

Supplementary Information

Isosteric O/S exchange in carbonyl(thio)ureides: molecular interactions, structure, and bioactivity assays

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Crystallographic Supplementary Information

Table S1. Crystal data and structure refinement results for compounds I and II.

Compound	I	II
Chemical formula	C ₁₃ H ₁₈ N ₂ O ₂ S	C ₁₃ H ₁₈ N ₂ O ₃
Formula weight	266.35	250.29
Temperature (K)	297(2)	297(2)
Wavelength (Å)	0.71073	0.71073
Crystal system	Monoclinic	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> 2 ₁ / <i>c</i>
Unit cell dimensions:		
<i>a</i> (Å)	6.3789(3)	4.8639(2)
<i>b</i> (Å)	23.739(1)	15.2441(7)
<i>c</i> (Å)	9.2761(5)	18.5621(8)
β (°)	98.122(5)	96.637(4)
Volume (Å ³)	1390.6(1)	1367.1(1)
<i>Z</i>	4	4
Density (calculated, Mg/m ³)	1.272	1.216
Absorption coeff. (mm ⁻¹)	0.229	0.087
<i>F</i> (000)	568	536
Crystal color	Colorless	Colorless
Crystal size (mm ³)	0.324 x 0.078 x 0.061	0.402 x 0.110 x 0.079
θ -range (°) for data collection	3.226 to 28.752	3.468 to 29.025
Index ranges	-8 ≤ <i>h</i> ≤ 8, -13 ≤ <i>k</i> ≤ 30, -12 ≤ <i>l</i> ≤ 10	-6 ≤ <i>h</i> ≤ 6, -20 ≤ <i>k</i> ≤ 20, -23 ≤ <i>l</i> ≤ 24
Reflections collected	5746	12946
Independent reflections	3009 [<i>R</i> (int) = 0.0362]	3150 [<i>R</i> (int) = 0.0446]
Obs. reflections [<i>I</i> > 2σ(<i>I</i>)]	1932	1672
Completeness (%)	99.9 (to θ = 25.242°)	99.8 (to θ = 25.242°)
Refinement method	Full-matrix least-squares on <i>F</i> ²	Full-matrix least-squares on <i>F</i> ²
Data / restraints / parameters	3009 / 0 / 235	3159 / 3 / 165

Goodness-of-fit on F^2	1.027	1.027
Final R indices ^a [$I > 2\sigma(I)$]	$R1 = 0.0515$, $wR2 = 0.1021$	$R1 = 0.0785$, $wR2 = 0.2118$
R indices (all data)	$R1 = 0.0937$, $wR2 = 0.1205$	$R1 = 0.1448$, $wR2 = 0.2587$
Larg. diff. peak / hole ($e.\text{\AA}^{-3}$)	0.161 / -0.239	0.472 / -0.335

$$^a R1 = \sum ||F_o| - |F_c|| / \sum |F_o|, wR2 = [\sum w(|F_o|^2 - |F_c|^2)^2 / \sum w(|F_o|^2)]^{1/2}$$

Table S2. Bond lengths [\AA] and angles [$^\circ$] in compound (I).

C(1)-O(1)	1.361(2)
C(1)-C(6)	1.378(3)
C(1)-C(2)	1.379(3)
C(2)-C(3)	1.373(3)
C(3)-C(4)	1.389(3)
C(4)-C(5)	1.379(3)
C(4)-C(8)	1.482(3)
C(5)-C(6)	1.377(3)
C(7)-O(1)	1.416(3)
C(8)-O(2)	1.224(2)
C(8)-N(1)	1.369(3)
C(9)-N(2)	1.319(3)
C(9)-N(1)	1.394(3)
C(9)-S	1.665(2)
C(10)-N(2)	1.458(3)
C(10)-C(11)	1.506(3)
C(11)-C(12)	1.525(3)
C(12)-C(13)	1.510(4)
O(1)-C(1)-C(6)	124.8(2)
O(1)-C(1)-C(2)	115.5(2)
C(6)-C(1)-C(2)	119.7(2)
C(3)-C(2)-C(1)	120.5(2)
C(2)-C(3)-C(4)	120.5(2)
C(5)-C(4)-C(3)	118.2(2)
C(5)-C(4)-C(8)	117.8(2)
C(3)-C(4)-C(8)	123.9(2)
C(6)-C(5)-C(4)	121.6(2)
C(1)-C(6)-C(5)	119.5(2)
O(2)-C(8)-N(1)	121.5(2)
O(2)-C(8)-C(4)	121.4(2)
N(1)-C(8)-C(4)	117.1(2)
N(2)-C(9)-N(1)	116.4(2)
N(2)-C(9)-S	124.4(2)
N(1)-C(9)-S	119.1(2)
N(2)-C(10)-C(11)	110.7(2)
C(10)-C(11)-C(12)	111.4(2)
C(13)-C(12)-C(11)	113.0(3)
C(8)-N(1)-C(9)	128.8(2)
C(9)-N(2)-C(10)	123.9(2)
C(1)-O(1)-C(7)	118.4(2)

Table S3. Bond lengths [Å] and angles [°] in compound II.

C(1)-O(1)	1.363(4)
C(1)-C(2)	1.378(5)
C(1)-C(6)	1.372(4)
C(2)-C(3)	1.371(5)
C(3)-C(4)	1.385(4)
C(4)-C(5)	1.374(4)
C(4)-C(8)	1.481(4)
C(5)-C(6)	1.381(4)
C(7)-O(1)	1.419(4)
C(8)-O(2)	1.222(3)
C(8)-N(1)	1.371(3)
C(9)-O(3)	1.232(3)
C(9)-N(2)	1.320(4)
C(9)-N(1)	1.393(4)
C(10)-N(2)	1.443(4)
C(10)-C(11)	1.490(7)
C(11)-C(12)	1.361(10)
C(12)-C(13)	1.509(12)
O(1)-C(1)-C(2)	116.0(3)
O(1)-C(1)-C(6)	124.7(3)
C(2)-C(1)-C(6)	119.3(3)
C(1)-C(2)-C(3)	120.7(3)
C(2)-C(3)-C(4)	120.7(3)
C(5)-C(4)-C(3)	117.9(3)
C(5)-C(4)-C(8)	124.8(2)
C(3)-C(4)-C(8)	117.3(3)
C(6)-C(5)-C(4)	121.8(3)
C(5)-C(6)-C(1)	119.6(3)
O(2)-C(8)-N(1)	121.8(3)
O(2)-C(8)-C(4)	121.2(2)
N(1)-C(8)-C(4)	117.1(2)
O(3)-C(9)-N(2)	123.3(3)
O(3)-C(9)-N(1)	118.1(3)
N(2)-C(9)-N(1)	118.5(2)
N(2)-C(10)-C(11)	112.6(3)
C(12)-C(11)-C(10)	119.3(7)
C(11)-C(12)-C(13)	111.2(13)
C(8)-N(1)-C(9)	127.8(2)
C(9)-N(2)-C(10)	122.0(3)
C(1)-O(1)-C(7)	118.0(2)

Table S4. Torsion angles [°] in I.

