

Supporting Information for

Hierarchically porous $\text{Ce}_x\text{Zr}_{1-x}\text{O}_2$ prepared by solvent volatilization for high-efficient synthesis of DMC from CO_2 and methanol

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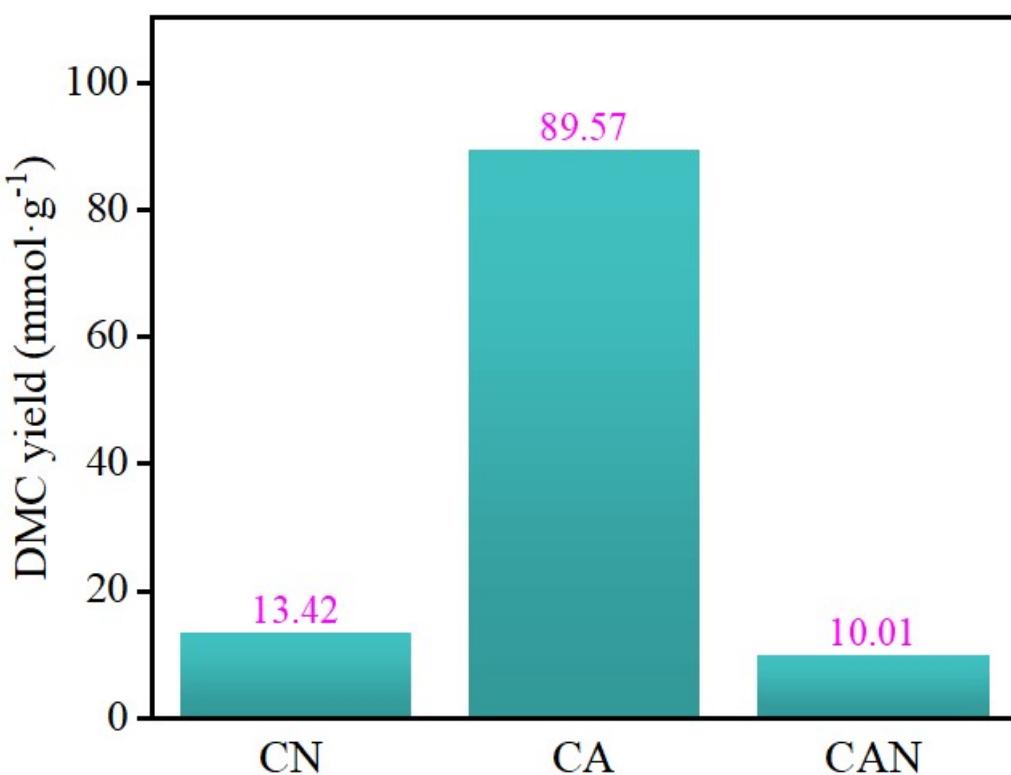


Fig. S1 Catalytic performance of CeO_2 catalysts prepared from different metal salts.
(CN: cerium nitrate, CAN: cerium ammonium nitrate, CA: cerium acetate)

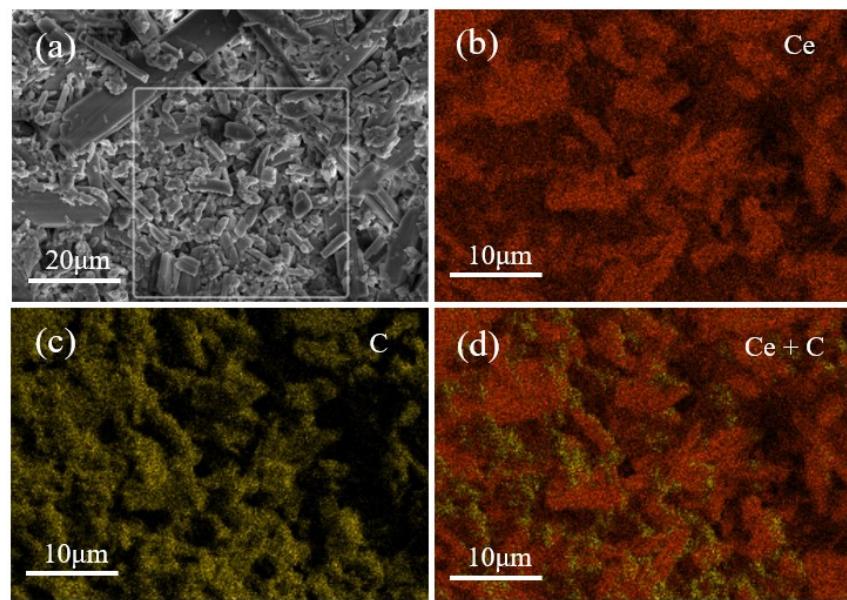


Fig. S2 The SEM mapping of the precursor of sample $\text{Ce}_{0.9}\text{Zr}_{0.1}\text{O}_2$ prepared with cerium acetate as cerium source.

