

Supporting Information

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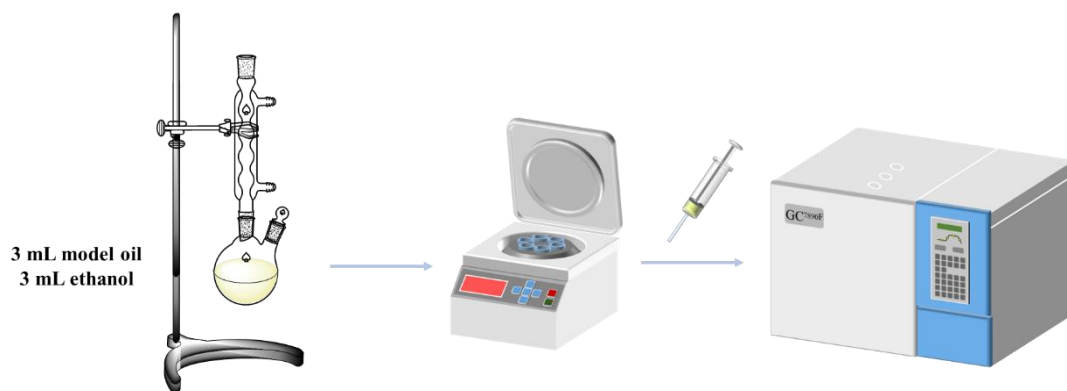


