

DFT Studies on the Effect of Additives on Stereoselectivity in the Polymerization of Styrene Catalyzed by Rare Earth Metal Complexes

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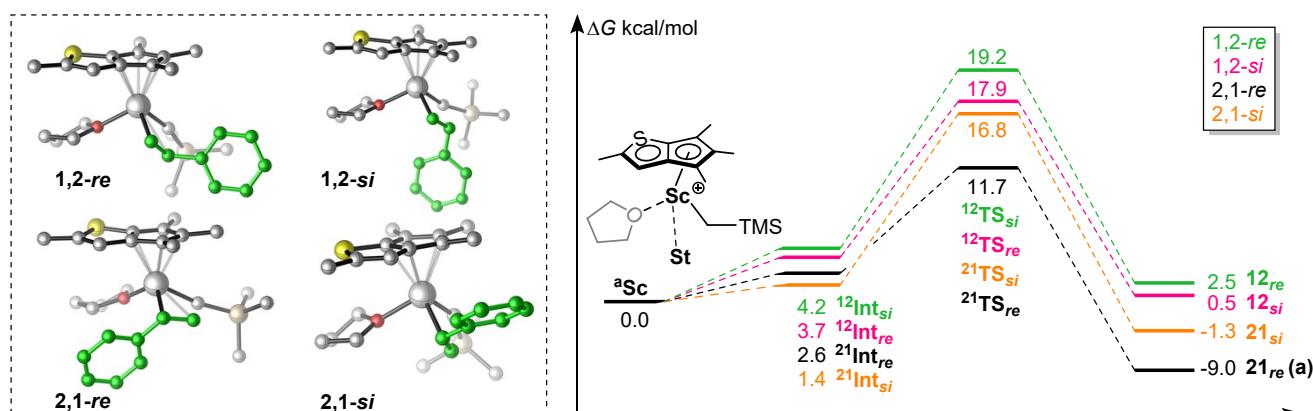


Fig. S1. Computed energy profiles for ${}^3\text{Sc}$ mediated various insertion manners of styrene. Energies are relative to the corresponding cationic species ${}^3\text{Sc}$ and styrene.

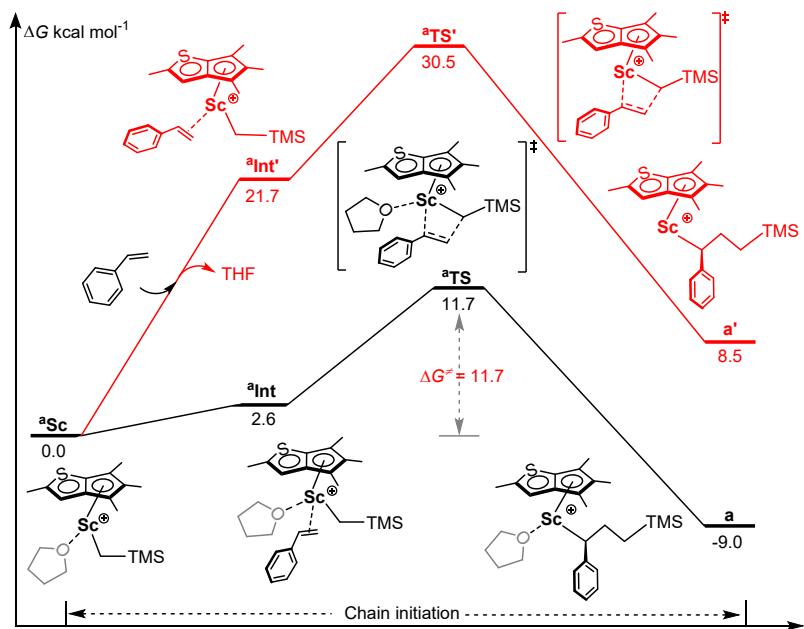


Fig. S2. Calculated energy profiles for styrene insertion into a^{Sc} with THF (black curve) and without THF (red curve) of the first molecule. Energies are relative to the corresponding cationic species a^{Sc} and styrene.

