

Supplementary information

Isostructural coordination polymers of tethering naphthalene anchored bis(2-methylpyridinecarboxamide) ligand: Single crystal, XPS, EDS and theoretical studies

Love Karan Rana^{1*}, Prabhjyot Kaur¹, Alborz Bavandsavadkouhi¹, Gurpreet Singh Selopal², and Adam Duong¹

Table of contents

1. Comparison of PXRD pattern of asynthesized compound 1 with the simulated pattern (Fig. S1).....	2
2. Comparison of PXRD pattern of asynthesized compound 2 with the simulated pattern (Fig. S2)	2
3. Comparison of PXRD pattern of asynthesized compound 3 with the simulated pattern (Fig. S2)	3
4. BFDH morphologies of L1 , compound 1 (Fig. S4).....	3
5. Diffraction data plots showing weak diffraction from the crystal at higher angles (Fig. S5)	
6. ¹ H-NMR of L1 (Fig. S6).....	4
7. ¹³ C-NMR of L1 (Fig. S7).	4
8. IR spectra comparison involving L1 and compounds 1 , 2 , and 3 (Fig. S8).	5
9. Table S1- I.R table.....	6-7
10. ORTEP diagram of L1 at 40% probability (Fig. S9).	7
11. Showing the dihedral angle between pyridine and naphthalene ring planes of L1 . (Fig. S10)	7
12. ORTEP diagram of compound 1 at 40% ellipsoidal probability (Fig. S11)	7
13. ORTEP diagram of compound 2 at 40% ellipsoidal probability (Fig. S12).	9
14. ORTEP diagram of compound 3 at 40% ellipsoidal probability (Fig. S13).....	9

15. Supramolecular synthons generated in a 2D sheet and 3D hydrogen bonded framework of compound 1 (Fig. S14).....	10
16. XPS spectral peaks for compound 2 (Fig. S15).....	10
17. XPS spectral peaks for compound 3 (Fig. S16).....	11
18. EDS spectra of compound 2 (Fig. S17).....	12
19. EDS spectra of compound 3 (Fig. S18).....	12
20. Thermogravimetry curves for L1 and 1-3 (Fig. S19).....	13
21. Energy gap between excited and ground states of compounds 1 and 3 (Fig. S20).....	13, 14
22. Hydrogen bonds for L1 [Å and deg.] (Table S2)	13
23. Hydrogen bonds for compound 1 [Å and deg.] (Table S3)	14
24. Hydrogen bonds for compound 2 [Å and deg.] (Table S4)	15
25. Hydrogen bonds for compound 3 [Å and deg.] (Table S5)	16

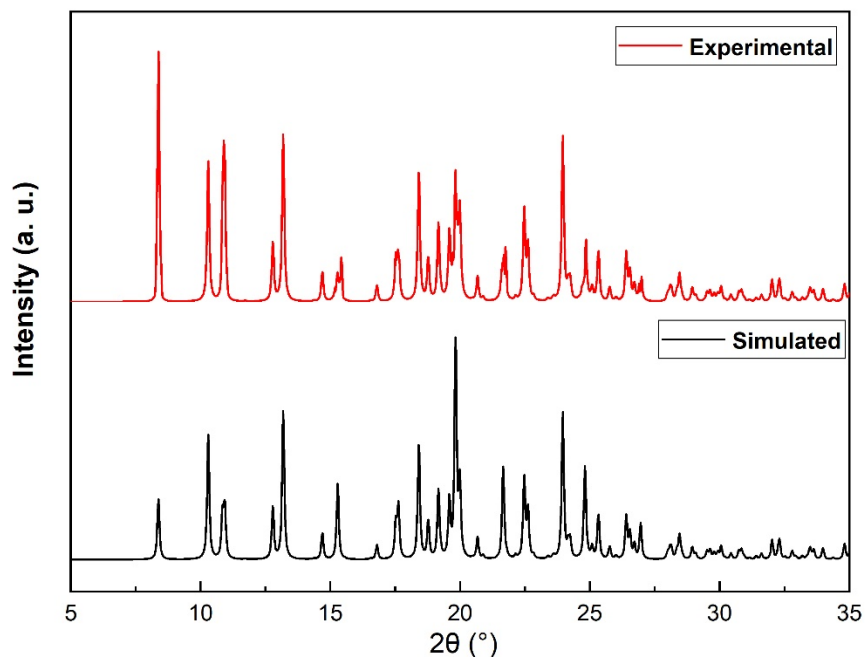


Figure S1 Shows the comparison of powder patterns of compound **1**.

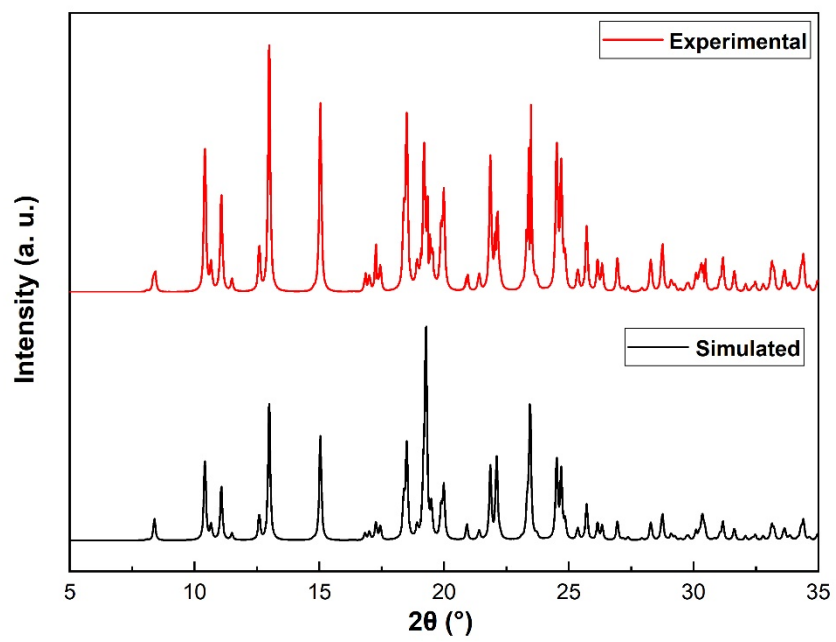


Figure S2 Shows the comparison of powder patterns of compound **2**.

