

Noncovalent interactions of antitumor cycloplatinated complexes containing trifluoroacetate ligand as leaving group with bovine serum albumin. Implications for drug design

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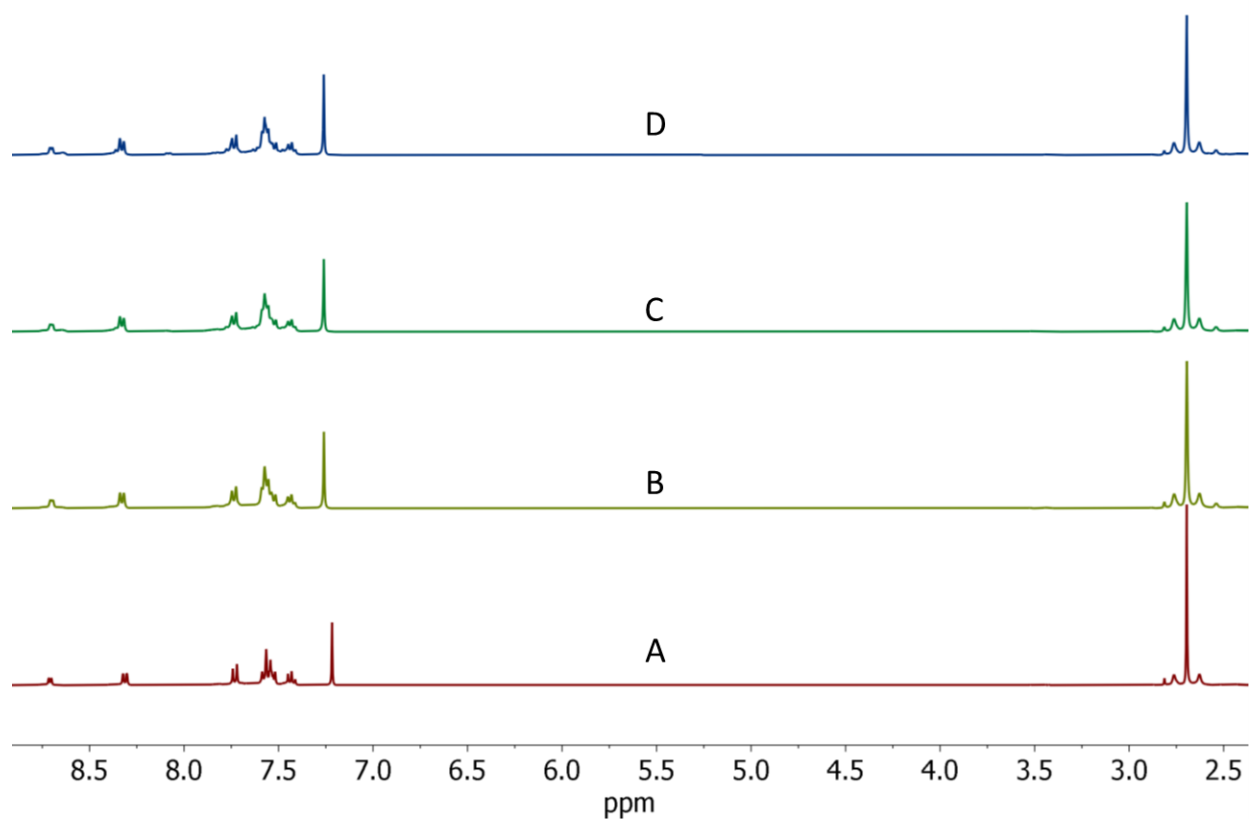


Figure S1. Changes in ^1H NMR spectrum of complex **1** (0.01 M) during its reaction with 10-fold excess of dimethyl sulfoxide- d_6 (0.1 M) in CDCl_3 . (A) Pure **1**, (B) 10 min (C) 1 h and (D) 24 h after addition of DMSO.

