

Synergy of highly dispersed Pd and oxygen vacancy promoted hydrodeoxygenation of lactic acid to propionic acid

Mei Zhao,^a Congming Tang, ^a Kai Ma,^b and Xinli Li^{a*}

(¹School of Chemistry and Chemical Engineering, Chongqing University of Technology, Chongqing 400054, PR China; ²Synthetic Lubricants Research Institute of Sinopec Lubricant Co., Ltd., Chongqing 400039, PR China)

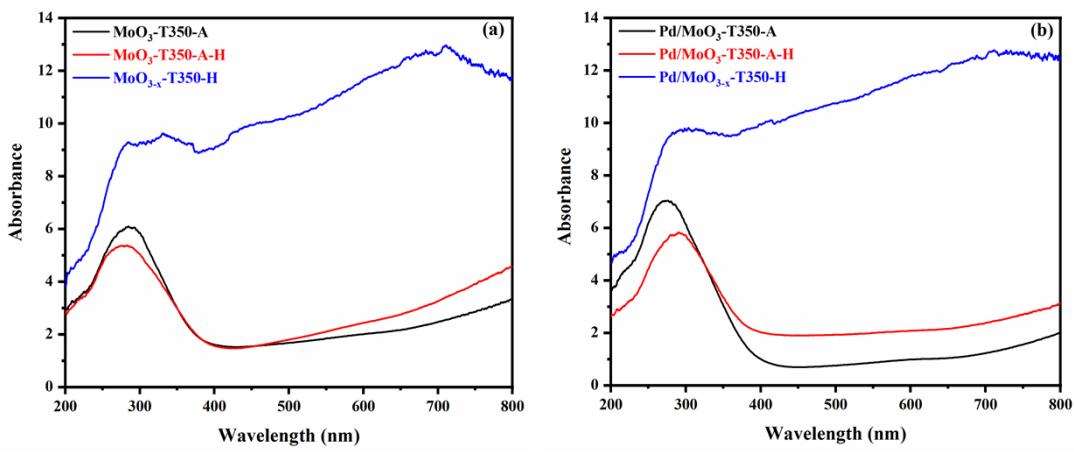


Figure S1. UV-visible absorption spectra of MoO_3 supports (a) and corresponding Pd/MoO_3 catalysts (b).

