

Supporting Information

Polyethyleneimine-based fluorescent polymeric nanoparticles: synthesis and application in fluorescence sensing of pH and para-nitrophenol

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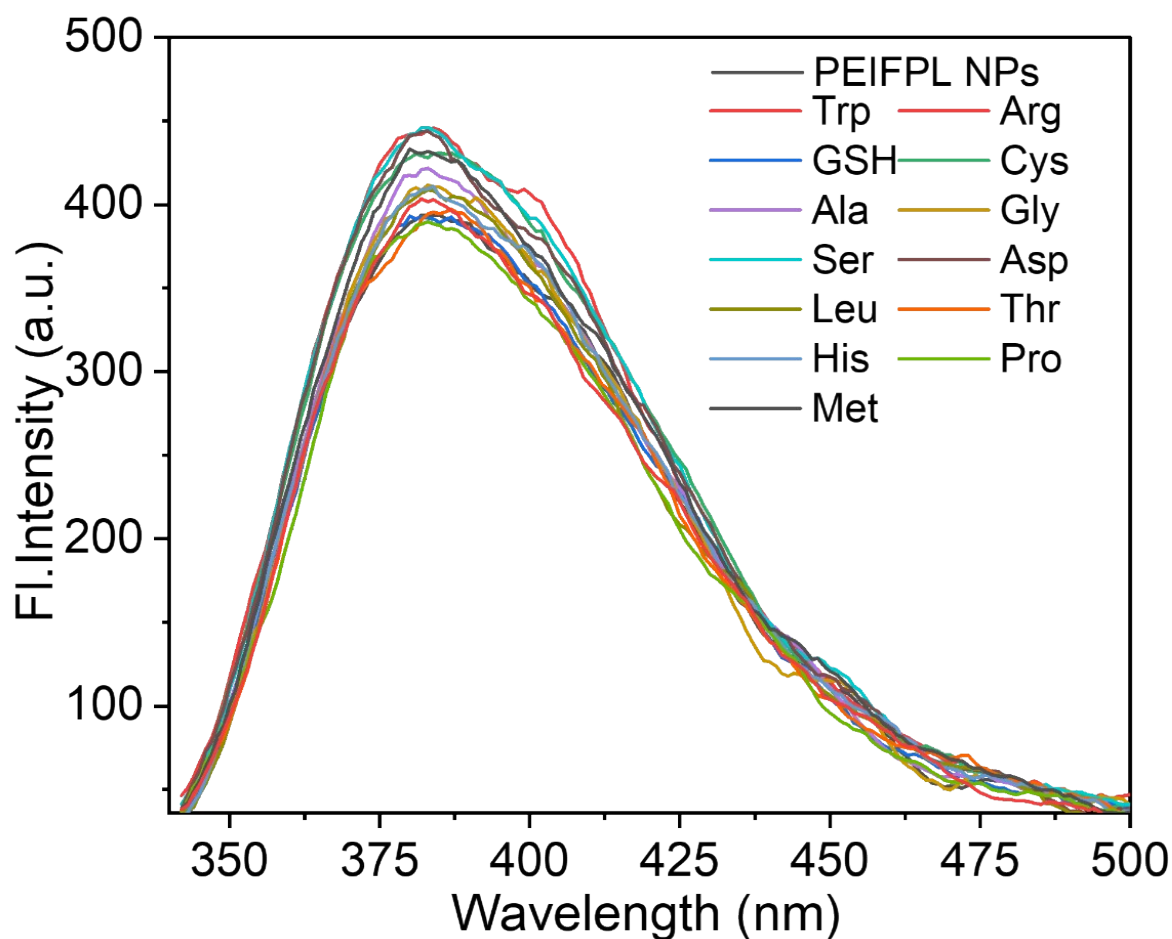


Fig. S1. Fluorescence spectra of PEIFPL NPs in the presence of different amino acids.