O(1)-C(1)-C(2)-C(3)	179.58(19)
C(6)-C(1)-C(2)-C(3)	-0.2(4)
C(1)-C(2)-C(3)-C(4)	0.0(3)
C(2)-C(3)-C(4)-C(5)	0.5(3)

C(2)-C(3)-C(4)-C(8)	177.69(19)
C(3)-C(4)-C(5)-C(6)	-0.8(3)
C(8)-C(4)-C(5)-C(6)	-178.1(2)
O(1)-C(1)-C(6)-C(5)	-179.8(2)
C(2)-C(1)-C(6)-C(5)	0.0(4)
C(4)-C(5)-C(6)-C(1)	0.6(4)
C(5)-C(4)-C(8)-O(2)	13.0(3)
C(3)-C(4)-C(8)-O(2)	-164.2(2)
C(5)-C(4)-C(8)-N(1)	-168.2(2)
C(3)-C(4)-C(8)-N(1)	14.7(3)
N(2)-C(10)-C(11)-C(12)	-174.2(2)
C(10)-C(11)-C(12)-C(13)	173.8(3)
O(2)-C(8)-N(1)-C(9)	6.3(3)
C(4)-C(8)-N(1)-C(9)	-172.52(19)
N(2)-C(9)-N(1)-C(8)	-1.2(3)
S-C(9)-N(1)-C(8)	178.18(18)
N(1)-C(9)-N(2)-C(10)	-178.7(2)
S-C(9)-N(2)-C(10)	2.0(3)
C(11)-C(10)-N(2)-C(9)	164.5(2)
C(6)-C(1)-O(1)-C(7)	0.8(4)
C(2)-C(1)-O(1)-C(7)	-179.0(3)

Table S5. Torsion angles [°] in II.

O(1)-C(1)-C(2)-C(3)	179.4(3)
C(6)-C(1)-C(2)-C(3)	-0.5(5)
C(1)-C(2)-C(3)-C(4)	0.9(6)
C(2)-C(3)-C(4)-C(5)	-0.4(5)
C(2)-C(3)-C(4)-C(8)	-179.6(3)
C(3)-C(4)-C(5)-C(6)	-0.7(4)
C(8)-C(4)-C(5)-C(6)	178.4(3)
C(4)-C(5)-C(6)-C(1)	1.2(5)
O(1)-C(1)-C(6)-C(5)	179.6(3)
C(2)-C(1)-C(6)-C(5)	-0.6(5)
C(5)-C(4)-C(8)-O(2)	-159.4(3)
C(3)-C(4)-C(8)-O(2)	19.8(4)
C(5)-C(4)-C(8)-N(1)	20.4(4)
C(3)-C(4)-C(8)-N(1)	-160.5(3)
N(2)-C(10)-C(11)-C(12)	167.8(8)
C(10)-C(11)-C(12)-C(13)	-173.9(8)
O(2)-C(8)-N(1)-C(9)	-4.3(5)
C(4)-C(8)-N(1)-C(9)	176.0(3)
O(3)-C(9)-N(1)-C(8)	-173.9(3)
N(2)-C(9)-N(1)-C(8)	5.2(4)
O(3)-C(9)-N(2)-C(10)	2.7(5)
N(1)-C(9)-N(2)-C(10)	-176.3(3)
C(11)-C(10)-N(2)-C(9)	-85.9(5)
C(2)-C(1)-O(1)-C(7)	177.1(3)
C(6)-C(1)-O(1)-C(7)	-3.1(5)

Table S6. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for (I). $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Atom	x	y	z	U(eq)
C(1)	-2264(3)	3351(1)	5560(2)	46(1)
C(2)	-1679(3)	3906(1)	5446(3)	45(1)
C(3)	2(3)	4047(1)	4742(2)	42(1)
C(4)	1141(3)	3632(1)	4136(2)	39(1)
C(5)	524(3)	3079(1)	4251(3)	50(1)
C(6)	-1155(4)	2934(1)	4959(3)	54(1)
C(7)	-4602(5)	2697(1)	6465(5)	73(1)
C(8)	2922(3)	3750(1)	3314(2)	42(1)
C(9)	5283(3)	4518(1)	2670(2)	41(1)
C(10)	7650(4)	4366(1)	842(3)	53(1)
C(11)	8586(4)	3863(1)	176(3)	48(1)
C(12)	10431(4)	4033(1)	-612(3)	54(1)
C(13)	11563(5)	3534(1)	-1155(4)	63(1)
N(1)	3773(3)	4280(1)	3434(2)	44(1)
N(2)	6034(3)	4191(1)	1714(2)	48(1)
O(1)	-3943(2)	3259(1)	6288(2)	64(1)
O(2)	3581(2)	3392(1)	2543(2)	58(1)
S	6003(1)	5186(1)	2994(1)	55(1)

Table S7. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for (II). $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Atom	x	y	z	U(eq)
C(1)	8807(6)	3606(2)	7308(2)	61(1)
C(2)	7581(8)	4112(3)	7798(2)	82(1)
C(3)	5698(7)	4749(3)	7562(2)	77(1)
C(4)	4957(6)	4889(2)	6828(1)	54(1)
C(5)	6179(6)	4374(2)	6347(2)	59(1)
C(6)	8111(6)	3741(2)	6579(2)	60(1)
C(7)	11887(8)	2433(3)	7118(2)	87(1)
C(8)	2925(6)	5594(2)	6620(2)	57(1)
C(9)	-538(6)	6115(2)	5624(2)	55(1)
C(10)	-3496(7)	7369(2)	5742(2)	75(1)
C(11)	-2631(13)	8081(4)	5267(4)	173(3)
C(12)	-4600(30)	8571(7)	4873(6)	299(7)
C(13)	-3330(40)	9189(6)	4369(5)	439(14)
N(1)	1562(5)	5563(2)	5931(1)	56(1)
N(2)	-1238(5)	6795(2)	6003(1)	64(1)
O(1)	10668(5)	2998(2)	7599(1)	78(1)
O(2)	2498(5)	6178(2)	7044(1)	79(1)
O(3)	-1671(5)	5936(2)	5014(1)	77(1)

Table S8. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for I. The anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^2U^{11} + \dots + 2hka^*b^*U^{12}]$.

Atom	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
C(1)	38(1)	49(2)	51(1)	3(1)	4(1)	-4(1)
C(2)	39(1)	42(1)	56(1)	-4(1)	9(1)	2(1)
C(3)	39(1)	35(1)	53(1)	-2(1)	7(1)	-2(1)
C(4)	36(1)	38(1)	42(1)	-4(1)	4(1)	0(1)
C(5)	50(1)	38(1)	65(2)	-5(1)	14(1)	0(1)
C(6)	57(1)	36(1)	70(2)	2(1)	10(1)	-9(1)
C(7)	63(2)	64(2)	97(3)	17(2)	26(2)	-11(2)
C(8)	41(1)	38(1)	47(1)	-6(1)	6(1)	-1(1)
C(9)	38(1)	43(1)	41(1)	-3(1)	8(1)	0(1)
C(10)	62(2)	49(2)	51(2)	-1(1)	24(1)	0(1)
C(11)	54(1)	47(2)	45(1)	-4(1)	15(1)	-2(1)
C(12)	62(2)	51(2)	53(2)	2(1)	23(1)	4(1)
C(13)	68(2)	64(2)	62(2)	-2(2)	23(2)	11(2)
N(1)	45(1)	40(1)	50(1)	-14(1)	17(1)	-4(1)
N(2)	51(1)	46(1)	51(1)	-10(1)	20(1)	-7(1)
O(1)	56(1)	59(1)	81(1)	6(1)	28(1)	-10(1)
O(2)	63(1)	44(1)	72(1)	-19(1)	29(1)	-7(1)
S	71(1)	40(1)	61(1)	-7(1)	28(1)	-8(1)

Table S9. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for II.

Atom	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
C(1)	61(2)	61(2)	59(2)	14(1)	-5(1)	-5(1)
C(2)	100(3)	96(3)	48(2)	6(2)	-5(2)	20(2)
C(3)	93(2)	84(2)	51(2)	-6(2)	-5(2)	18(2)
C(4)	56(2)	57(2)	47(2)	0(1)	-4(1)	-8(1)
C(5)	61(2)	69(2)	44(2)	4(1)	-2(1)	-4(2)
C(6)	60(2)	65(2)	55(2)	0(1)	2(1)	2(1)
C(7)	84(2)	87(3)	88(3)	21(2)	10(2)	19(2)
C(8)	59(2)	60(2)	51(2)	-7(1)	-3(1)	-8(1)
C(9)	63(2)	52(2)	49(2)	-1(1)	-2(1)	-4(1)
C(10)	84(2)	63(2)	74(2)	-1(2)	-5(2)	11(2)
C(11)	158(5)	95(4)	243(8)	60(4)	-71(5)	2(3)
C(12)	500(20)	170(9)	206(9)	83(8)	-40(11)	-3(12)
C(13)	1070(40)	131(7)	154(7)	62(6)	223(15)	62(14)
N(1)	61(1)	55(1)	48(1)	-7(1)	-3(1)	2(1)
N(2)	70(2)	61(2)	56(2)	-9(1)	-7(1)	7(1)
O(1)	85(2)	79(2)	68(1)	18(1)	-4(1)	14(1)
O(2)	93(2)	78(2)	62(1)	-23(1)	-16(1)	15(1)
O(3)	98(2)	71(2)	56(1)	-12(1)	-22(1)	16(1)

Table S10. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for I.

