

Supplementary Information

Metal-organic Framework-derived Hierarchical Porous N-doped Carbon Supported Sponge-like Pd-SnO<sub>2</sub> Nanostructures for Low-temperature CO Oxidation

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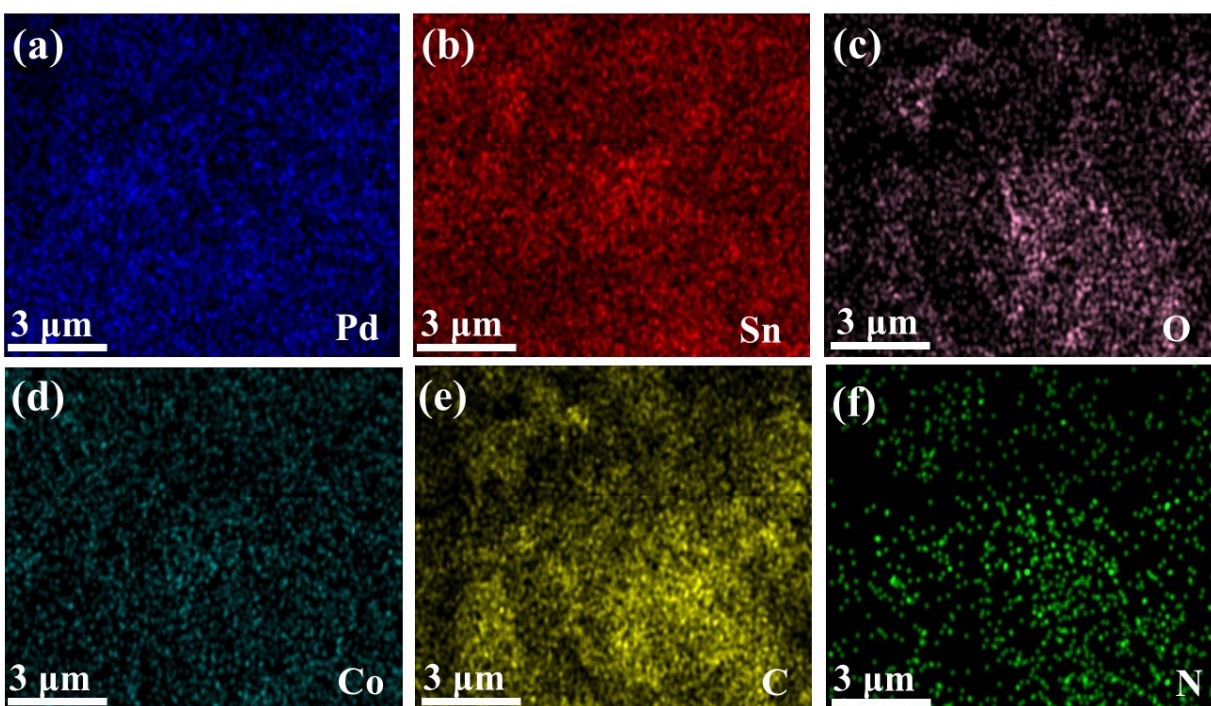
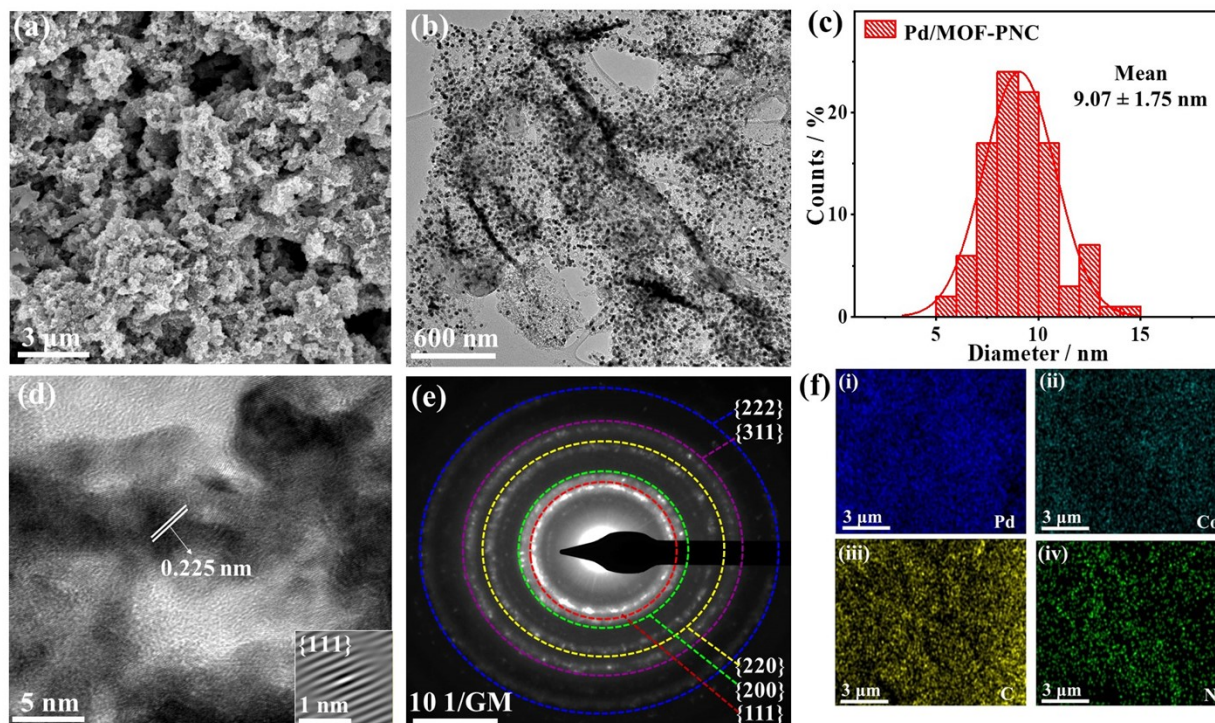
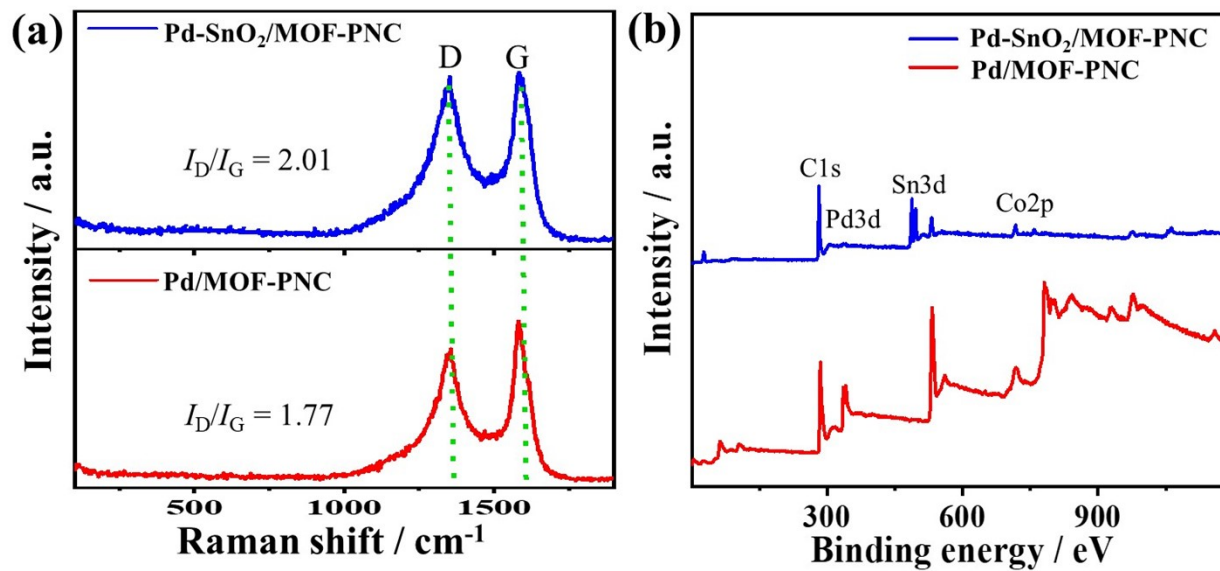