Fig. S1 ODS experimental device diagram

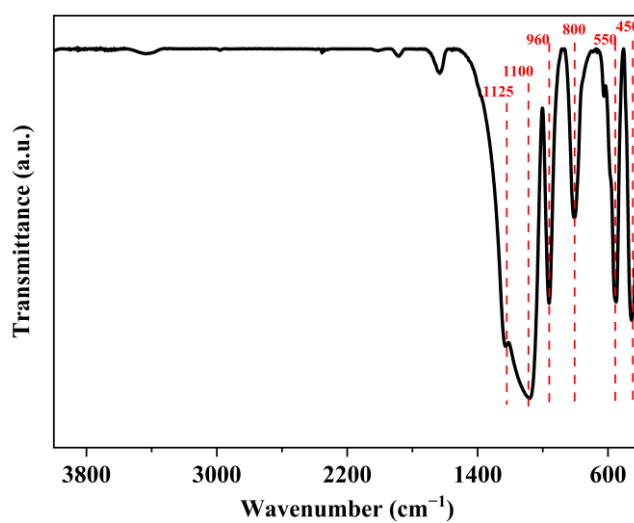


Fig. S2 FT-IR spectra of HPW/TS-1@TiO₂-0.05

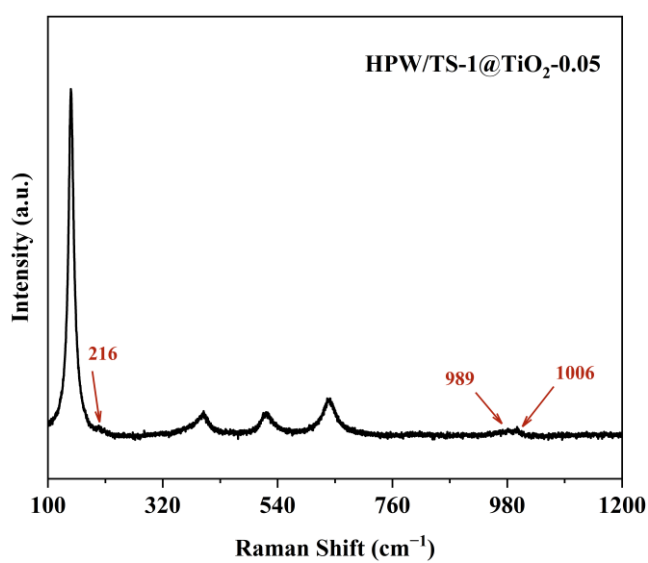


Fig. S3 Raman spectra of HPW/TS-1@TiO₂-0.05

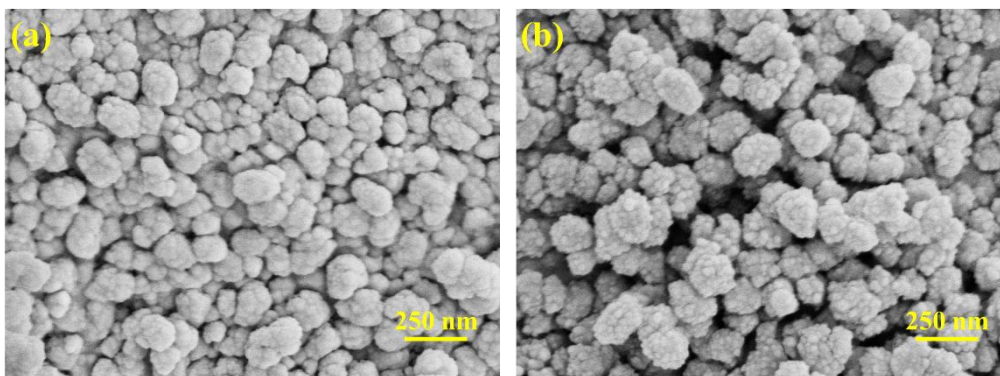


Fig. S4 Scanning electron micrographs of (a)TS-1@TiO₂-0.025 and (b) TS-1@TiO₂-1

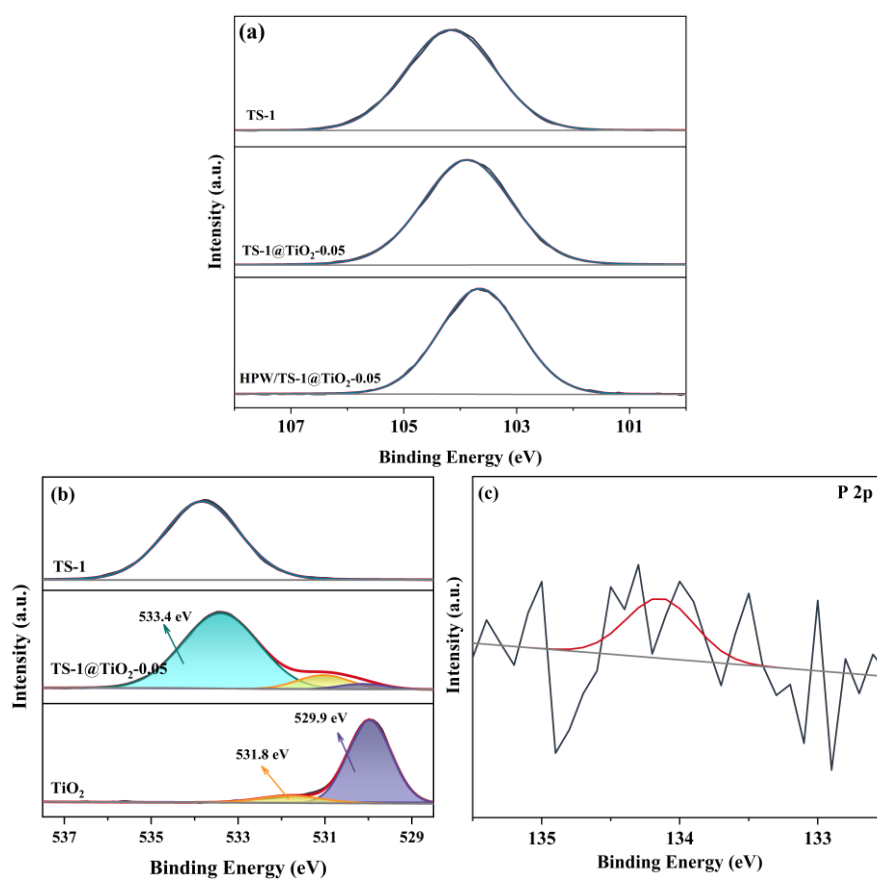


Fig. S5 XPS spectra of samples: (a) Si 2p, (b) O 1s and (c) P 2p

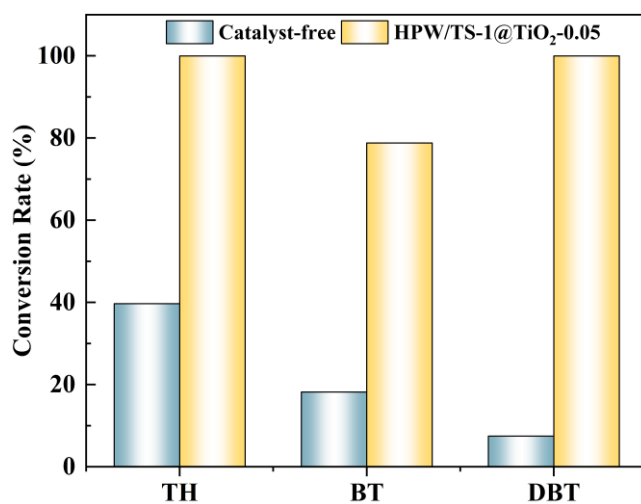


Fig. S6 The ODS performance on different sulfides using HPW/Ts-1@TiO₂-0.05 and catalyst-free. Reaction conditions: $t = 90$ min, $T = 60$ °C, HPW/Ts-1@TiO₂-0.05 dosage = 0.06 g, O/S molar ratio = 12.