Atom	x	y	z	U(eq)
H(1N)	3470(30)	4487(10)	4040(20)	41(7)
H(2)	-2430(30)	4178(11)	5850(20)	63(7)
H(2N)	5550(40)	3866(11)	1670(30)	56(8)
H(3)	320(30)	4431(10)	4680(20)	40(6)
H(5)	1220(30)	2793(11)	3800(30)	63(7)
H(6)	-1500(30)	2570(10)	5050(20)	51(6)
H(7A)	-4990(40)	2523(13)	5530(30)	89(11)
H(7B)	-5840(40)	2737(13)	6960(30)	89(9)
H(7C)	-3440(40)	2475(13)	6990(30)	91(10)
H(10A)	8780(40)	4582(13)	1500(30)	88(9)
H(10B)	7070(30)	4616(11)	180(30)	59(8)
H(11A)	9060(30)	3594(10)	880(30)	53(7)
H(11B)	7560(30)	3653(10)	-530(30)	58(7)
H(12A)	9920(30)	4294(11)	-1370(30)	56(7)
H(12B)	11430(40)	4274(12)	30(30)	71(8)
H(13A)	12200(40)	3301(12)	-400(30)	65(8)
H(13B)	10670(40)	3311(13)	-1810(30)	81(10)
H(13C)	12720(50)	3641(13)	-1640(30)	97(10)

Table S11. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for II.

Atom	x	y	z	U(eq)
H(2)	80364021	829299		
H(3)	49095091	789993		
H(5)	56924455	585270		
H(6)	89353408	624373		
H(7A)	104702088	6849130		
H(7B)	131922053	7391130		
H(7C)	128222778	6788130		
H(10A)	-42527629	615490		
H(10B)	-49487025	547390		
H(11A)	-14528480	5571207		
H(11B)	-14937817	4930207		
H(12A)	-56178907	5199359		
H(12B)	-59028186	4591359		
H(13A)	-14469022	4339658		
H(13B)	-33849777	4552658		
H(13C)	-43579161	3896658		
H(1)	20635154	565667		
H(2A)	-3206900	641976		

Table S12. Main bond lengths [Å], angles [°] and torsion angles [°] for I and II.

Bond/angles	I		II	
	Exp.	Calc.	Exp	Calc.
C(8)-O(2)	1.224(2)	1.2288	1.222(3)	1.2277
C(8)-N(1)	1.369(3)	1.3828	1.371(3)	1.3794
C(9)-N(2)	1.319(3)	1.3349	1.320(4)	1.3484
C(9)-N(1)	1.394(3)	1.4064	1.393(4)	1.4264
C(9)-S	1.665(2)	1.6772	---	---
C(9)-O(3)	---	---	1.232(3)	1.2209
C(10)-N(2)	1.458(3)	1.4594	1.443(4)	1.4581
C(10)-C(11)	1.506(3)	1.5277	1.490(7)	1.5337
C(11)-C(12)	1.525(3)	1.5326	1.361(10)	1.5323
C(12)-C(13)	1.510(4)	1.5312	1.509(12)	1.5314
O(2)-C(8)-N(1)	121.5(2)	122.24	121.8(3)	122.37
N(2)-C(9)-N(1)	116.4(2)	115.08	118.5(2)	115.76
O(3)-C(9)-N(2)	---	---	123.3(3)	126.14
N(2)-C(9)-S	124.4(2)	125.73	---	---
O(3)-C(9)-N(1)	---	---	118.1(3)	118.10
N(1)-C(9)-S	119.1(2)	118.18	---	---
N(2)-C(10)-C(11)-C(12)	-174.2(2)	179.91	167.8(8)	179.50
C(10)-C(11)-C(12)-C(13)	173.8(3)	170.92	-173.9(8)	-179.86
O(2)-C(8)-N(1)-C(9)	6.3(3)	-3.38	-4.3(5)	-178.81
C(4)-C(8)-N(1)-C(9)	-172.52(19)	177.21	176.0(3)	-178.15
N(2)-C(9)-N(1)-C(8)	-1.2(3)	-0.08	5.2(4)	1.83
O(3)-C(9)-N(1)-C(8)	---	---	-173.9(3)	-178.81
S-C(9)-N(1)-C(8)	178.18(18)	-179.92	---	---
C(11)-C(10)-N(2)-C(9)	164.5(2)	177.70	-85.9(5)	91.01
O(3)-C(9)-N(2)-C(10)	---	---	2.7(5)	2.19
S-C(9)-N(2)-C(10)	2.0(3)	0.18	---	---
N(1)-C(9)-N(2)-C(10)	-178.7(2)	-179.64	-176.3(3)	-178.51

Table S13. Stabilization energies for the **I** and **II** dimers, calculated by NBO (B3LYP/6-31g(d,p)).

Dim.	Donor		Acceptor	E ⁽²⁾	
	NBO (i)	NBO (j)		kJ mol ⁻¹	
				I	II
A	Lp (S)	→	σ*N1-H	5.65	
B	Lp (S)	→	σ*C2-H	5.44	
C	Lp (O)	→	σ*C7-H	0.25	
A	Lp (O3)	→	σ*N1-H	25.9	
	Lp (O3)	→	σ*C5-H	2.85	
B	Lp (O2)	→	σ*C7-H	2.72	
C	Lp (O1)	→	σ*N2-H	4.60	

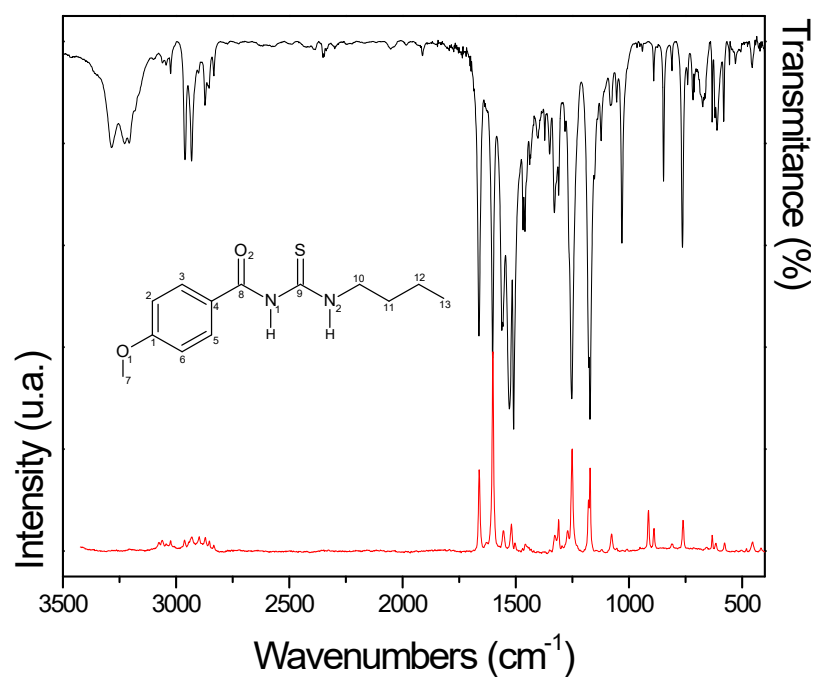


Figure S1. Infrared spectrum (upper trace, KBr pellets) and Raman spectrum (solid, lower trace) of **I**.

