

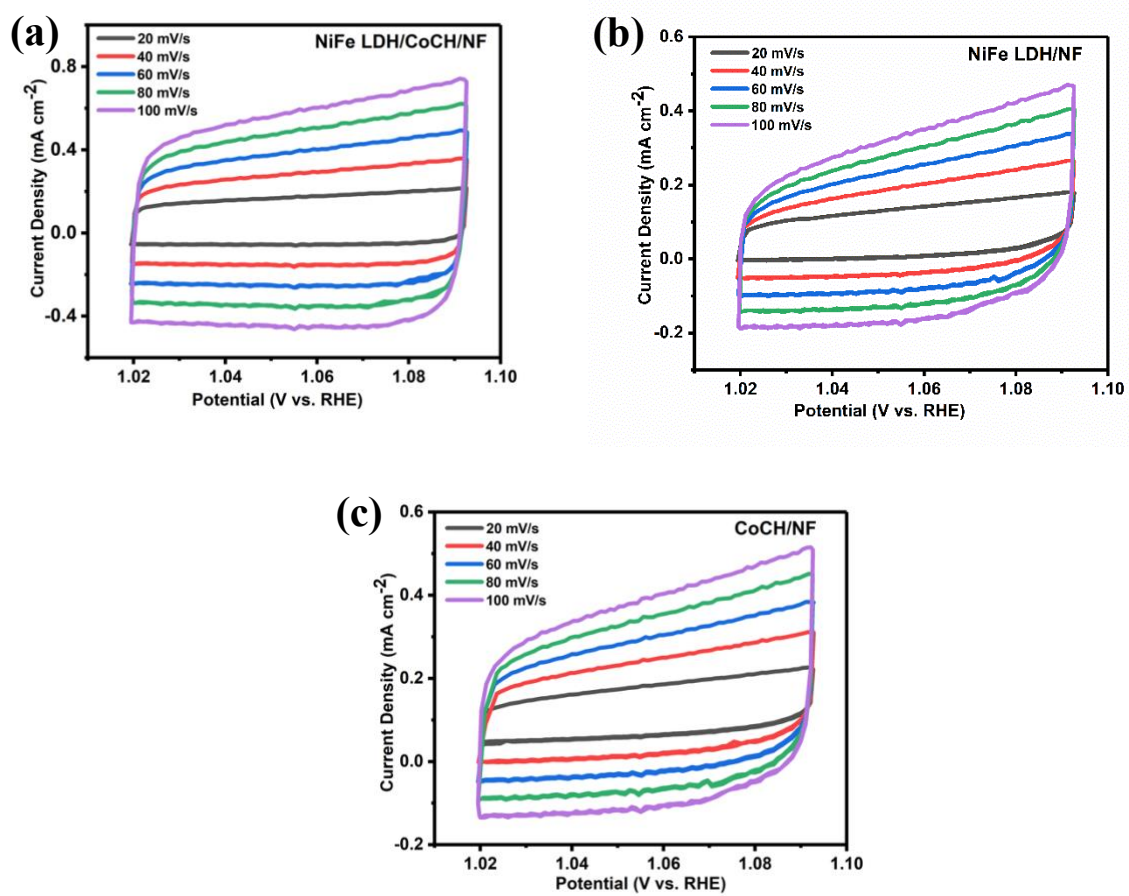
## Supplementary Information

# Highly Active NiFe LDH anchoring on Cobalt Carbonate Hydroxide for Efficient Electrocatalytic 5-hydroxymethylfurfural oxidation towards 2,5-furandicarboxylic acid