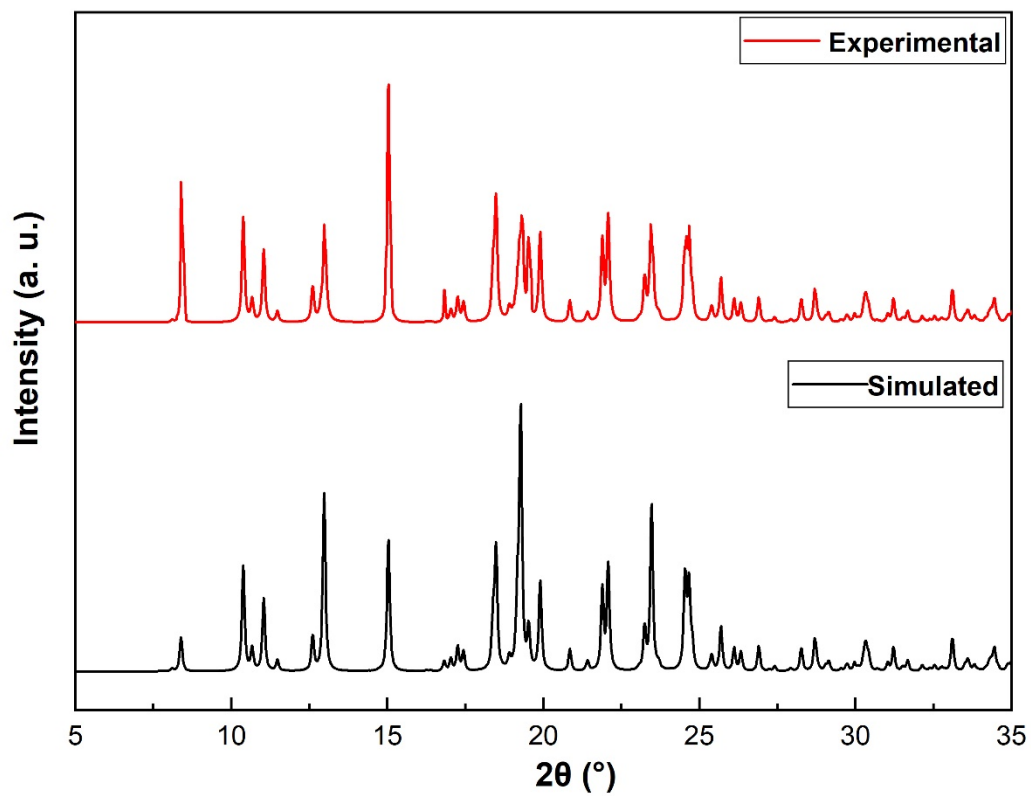


Figure S3 Shows the comparison of powder patterns of compound **3**.

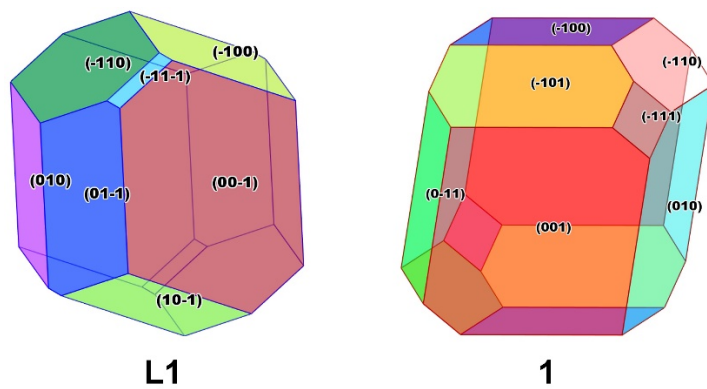


Figure S4 BFDH morphologies of **L1** and compound **1**.

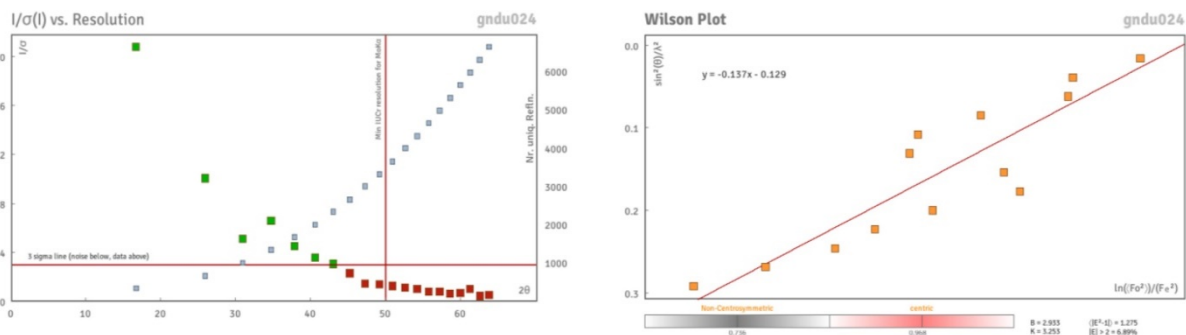


Figure S5 Diffraction data plots showing noise at higher angles ($I/\sigma(I)$ vs resolution plot) and poor quality of data (large deviation from straight line in Wilson plot).