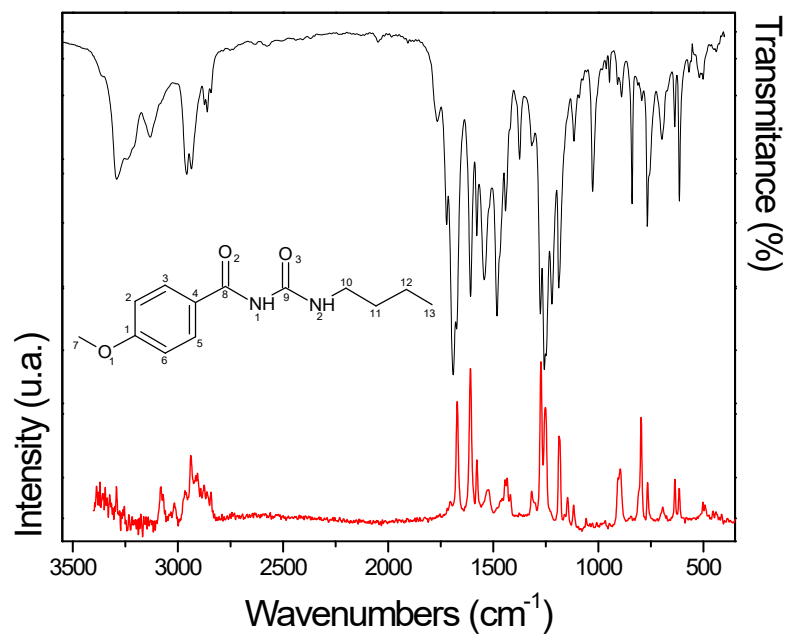


Figure S2. Infrared spectrum (upper trace, KBr pellets) and Raman spectrum (solid, lower trace) of II.

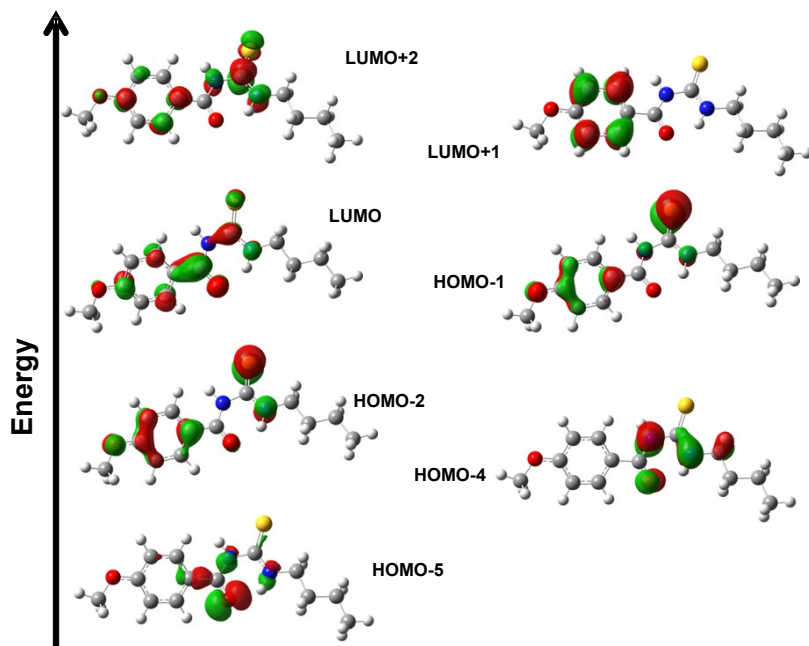


Figure S3. Calculated molecular orbitals for I.

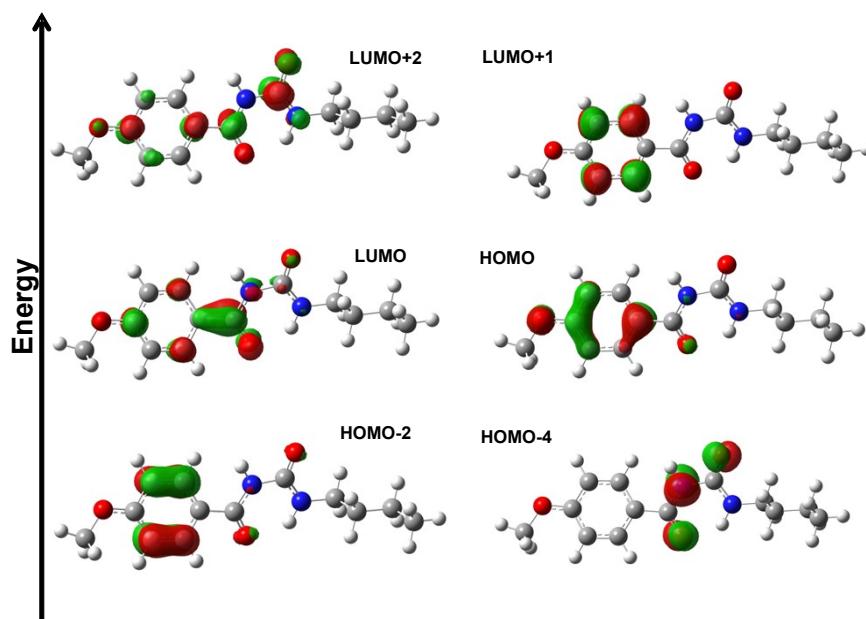


Figure S4. Calculated molecular orbitals for II.