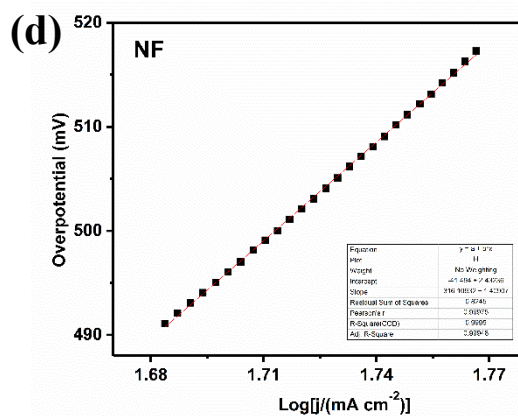
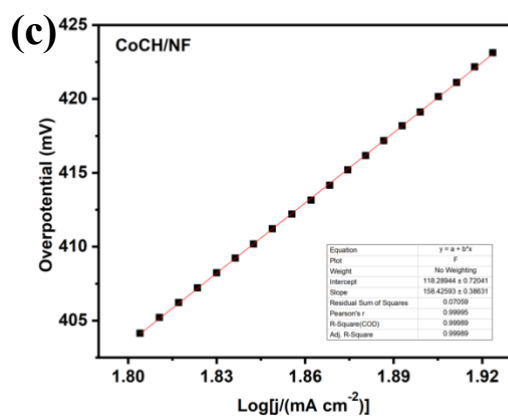
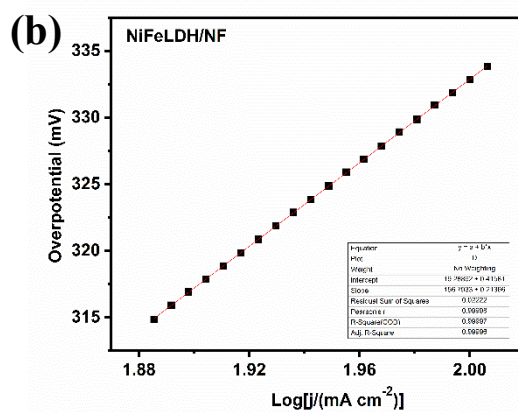
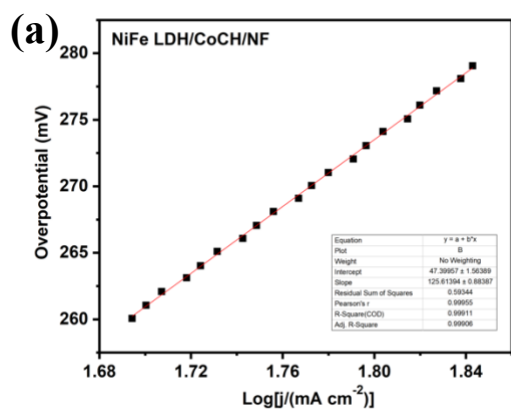
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**Figure S1.** Cyclic voltammety curves of different catalysts in 5 mM HMF solution with different sweep speeds



