

# **One pot synthesis and characterization of binary and ternary metal organic frameworks (MOFs) as tri-modal catalysts for thiophene electrooxidation, water splitting and 4-nitrophenol reduction**

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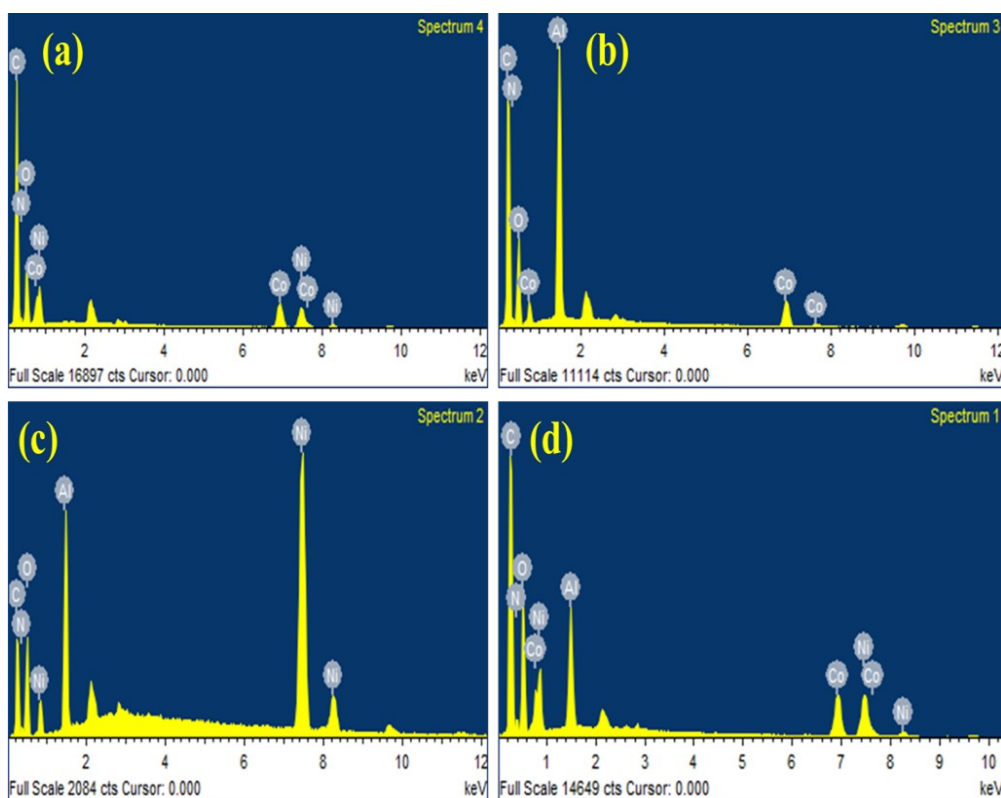