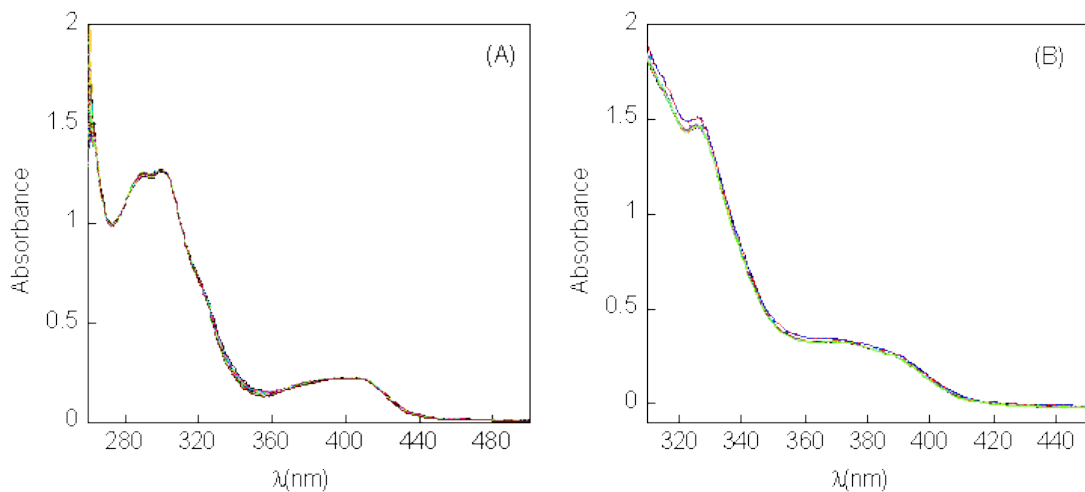


Figure S2. The changes in UV/Vis spectra of (A) complex 1 (3×10^{-4} M) and (B) complex 2 (3×10^{-4} M) in CHCl_3 after the addition of 10-fold excess of DMSO (3.0 mM).

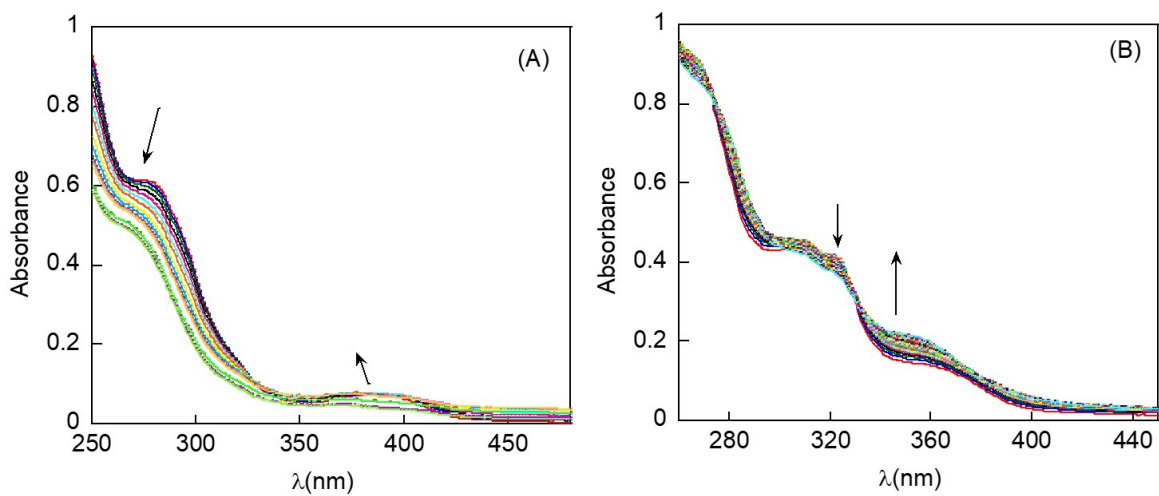


Figure S3. Changes in UV/Vis spectra of (A) complex 1 (90 μM) and (B) complex 2 (75 μM) in 10 % v/v DMSO/ H_2O solution for 5 h.