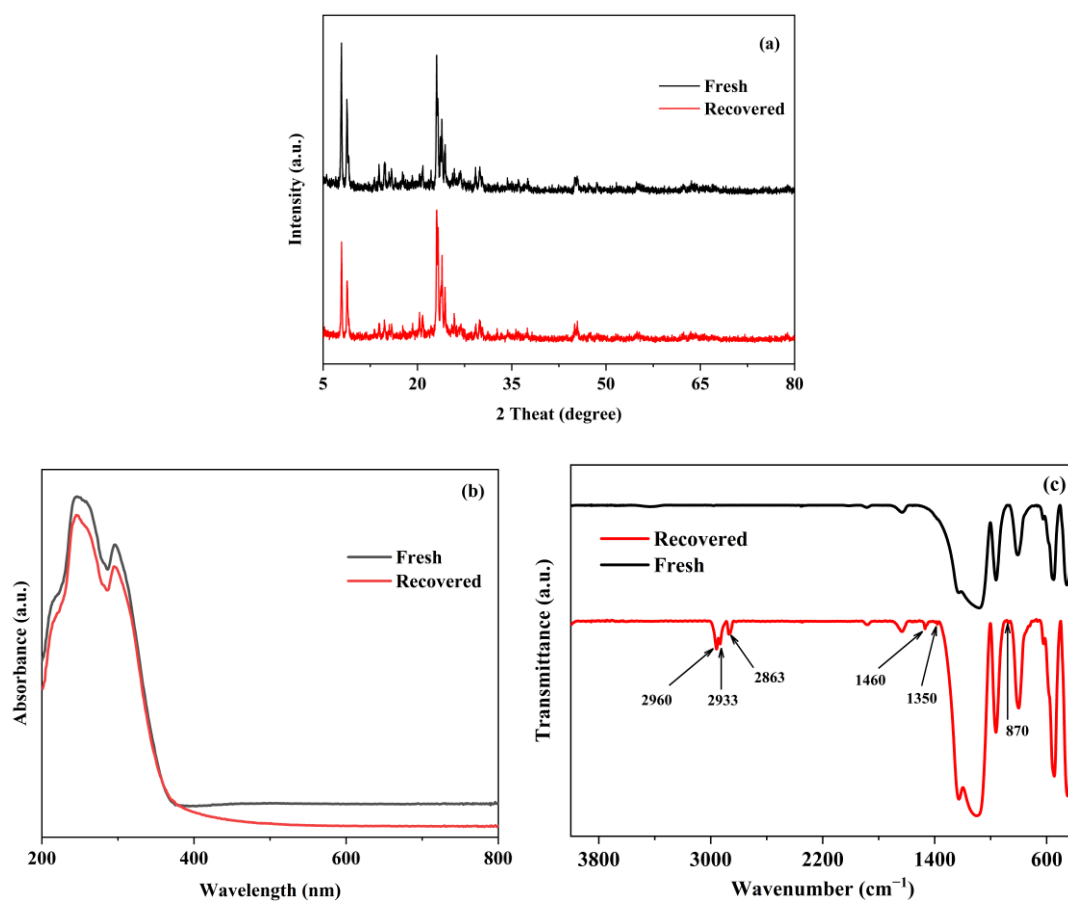


Fig. S7 XRD pattern (a) UV-vis spectra (b) and FT-IR spectra (c) of the recovered HPW/Ts-1@TiO₂-0.05

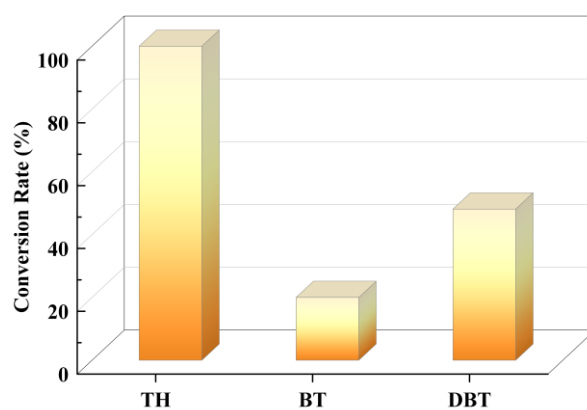


Fig. S8 ODS efficiencies on different sulfides over TS-1 catalyst.

Reaction conditions: $t = 120$ min, $T = 70$ °C, TS-1 dosage = 0.06 g, O/S molar ratio = 12.

