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Supporting Information

Fabrication of hybrid phase TiO₂/g-C₃N₄ heterojunction composite with enhanced adsorption and photocatalytic degradation of MB under visible light

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Fig. S1. Rietveld refinement of (a)TiO₂, (b) TiO₂/g-C₃N₄(1:5) and (c)TiO₂/g-C₃N₄(1:15) composite.



Fig. S2 SEM of (a, b) TiO_2 , (c, d) $TiO_2/g-C_3N_4(1:5)$, (e, f) $TiO_2/g-C_3N_4(1:15)$.



Fig. S3. (a) Pseudo-first order kinetics plot, (b) pseudo-second order kinetics plot for the adsorption of MB onto $TiO_2/g-C_3N_4(1:10)$.



Fig. S4. N₂ adsorption-desorption isotherms of TiO₂, TiO₂/g-C₃N₄(1:5), TiO₂/g-C₃N₄(1:10), TiO₂/g-C₃N



Fig. S5. The photodegradation performance of $TiO_2/g-C_3N_4(1:10)$.



Fig. S6. Cycling experiments of MB degradation by TiO₂/g-C₃N₄(1:10) under visible light..



Fig. S7. XRD diffraction pattern (a), FT-IR spectra (b), SEM (c-pristine, d-used) of the $TiO_2/g-C_3N_4$ (1:10) composite before and after photodegradation of MB.







Fig. S9. Photoluminescence (PL) spectra of different samples.

Table S1 Microstructural parameters deduced from the Rietveld refinement analysis of the XRD patterns for the TiO_2 and $TiO_2/g-C_3N_4$ catalysts.

Samples	Component (TiO ₂)	Space Group	Mass content/wt%	a/Å	b/Å	c/Å	Rw/%
TiO ₂	Brookite Anatase	C2/m I4 ₁ /amd	74.0% 26.0%	12.05 3.78254	3.770 3.78254	6.604 9.61502	7.13
TiO ₂ /g-	Brookite	C2/m	43.5%	12.10141	3.8099	6.5557	6.74
C ₃ N ₄ (1;5)	Anatase	I4 ₁ /amd	56.5%	3.78254	3.78254	9.61502	
TiO ₂ /g-	Brookite	C2/m	93.6%	12.18	3.761	6.495	6.86
C ₃ N ₄ (1;10)	Anatase	I4 ₁ /amd	6.4%	3.78254	3.78254	9.61502	
TiO ₂ /g-	Brookite	C2/m	91.9%	12.19409	3.75433	6.51884	4.538
C ₃ N ₄ (1;15)	Anatase	I4 ₁ /amd	8.1%	3.78254	3.78254	9.61502	