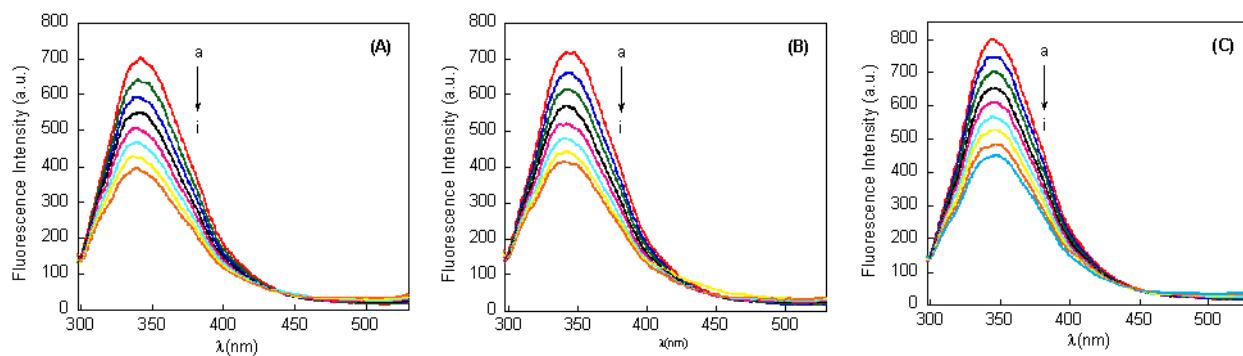


Figure S4. Changes in the fluorescence spectra of BSA upon titration with **1** at (A) 303 K, (B) 310 K and (C) 313 K. The concentration of BSA is 0.1×10^{-6} M and **1** concentration was varied from (a) 0.00 to (i) 0.15×10^{-6} M; pH 7.4 and λ_{ex} : 280 nm.

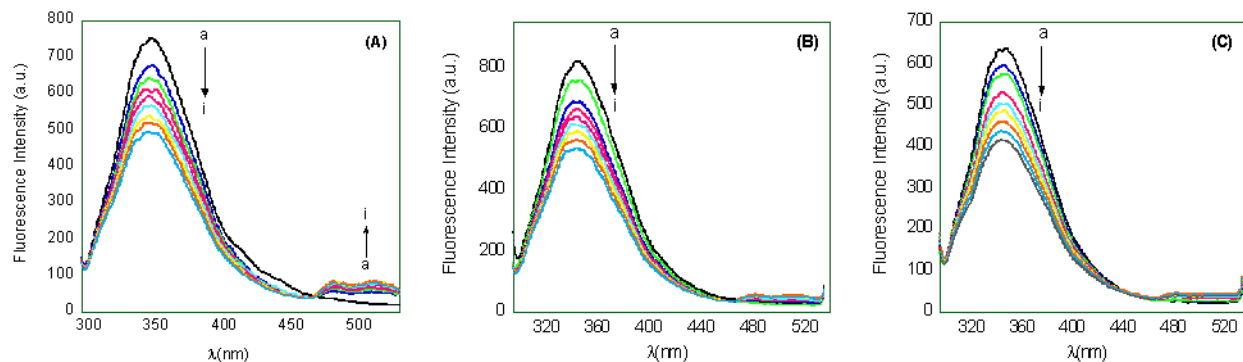


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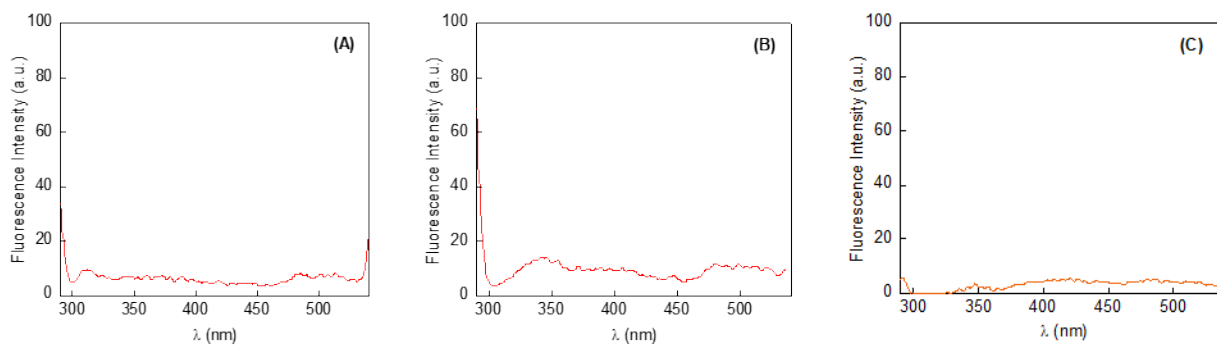


Figure S6. Fluorescence emission spectra of **2**, 0.15 μM (A) 1.5 μM (B) and 2-phenylpyridine, 0.15 μM (C), T = 298 K, pH = 7.40, λ_{ex} = 280 nm, in 0.1M Tris-HCl.