**Figure S2.** Tafel plot of NiFe LDH/CoCH/NF and other catalysts in 5 mM HMF solution.

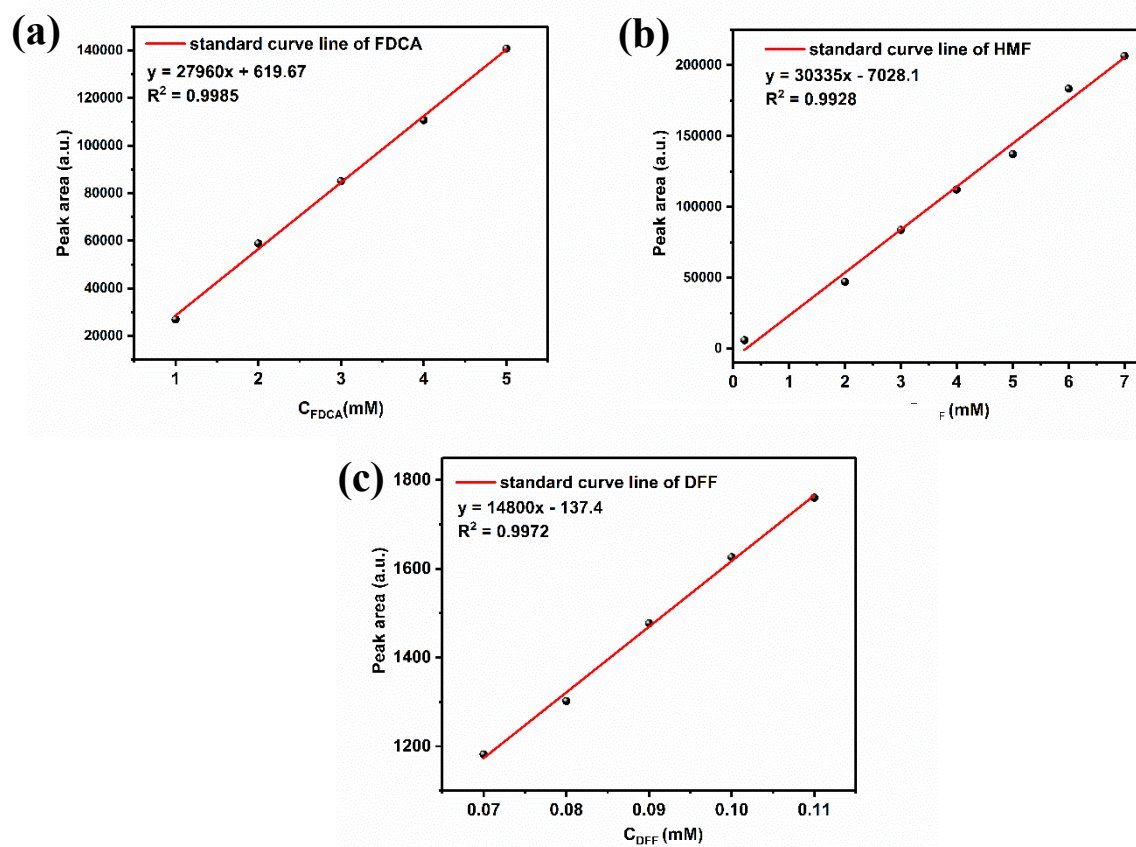
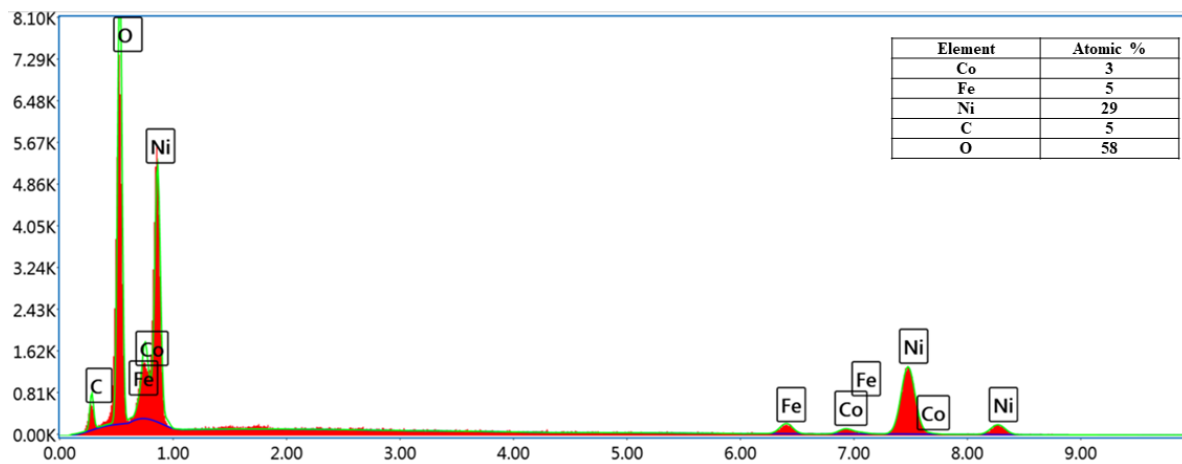
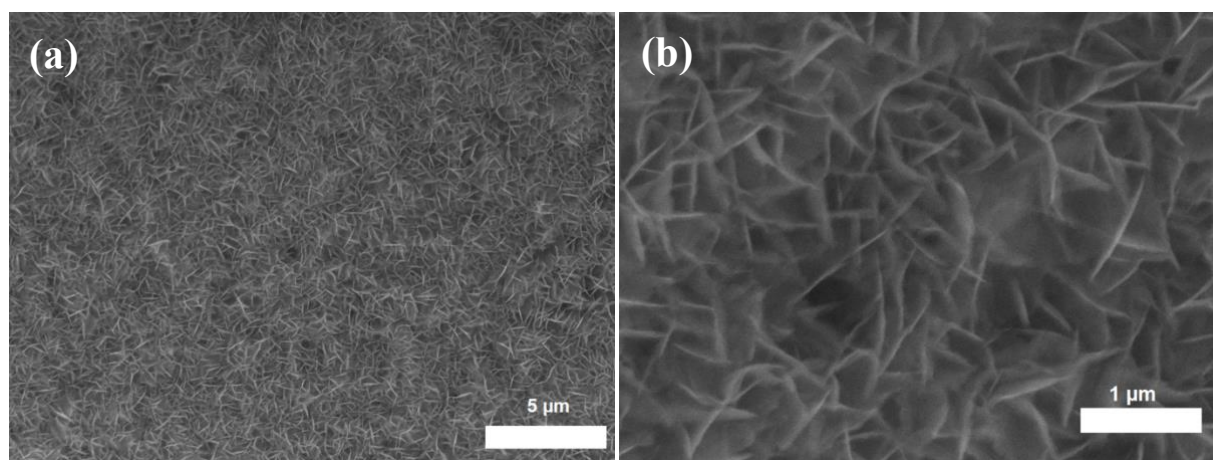


Figure S3. HPLC Calibration curves of (a) FDCA, (b) HMF, and (c) DFF.