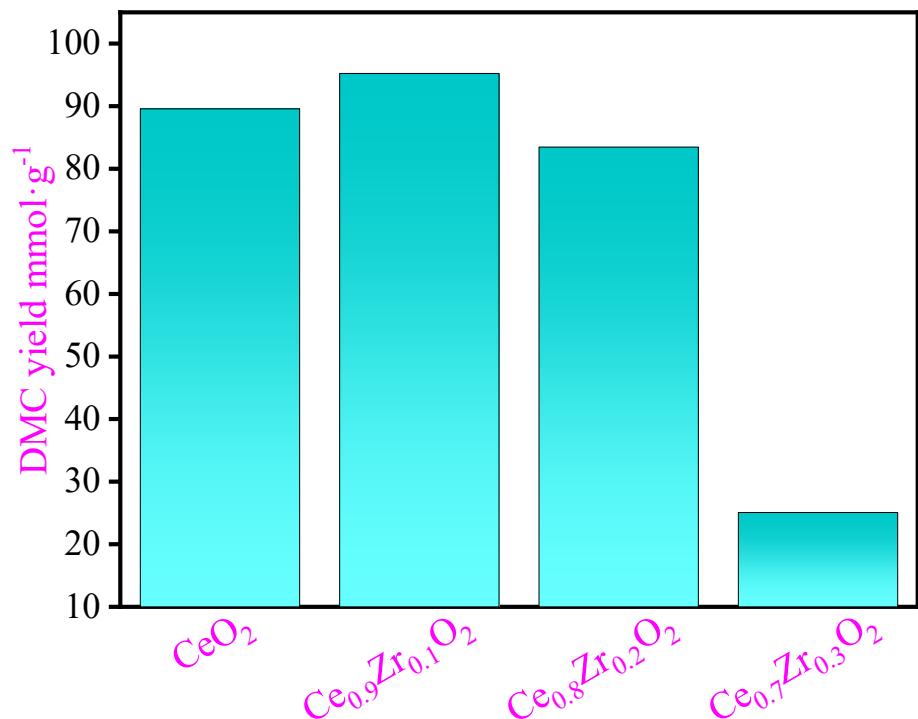


Fig. S3 Catalytic performance of $\text{Ce}_x\text{Zr}_{1-x}\text{O}_2$ samples. Reaction conditions: catalyst 0.1 g, CH_3OH 15mL, 2-CP 3g, 5h.

Tab. S1 The acid-basic properties of synthesized $\text{Ce}_x\text{Zr}_{1-x}\text{O}_2$ catalysts.

Catalyst	Basicity ^a ($\mu\text{mol/g}$)				Acidity ^b ($\mu\text{mol/g}$)			
	B_W (<200 ^c)	B_M (200~400)	B_S (>400)	B_T	A_W (<200)	A_M (200~400)	A_S (>400)	Total
CeO_2	192.04	64.28	13.66	269.98	54.24	40.39	15.83	110.47
$\text{Ce}_{0.9}\text{Zr}_{0.1}\text{O}_2$	235.85	80.87	5.63	322.35	63.3	40.69	17.84	113.03
$\text{Ce}_{0.8}\text{Zr}_{0.2}\text{O}_2$	196.37	32.1	5.82	234.28	42.21	26.81	11.95	80.96
$\text{Ce}_{0.7}\text{Zr}_{0.3}\text{O}_2$	190.86	24.06	6.42	221.33	27.12	18.01	8.01	53.14

^a Calculated by CO_2 -TPD; ^b Calculated by NH_3 -TPD; ^c The unit of temperature is $^{\circ}\text{C}$.



Fig. S4 The digital photographs of samples: (a₁-a₄) $\text{Ce}_x\text{Zr}_{1-x}\text{O}_2$ precursors and (b₁-b₄) $\text{Ce}_x\text{Zr}_{1-x}\text{O}_2$ materials ($x=1-0.7$, for the sample number from 1 to 4).