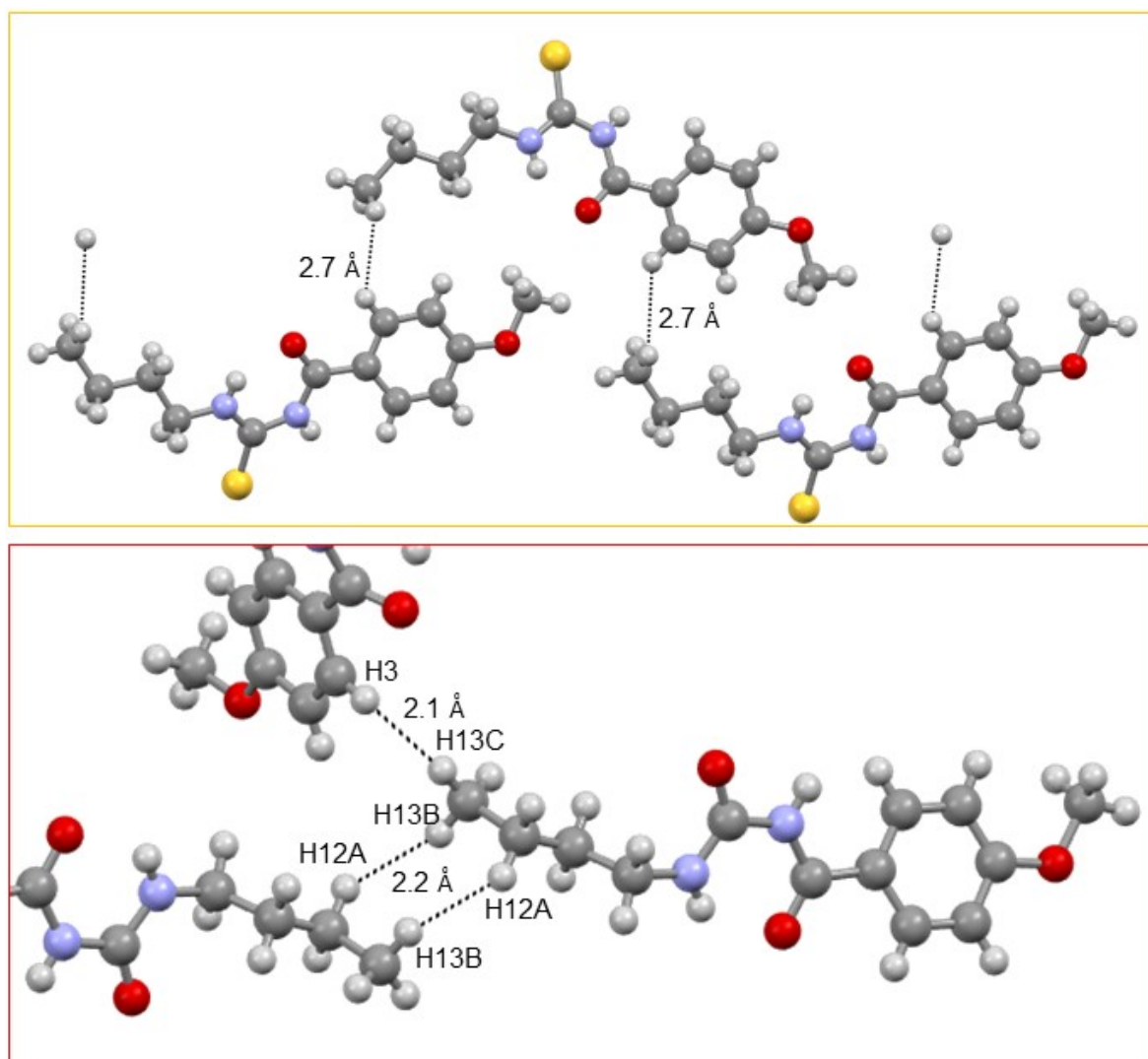


Figure S5. H...H contacts in I (top) and II (bottom).

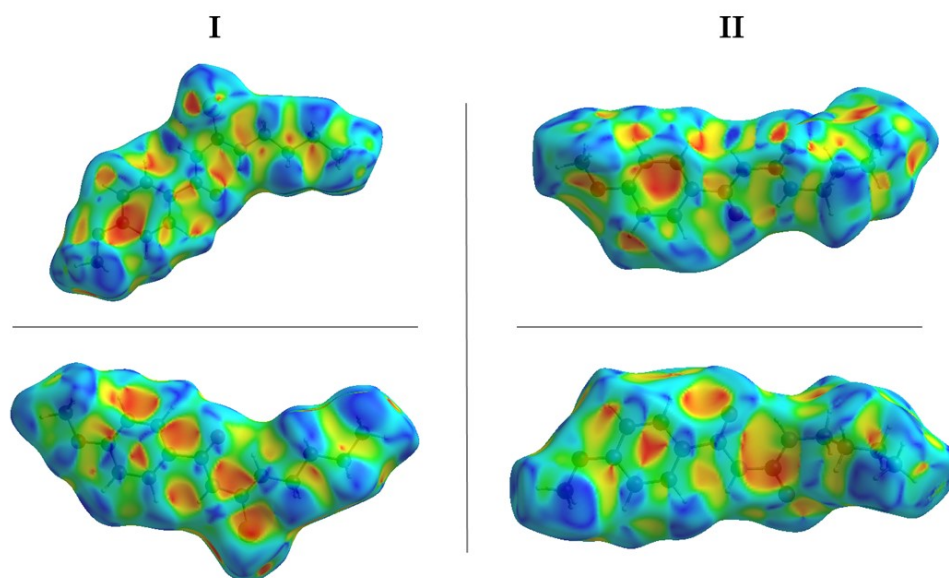


Figure S6. Anterior (top) and posterior (bottom) views of the Hirshfeld surface of **I** (left) and **II** (right) evaluated with the Shape Index descriptor.

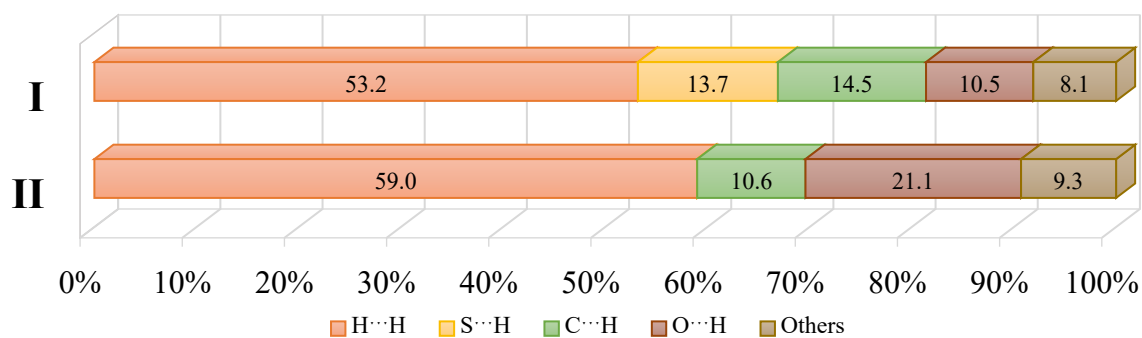


Figure S7. Contribution (%) of the intermolecular contacts of **I** and **II**.

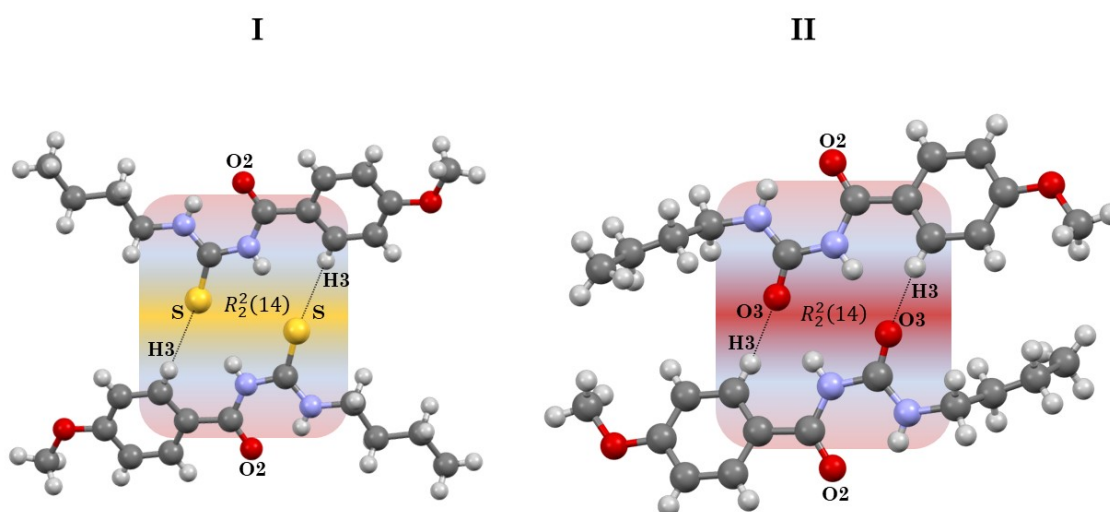


Figure S8. Dimers formed by the interaction C3-H3...S in **I** and C3-H3...O3 **II**.