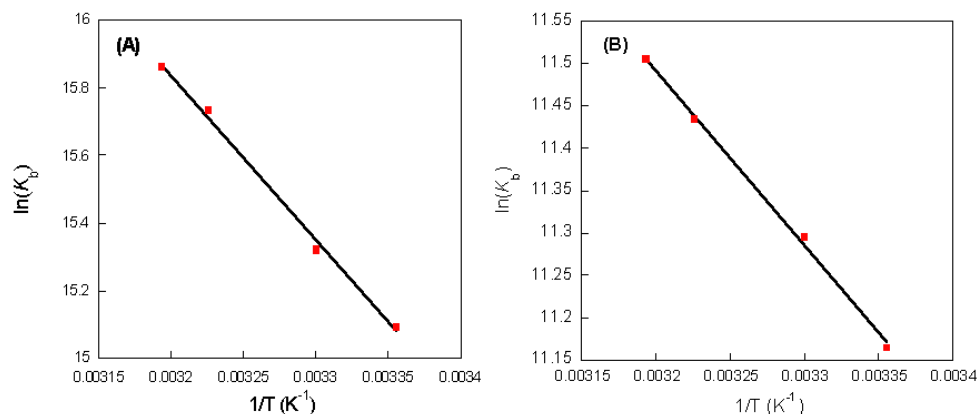


Figure S7. Van't Hoff plot obtained from the interaction between BSA and complex 1 (A) and 2 (B).

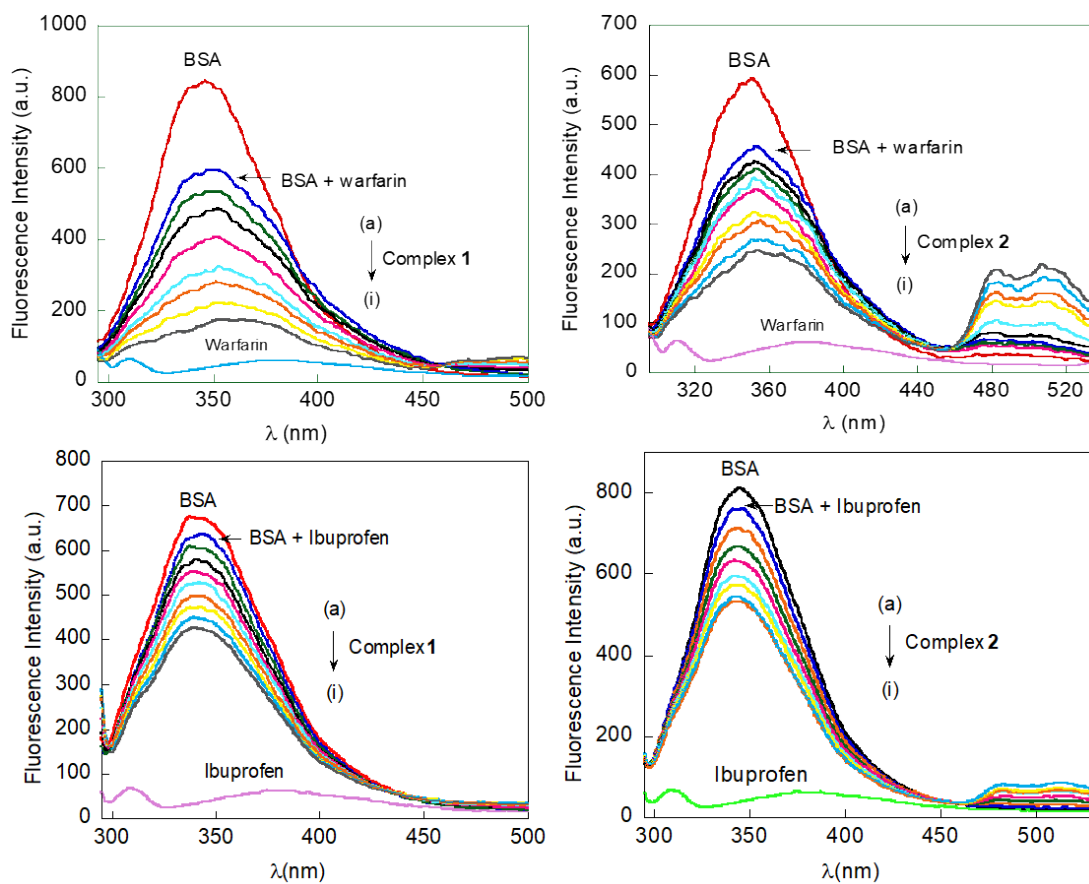


Figure S8. Effects of Pt(II) complexes 1 or 2, on two BSA-site marker probes systems. $[BSA] = [warfarin] = [ibuprofen] = 0.1 \mu M$; the concentration of Pt(II) complexes was varying from 0.0 (a) to 0.15 (i) μM , $\lambda_{ex} = 280$ nm.