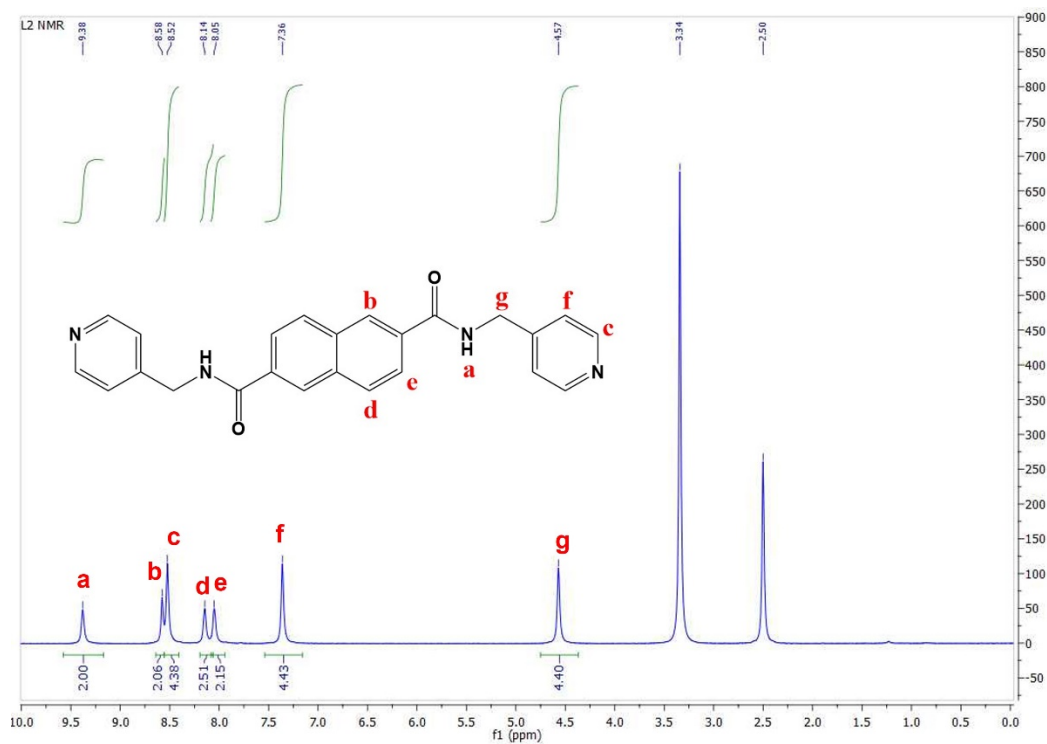


Figure S6 $^1\text{H-NMR}$ of L1.

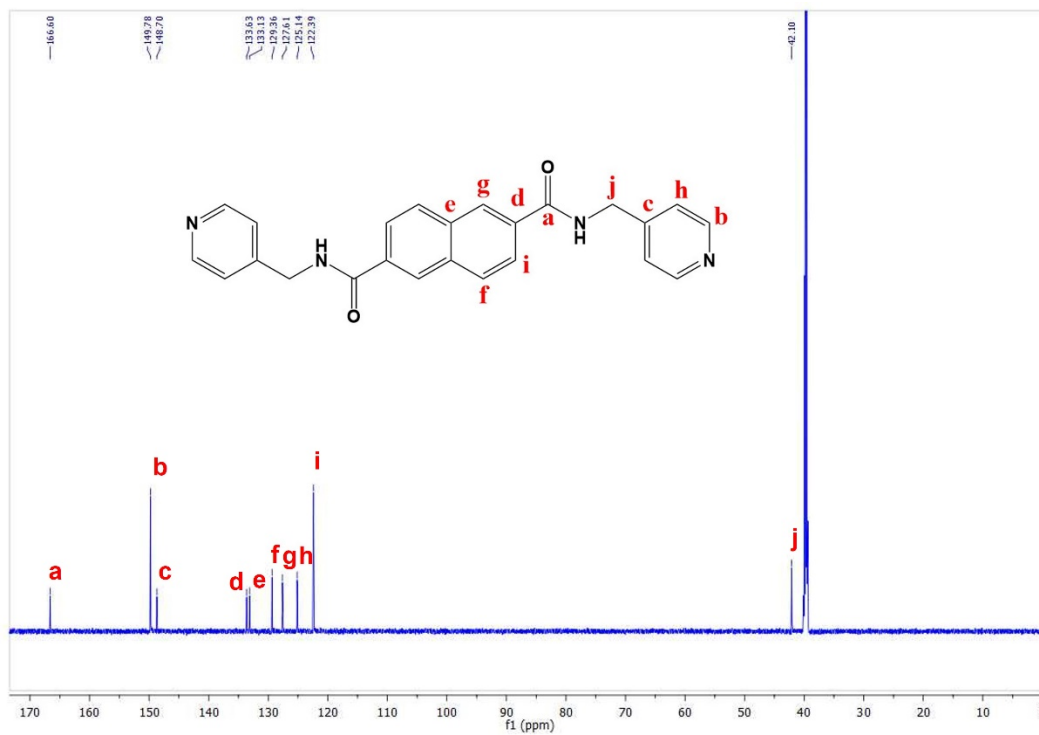


Figure S7 ^{13}C -NMR of L1.

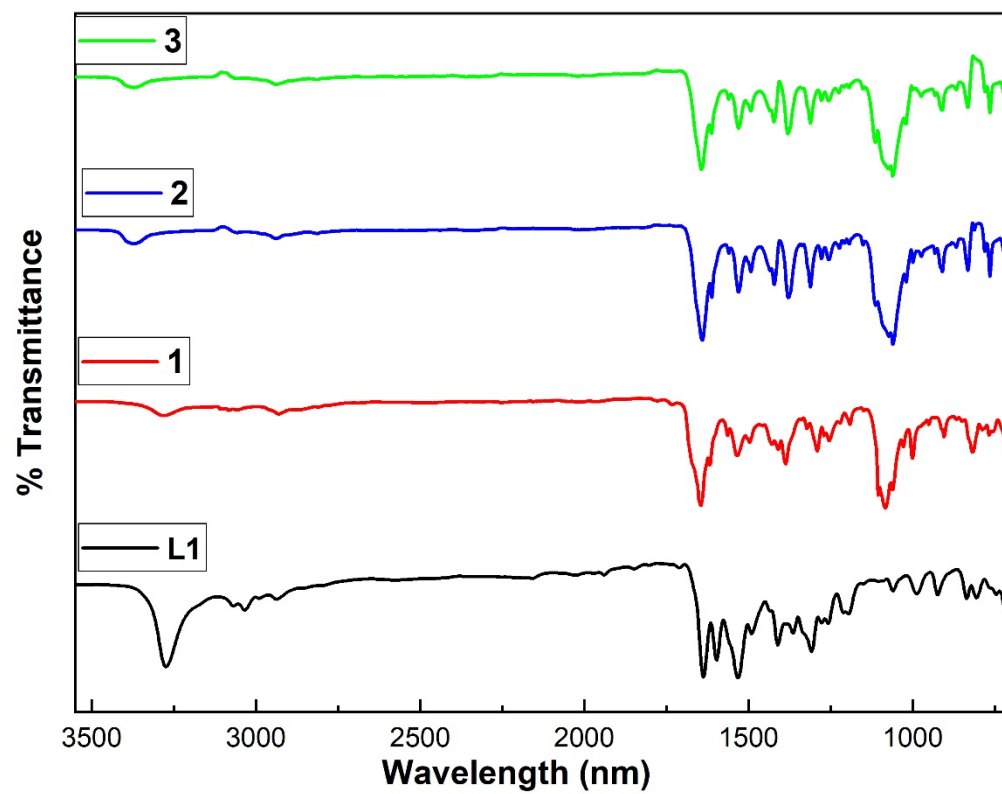


Figure S8 IR spectra comparison involving **L1** and compounds **1**, **2**, and **3**.