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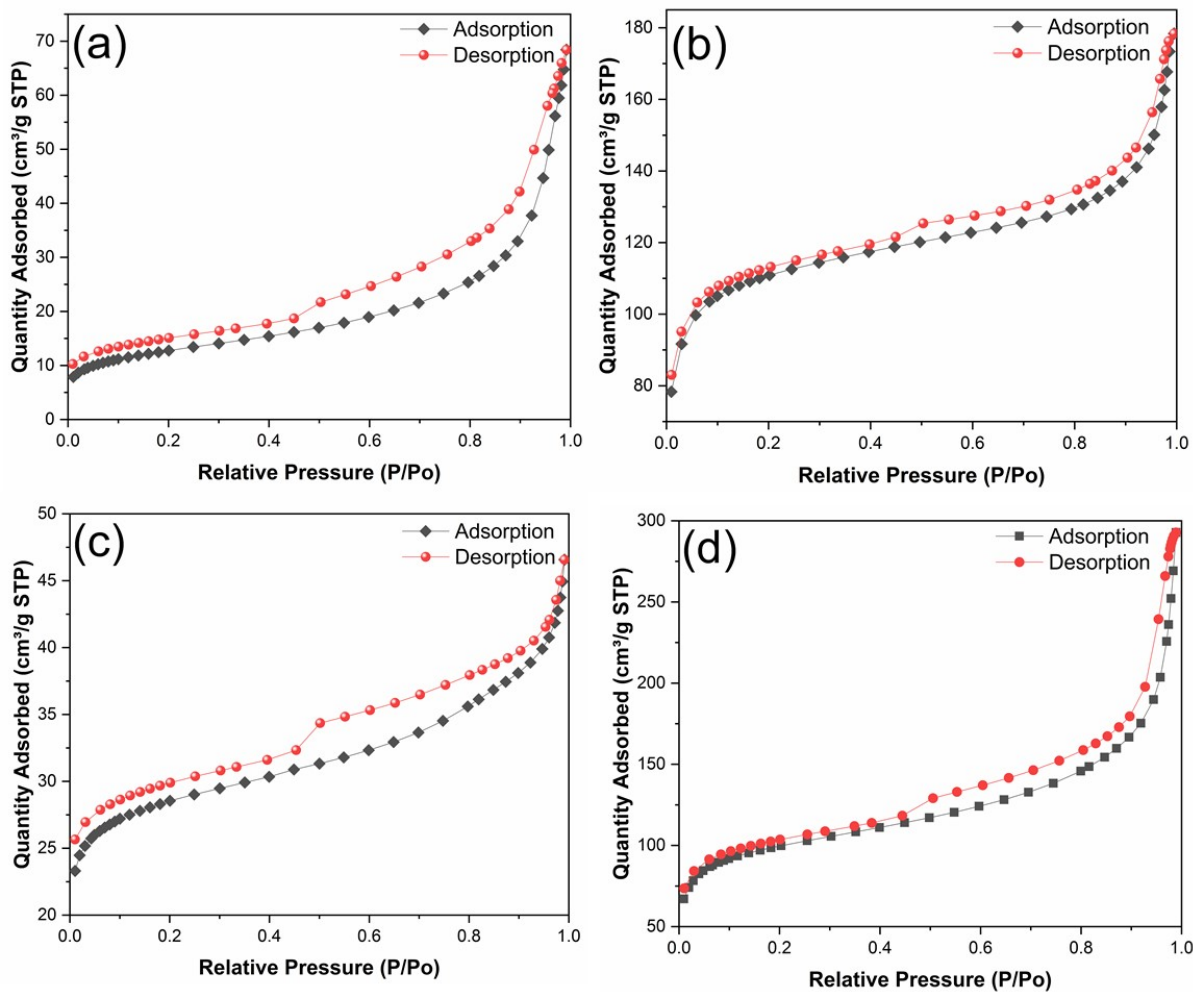
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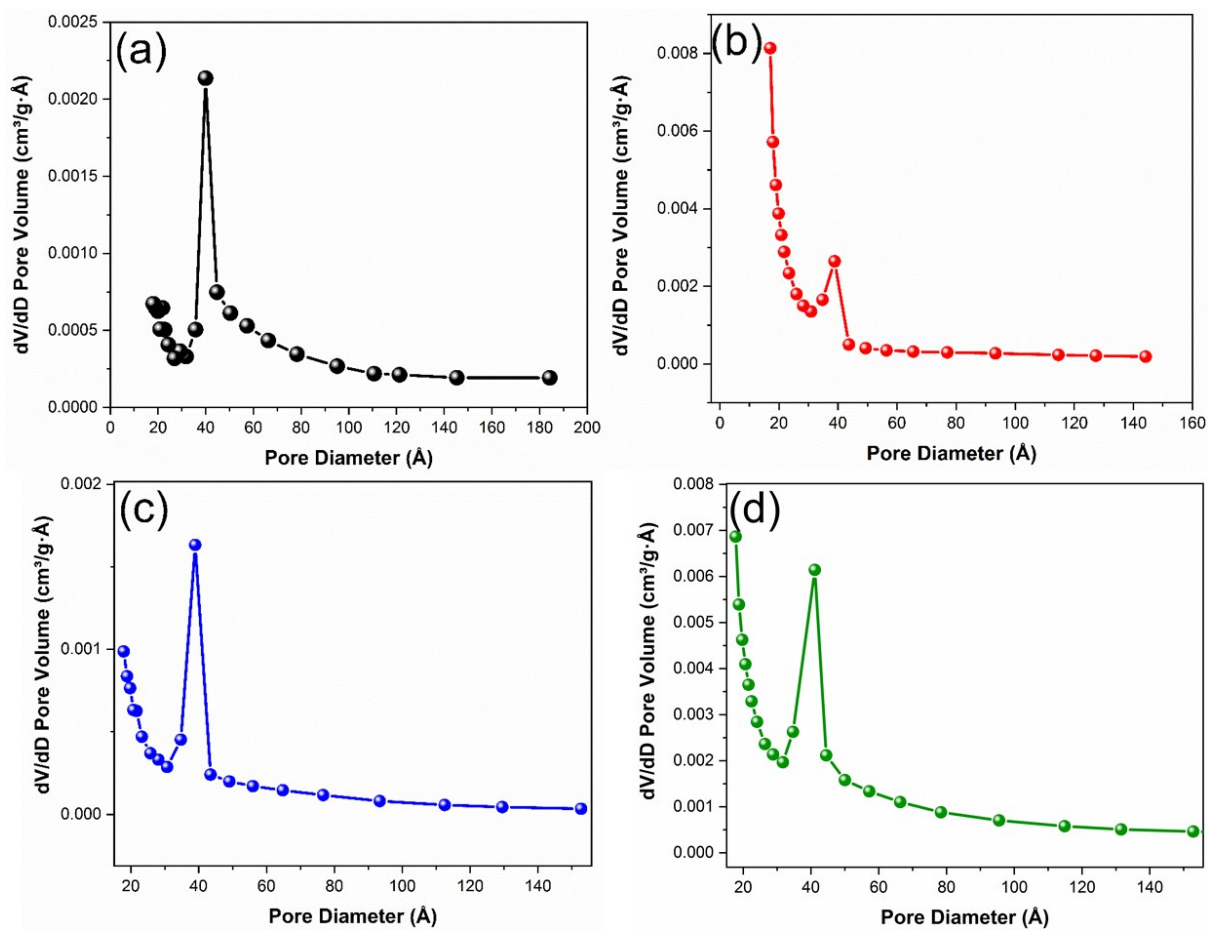
**Figure S1.** EDAX images of a) Co-Ni-MOF, b) Co-Al-MOF c) Ni-Al-MOF, and d)Co-Al-Ni-MOF.

