## **Supplementary Informations**

## Designing inorganic-organic hybrid molecules based on carbazole/indole appended cyclotriphosphazenes and the investigation of their photophysical properties

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**Figure S1:** <sup>31</sup>P-decoupled NMR spectrum of the mixture of **1** and **I** at 1:3 mole ratio in THF solution; the reaction mixture was filtered and the solvent removed prior to dissolving in CDCl<sub>3</sub> solution.



Figure S2: <sup>31</sup>P-decoupled NMR spectrum of bis nongeminal (*cis/trans*) products (2a/2b).



Figure S3: The mass spectrum of compounds 2a/2b.



Figure S4: <sup>31</sup>P-decoupled NMR spectrum of tris nongeminal products (3a, b).



Figure S5: The mass spectrum of compounds 3a, b.



Figure S6: <sup>31</sup>P-decoupled NMR spectrum of tetrakis (*cis/trans*) products (4a/4b).



Figure S7: The mass spectrum of compounds 4a/4b.



**Figure S8:** <sup>31</sup>P-decoupled NMR spectrum of the mixture of **1** and **II** at 1:3.1 mole ratio in THF solution; the reaction mixture was filtered and the solvent removed prior to dissolving in CDCl<sub>3</sub> solution.



Figure S9: The mass spectrum of compounds 5a, b.



**Figure S10:** <sup>31</sup>P-decoupled NMR spectrum of the reaction mixture of **5a** and **5b** with **I** in THF solution; the reaction mixture was filtered and the solvent removed prior to dissolving in CDCl<sub>3</sub> solution.



**Figure S11:** <sup>31</sup>P-decoupled NMR spectra of compounds a) **7a** b) **8** c) **9**.



Figure S12: The mass spectrum of compound 7a.





Figure S14: The mass spectrum of compound 8.



Figure S15: <sup>31</sup>P-decoupled NMR spectrum of the reaction mixture of 1 with II at 1:8 mole

ratio.



Figure S16: The mass spectrum of compound 9.



**Figure S17: a)** Crystal structure of **II** with atom numbering scheme, displacement ellipsoids are drawn at the 30% probability level and H atoms are omitted for clarity. **b**) A view of classical intermolecular O-H/O (green dotted lines along the a-axis) hydrogen bonding. **c**) Perspective view of intermolecular  $\pi$ - $\pi$  stacking interactions. **d**) Crystal packing illustration of **II**.



**Figure S18:** Crystal structure of compound **8**. Displacement ellipsoids are drawn at the 30% probability level and H atoms are omitted for clarity. The grey, blue-, red- and orange-coloured atoms represent C, N, O and P atoms, respectively.



Figure S19: UV-Vis absorption spectra for 7a in various solvents (a: n-hexane; b: toluene; c: dichloromethane; d: 1,4-dioxane; e: ACN; f: EtOH).



Figure S20: UV-Vis absorption spectra for 8 in various solvents (a: n-hexane; b: toluene; c:

dichloromethane; d: 1,4-dioxane; e: ACN; f: EtOH).



Figure S21: UV-Vis absorption spectra for 9 in various solvents (a: n-hexane; b: toluene; c: dichloromethane; d: 1,4-dioxane; e: ACN; f: EtOH).



**Figure S22:** Fluorescence spectra for **7a** in various solvents (**a**: n-hexane; **b**: toluene; **c**: dichloromethane; **d**: 1,4-dioxane; **e**: ACN; **f**: EtOH).



**Figure S23:** Fluorescence spectra for **8** in various solvents (**a**: n-hexane; **b**: toluene; **c**: dichloromethane; **d**: 1,4-dioxane; **e**: ACN; **f**: EtOH).



Figure S24: Fluorescence spectra for 9 in various solvents (a: n-hexane; b: toluene; c: dichloromethane; d: 1,4-dioxane; e: ACN; f: EtOH).