Table S1 – Characteristic IR peaks for **L1** and compounds **1**, **2**, and **3**.

Complex	Group	Functionality	Wavenumber (cm⁻¹)
L1	ν (O-H) _{Stretching}	Water	3272 (s)
	ν (N-H) _{Stretching}	Amide	3068 (w,b)
	ν (C=O)	Amide	1639 (s)
	ν (C=N)	Pyridine	1597 (s)
	ν (C=C)	Pyridine	1533 (b)
	ν (C-H) _{Stretching}	Methylene	2937(m)
	ν (C=C)	Naphthalene	1487 (w)
1	ν (N-H)	Amide	3076 (w)
	ν (C=O)	Amide	1645 (s)
	ν (C=N)	Pyridine	1615 (w)
	ν (C=C)	Pyridine	1532 (m, b)
	ν (C-H) _{Stretching}	Methylene	2932 (vw)
	ν (C=C)	Naphthalene	1494 (w)
		ν (Cl-O)	Perchlorate
2	ν (N-H)	Amide	3069 (w)
	ν (C=O)	Amide	1640 (s)
	ν (C=N)	Pyridine	1611 (w)
	ν (C=C)	Pyridine	1533 (s)
	ν (C-H) _{Stretching}	Methylene	2940 (vw)
	ν (C=C)	Naphthalene	1493 (w)
		ν (Cl-O)	Perchlorate
3	ν (N-H)	Amide	3071 (w)
	ν (C=O)	Amide	1643 (s)
	ν (C=N)	Pyridine	1612 (w)
	ν (C=C)	Pyridine	1529 (s)
	ν (C-H) _{Stretching}	Methylene	2935 (vw)
	ν (C=C)	Naphthalene	1492 (w)
		ν (Cl-O)	Perchlorate

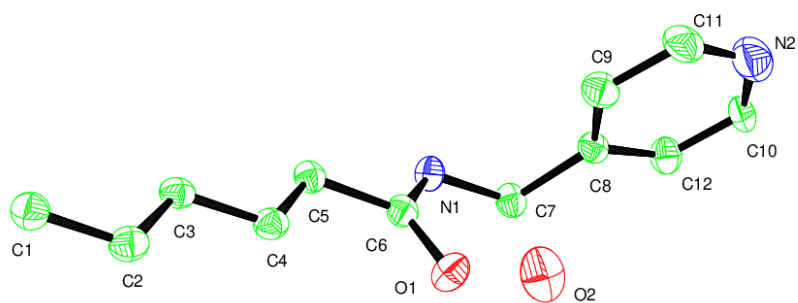


Figure S9 ORTEP diagram of **L1** drawn at 40% ellipsoidal probability (Hydrogen atoms have not been shown for clarity).

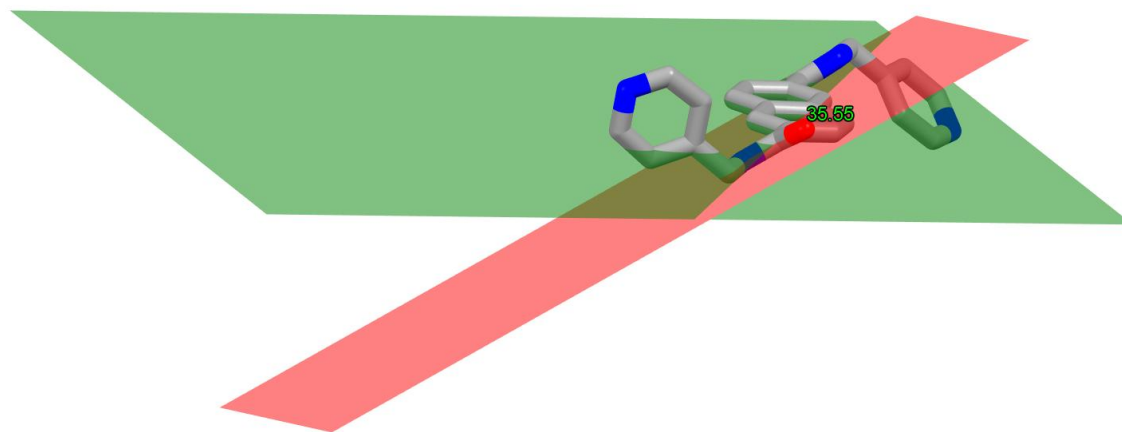


Figure S10 Showing the dihedral angle between pyridine and naphthalene ring planes of **L1**.

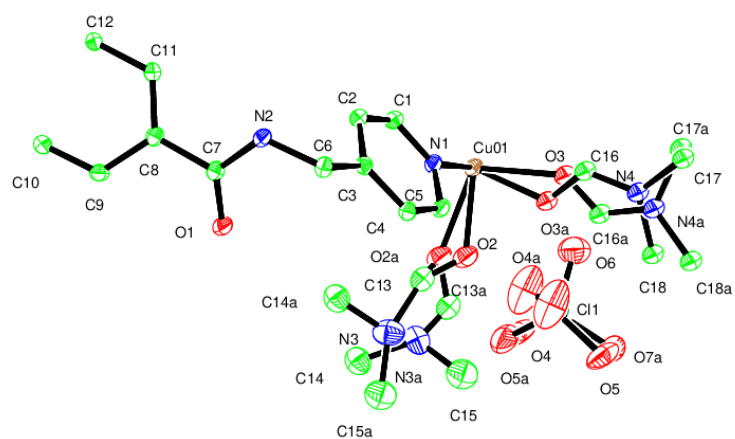


Figure S11 ORTEP diagram of compound **1** at 40% ellipsoidal probability.

