Supporting Information

Effect of electrochemically active element species on the stability of the

layered cathode-sulfide electrolyte interface

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Fig. S1. Scheme of precursor and cathode material preparation.



Fig. S2. Schematic diagram of assembled ASSLB.



Fig. S3 Particle size distribution of the precursors.



Fig. S4 SEM images of precursors of (a1-d2) LNO, N75, N50, and N25.



Fig. S5 X-ray photoelectron spectroscopy raw spectra of (a)Ni 2p and (b)Co 2p for cathode materials.



Fig. S6 XPS spectra of (a) LNO/LSPSC, (b) N50/LSPSC and (c) LCO/LSPSC mixture after 150h standing time.



Fig. S7 Voltage-time patterns for the ASSLB assembled with (a) LNO, (b) N50 and (c) LCO cathode materials at a

current density of 0.1C using the GITT test method.



Fig. S8 SEM images of precursors of (a1-c3) NCM442, NCM622 and NCM811.



Fig. S9 SEM images of (a1-c3) NCM442, NCM622 and NCM811.



Fig. S10 XRD patterns of NCM442, NCM622 and NCM811.



Fig. 11 Rate performance of LiNi_xCo_yMn_{1-x-y}O₂ cathodes in ASSLB.





Fig. S12 (a-c) Electrochemical impedance spectroscopy and (d) impedance increase comparison of LiNi_xCo_yMn_{1-x-}

 $_{y}O_{2}/LGPS$ mixture with standing time.

Table S1 Particle size analysis of $Ni_xCo_{1-x}(OH)_2$ precursors.

Precursors	D ₁₀ (µm)	D ₅₀ (µm)	D ₉₀ (µm)
Ni(OH)2	5.25	8.23	10.88
Ni _{0.75} Co _{0.25} (OH) ₂	3.45	6.5	9.73
Ni _{0.5} Co _{0.5} (OH) ₂	4.76	7.8	10.68
Ni _{0.25} Co _{0.75} (OH) ₂	5.75	8.37	10.64

Table S2 Transition metal element content of $Ni_xCo_{1-x}(OH)_2$ precursors.

Precursors –	Designed ato	Designed atomic ratio (%)		Measured atomic ratio (%)	
	Ni	Со	Ni	Со	
Ni _{0.75} Co _{0.25} (OH) ₂	0.75	0.25	0.744	0.256	
Ni _{0.5} Co _{0.5} (OH) ₂	0.5	0.5	0.503	0.497	
Ni _{0.25} Co _{0.75} (OH) ₂	0.25	0.75	0.246	0.754	

Table S3 Lattice parameters of LiNi_xCo_{1-x}O₂.

Samples	Lattice	Lattice parameters		т
	a[Å]	c[Å]	C/a	I (003)/ (104)
LNO	2.8847	14.2217	4.9301	1.001
N75	2.815	14.045	4.9893	1.20
N50	2.815	14.046	4.9896	1.23
N25	2.816	14.048	4.9886	1.42
LCO	2.8156	14.0564	4.9994	1.304

Table S4 Initial charge/discharge capacities and coulombic efficiencies of $LiNi_xCo_{1-x}O_2$ materials in liquid coin-cells.

- Electro de motoriole -	Initial capacity (mAh g ⁻¹)		$C_{1} = 1 = 1 = 2$
Electrode materials	Charge	Discharge	- Coulomb efficiency (%)
LNO	249.9	194.9	78.0
N75	216.4	175.9	81.3
N50	179.1	170.0	94.9
N25	168.6	153.4	91.0
LCO	159.3	145.6	91.4

Table S5 Cycle performance of LiNi_xCo_{1-x}O₂ materials in liquid coin-cells.

Electrode materials	6th Discharge capacity (mAh g ⁻¹)	100th Discharge capacity (mAh g ⁻¹)	Capacity retention (%)
LNO	164.2	124.3	78.0
N75	172.7	127.6	73.8
N50	160.3	129.8	80.9
N25	151.7	133.3	87.8
LCO	155.9	137.2	88.0

 $\label{eq:constraint} \begin{array}{l} \textbf{Table S6} \mbox{ Initial charge/discharge capacities and coulombic efficiencies of $LiNi_xCo_{1-x}O_2$ materials} \\ \mbox{ in ASSLB} \end{array}$

Electrodo motoriala -	Initial capacity (mAh g ⁻¹)		Coulomb officianay (9/)
	Charge	Discharge	- Coulonib enficiency (76)
LNO	169.9	115.4	67.9
N75	216	181.5	84.0
N50	186.6	165.8	88.8
N25	169.4	150.3	88.7
LCO	160.8	147	91.4

Table S7 <u>Cycle performance and capacity retention of $LiNi_xCo_{1-x}O_2$ materials in ASSLB</u>

Electrode materials	6th discharge capacity (mAh g ⁻¹)	100th discharge capacity (mAh g ⁻¹)	Capacity retention (%)
LNO	107.7	55	51.0
N75	181.6	95.1	52.3
N50	153	119.4	78.0
N25	144.1	122.4	84.9
LCO	144.8	122	84.2

Stable S8 The fitted EIS values of cycled ASSLB.

Electr	ode materials	Cycle numbers	Rs	Rf	Rct2
		1	24.49	15.41	146.4
	LINO	100	25.91	11.51	648.5
	N175	1	30.88	9.08	24.99
	IN / 3	100	27.86	12.94	104.3
	N150	1	42.43	11.89	19.45
	N30	100	40.87	10.42	54.87
	N125	1	22.61	11.18	48.92
	IN23	100	25.52	11.51	78.36
	LCO	1	27.01	14.89	10.58
	LCO	100	27.59	13.37	20.85