Fig. S1. Elemental mapping (a) Pd, (b) Sn, (c) O, (d) Co, (e) C and (f) N of Pd-SnO<sub>2</sub>/MOF-PNC



**Fig. S2.** (a) SEM, (b) TEM, (c) Nanoparticles size distribution, (d) HRTEM, (e) SAED, and (f) Elemental mapping (i) Pd, (ii) Co, (iii) C and (iv) N of Pd-of Pd/MOF-PNC.



**Fig. S3.** (a) Raman spectra and XPS wide scans of Pd-SnO<sub>2</sub>/MOF-PNC and Pd/MOF-PNC

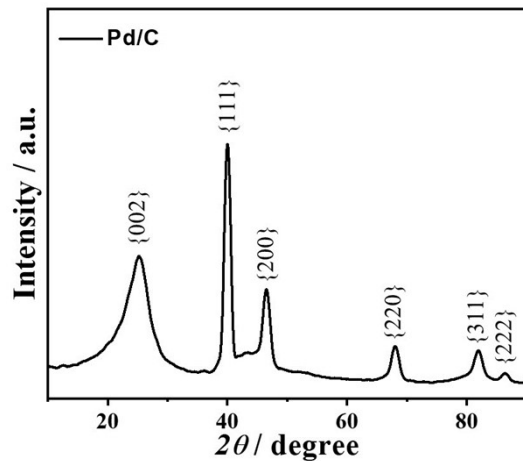


Fig. S4. XRD of Pd/C taken from the JCPDS database

Table S1. XPS and EDX atomic contents of elements in Pd-SnO<sub>2</sub>/MOF-PNC and Pd/MOF-PNC

	XPS		EDX		ICP-OES		
	Pd-SnO <sub>2</sub> /MOF-PNC	Pd/MOF-PNC	Pd-SnO <sub>2</sub> /MOF-PNC	Pd/MOF-PNC	Pd-SnO <sub>2</sub> /MOF-PNC	Pd/MOF-PNC	Pd/C
Pd	2.16 At.%	2.71 At.%	2.31 At.%	2.12 At.%	15.90 wt.%	17.70 wt.%	19.78 wt.%
Sn	3.57 At.%	-	1.31 At.%	-	18.12 wt.%	-	-
Co	4.50 At.%	15.88 At.%	2.52 At.%	5.29 At.%	3.16 wt.%	11.24 wt.%	-
C	78.86 At.%	80.97 At.%	66.21 At.%	49.81 At.%	-	-	-
O	10.42 At.%	-	11.93 At.%	23.24 At.%	-	-	-
N	0.39 At.%	0.44 At.%	15.72 At.%	20.54 At.%	-	-	-

