

# Copper(I) as reducing agent for the synthesis of bimetallic PtCu catalytic nanoparticles

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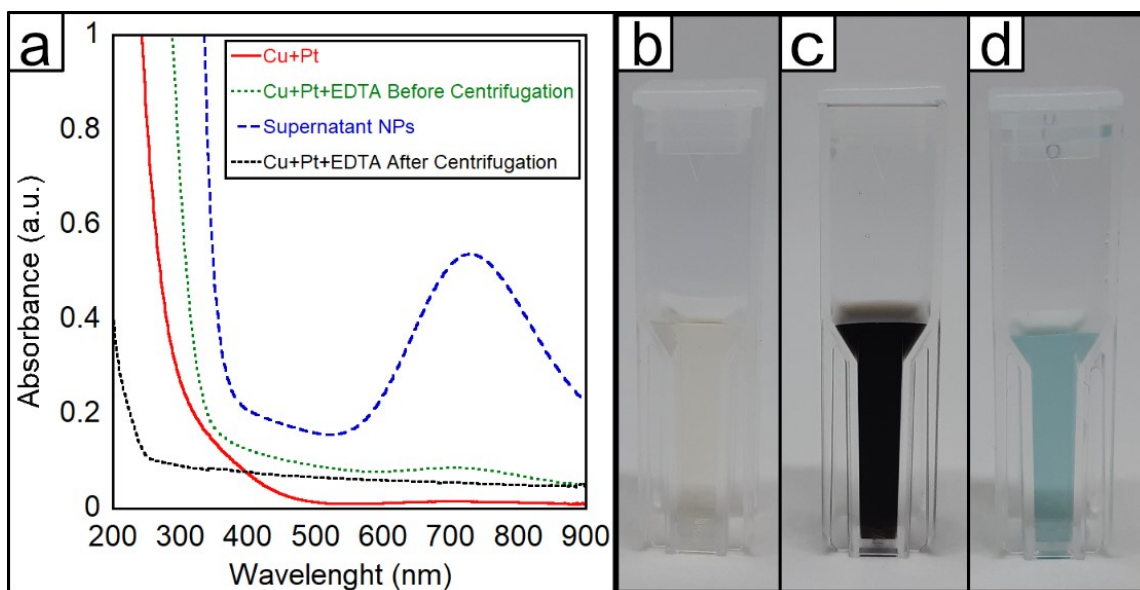
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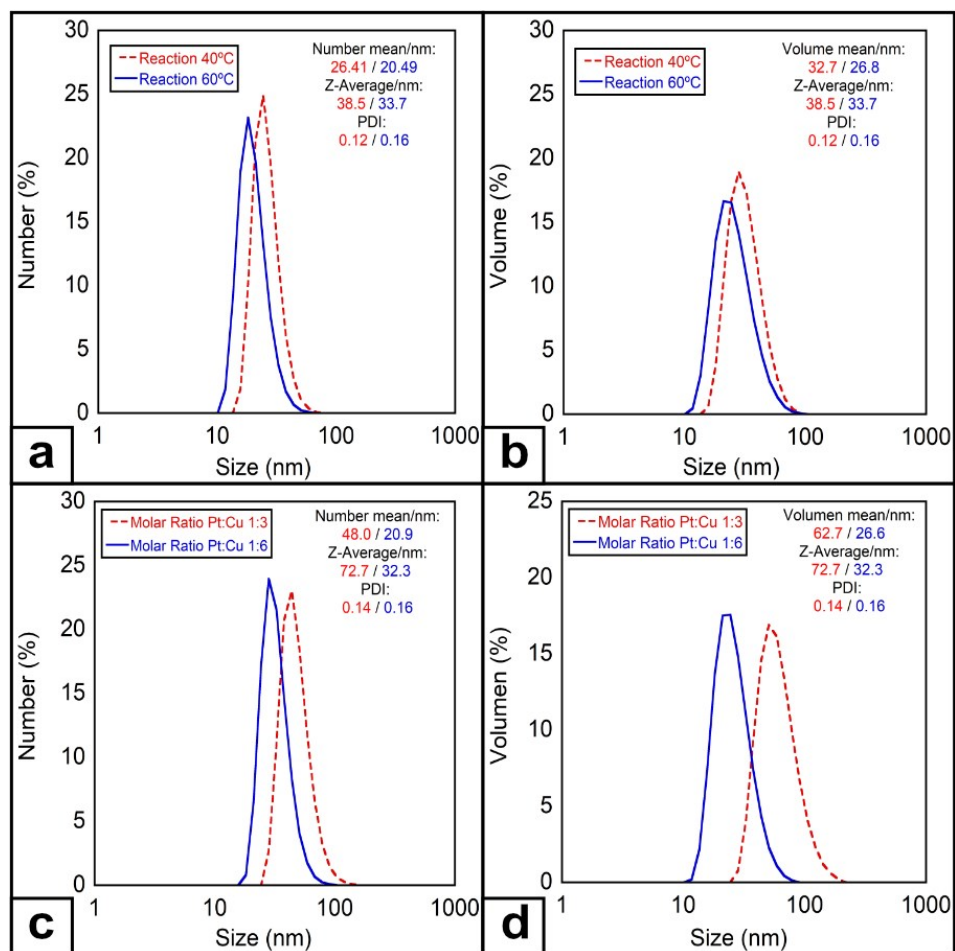
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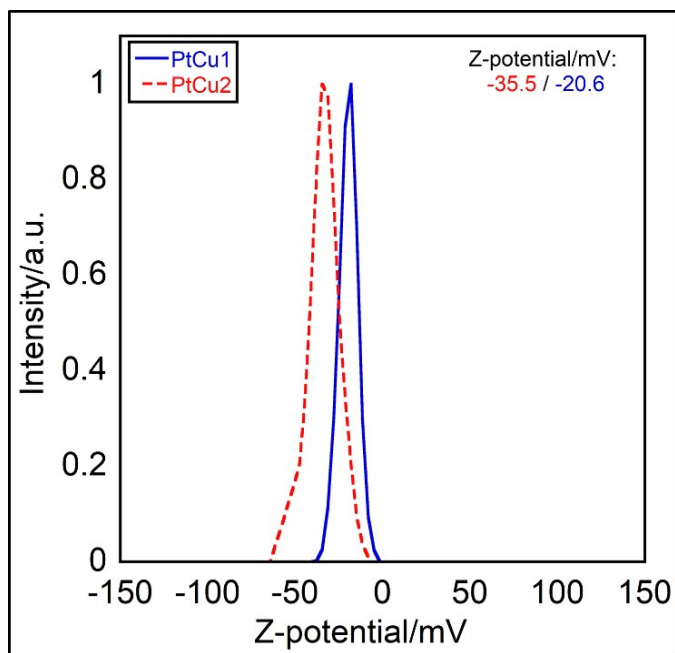