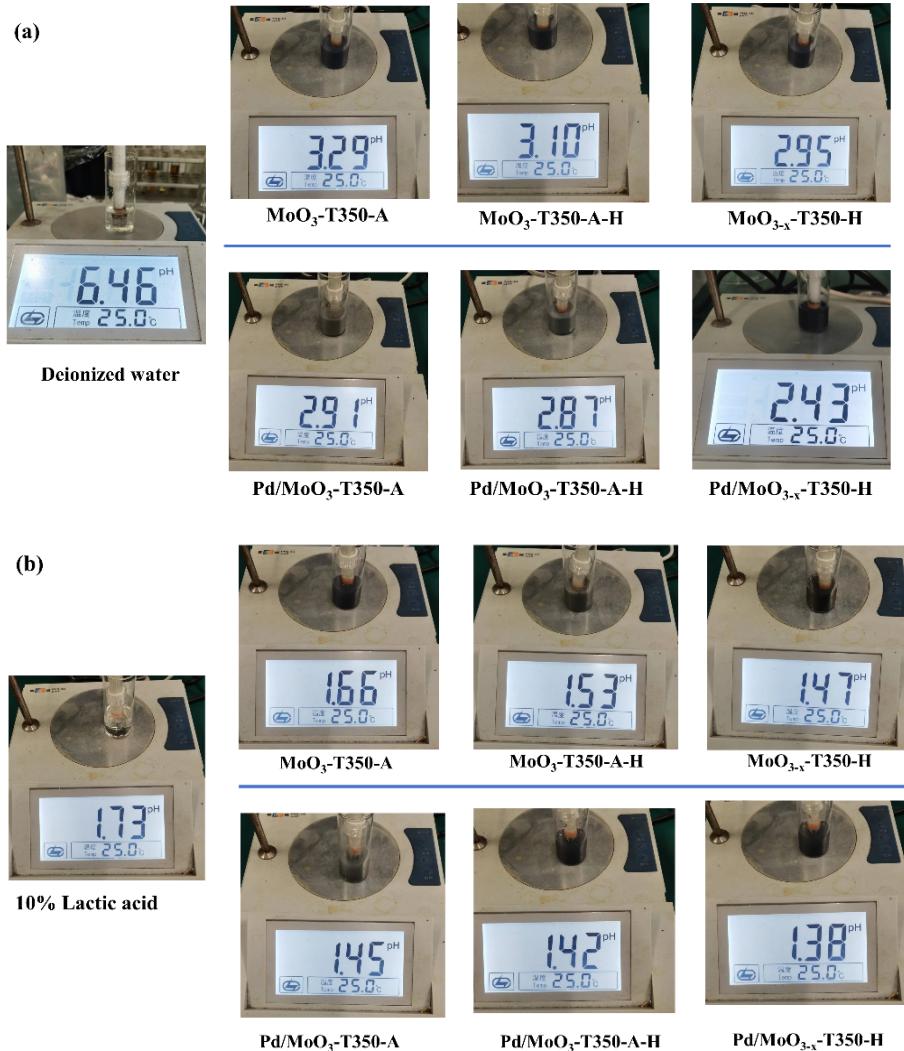


Fig. S2 pH change of pristine supports (a) and their corresponding supported Pd (b) in deionized water and 10 wt% LA aqueous solution.