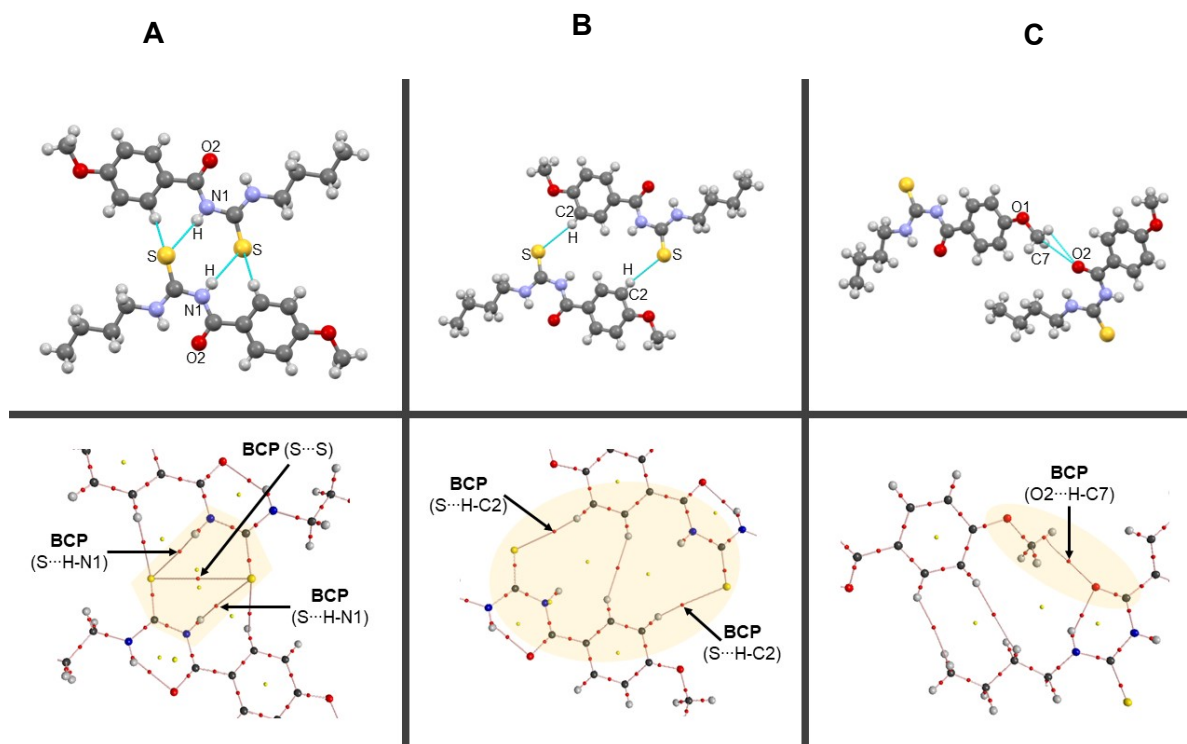


Figure S9. Dimers formed by the interaction $N1-H \cdots S$ (A), $C2-H2 \cdots S$ (B) and $C7H_3 \cdots O2$ (C) in I.

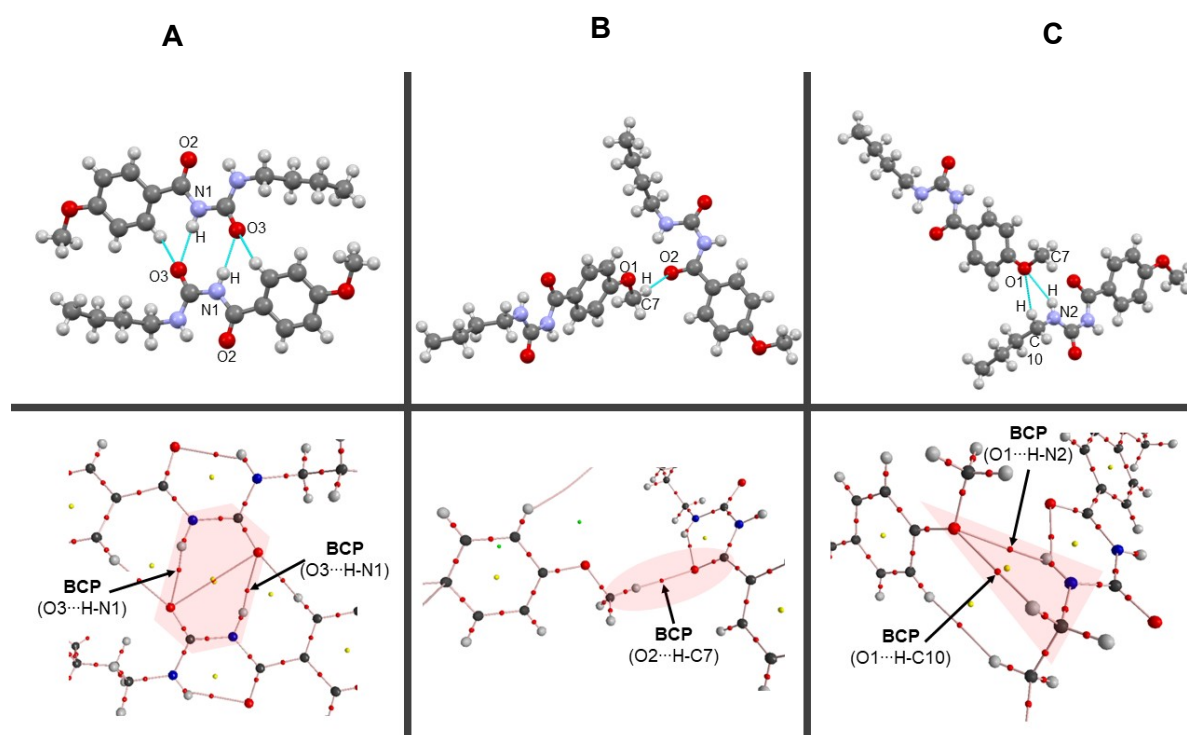


Figure S10. Dimers formed by the interaction $N1-H \cdots O3$ (A), $C7H_3 \cdots O2$ (B) and $N2-H2 \cdots O1$ (C) in II.

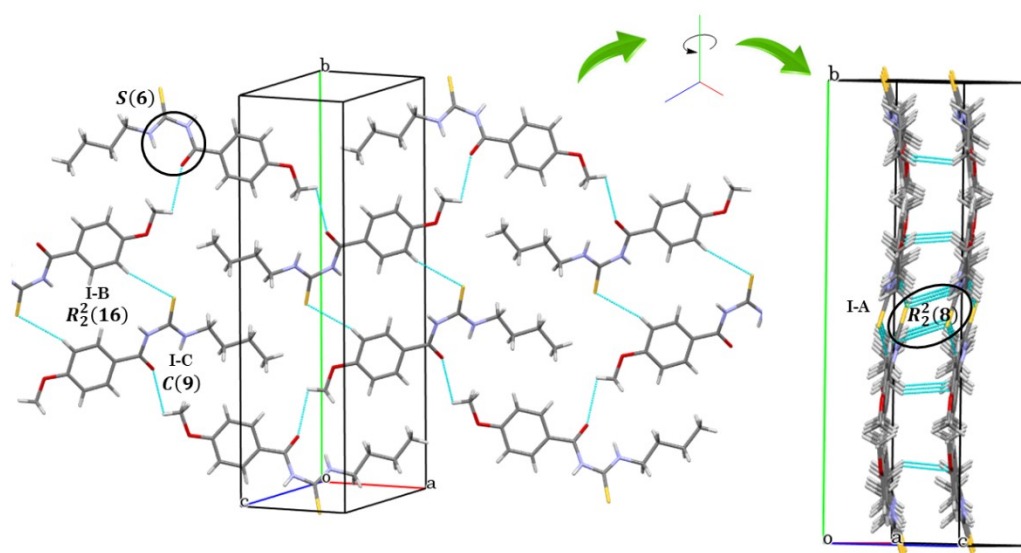


Figure S11. Motifs that form sheets in the crystalline packing of **I**.

In addition to the $R_2^2(8)$ motif associated with the I-A dimer, whose energy contributes to the stabilization of the layers in the packing (see Figure 7), $S(6)$, $R_2^2(16)$ y $C(9)$ motifs that form the sheets were identified in the crystalline packing of **I** (see Figure S11). These motifs correspond to the O2C8N1C9N2H pseudo ring, the ring found in the I-B dimer, and the chain established by the C7-H \cdots O2 interaction displayed in the I-C dimer, respectively. These interactions are depicted in Figure S11. The sheets have a small curvature, as shown on the left side of Figure S11.