**Figure S4.** EDX result of the NiFe LDH/CoCH/NF sample.



**Figure S5.** SEM images of NiFe LDH/CoCH/NF at low and high magnifications after 5 cycles of electrolytic HMF oxidations

**Table S1.** The values of Cdl and ECSA for of different catalysts in 5 mM HMF

Catalyst	Cdl (mF cm <sup>-2</sup> )	ECSA(cm <sup>2</sup> )
NiFe LDH/CoCH/NF	10	250
NiFe LDH/NF	4.8	120
CoCH/NF	4.6	115

**Table S2.** Comparison of the NiFe LDH/CoCH/NF with other reported catalysts for electrocatalytic oxidation of HMF in alkaline electrolytes.

Catalyst	Electrolyte	Reaction time (min)	FDCA yield (%)	Faradaic Efficiency (%)	Ref.
NiFe LDH/CoCH/NF	1 M KOH + 5 mM HMF	90	98.6	98.1	This work
d-NiFe LDH/CP	1 M NaOH + 10 mM HMF	300	96.8	84.5	[1]
NiCoFe LDHs	1 M NaOH + 10 mM HMF	60	84.9	90	[2]
NiFe LDH	1 M KOH + 10 mM HMF	90	98	99.4	[3]
NiCoMn-LDHs/NF	1 M NaOH + 1 mM HMF	150	91.7	65	[4]
CoP-CoOOH	1M KOH + 150 mM HMF	1000	96.3	96.3	[5]
CoOOH	1 M KOH + 5 mM HMF	1320	35.1	35.1	[6]
NiCO <sub>2</sub> O <sub>4</sub> /NF	1M KOH + 5mM HMF	55	90.8	87.5	[7]
N-NiMoO <sub>4</sub> /NF	1M KOH + 10mM HMF	120	97	91	[8]
Ni <sub>2</sub> S <sub>3</sub> /NF	1M KOH + 10mM HMF	120	98	94	[9]
S-Ni@C	1M KOH + 10mM HMF	270	96	96	[10]
NiO-N/C	1M KOH + 10mM HMF	120	84	96	[11]
Ni <sub>0.9</sub> Cu <sub>0.1</sub> (OH) <sub>2</sub>	1M KOH + 5mM HMF	120	91.2	91.2	[12]

## References

1. Y.-F. Qi, K.-Y. Wang, Y. Sun, J. Wang and C. Wang, *ACS Sustain. Chem. Eng.*, 2021, 10, 645-654.
2. M. Zhang, Y. Liu, B. Liu, Z. Chen, H. Xu and K. Yan, *ACS Catal.*, 2020, 10, 5179-5189.
3. W.-J. Liu, L. Dang, Z. Xu, H.-Q. Yu, S. Jin and G. W. Huber, *ACS Catal.*, 2018, 8, 5533-5541.
4. B. Liu, S. Xu, M. Zhang, X. Li, D. Decarolis, Y. Liu, Y. Wang, E. K. Gibson, C. R. A. Catlow and K. Yan, *Green Chem.*, 2021, 23, 4034-4043.
5. H. Wang, Y. Zhou and S. Tao, *Appl. Catal. B: Environ.*, 2022, 315.
6. B. J. Taitt, D.-H. Nam and K.-S. Choi, *ACS Catal.*, 2018, 9, 660-670.
7. M. J. Kang, H. Park, J. Jegal, S. Y. Hwang, Y. S. Kang and H. G. Cha, *Appl. Catal. B: Environ.*, 2019, 242, 85-91.
8. W. Wang and M. Wang, *Catal. Sci. Technol.*, 2021, 11, 7326-7330.
9. W. Wang, F. Kong, Z. Zhang, L. Yang and M. Wang, *Dalton Trans.*, 2021, 50, 10922-10927.
10. F. Kong and M. Wang, *ACS Appl. Energy Mater.*, 2021, 4, 1182-1188.
11. W. Wang, Z. Zhang and M. Wang, *Biomass Convers. Biorefinery*, 2022, 1-8.
12. J. Zhang, P. Yu, G. Zeng, F. Bao, Y. Yuan and H. Huang, *J. Mater. Chem. A*, 2021, 9, 9685-9691.