

Supporting Information

Insights into the development and proceeding of $\text{CuO/CuFe}_2\text{S}_x\text{O}_{4-x}$ catalysts: An effective approach for hydrogen generation

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Table S1: Chemical composition and atomic percentages of CuFe_2O_4 determined using energy-dispersive X-ray (EDX) analysis.

Elements	Weight %	At %
C	0.66	1.68
O	25.28	54.75
Fe	47.92	28.56
Cu	26.14	15.01
Totals	100	100

Table S2: Energy-Dispersive X-ray (EDX) analysis was used to estimate the elemental weight and atomic percentages in the as-prepared $\text{CuO}/\text{CuFe}_2\text{S}_x\text{O}_{4-x}$.

Elements	Weight %	At %
C	0.61	1.66
O	25.88	56.74
S	2.2	1.24
Fe	47.95	29.12
Cu	23.36	11.24
Totals	100	100

Table S3: Magnetic parameters Magnetic saturation, Residual magnetic intensity, Coercivity and Squareness ratio of as-synthesized catalysts.

Magnetic Parameters	CuFe₂O₄	CuO/CuFe₂O₄	CuO/CuFe₂S_xO_{4-x}
Magnetic saturation (MS) (emu/g)	21.4	40.7	31.6
Residual magnetic intensity (Mr) (emu/g)	4.5	10.1	7.04
Coercivity (Hc) (G)	218	346	294
Squareness ratio	0.021	0.029	0.023

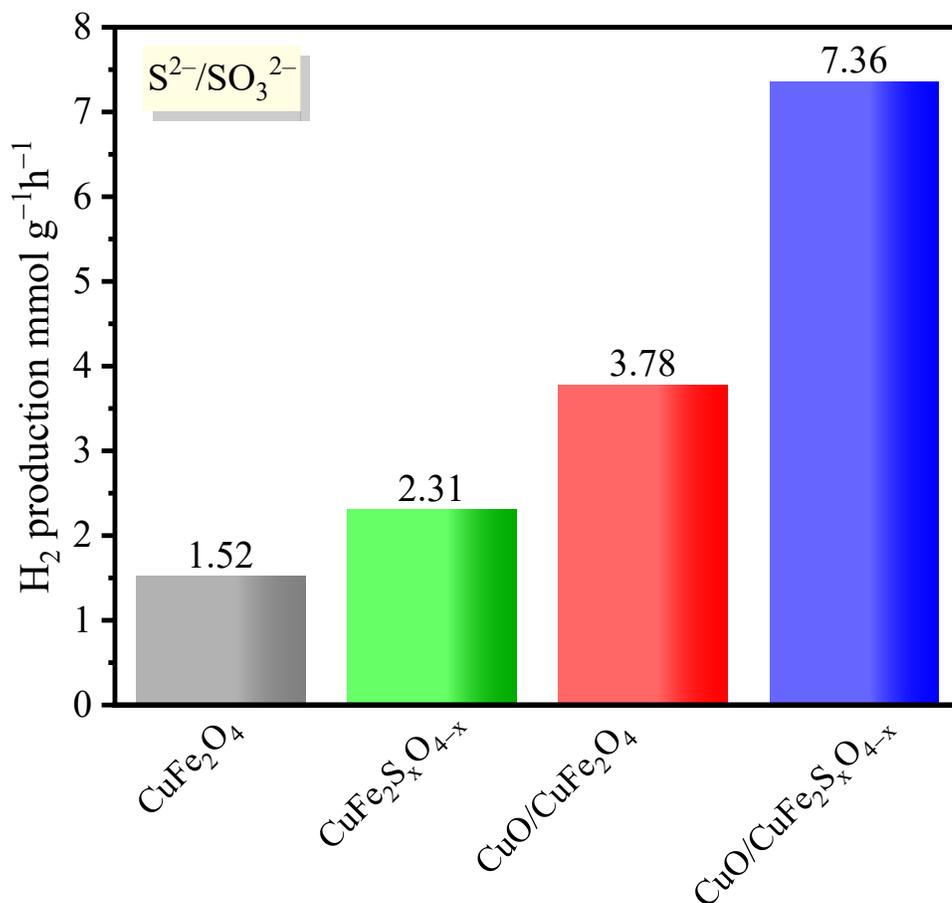
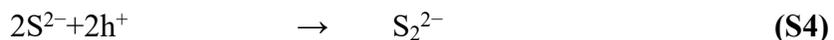
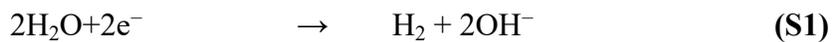


