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Electronic supplementary information (ESI)

Synthesis and characterization of Y-shaped optical nonlinear chromophores with strong acceptor

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1. Optical properties



Figure S1 UV absorption spectra of chromophore L1 in solvents



Figure S2 UV absorption spectra of chromophore L2 in solvents



Figure S3 UV absorption spectra of chromophore L3 in solvents

2. Theoretical calculations



Figure S4. Hyperpolarizability in toluene, chloroform and tetrahydrofuran solution for chromophores L1-L3 calculated by DFT (M062X/6-31+G(d))



Figure S5. Chemical structure of monodonor chromophore L4

Table S1 Summary of DFT and EO coefficients of chromophores

Cmpd	$\Delta E(DFt)$	μ^{b}	Cam ^c	Chl ^d	Thf ^e	Tol ^f	max.r ₃₃
	^a (eV)	(D)					(pm/V)
L4	3.77	21.15	987	3077	2661	1911	173
^a calculated from DFT calculations. ^b the total dipole moment. ^{cdef} First order							
hyperpolarizability calculated in vacuum, Chloroform, Tetrahydrofuran and							
Toluene, respectively.							

3. ¹H and ¹³C NMR spectra



Figure S6. ¹H NMR spectra of chromophore L1



Figure S7. ¹³C NMR spectra of chromophore L1



Figure S9. ¹³C NMR spectra of chromophore L2



Figure S10. ¹H NMR spectra of chromophore L3



Figure S11. ¹³C NMR spectra of chromophore L3

4. Poling and electro-optical coefficient parameters of chromophores



Figure S12. The Number density, Poling efficiency and Maximum electro-optic coefficient of chromophores L1-L3.