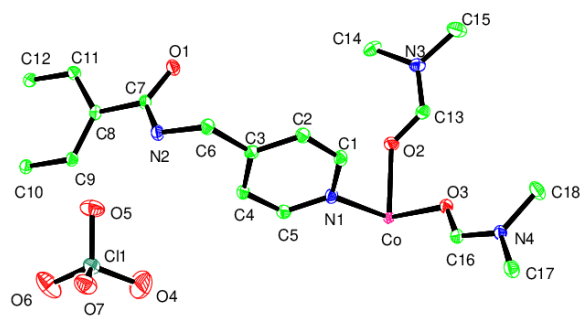


Figure S12 ORTEP diagram of compound **2** at 40% ellipsoidal probability.

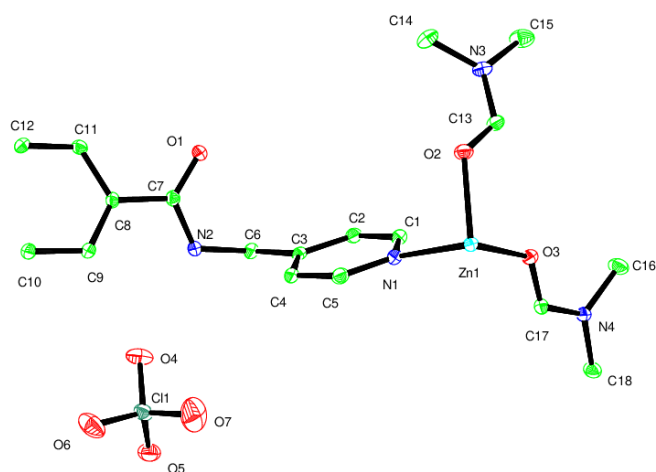


Figure S13 ORTEP diagram of compound **3** at 40% ellipsoidal probability.

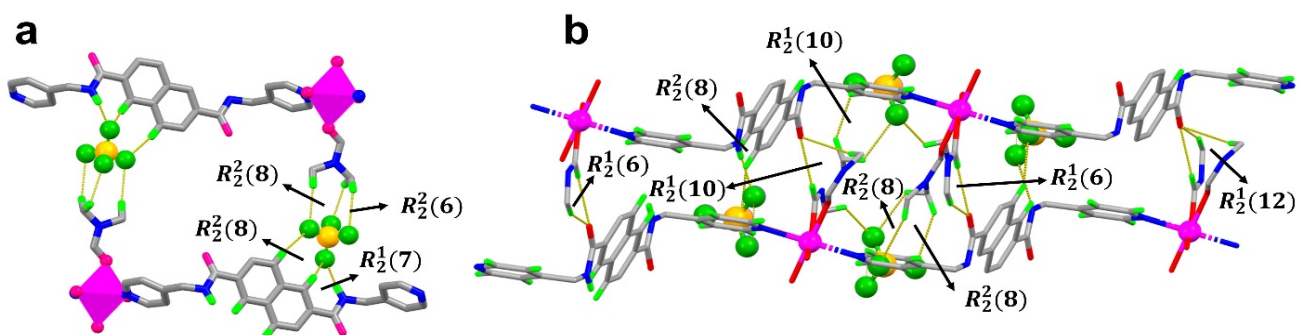


Figure S14 Supramolecular synthons generated in a 2D sheet and 3D hydrogen bonded framework of compound **1**.

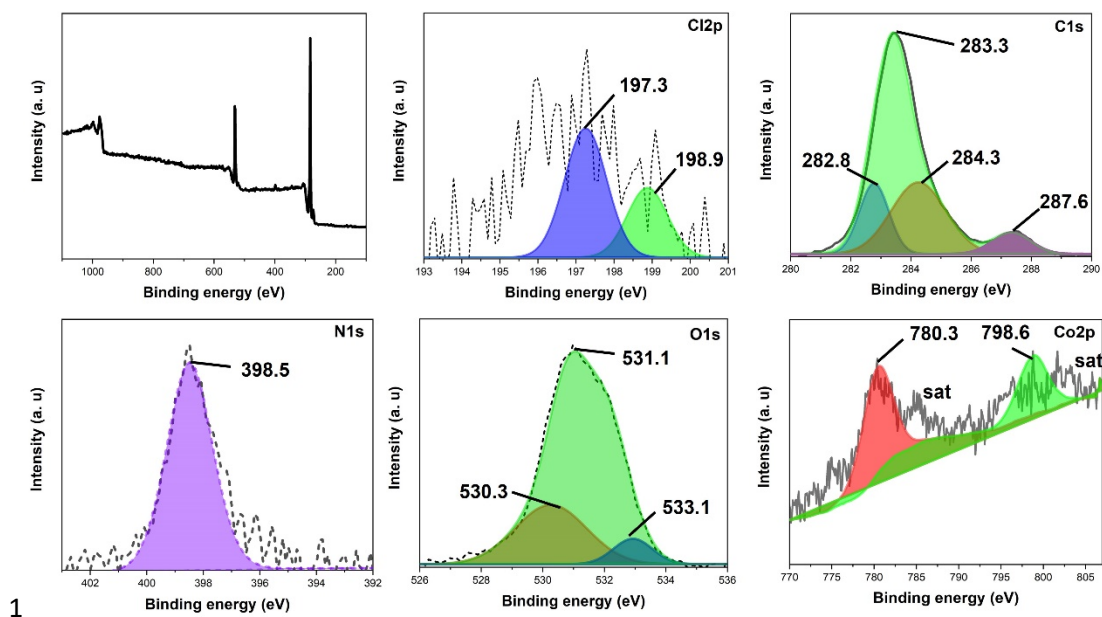


Figure S15 XPS spectral peaks for compound 2.