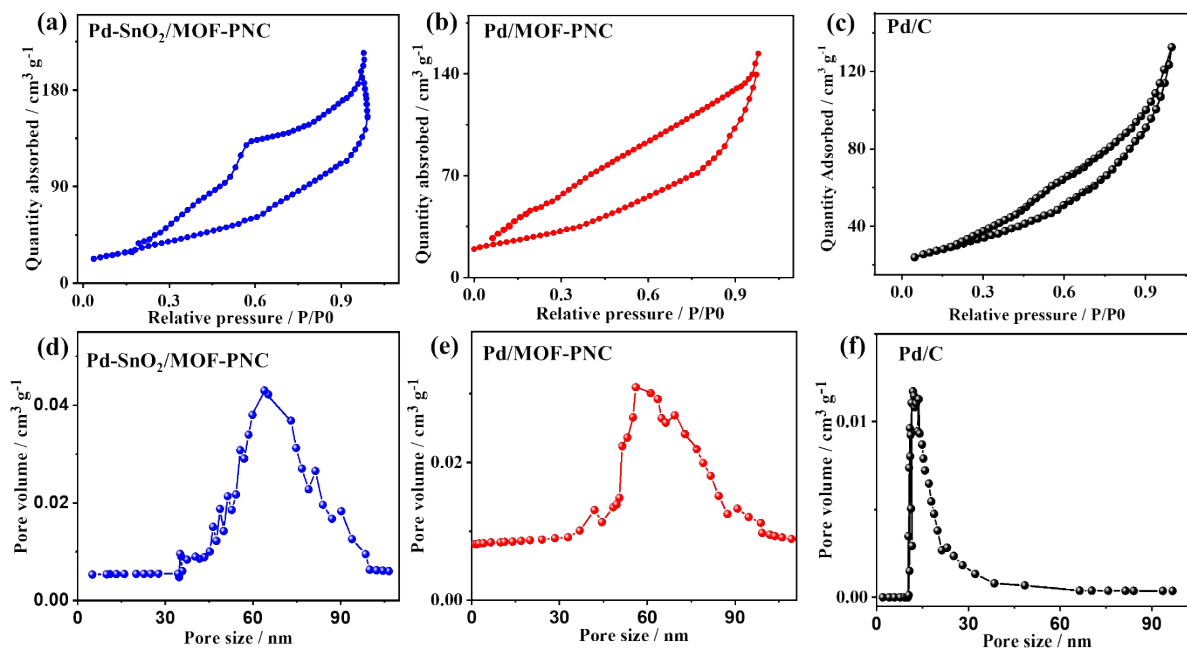


Fig. S5. (a-c) N<sub>2</sub>-adsorption/desorption isotherm and (d-e) pore size distribution of Pd-SnO<sub>2</sub>/MOF-PNC, Pd/MOF-PNC, and Pd/C.

Table S2: Binding energies of Pd 3d spectra of Pd-SnO<sub>2</sub>/MOF-PNC and Pd/MOF-PNC catalysts

Catalysts	Pd 3d <sub>5/2</sub>			Pd 3d <sub>3/2</sub>		
	Pd <sup>0</sup>	Pd <sup>2+</sup>	Pd <sup>4+</sup>	Pd <sup>0</sup>	Pd <sup>2+</sup>	Pd <sup>4+</sup>
Pd-SnO <sub>2</sub> /MOF-PNC	335.42	336.25	337.92	341.25	342.75	343.86
Pd/MOF-PNC	335.34	336.09	337.58	340.95	341.82	343.61