**Table S1.** EDAX analysis of the synthesized composite weight percentage

| S.No. | Catalyst     | Element weight (%) |      |       |       |      |       |
|-------|--------------|--------------------|------|-------|-------|------|-------|
|       |              | Al                 | Co   | Ni    | C     | N    | O     |
| 1     | Co-Ni-MOF    | -                  | 7.71 | 7.22  | 53.59 | 9.20 | 22.29 |
| 2     | Co-Al-MOF    | 10.49              | 6.22 | -     | 51.85 | 7.15 | 24.30 |
| 3     | Ni-Al-MOF    | 8.70               | -    | 49.57 | 25.92 | 3.72 | 12.09 |
| 4     | Co-Al-Ni-MOF | 3.94               | 7.41 | 8.61  | 44.66 | 8.83 | 26.54 |



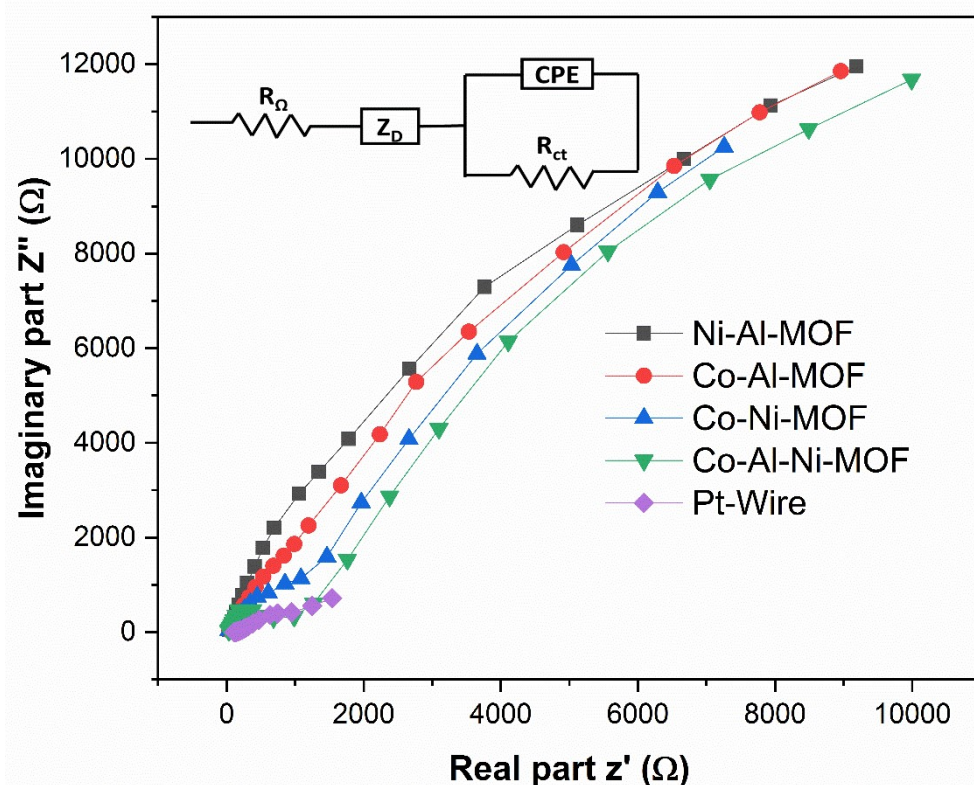
**Figure S2.** BET results of a) Ni-Al-MOF, b) Co-Al-MOF, c) Co-Ni-MOF and d) Co-Al-Ni-MOF.