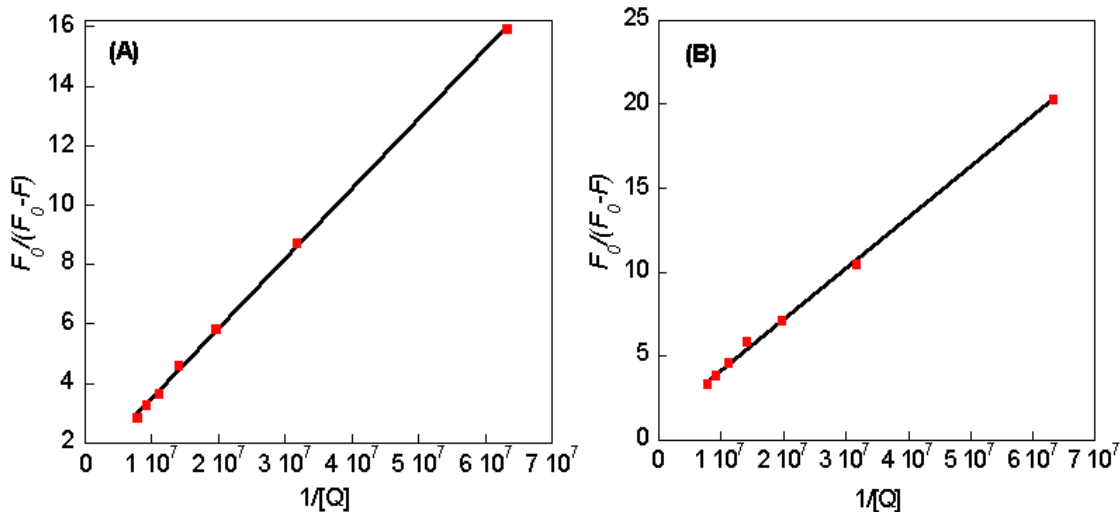


Figure S9. The modified Stern–Volmer plots of (A) BSA-warfarin and (B) BSA-ibuprofen in the presence of various concentration of Pt(II) complex **1**, at 310 K. $\lambda_{\text{ex}} = 280$ nm; pH = 7.4, [HSA] = [warfarin or ibuprofen] = 0.1 μM . **1** Concentration was from 0.0 to 0.15 μM .

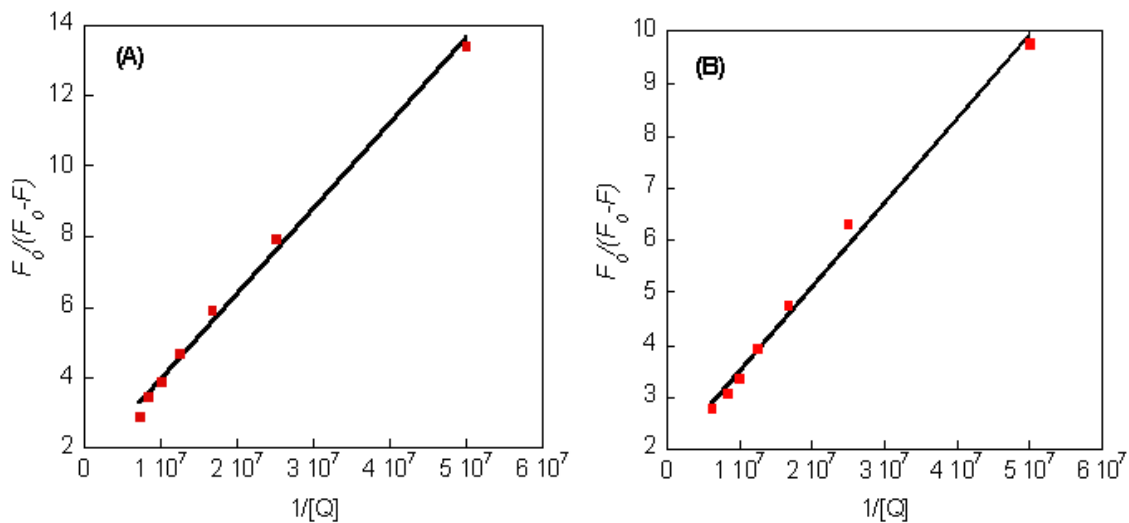


Figure S10. The modified Stern–Volmer plots of (A) BSA-warfarin and (B) BSA-ibuprofen in the presence of various concentration of Pt(II) complex **2**, at 310 K. $\lambda_{\text{ex}} = 280$ nm; pH = 7.4, [HSA] = [warfarin or ibuprofen] = 0.1 μM . **2** Concentration was from 0.0 to 0.15 μM .