Bond length	<b>3</b> a	Bond / Torsion angles	<b>3</b> a
P1-N1	1.593(4)	N1-P1-N3	117.0(2)
N1-P2	1.567(4)	N1-P2-N2	118.7(2)
P2-N2	1.579(4)	N2-P3-N3	117.4(2)
N2-P3	1.574(4)	P1-N1-P2	118.9(2)
P3-N3	1.592(4)	P2-N2-P3	122.2(2)
P1-N3	1.589(4)	P1-N3-P3	118.6(2)
P1-O1	1.559(3)	P1-01-C1	118.0(3)
P2-O2	1.551(3)	P2-O2-C15	123.8(3)
P3-O3	1.557(3)	P3-O3-C29	119.5(3)
P1-Cl1	2.021(2)	P1-N1-P2-N2	-13.2(4)
P2-Cl2	2.024(17)	P2-N2-P3-N3	-2.7(4)
P3-Cl3	1.990(18)	P3-N3-P1-N1	-31.8(4)
Puckering amplitude, Q for P3N3 rings	0.261(3)	Max. Deviation for P <sub>3</sub> N <sub>3</sub> rings	0.1826(18) (P1)

 Table S1. Some bond and conformational parameters of compound 3a.

Intermolecular interactions							
			]	I			
I…J			D(I···J)			Symmetry code	
0102			2.682(3)			-1/2+x,1/2-y,z	
С6Н4			2.82			1/2+x,1/2-y,z	
H1H2			2.39			-1/2+x,1/2-y,z	
H1O2			1.90			-1/2+x,1/2-y,z	
			3	Ba			
I…J			D(I···J)			Symmetry code	
Cl1H11			2.94			x,-1+y,z	
Cl2H24			2.94			x,-1+y,z	
Cl2H35			2.88			x,1/2-y,1/2+z	
C7H1B			2.88			1-x,1/2+y,3/2-z	
С9Н18			2.76			x,1/2-y,-1/2+z	
H6H15A			2.35			1-x,1/2+y,3/2-z	
			]	Ι			
I…J				$I(\mathbf{I}) \cdots \mathbf{C} \mathbf{g}(\mathbf{J})$	Symmetry code		
C1—H1BCg1				7(4)	-1/2+x,1/2-y,z		
C11—H11CCg5				4(5)	1/2-x,-1/2+y,-1/2+z		
C11—H11CCg6				0(5)	1/2-x,-1/2+y,-1/2+z		
C13—H13BCg5				2(4)	-1/2+x,1/2-y,z		
C22—H22CCg2			3.552(4)			1/2-x,1/2+y,1/2+z	
C22—H22CCg3			3.561(4)			1/2-x,1/2+y,1/2+z	
			3	Ba			
X—H(I)Cg(J)	HCg	H-Perp	)	XCg	X-HCg	Symmetry code	
C2—H2ACg7	2.88	-2.86		3.664(6)	137	x,y,z	
C18—H18Cg1	2.82	-2.72		3.506(5)	130	x,1/2-y,1/2+z	
C19—H19Cg5	2.81	-2.81		3.600(5)	141	x,1/2-y,1/2+z	
C30—H30ACg9	2.95	-2.83		3.709(7)	135	-x,1-y,1-z	

Table S2. Selected intermolecular interactions for II and 3a.

 Intromologular interactions					
I…J	D(I···J)				
01N1	3.054(4)				
N1C9	2.230(5)				
C2C4	3.074(6)				
C3C9	2.267(5)				
C4C7	2.809(6)				
C2C11	3.056(6)				
C1C3	3.307(5)				
C3C6	2.742(6)				
C5C8	2.777(5)				
C8C10	2.255(5)				
<u>3a</u>					
I…J	D(I···J)				
Cl1C1	3.278(6)				
Cl2C15	3.343(5)				
P1N2	3.110(4)				
P2N3	3.149(4)				
O1N4	3.025(5)				
O3N6	2.876(5)				
N4C9	2.260(6)				
C14C8	2.290(9)				
C14C11	2.751(9)				
C3C9	2.280(7)				
C4C7	2.809(7)				
C15N1	3.177(6)				
N5C23	2.273(6)				
C37N6	2.264(6)				

Table S3. Some selected intramolecular interactions for II and 3a.