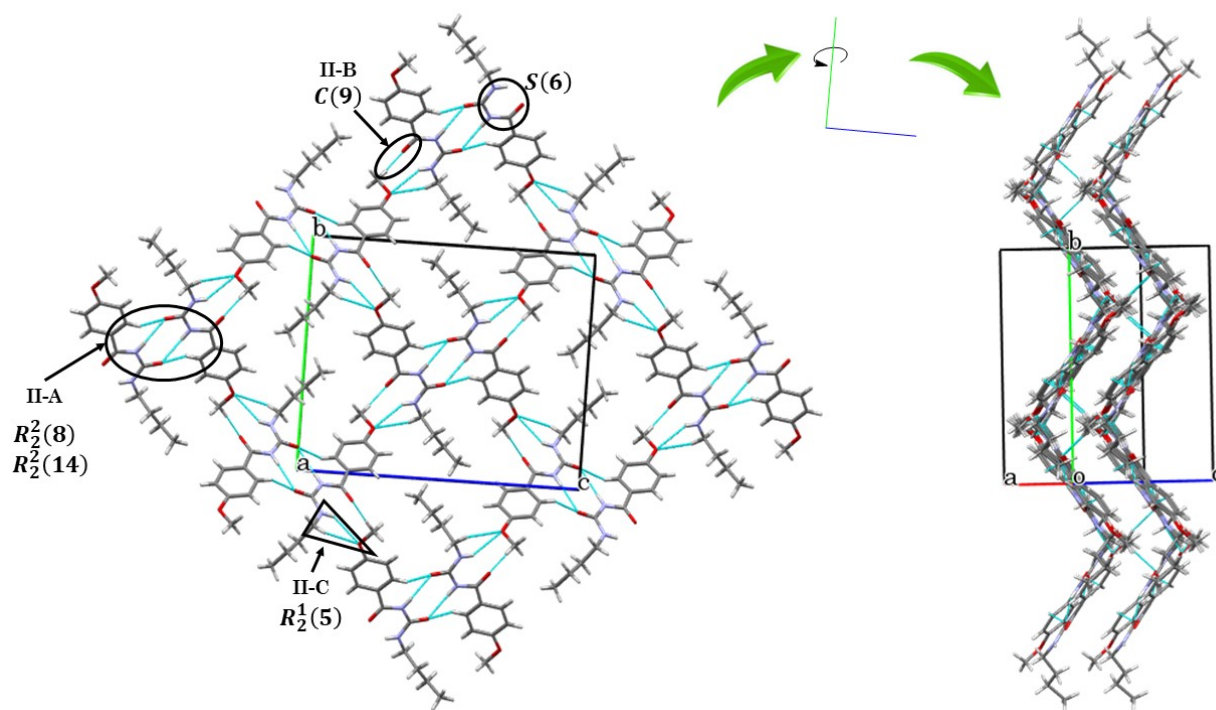


Figure S12. Motifs that form sheets in the crystal packing of **II**.

Likewise, motifs S(6) (of the intramolecular hydrogen bond N2H \cdots O2), C(9) (II-B dimer) and $R_2^1(5)$ (II-C dimer) were identified in **II**. In addition, $R_2^2(8)$ is observed in dimer II-A, in which the $R_2^2(14)$ motif is also formed, as shown in Figure S12. In this case, the indicated motifs are oriented in such a way that favor the formation of sheets, now in the zig-zag form, as represented on the right side of figure S12. The lattice energies (calculated as energy Frameworks) are studied later, as well as the interactions that stabilize the formation of layers in the crystal packing of **II**.

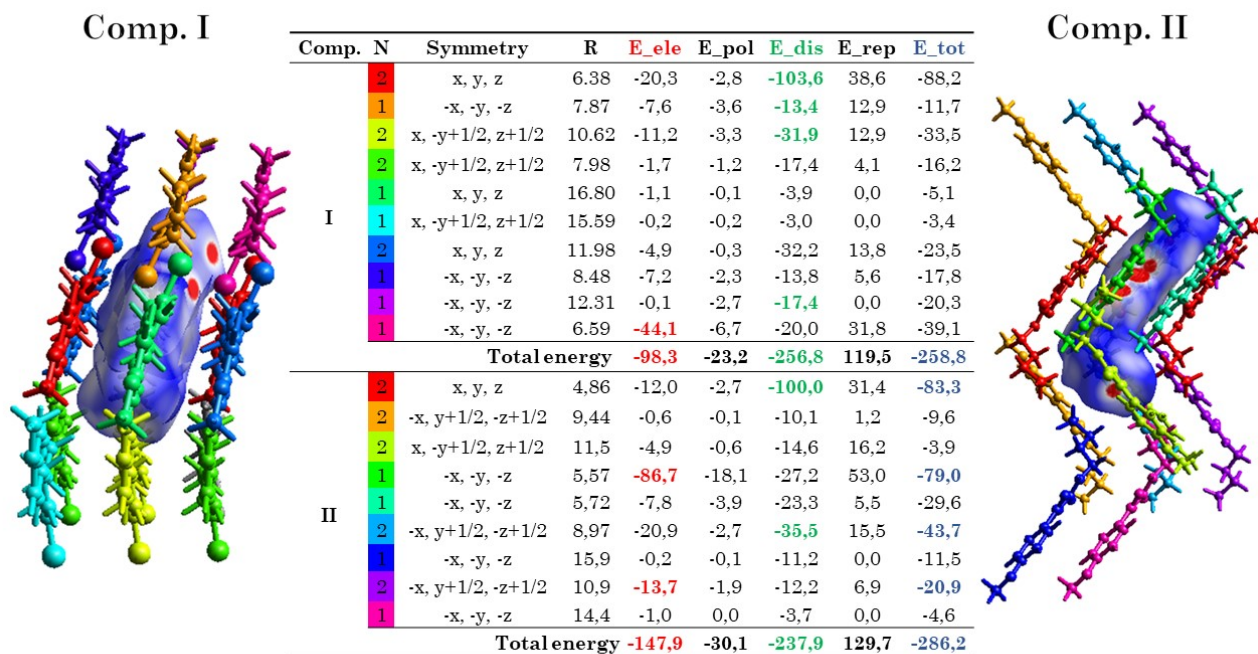


Figure S13. Interaction Energies (kJ/mol). Total energies, only reported for two benchmarked energy models, are the sum of the four energy components, scaled appropriately (CE-B3LYP ... B3LYP/6-31G(d,p) electron densities: k_{ele} , 1.057; k_{pol} , 0.740; k_{disp} , 0.871 and k_{rep} , 0.618).¹

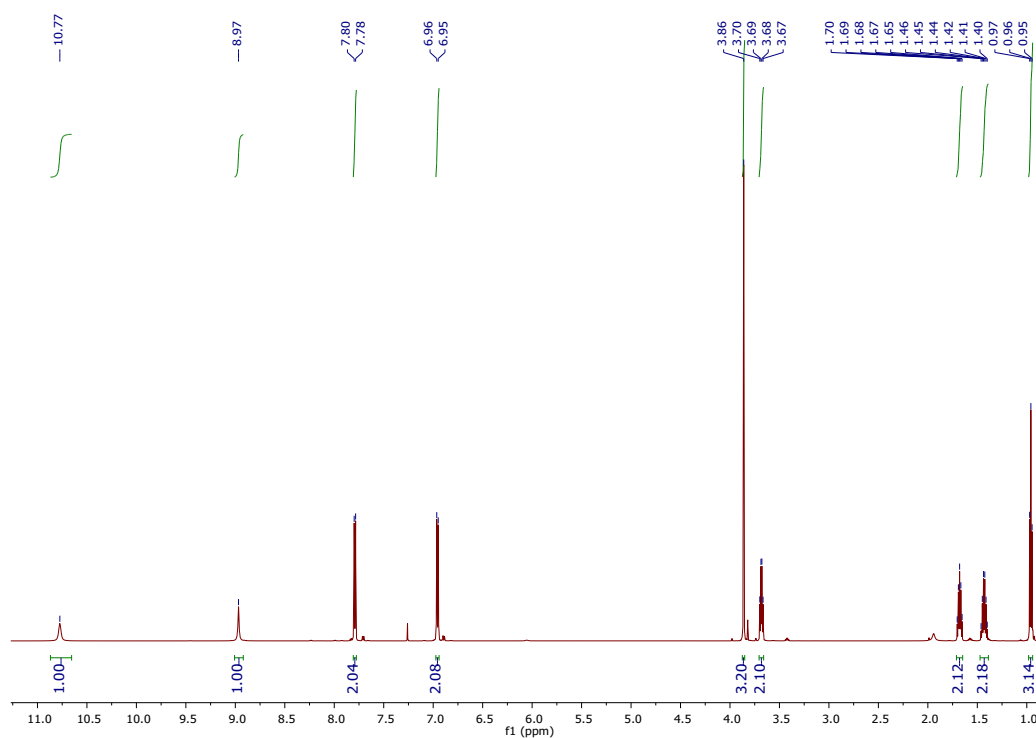


Figure S14. ¹H NMR of *N*-(butylcarbamothioyl)-4-methoxybenzamide (I).