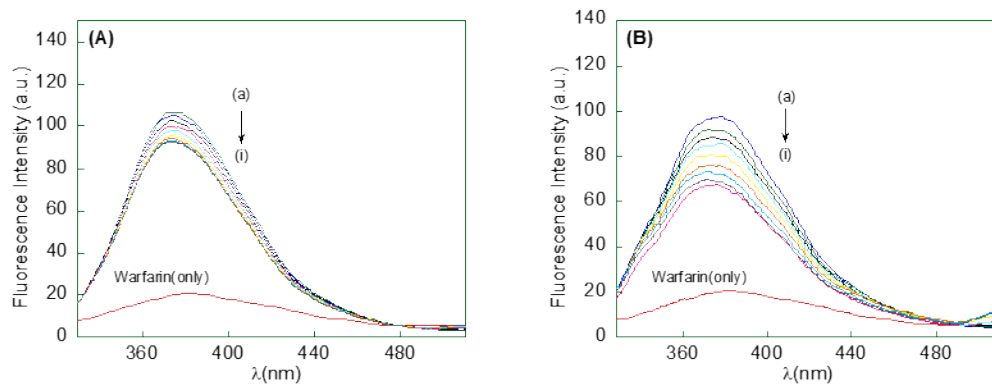


Figure S11. Effect of **1** (A) and **2** (B) to warfarin–BSA system ($\lambda_{\text{ex}} = 310 \text{ nm}$). $[\text{warfarin}] = [\text{BSA}] = 0.1 \mu\text{M}$; **1** and **2** concentrations were from (a) 0.0 to (i) $0.15 \mu\text{M}$, $\lambda_{\text{ex}} = 310 \text{ nm}$.

Table S1. List of cycloplatinated complexes and their $\text{Log } K_b$ analyzed in this work

Comp	Structure	$\text{Log } K_b$	Comp	Structure	$\text{Log } K_b$	Comp	Structure	$\text{Log } K_b$
C1 = 1		6.55	C6 Ref: 45		4.56	C11 Ref: 48		5.51
C2 = 2		4.84	C7 Ref: 47		5.16	C12 Ref: 13		5.48
C3 Ref:46		0.11	C8 Ref: 45		4.82	C13 Ref: 49		7.02
C4 Ref:46		0.65	C9 Ref: 13		5.78	C14 Ref: 49		5.13
C5 Ref: 47		5.14	C10 Ref: 48		5.47			