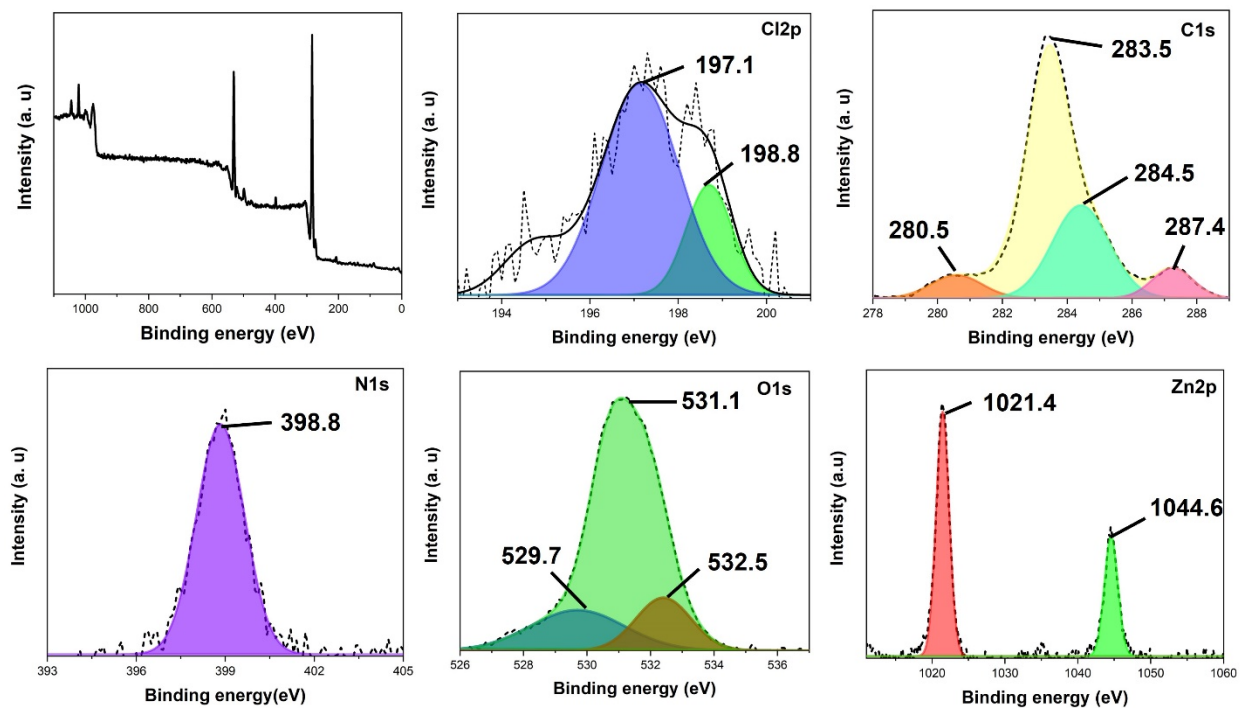


Figure S16 XPS spectral peaks for compound 3.