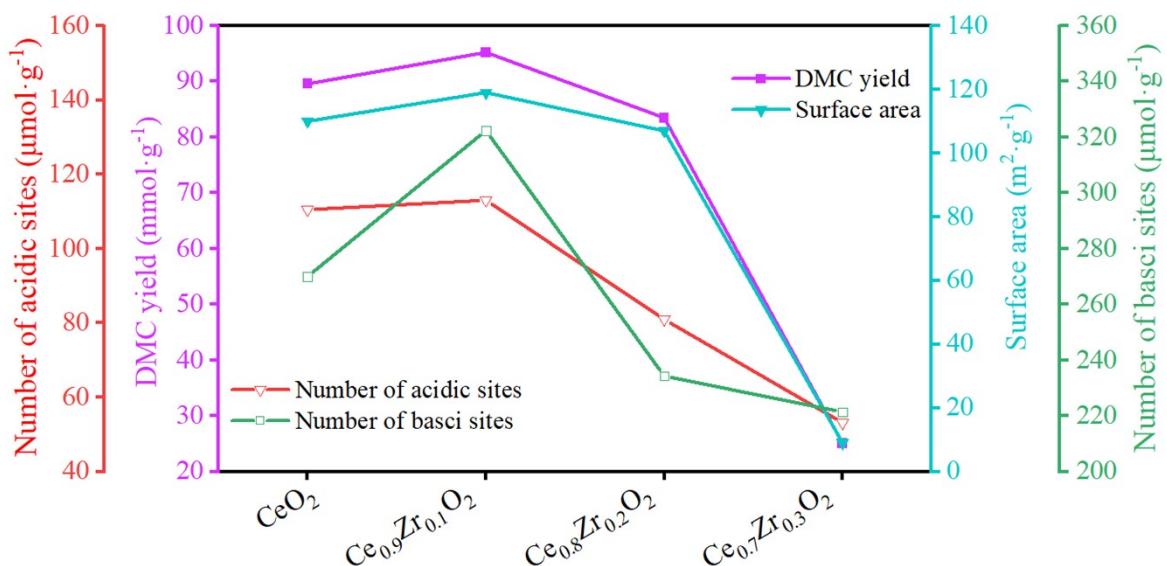


Fig. S5 The relationship between catalytic performance and acid-base site content of $\text{Ce}_x\text{Zr}_{1-x}\text{O}_2$.

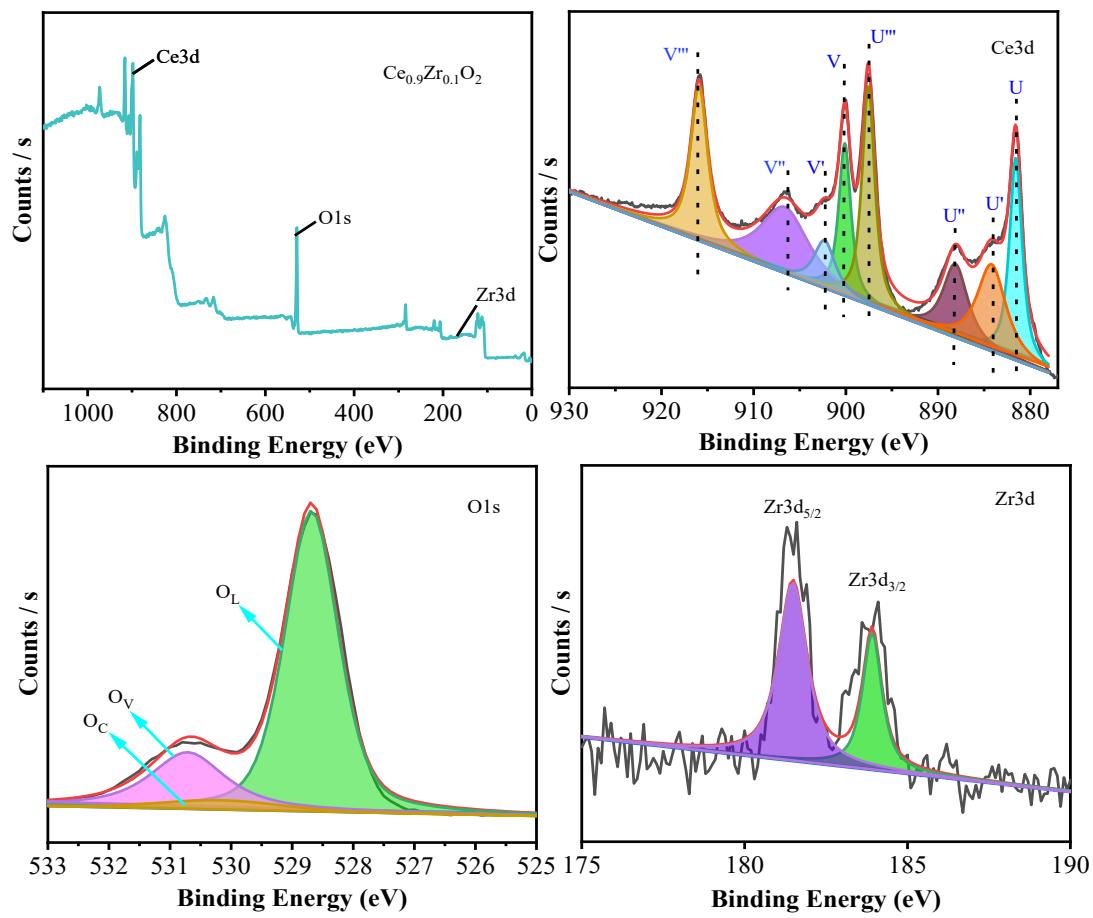


Fig. S6 XPS spectra of (a) full spectra; (b) Ce3d; (c) O1s and (d) Zr3d of $\text{Ce}_{0.9}\text{Zr}_{0.1}\text{O}_2$.