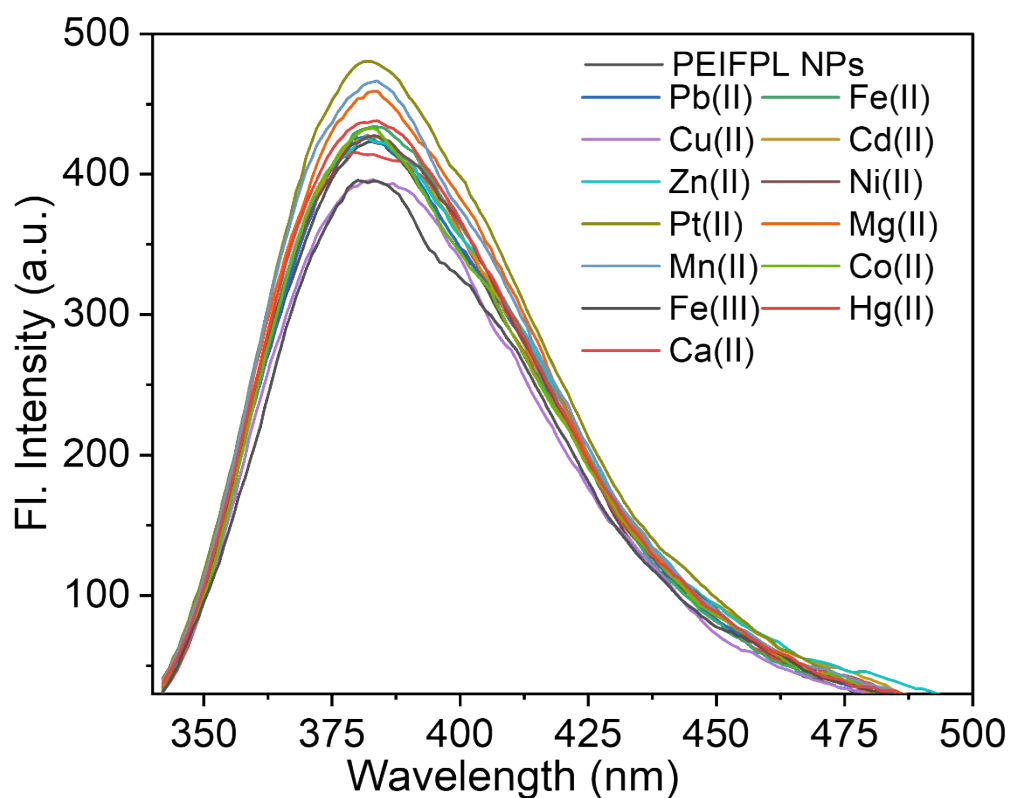


Fig. S2. Fluorescence spectra of PEIFPL NPs in the presence of different metal cations.