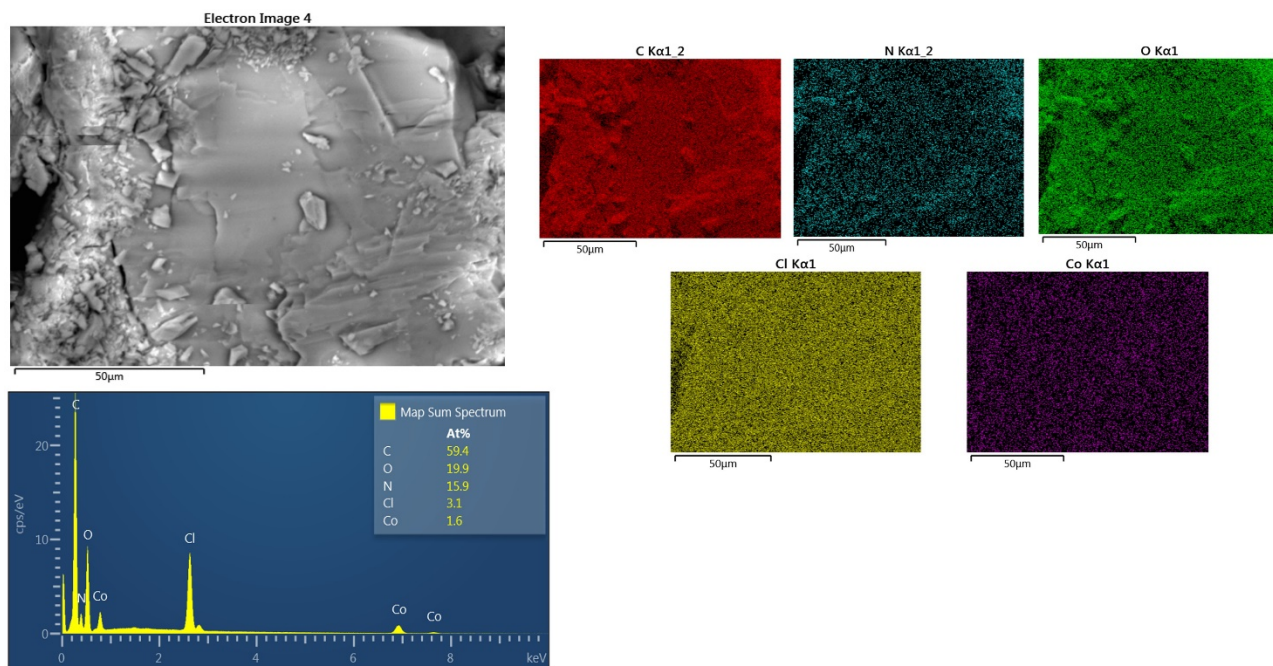


Figure S17 EDS spectra of compound 2

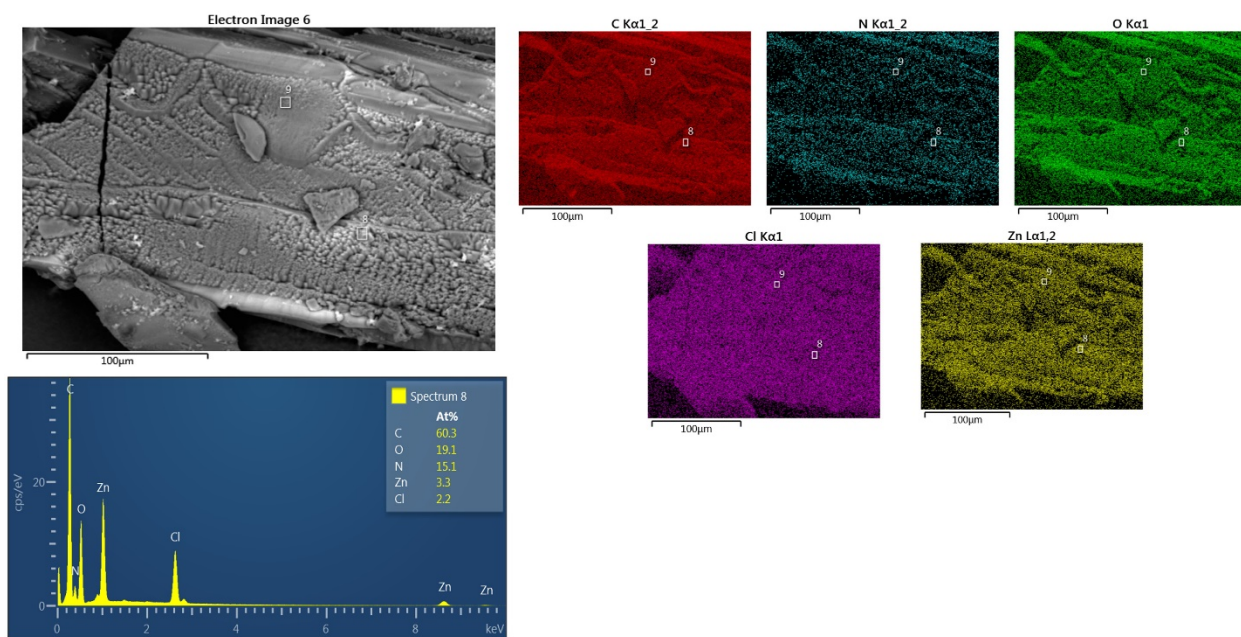


Figure S18 EDS spectra of compound 3

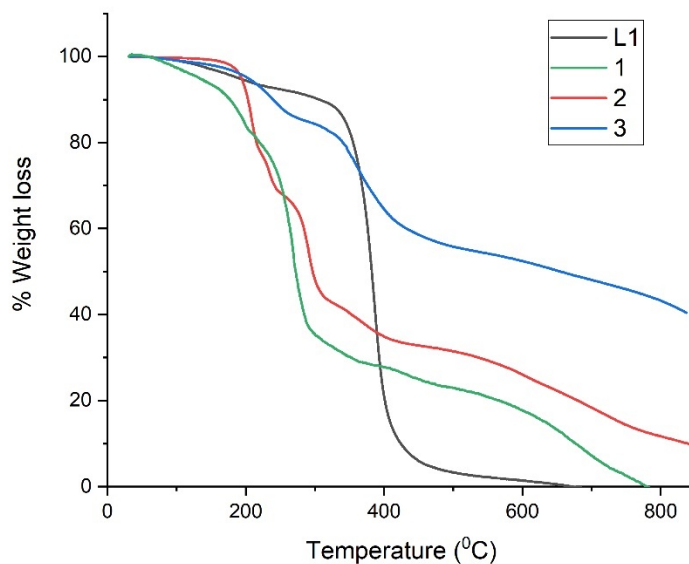


Figure S19 Thermogravimetry curves for **L1** and **1-3**.

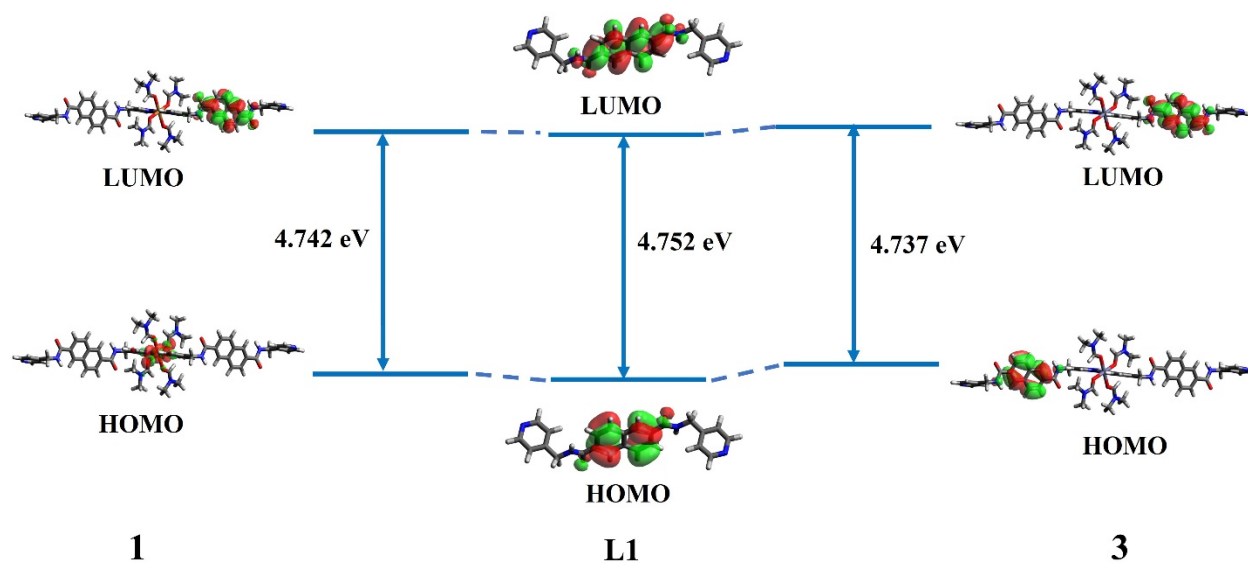


Figure S20 Variation of HOMO-LUMO energy gap in compounds **1** and **3** relative to **L1**. (Note- This figure is showing speculative anomaly in HOMO-LUMO energies of compounds **1** and **3** due to solvent effect of DMF. In both cases, HOMO orbitals are more affected by solvent interactions than LUMO orbitals.)

