.84	62.66	C_6 -H ₆ of β -D-mannopyranoside b	

100.02 C₁-H₁ of β -D-arabinofranoside

5.17

M6`	3.49	67.83	LCC C ₆ ·H ₆ of β-D-xylopyranoside
3- <i>0</i> -Ac-M1	4.79	98.87	C ₁ -H ₁ of 3- <i>O</i> -acetyl-β-D- mannopyranoside
2- <i>0</i> -Ac-M2	4.98	72.66	C ₂ -H ₂ of 2- <i>O</i> -acetyl-β-D- mannopyranoside
3- <i>0</i> -Ac-M2	3.78	68.96	C_2 - H_2 of 3-O-acetyl- β -D-mannopyranoside
3- <i>0</i> -Ac-M3	4.78	77.25	C₃-H₃ of 3- <i>O</i> -acetyl-β-D- mannopyranoside
2- <i>0</i> -Ac-M4	3.70	76.55	C₄-H₄ of 2- <i>O</i> -acetyl-β-D- mannopyranoside
3- <i>0</i> -Ac-M4	3.92	75.54	C₄-H₄ of 3- <i>O</i> -acetyl-β-D- mannopyranoside
Gal1-NR	4.91	92.36	$C_1\text{-}H_1$ of $\beta\text{-}D\text{-}galactopyranoside at non-reducing end$
Gal1-I	4.32	105.16	$C_1\text{-}H_1$ of $\beta\text{-}D\text{-}galactopyranoside in internal$
Gal1-Rβ	4.28	97.49	$C_1\text{-}H_1$ of $\beta\text{-}D\text{-}galactopyranoside at reducing end$
Gal1-Rα*¹	4.91	92.36	C_1 -H ₁ of α -D-galactopyranoside at reducing end

	HSQC			Wood	
Labels	δHδC(ppm)(ppm)		Compositions		
S 2/6	6.65	103.50	C ₂ -H ₂ /C ₆ -H ₆ of syringyl units		
S` 2/6	7.30	106.30	C ₂ -H ₂ /C ₆ -H ₆ of syringyl units with α linkage		
G 2	6.98	110.84	C ₂ -H ₂ of guaiacyl units		
G` 2	7.47	111.66	C_2 -H ₂ /C ₆ -H ₆ of guaiacyl units with α linkage	Aromatic units	
	6.92	115.31			
G 5+G 6	6.78	118.90	C₅-H₅+C₀-H₀ of guaiacyl units		
G` 6	7.34	124.06	$\textbf{C}_{\textbf{6}}\textbf{-}\textbf{H}_{\textbf{6}}$ of guaiacyl units with α linkage		
Αα	4.86	71.51	C_a -H _a of β -O-4 linkages	Linkages of ligin and LCC	
α,γ-F-Αα	6.02	73.69	$C_{\alpha}\text{-}H_{\alpha}$ of $\beta\text{-}\textit{O}\text{-}4$ formylated linkages	Formylated β-O-4 of lignin	
Aβ(<i>t</i>)	4.01	86.62	C_{β} -H _{β} of β -O-4 linkages linked to a syringyl unit (<i>threo</i> form)		
Αβ(<i>e</i>)	4.10	85.93	C_{β} -H _{β} of β -O-4 linkages linked to a syringyl unit (<i>erythro</i> form)	Linkages of ligin	
Αγ1	3.21	62.00	$C_{\gamma 1}$ -H _{$\gamma 1$} of β -O-4 linkages	and LCC	
Αγ2	3.53	60.69	$C_{\gamma 2}$ - $H_{\gamma 2}$ of β - O -4 linkages		

Table S7: Assignment of 2D ¹H-¹³C HSQC NMR data for lignin and LCC of Eucalyptus and Japanese cedar films. The names of lables are corresponding to those in Fugure S5.

α,γ-F-Αγ1	3.92	63.27	$\textbf{C}_{\gamma 1} \textbf{-} \textbf{H}_{\gamma 1}$ of formylated $\beta \textbf{-} \textbf{O} \textbf{-} \textbf{4}$ linkages	Formylated β-O-4
α,γ-F-Αγ2	4.18	63.07	$C_{\gamma 2}$ - $H_{\gamma 2}$ of formylated β -O-4 linkages	of light
Βα	5.44	86.82	$C_{\alpha}\text{-}H_{\alpha}$ of $\beta\text{-}5$ linkages	
Ββ	3.49	53.45	C_{β} - H_{β} of β -5 linkages	
Cα	4.65	84.89	C_{α} -H _a of β - β linkages	
Сβ	3.05	53.57	C_{β} - H_{β} of β - β linkages	
Ϲγ	41.17	71.14	C_{γ} - H_{γ} of β - β linkages	Linkages of lignin
α	4.55	80.20	LCC C_{α} -H _{α} of β -O-4 linkages	
β	4.30	83.19	LCC C _{β} -H _{β} of β -O-4 linkages	
γ1	3.34	61.62	LCC $C_{\gamma 1}$ -H _{$\gamma 1$} of β - <i>O</i> -4 linkages	
γ2	3.61	60.15	LCC C _{$\gamma 2$} -H _{$\gamma 2$} of β - <i>O</i> -4 linkages	