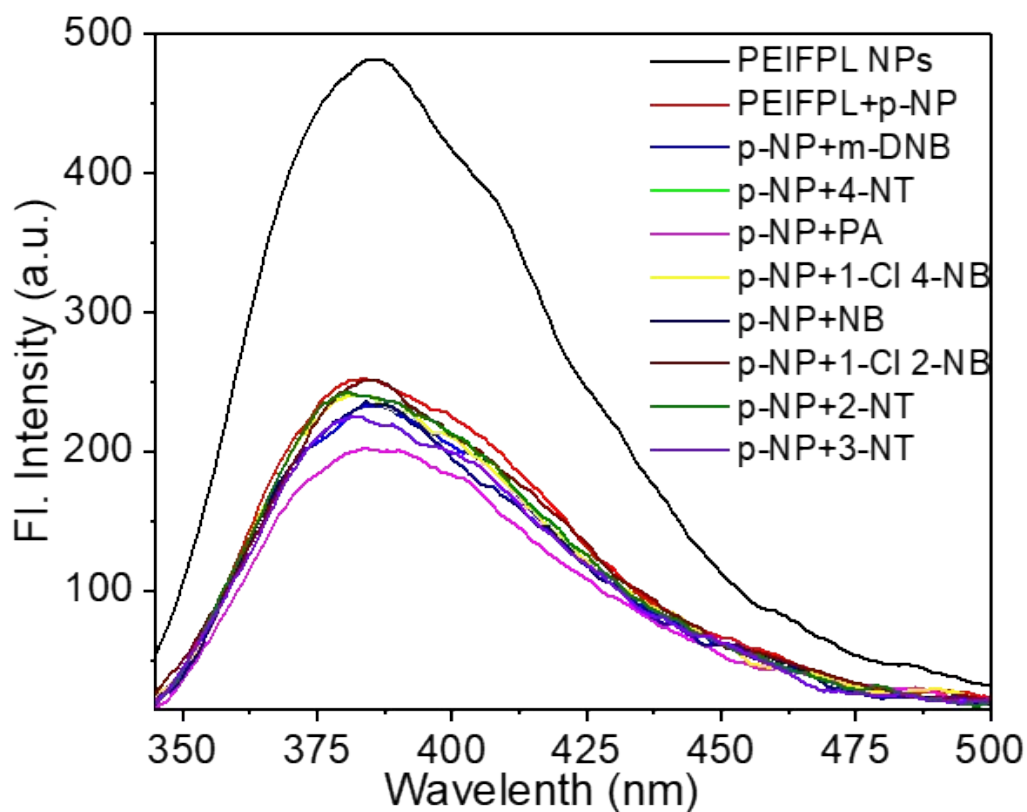


Fig. S3. Fluorescent spectra of PEIFPL NPs in the presence of p-NP and co-presence of equimolar amount of different interfering nitroaromatic compounds.

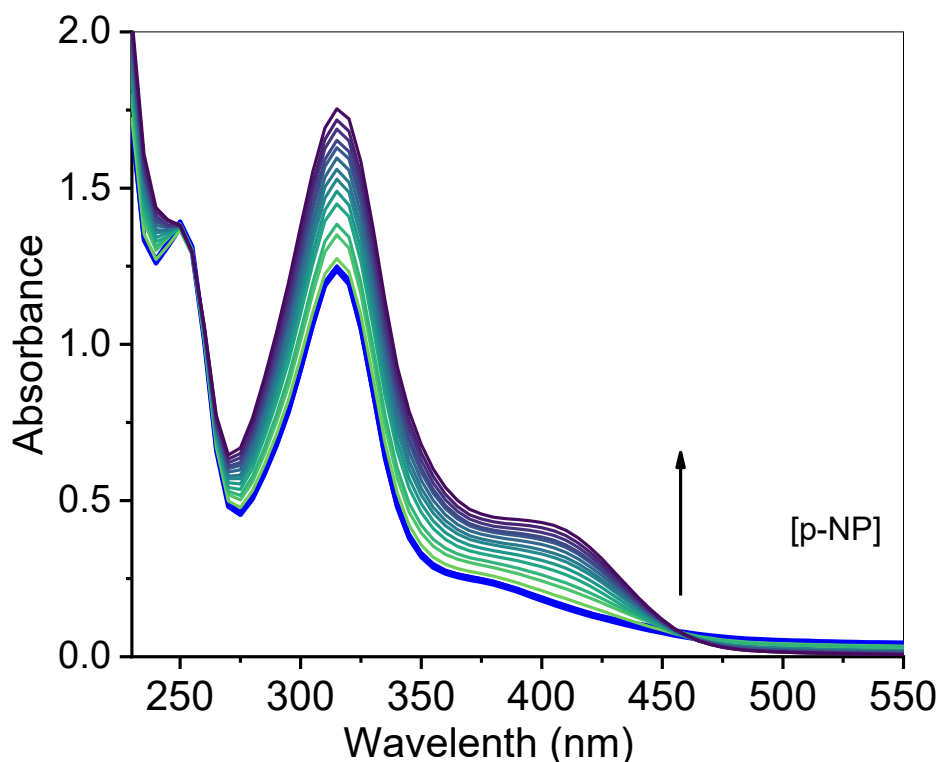


Fig. S4. UV-Vis absorption spectra of PEIFPL NPs with the successive incremental addition of p-NP from 0 to 36 μM .

Table S1. Comparison of reported fluorescent p-NP probes with PEIFPL NPs.

Probes	Response	Linear range	LOD	Applications	Ref.
Ag-doped ZnO	“Turn off”	0 to 9 μM	0.66 μM	-	1
Gd ₂ O ₃ NPs	“Turn off”	1 to 10 μM	1.52 μM	Real water sample	2
NOCDs	“Turn off”	2 to 100 μM	2 μM	Tap water and industrial samples	3
Bluish-green, fluorescent histidine	“Turn off”	0 to 61.1 μM	352.6 μM	Tap and river water	4
PEIFPL NPs	“Turn off”	0.5 to 50 μM	0.42 μM	Tap and river water	This paper

References

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