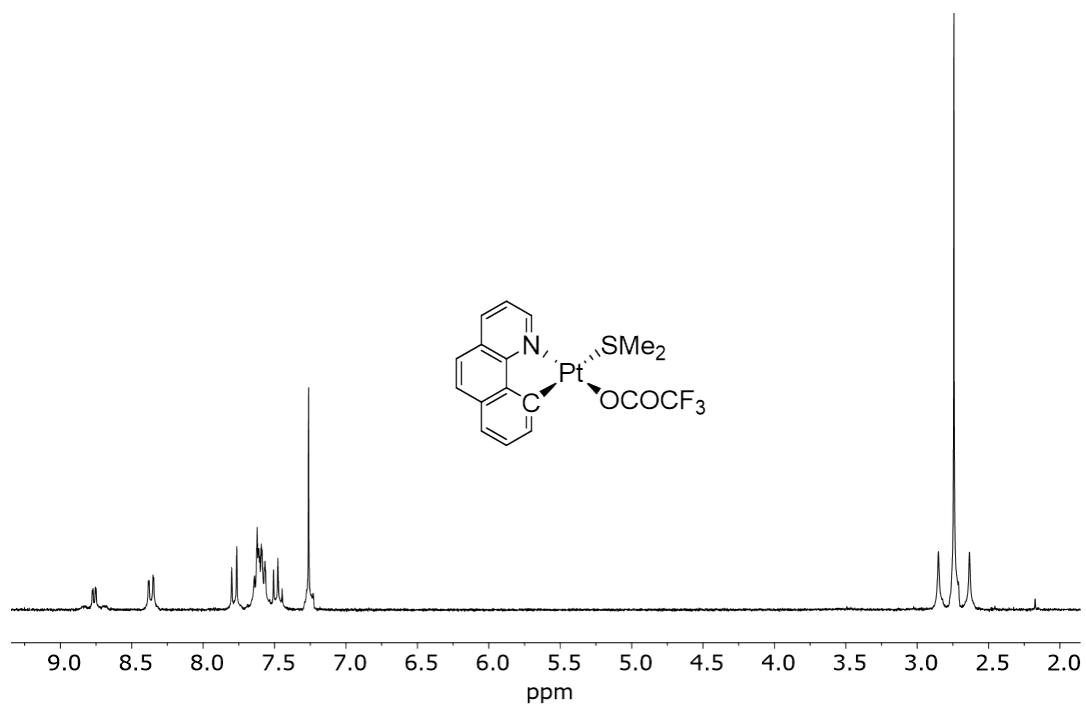


Figure S12. ¹H NMR of complex 1 in CDCl₃.

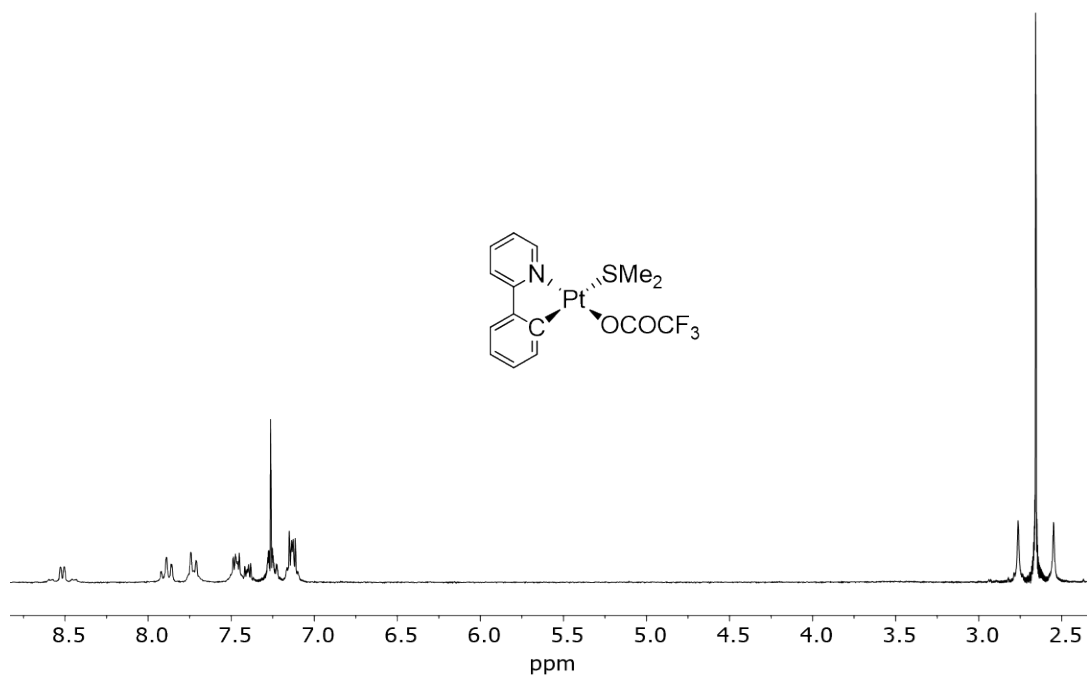


Figure S13. ¹H NMR of complex 2 in CDCl₃.