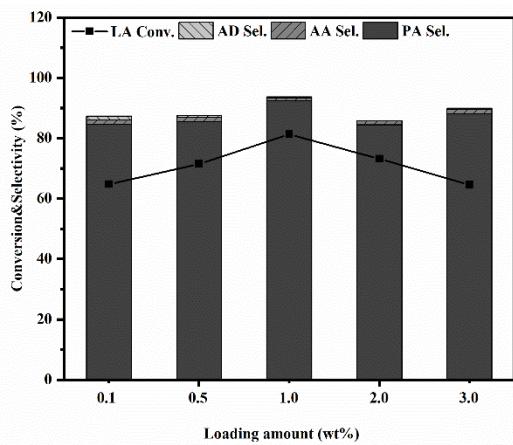


Figure S3. Effect of Pd loading amount on catalytic performance

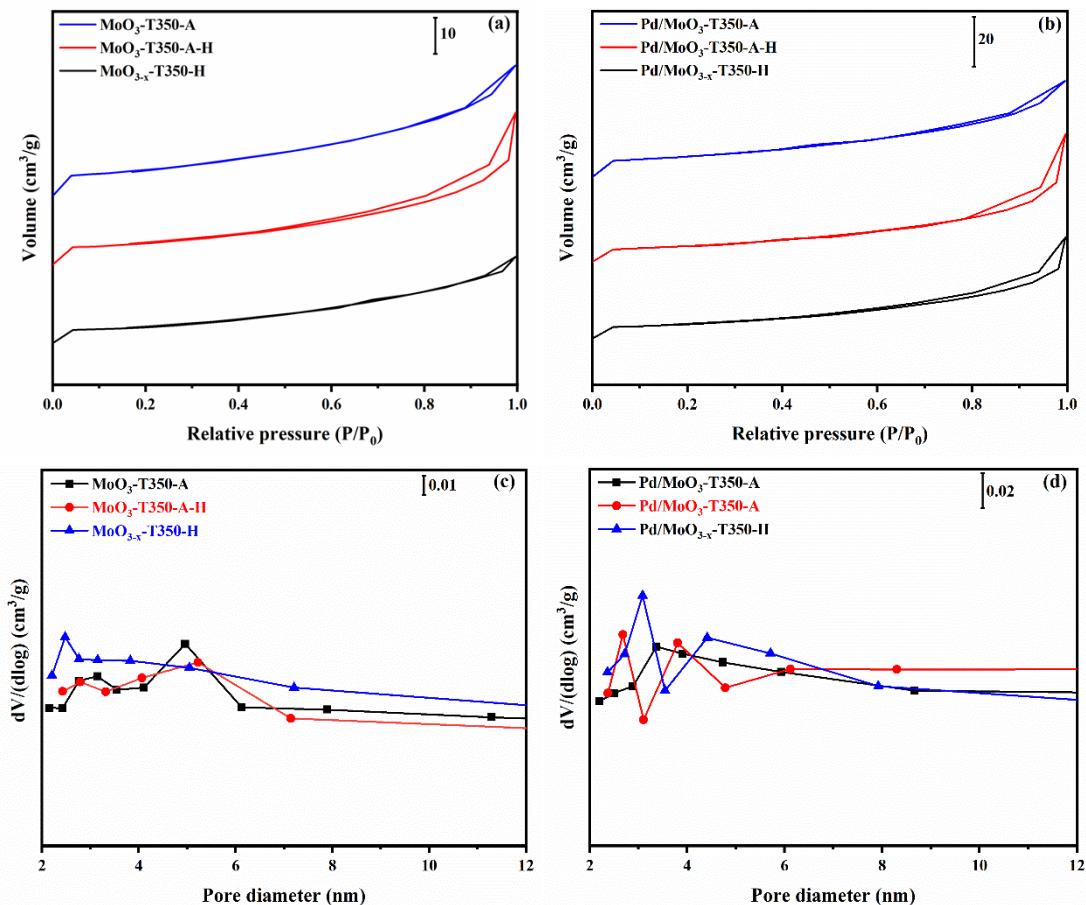


Figure S4. Nitrogen adsorption-desorption isotherm of MoO₃ supports (a) and Pd/MoO₃ catalysts (b); Pore size distribution of MoO₃ supports (c) and Pd/MoO₃ catalysts (d).

Table S1 Energy bands of MoO₃ supports and Pd/MoO₃ catalysts.

Catalyst	E _g (eV)	E _{CB} (eV)	E _{VB} (eV)
MoO ₃ -T350-A	3.12	-0.01	3.11
MoO ₃ -T350-A-H	2.97	-0.08	2.89
MoO _{3-x} -T350-H	2.16	-0.18	1.98
Pd/MoO ₃ -T350-A	2.91	-0.07	2.84
Pd/MoO ₃ -T350-A-H	2.68	-0.21	2.47
Pd/MoO _{3-x} -T350-H	1.38	-0.3	1.08

Table S2. Comparison of HDO Reactivity in the Conversion of Lactic Acid to Propionic Acid for Pd/MoO₃ catalysts, along with Other HDO Catalysts from the Literature.

Entry	Catalyst	Reaction system	Reaction temp. (°C)	LA Conv. (%)	PA (%)	Sel.	Ref.
1	Pd/MoO _{3-x} -T350-H	heterogeneous	215	81.3	92.4		This work
2	MoO ₂ (acac) ₂	homogeneous	270	85.0	52.0		¹
3	Fe _x O _y	heterogeneous	390	96.7	46.7		²
4	Fe ₃ O ₄	heterogeneous	390	93.2	58.4		³
5	MoO ₂	heterogeneous	390	96.8	62.8		⁴
5	Co-Zn	heterogeneous	250	88.0	66.8		⁵
6	NaI	homogeneous	220	100	99.0		⁶
7	^{ss} MoS ₂	heterogeneous	215	96.8	96.3		⁷

Table S3 Pd Leaching rate in the conversion of lactic acid to propionic acid for Pd/MoO_{3-x} tested by ICP-MS.

Catalyst	Concentration of Pd in reacted solution (mg/L)	Pd leaching rate (%)
Pd/MoO _{3-x}	0.58	0.26

Catalysts, 0.5g; LA feedstock: 20 g, 10 wt%, H₂ initial pressure, 3.0 MPa; Reaction temperature, 215°C; reaction time, 12 h.

Table S4 Effect of the amount of 1,1-Diphenylethylene on the catalytic activity of HDO reactivity in the conversion of lactic acid to propionic acid over Pd/MoO_{3-x}.

1,1-Diphenylethylene : LA (mol)	Conv. of LA (%)	Sel. of PA (%)
1:1	78.3	88.9
2:1	79.7	88.2

Catalyst, 0.5 g; LA feedstock, 20 g (5 wt% LA); H₂ initial pressure, 3.0 MPa; Reaction temperature, 215°C; reaction time, 12 h.

References

1. T. J. Korstanje, H. Kleijn, J. Jastrzebski and R. Gebbink, *Green Chem.*, 2013, **15**, 982-988.
2. X. L. Li, Z. J. Zhai, C. M. Tang, L. W. Sun, Y. Zhang and W. Bai, *RSC Adv.*, 2016, **6**, 62252-62262.
3. Y. Zhang, X. Li, L. Sun and C. Tang, *ChemistrySelect*, 2016, **1**, 5002-5007.
4. X. Li, J. Pang, J. Zhang, C. Yin, W. Zou, C. Tang and L. Dong, *Ind. Eng. Chem. Res.*, 2019, **58**, 101-109.
5. Z. B. Huo, J. F. Xiao, D. Z. Ren, F. M. Jin, T. Wang and G. D. Yao, *Green Chem.*, 2017, **19**, 1308-1314.
6. S. Liu, H. Feng, T. Li, Y. Wang, N. Rong and W. Yang, *Green Chem.*, 2020, **22**, 7468-7475.
7. R. Liu, X. Li, Z. Jia, Y. Wang, Y. Peng, C. Tang, Z. Chen and L. Dong, *ACS Sustain. Chem. Eng.*, 2022, **10**, 5463–5475.