Table S2. Hydrogen bonds for **L1** [\AA and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
N(1)-H(1)...O(2)#1	0.88	2.02	2.801(4)	147.1
N(3)-H(3)...O(1)#2	0.88	2.05	2.821(4)	146.5
C(3)-H(3A)...O(2)#3	0.95	2.73	3.382(5)	126.6
C(6)-H(6)...N(2)#4	0.95	2.71	3.556(5)	149.3
C(4)-H(4)...N(2)#4	0.95	2.83	3.657(6)	145.8
C(1)-H(1A)...N(4)#5	0.95	2.68	3.538(5)	150.7
C(9)-H(9)...N(4)#5	0.95	2.86	3.681(6)	145.4
C(16)-H(16)...O(1)#6	0.95	2.84	3.640(6)	142.0

Symmetry transformations used to generate equivalent atoms:

#1 $-x+1, -y+2, -z+1$ #2 $-x+1, -y+1, -z+1$ #3 $x+1, y, z$ #4 $x-1, y-1, z-1$ #5 $x+1, y+1, z+1$ #6 $-x+2, -y+2, -z+2$

Table S3. Hydrogen bonds for **1** [\AA and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
C(3)-H(3)...O(4 ^a)#1	0.950	2.574	3.197	123.3
C(3)-H(3)...O(2 ^a)#2	0.950	2.879	3.649	137.16
C(7)-H(7b)...O(5)#3	0.990	2.989	3.590	120.21
C(9)-H(9)...O(4 ^a)#4	0.950	2.862	3.687	145.88
N(2)-H(2)...O(1 ^a)#4	0.880	2.081	2.930	160.64

C(16 ^b)-H(17 ^a)...O(2 ^a)#5	0.880	2.177	2.904	139.64
C(17 ^b)-H(17e ^a)...O(3a ^b)#6	0.879	2.675	3.510	159.21
C(17a ^a)-H(17e ^a)...O(1 ^a)#6	0.980	2.977	3.899	156.4
C(11)...O(11)#6	0.950	2.439	3.292	149.34
C(18b ^b)-H(18a ^b)...O5 #7	0.980	2.418	3.314	151.68
C(18a ^b)-H(18e ^a)...O5 #7	0.882	2.519	3.314	150.25
C(14a ^a)-H(14c ^b)...O4a ^b #7	1.080	2.589	3.686	140
C(18a ^a)-H(18e)...O#7	0.980	2.519	3.182	160.5
C(15a ^a)-H(14b ^b)...N3 ^b #8	1.416	2.660	3.753	131.59
C(15a ^a)-H(15f ^a)...N3a ^a #8	0.980	2.770	3.744	172.91
C(14 ^b)-H(14b ^b)...N3 ^b #8	0.980	2.660	3.631	170.66
C(15 ^b)-H(15c ^b)...N(3a ^a)#9	0.980	2.874	3.795	156.84
C(14a ^a)-H(14e ^b)...O(3 ^b)#9	0.98	2.597	3.537	160.9

Symmetry transformations used to generate equivalent atoms:

#1 -x+2,-y,-z #2 -x+1,+y,+z #3 -x+2,-y+1,-z #4 -x+1,-y,-z #5 -x+1,-y+1,-z+1 #6 x,y+1,z
 #7 x-1,+y,+z #8 -x,y+1,-z+1 #9 -x+1,-y,-z+1

Table S4. Hydrogen bonds for **2** [Å and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
C(17)-H(17C)...O(1)#3	0.96	2.49	3.412(3)	160.0
C(16)-H(16)...O(1)#3	0.93	2.52	3.383(3)	154.6

C(4)-H(4)...O(6)#4	0.93	2.88	3.508(3)	126.2
C(5)-H(5)...O(6)#4	0.93	2.90	3.536(3)	126.4
C(1)-H(1)...O(5)#5	0.93	2.88	3.527(3)	127.5
C(1)-H(1)...O(7)#5	0.93	2.56	3.401(3)	149.9
C(2)-H(2A)...O(4)#5	0.93	2.83	3.601(4)	141.2
C(12)-H(12)...O(6)#2	0.93	2.93	3.740(4)	147.1
C(14)-H(14C)...N(4)#6	0.96	2.91	3.767(3)	149.3
C(18)-H(18B)...O(7)#7	0.96	2.58	3.283(3)	130.0
C(14)-H(14A)...O(1)#8	0.96	2.57	3.383(3)	142.4
C(14)-H(14B)...O(6)#9	0.96	2.75	3.677(4)	162.4
C(15)-H(15A)...O(7)#9	0.96	2.64	3.290(4)	125.7
C(15)-H(15B)...O(4)#9	0.96	2.84	3.628(5)	139.6

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y+1,-z+2 #2 -x+2,-y,-z+1 #3 x-1,y,z #4 -x+1,-y,-z+1 #5 -x+1,-y+1,-z+1 #6
x+1,y,z #7 x,y+1,z+1 #8 -x+2,-y+1,-z+2 #9 x+1,y+1,z+1

Table S5. Hydrogen bonds for **3** [A and deg.].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
C(17)-H(17)...O(1)#3	0.93	2.53	3.391(3)	154.4
C(18)-H(18A)...O(1)#3	0.96	2.50	3.411(3)	158.8
C(4)-H(4)...O(6)#4	0.93	2.88	3.505(3)	125.9

C(5)-H(5)...O(6)#4	0.93	2.91	3.535(3)	126.2
C(1)-H(1)...O(4)#5	0.93	2.90	3.538(3)	127.0
C(1)-H(1)...O(5)#5	0.93	2.56	3.390(3)	149.1
C(2)-H(2A)...O(7)#5	0.93	2.83	3.600(4)	140.8
C(12)-H(12)...O(6)#2	0.93	2.94	3.755(3)	147.3
C(14)-H(14A)...N(4)#6	0.96	2.89	3.769(3)	151.9
C(16)-H(16B)...O(5)#7	0.96	2.58	3.297(3)	131.1
C(14)-H(14B)...O(6)#8	0.96	2.77	3.677(4)	158.8
C(15)-H(15B)...O(7)#8	0.96	2.87	3.662(4)	140.1
C(15)-H(15C)...O(5)#8	0.96	2.64	3.283(3)	125.0
C(14)-H(14C)...O(1)#9	0.96	2.59	3.382(3)	139.6

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y+1,-z #2 -x,-y+2,-z+1 #3 x+1,y,z #4 -x+1,-y+2,-z+1 #5 -x+1,-y+1,-z+1 #6 x-
1,y,z #7 x,y-1,z-1 #8 x-1,y-1,z-1 #9 -x,-y+1,-z