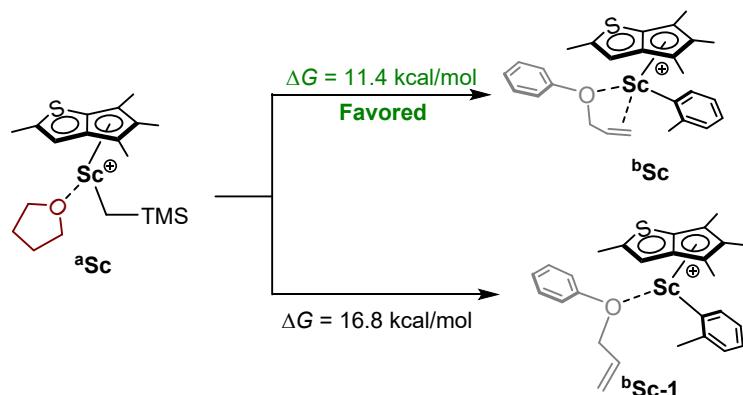


Fig. S3 Computed energy difference between a^{Sc} and b^{Sc} or b^{Sc-1} .

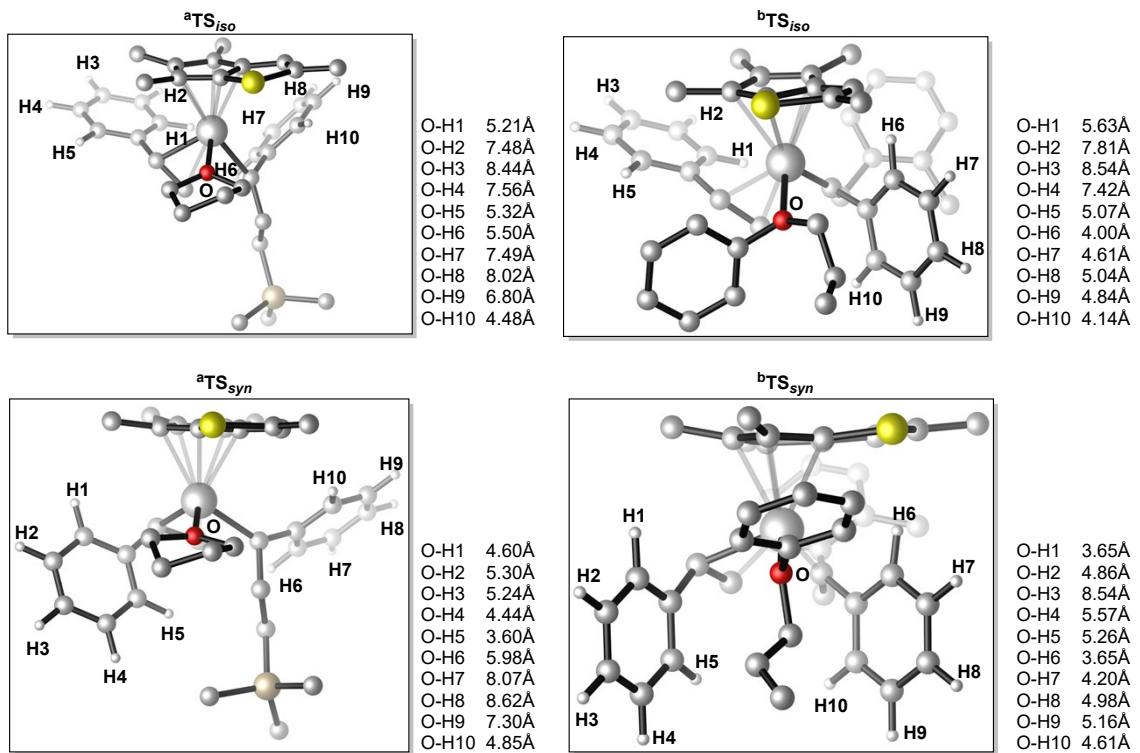
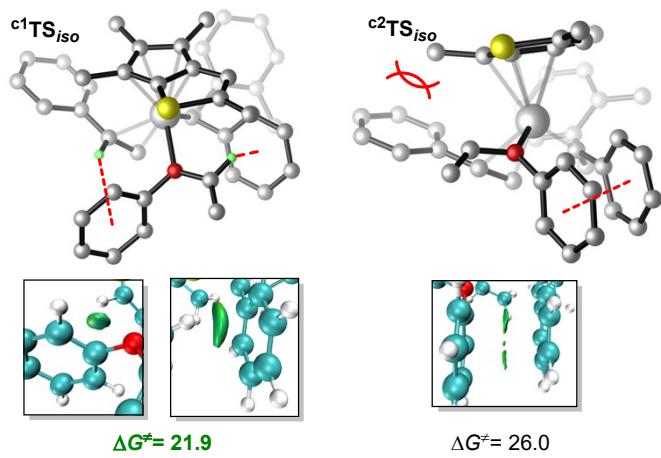