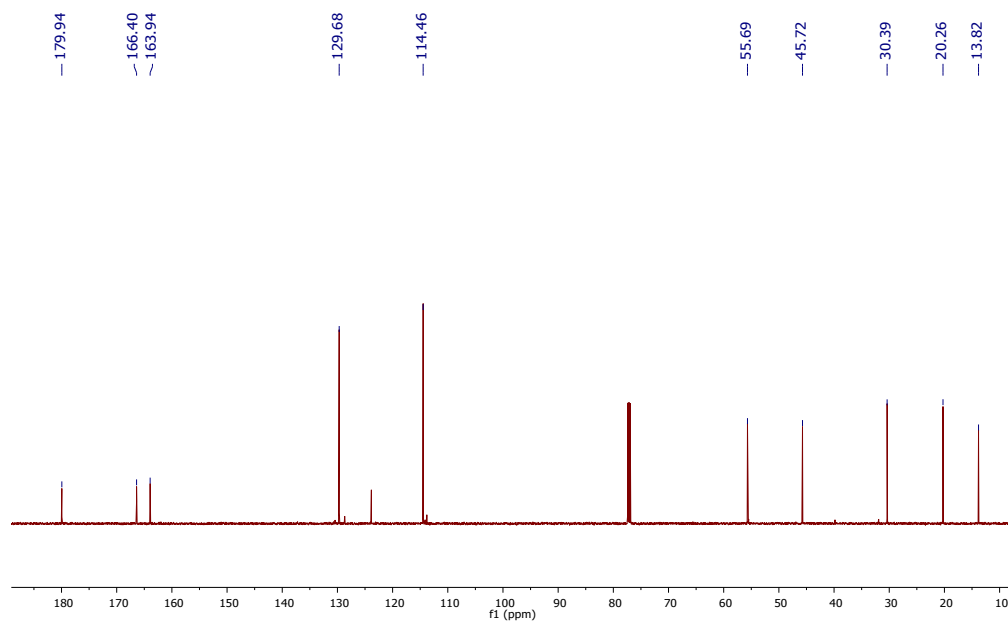


Figure S15. ^{13}C NMR of *N*-(butylcarbamothioyl)-4-methoxybenzamide (**I**).

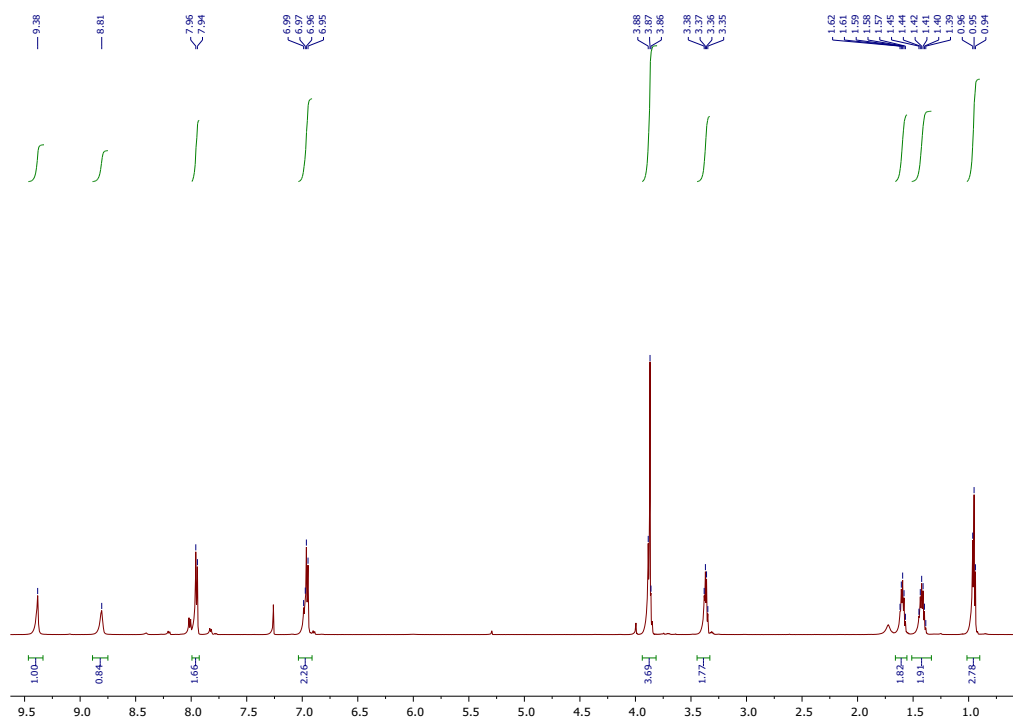


Figure S16. ^1H NMR of *N*-(butylcarbamoyl)-4-methoxybenzamide (**II**).

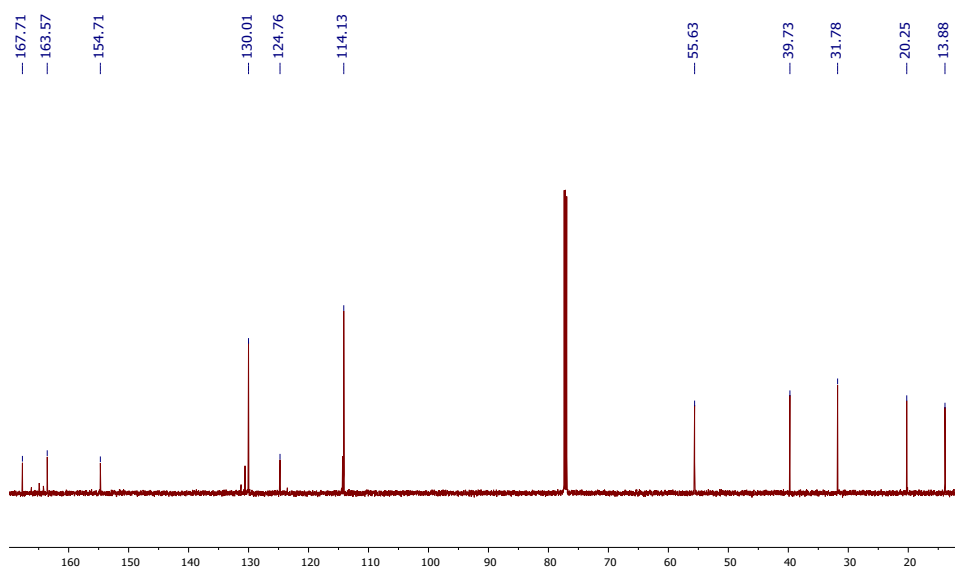


Figure S17. ^{13}C NMR of *N*-(butylcarbamoyl)-4-methoxybenzamide (**II**).

Reference

- 1 M. A. Spackman, P. R. Spackman and S. P. Thomas, in *Complementary Bonding Analysis*, De Gruyter, 2021, pp. 329–352.