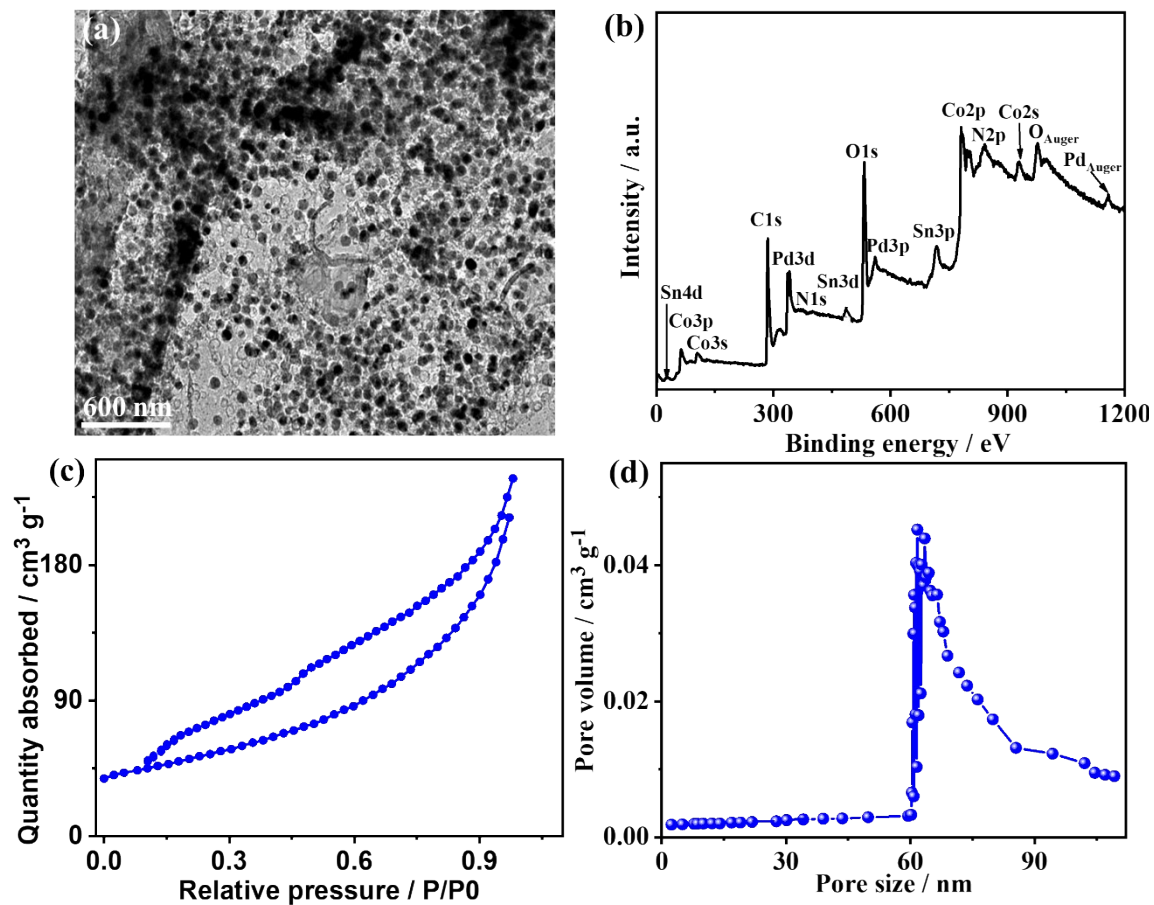
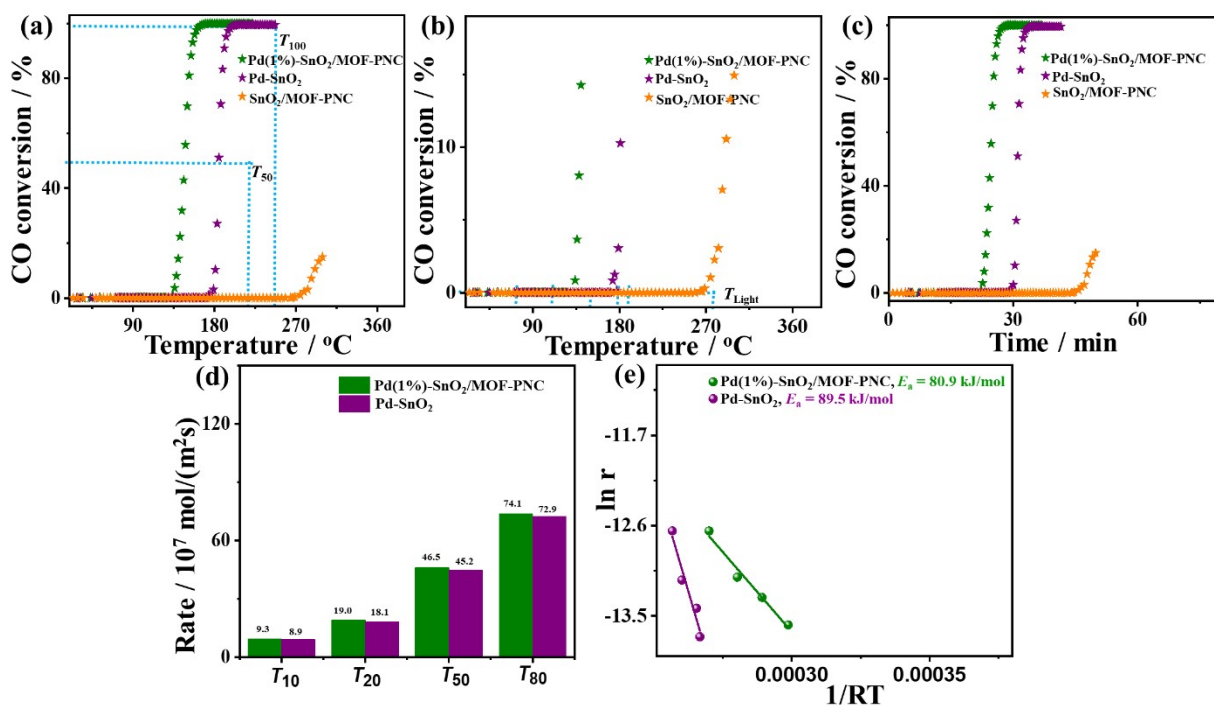


Fig. S6. High-resolution XPS (a) Pd3d and (b) Sn3d, (c) N<sub>2</sub>-adsorption/desorption isotherm and (d) pore size distribution of Pd-SnO<sub>2</sub>/MOF-PNC after the stability test for 108 h.



**Fig. S7.** (a,b) Temperature-dependent CO conversion, (c) Time-dependent CO conversion, (d) Rate at different CO conversion temperatures, and (e) Arrhenius plots of Pd(1%)-SnO<sub>2</sub>/MOF-PNC, Pd-SnO<sub>2</sub>, and SnO<sub>2</sub>/MOF-PNC.

Table S3. Comparative thermal CO oxidation performance of Pd-based catalysts.