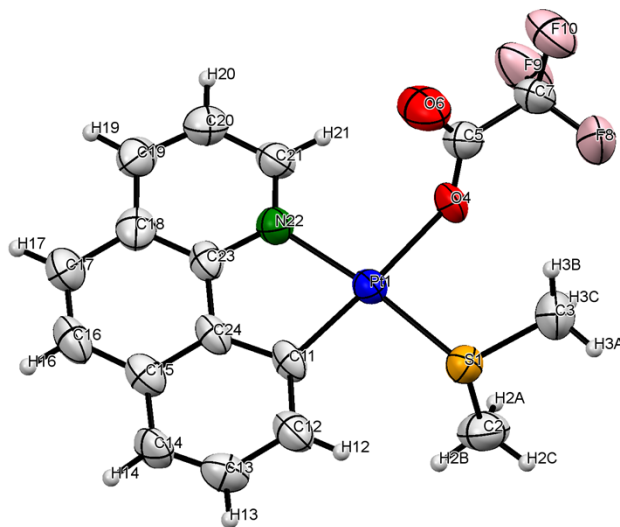


Figure S14. Structure of complex 1 (see ref. 42). Selected geometrical parameters (\AA , $^\circ$): Pt1–O4 2.113(5); Pt1–C11 1.972(8); Pt1–N22 2.062(8); Pt1–S1 2.263(3); O(4)–C(5) 1.265(17); O(6)–C(5) 1.223(15); N22–Pt1–O4 94.9(5); N22–Pt1–C11 82.0(6); O4–Pt1–S1 91.6(4); C11–Pt1–S1 91.5(5); S(1)–Pt(1)–N(22) 173.4(2); O(4)–Pt(1)–C(11) 176.8(8).

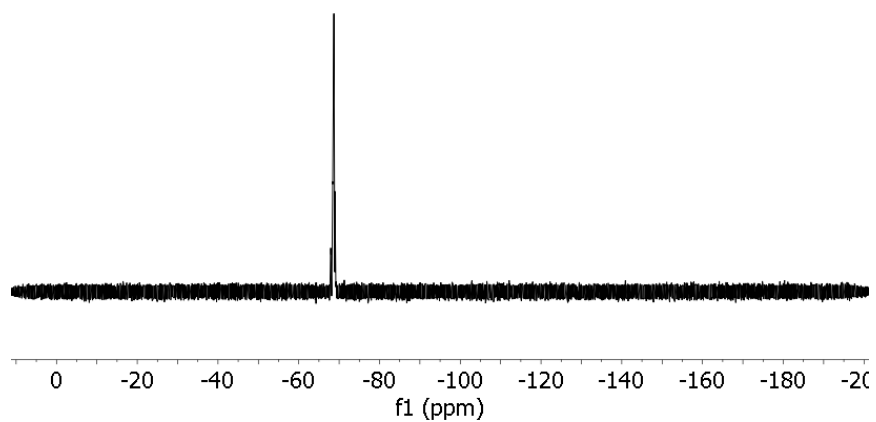


Figure S15. ^{19}F NMR of complex 2 in DMSO-d_6 .

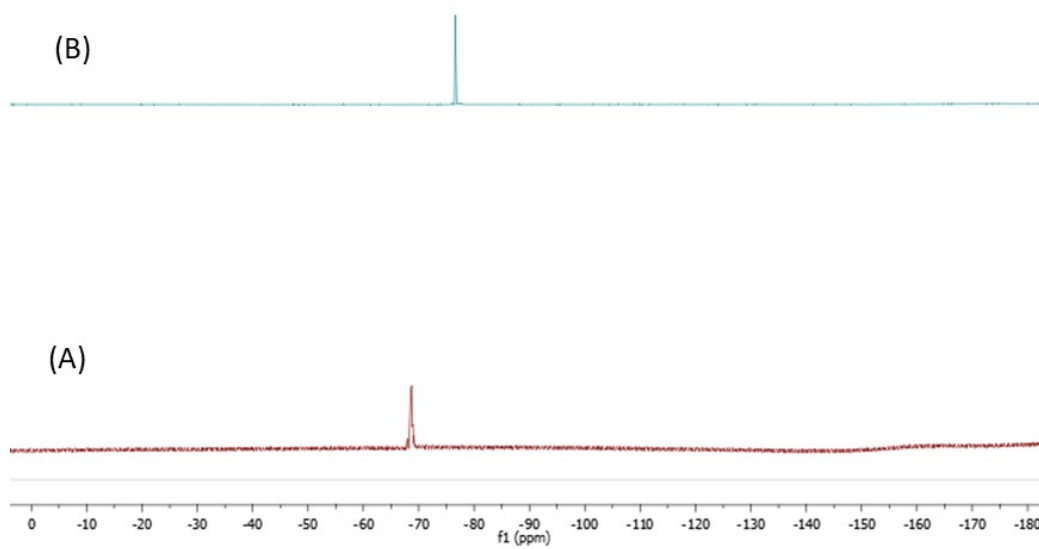


Figure S16. The changes in ^{19}F NMR spectrum of **2** (0.01 M) in DMSO (A) and in 10% DMSO/ H_2O solution after (B) 2h.