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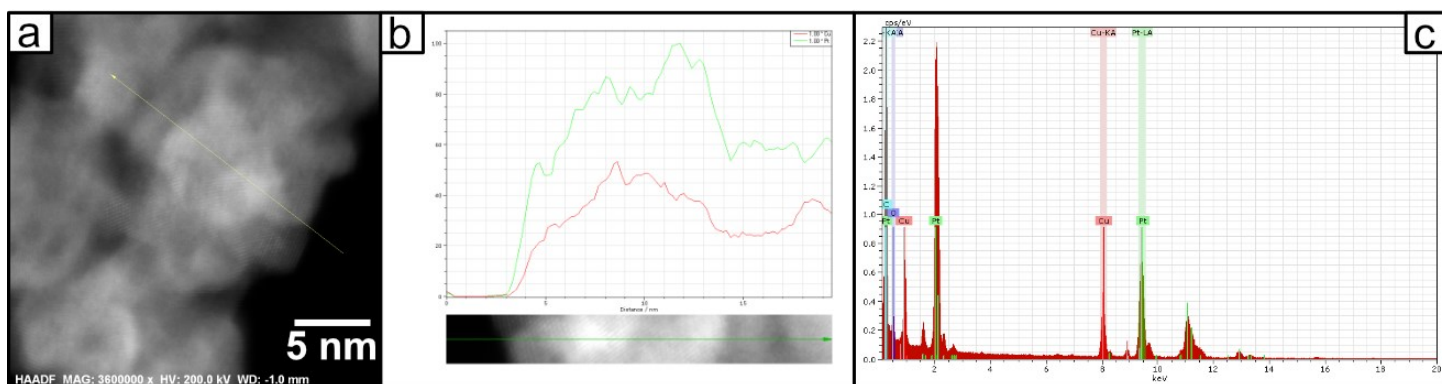
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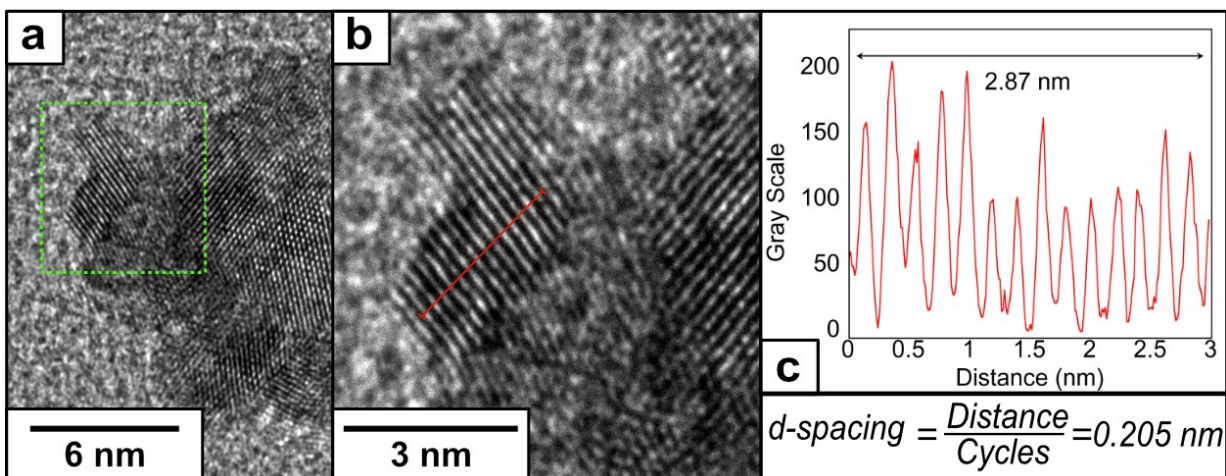
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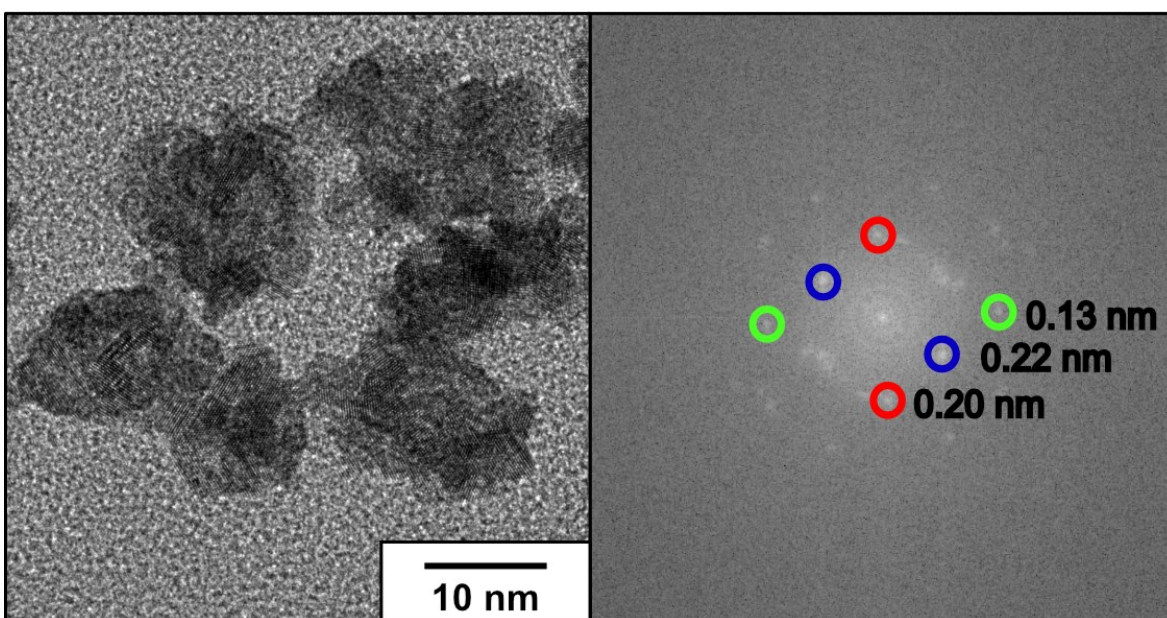
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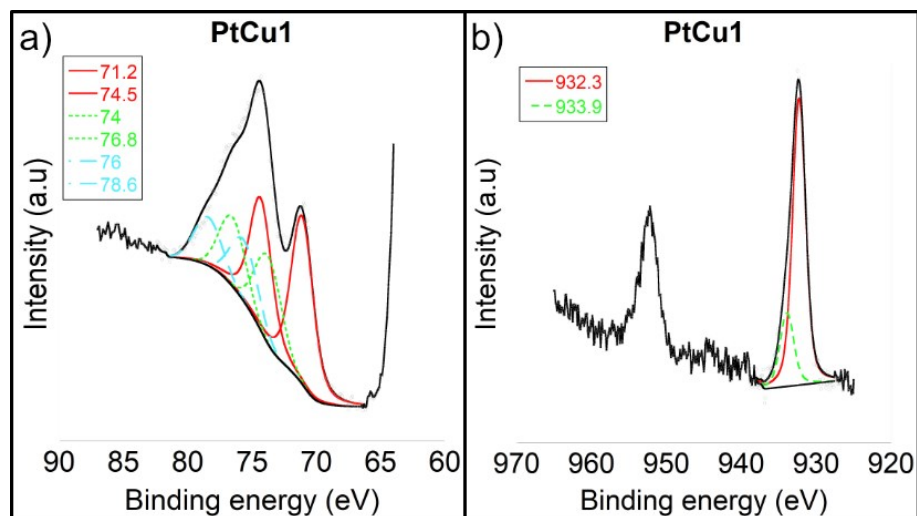
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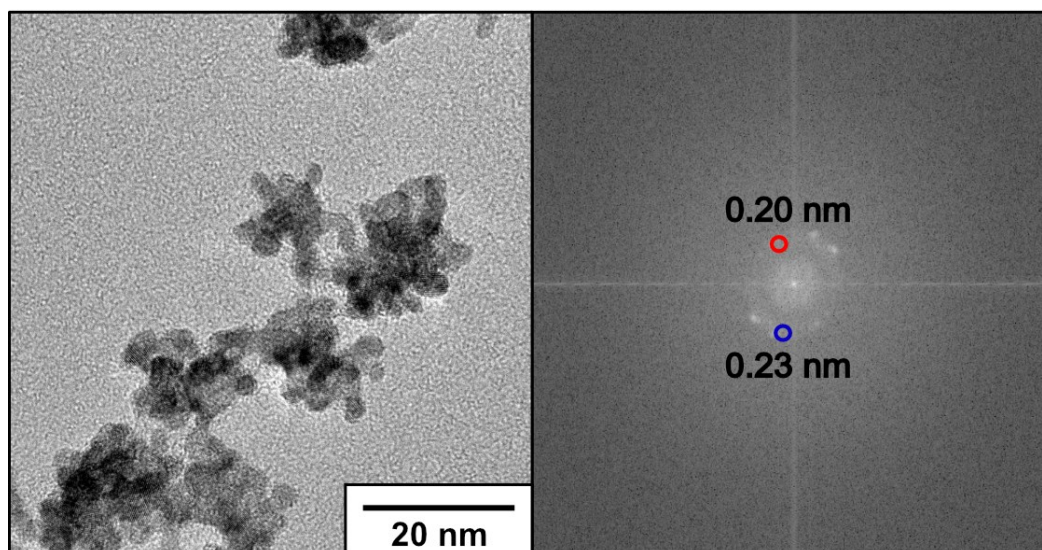
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**Figure S8.** HRTEM of PtCu2 and the corresponding FFT showing spots assigned to Pt (111) (marked as blue) and Cu (111) (marked as red).





## ICP analysis

### Digestion procedure

1 mL of sample was added to 50 mL of Mili-Q water. The sample was kept in agitation during 16 h. It was added  $1.0 \pm 0.1$  mL conc.  $\text{HNO}_3$  and  $0.50 \pm 0.05$  mL conc.  $\text{HCl}$  to each sample. The samples were digested for 3 hours  $95 \pm 5^\circ\text{C}$ . After 3 hours at  $95 \pm 5^\circ\text{C}$ , the samples were taken from heat source and let cool for at least 30 minutes. The volume of the samples was reconstituted to 50 mL with Mili-Q water.

**Table S1.** ICP results of PtCu1 and PtCu2 nanomaterials

Sample	Cu [ppm]	Pt [ppm]
PtCu1	52	150
PtCu2	48	110