MOF-derived porous N-doped carbon (MOF-PNC), high-entropy fluorite oxide (HEFO), hexagonal boron nitride (h-BN), calcination at 500 °C for 18 h (PC3), hydrothermal-like (HT), SSZ-13 zeolites (SSZ-13), rod-like (R)

Catalysts	Preparation methods	Morphology	T <sub>100</sub> (°C)	Refs
Pd-SnO <sub>2</sub> /MOF-PNC	Microwave-irradiation/annealing/etching/microwave-irradiation	Spongy-like	65.6	This work
Pd/MOF-PNC	Microwave-irradiation/annealing/etching/microwave-irradiation	Spongy-like	107.9	This work
Pd/CeSn75-800	Counter precipitation/calcination	Core-shell	~ 100	1
Pd@SiO <sub>2</sub> /TiO <sub>2</sub> -500	Precipitation/ calcination	Core-shell	~ 400	2
CeO <sub>2</sub> -Pd/S-800-5h	Impregnation/ calcination	2d hexagonal mesopores	~ 75	3
Pd <sub>1</sub> @HEFO	Ball milling/ annealing/etching	Cubic	170	4
Pd@CeO <sub>2</sub>			253	
Pd/MgO(5)-h-BN	Impregnation/calcination	Nanosheets	140	5
Ce <sub>1-x</sub> Pd <sub>x</sub> O <sub>2-δ</sub> (PC3)	hydrothermal/reduction/calcination	Nanocrystals	~ 95	6
Pd/MgAl-HT	Deposition-precipitation	Nanocrystals	~ 90	7
Pd-1%P	Wet-impregnation/ calcination	Fiber-like lamellar	~ 270	8

LaAlPd(0.8)O <sub>3</sub> -600	Impregnation/ Calcination	Perovskite	~ 325	9
Pd@SiO <sub>2</sub> -673	Polymerization/calcination	Core-shell	~ 130	10
Pd <sub>0.83</sub> Co <sub>0.17</sub> /C	Wet impregnation	Nanocrystals	150	11
Pd <sub>0.5</sub> /CeHfZrSnErO <sub>x</sub> Pd <sub>1.0</sub> /CeHfZrSnErO <sub>x</sub>	Ultrasound-mediated co-precipitation	Cubic	140 150	12
Pd-SSZ-13	Ion exchange/ calcination	Cubic particles	~ 175	13
PdO <sub>x</sub> /CeO <sub>2</sub>	Radio frequency sputtering	Dendrite-like	250	14
4%Pd/R-CeO <sub>2</sub>	Impregnation/ annealing	Rod, cubic, and octahedral	50	15
Pd-Cu/gC <sub>3</sub> N <sub>4</sub> NWs	Protonation/annealing	Nanowires	149	16
Pd/Cu/gC <sub>3</sub> N <sub>4</sub> NTs	Protonation/annealing	Nanotubes	154	17
Au/Pd/gC <sub>3</sub> N <sub>4</sub> NFs	Protonation/annealing	Nanofibers	144	18
Pd-impeded 3D porous graphene (3D Pd-E-PG)	Low-power microwave radiation	3D porous nanosheets	190	19
AuPd/TiO <sub>2</sub>	Incipient wetness method	Nanospheres	190	20
Pd/Ni-MOF-HNC Pd/Ni-MOF-NC	Microwave-irradiation/ annealing/etching/microwave- irradiation	Hollow nanosheets	114.5 153.8	21
5wt%Pd/Ce-MOF 7wt%Pd/Ce-MOF 3wt%Pd/Ce-MOF	Hydrothermal/microwave- irradiation	Spherical Pd on needle shaped support	92 148 190	22
2.9%Pd@MIL-101 4.9%Pd@MIL-101 2.2%Pd@MIL-101 1.1%Pd@MIL-101 0.6%Pd@MIL-101 0.3%Pd@MIL-101 0.1%Pd@MIL-101	Hydrothermal/microwave- irradiation	Spherical Ps on nanocubes support	107 147 123 151 151 140 181	23
Pd <sub>n</sub> (1-Cys) <sub>m</sub> /CeO <sub>2</sub> Pd <sub>n</sub> (1-Cys) <sub>m</sub> /TiO <sub>2</sub> Pd <sub>n</sub> (1-Cys) <sub>m</sub> /Fe <sub>3</sub> O <sub>4</sub> Pd <sub>n</sub> (1-Cys) <sub>m</sub> /ZnO	Chemical reduction/impregnation	Monodispersed spherical nanoclusters	110 112 115 155	24

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