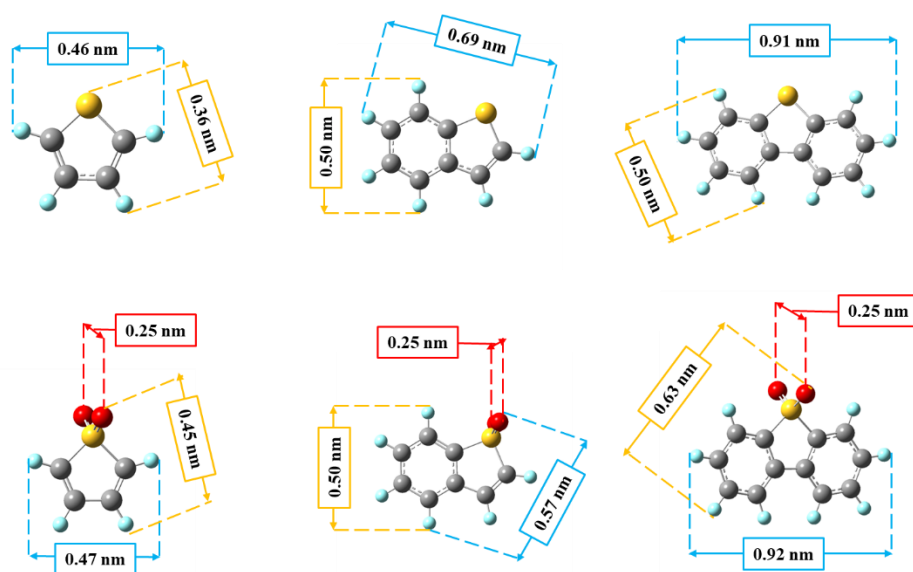


Fig. S9 Ball and stick models of TH, BT, DBT and corresponding sulphone.

Blue: H, Yellow: S, Gray: C and Red: O.

Table S1 Comparison of the reaction data with other reported methods

Catalysts	Model oil	Solvent	Reaction conditions	Conversion of TH	Conversion of BT	Conversion of DBT	Ref
TS-1	<i>n</i> -octane	H ₂ O	$C_{\text{Cat.}} = 10 \text{ g/L}$ $T = 60 \text{ }^\circ\text{C}$ $t = 30 \text{ min}$ $n (\text{H}_2\text{O}_2/\text{sulfur}) = 4:1$	97%	—	—	20
Hierarchical TS-1	<i>n</i> -octane	CH ₃ OH	$C_{\text{Cat.}} = 10 \text{ g/L}$ $T = 60 \text{ }^\circ\text{C}$ $t = 120 \text{ min}$ $n (\text{H}_2\text{O}_2/\text{sulfur}) = 4:1$	81%	69%	—	24
TiO ₂	<i>n</i> -octane	CH ₃ OH	$C_{\text{Cat.}} = 6.7 \text{ g/L}$ $T = 60 \text{ }^\circ\text{C}$ $t = 60 \text{ min}$ $n (\text{H}_2\text{O}_2/\text{sulfur}) = 4:1$	—	—	99.6%	36
TiO ₂ /GC	<i>n</i> -octane	—	$C_{\text{Cat.}} = 20 \text{ g/L}$ $T = 50 \text{ }^\circ\text{C}$ $t = 60 \text{ min}$ $n (\text{H}_2\text{O}_2/\text{sulfur}) = 4:1$	—	—	100%	31
H ₃ PW ₁₂ O ₄₀	<i>n</i> -octane	CH ₃ CN	$C_{\text{Cat.}} = 70 \text{ g/L}$ $T = 60 \text{ }^\circ\text{C}$ $t = 60 \text{ min}$ $n (\text{H}_2\text{O}_2/\text{sulfur}) = 2:1$	—	—	99.2%	78
HPMoV ₂ /TS-1-TPAOH@SiO ₂ -in situ	<i>n</i> -octane	C ₂ H ₅ OH	$C_{\text{Cat.}} = 20 \text{ g/L}$ $T = 70 \text{ }^\circ\text{C}$ $t = 180 \text{ min}$ $n (\text{H}_2\text{O}_2/\text{sulfur}) = 13:1$	99.3%	45.9%	84.1%	52
HPW/TS-1@TiO ₂ -0.05	<i>n</i> -octane	C ₂ H ₅ OH	$C_{\text{Cat.}} = 20 \text{ g/L}$ $T = 60 \text{ }^\circ\text{C}$ $t = 90 \text{ min}$ $n (\text{H}_2\text{O}_2/\text{sulfur}) = 12:1$	99.9%	78.8%	99.9%	This work