Figure S1: Photocatalytic H₂ evolution of CuFe₂O₄, CuFe₂S_xO_{4-x}, CuO/CuFe₂O₄ and CuO/CuFe₂S_xO_{4-x} photocatalysts in the presence of S²⁻/SO₃²⁻ sacrificial reagents.

The suggested reaction mechanism for the photocatalytic H₂ evolution in the presence of S²⁻/SO₃²⁻ mixtures is given in equation S1-S6¹,



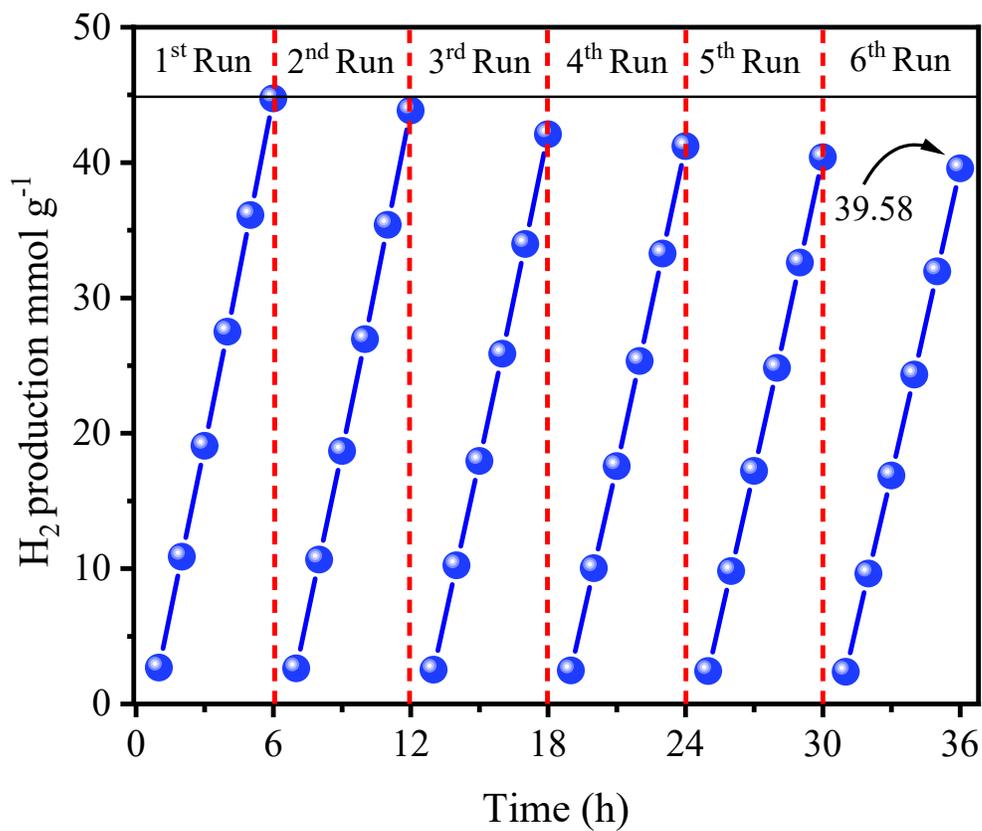


Figure S2: Recyclability test of catalysts.

References:

1. J. Schneider and D. W. Bahnemann, *The Journal of Physical Chemistry Letters*, 2013, **4**, 3479-3483.