**Figure S3.** BET pore size distribution plot of a) Ni-Al-MOF, b) Co-Al-MOF, c) Co-Ni-MOF and d) Co-Al-Ni-MOF.

**Table S2.** BET surface area ( $\text{m}^2 \text{g}^{-1}$ ), pore volume ( $\text{cm}^3 \text{g}^{-1}$ ) and mean pore diameter ( $\text{\AA}$ ) of Ni-Al-MOF, Co-Al-MOF, Co-Ni-MOF and Co-Al-Ni-MOF obtained using the  $\text{N}_2$  absorption-desorption isotherms.

| S.No. | Catalyst     | BET surface area ( $\text{m}^2 \text{g}^{-1}$ ) | pore volume ( $\text{cm}^3 \text{g}^{-1}$ ) | mean pore diameter ( $\text{\AA}$ ) |
|-------|--------------|---|---|-------------------------------------|
| 1     | Ni-Al-MOF    | 7.8144  | 0.1058                                      | 40                                  |
| 2     | Co-Al-MOF    | 19.8026   | 0.2758                                      | 38.77                               |
| 3     | Co-Ni-MOF    | 6.0303  | 0.0720                                      | 39.26                               |
| 4     | Co-Al-Ni-MOF | 38.777  | 0.4528                                      | 41.58                               |



**Figure S4.** Impedance curves for synthesized MOFs for electrooxidation of thiophene

**Table S3** Summary of comparative performance of Ni and Co based MOF catalysts recently reported for water splitting along with the present one.

| No. | Catalyst  | Electrolyte                          | Overpotential [mV] | Tafel slope [mV dec <sup>-1</sup> ] | References |
|-----|---|--------------------------------------|--------------------|-------------------------------------|------------|
| 1   | Ni-MOF  | 0.1 M KOH                            | 296                | 45                                  | 1          |
| 2   | Ni-BTC  | 0.1 M KOH                            | 330                | 63                                  | 2          |
| 3   | Ni-MOF-74   | 1.0 M KOH                            | 313                | 134.1                               | 3          |
| 4   | Ni-MOF  | 1.0 M KOH                            | 268                | 132.5                               | 4          |
| 5   | 3D NibpyfcdHp   | 0.5 M H <sub>2</sub> SO <sub>4</sub> | 350                | 60                                  | 5          |
| 6   | 3D CobpyfcdHp   | 0.5 M H <sub>2</sub> SO <sub>4</sub> | 400                | 65                                  | 5          |
| 7   | Co <sub>2</sub> (Hpycz) <sub>4</sub> ·4H <sub>2</sub> O | 0.5 M H <sub>2</sub> SO <sub>4</sub> | 223                | 121                                 | 6          |
| 8   | THTA–Co<br>H <sub>3</sub> [Co <sub>3</sub> (tht)(tha)]  | 0.5 M H <sub>2</sub> SO <sub>4</sub> | 283                | 71                                  | 7          |
| 9   | H <sub>3</sub> [Ni <sub>3</sub> (tht)(tha)]             | 0.5 M H <sub>2</sub> SO <sub>4</sub> | 315                | 76                                  | 7          |
| 10  | CoNi-MOF  | 1.0 M KOH                            | 265                | 56                                  | 8          |
| 11  | Ni/Co(10:1)-MOFs  | 1.0 M KOH                            | 248                | 40.92                               | 9          |
| 12  | Co, 0.3Ni   | 1.0 M KOH                            | 330                | 66                                  | 10         |
| 13  | CoNi-MOF/rGO  | 1.0 M KOH                            | 318                | 48                                  | 11         |
| 14  | NiCo-UMOFNs   | 1.0 M KOH                            | 250                | 42                                  | 12         |
| 15  | CoNi-MOF  | 1.0 M KOH                            | 304                | 89.7                                | 13         |
| 16  | 3D Co/Ni-MOFs   | 0.5 M H <sub>2</sub> SO <sub>4</sub> | 357                | 107                                 | 14         |
| 17  | Ni/Co(II)<br>MOFs/PPPT                                  | 0.5 M H <sub>2</sub> SO <sub>4</sub> | 369                | 127.1                               | 15         |
| 18  | Ni-Co-Al -MOF   | 1.0 M KOH (OER)                      | 220                | 97                                  | This work  |
|     |   | 1.0 M KOH (HER)                      | 174                | 108                                 | This work  |

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