Fig. S4 Geometric structure analysis of **aTS_{iso}**, **aTSS_{syn}**, **bTS_{iso}** and **bTSS_{syn}**.

(a) Noncovalent interaction (NCI) analyses of *iso* conformation



(b) Noncovalent interaction (NCI) analyses of *syndio* conformation

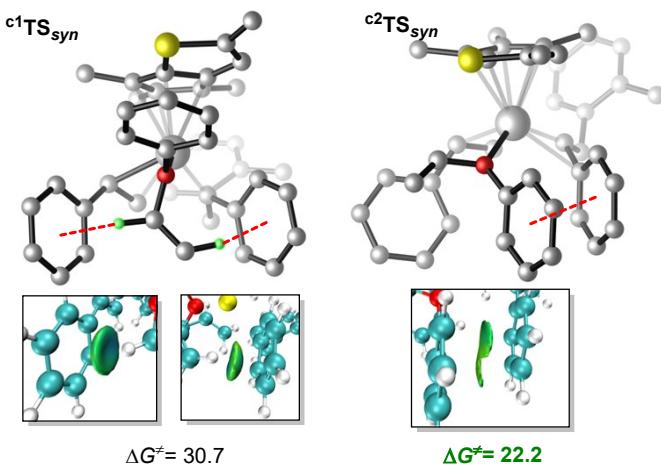


Fig. S5 Noncovalent interaction (NCI) analyses of different conformations in the presence of VOB.

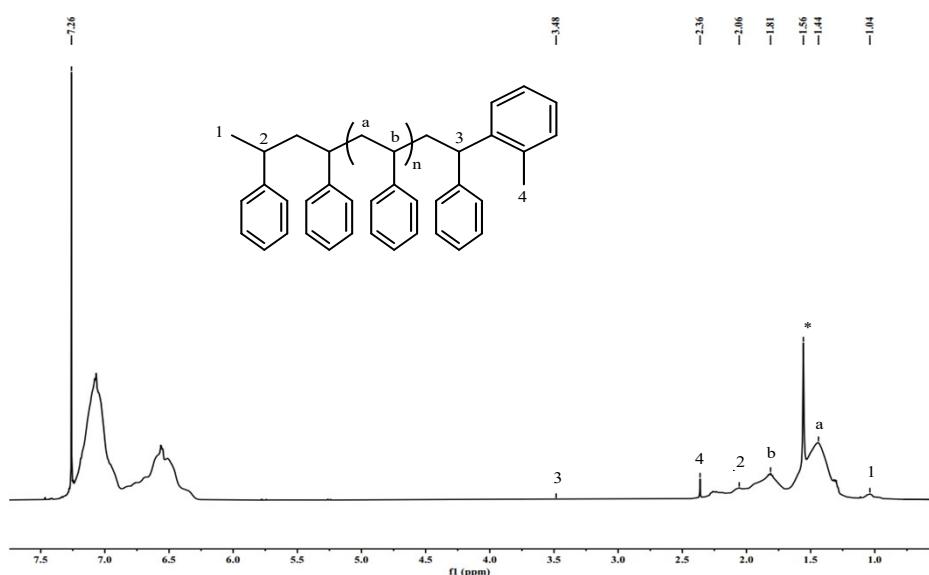


Fig. S6. ^1H NMR spectrum of *a*PS (Table 1, entry 4) obtained in toluene. (25°C , CDCl_3 , ${}^*\text{H}_2\text{O}$).

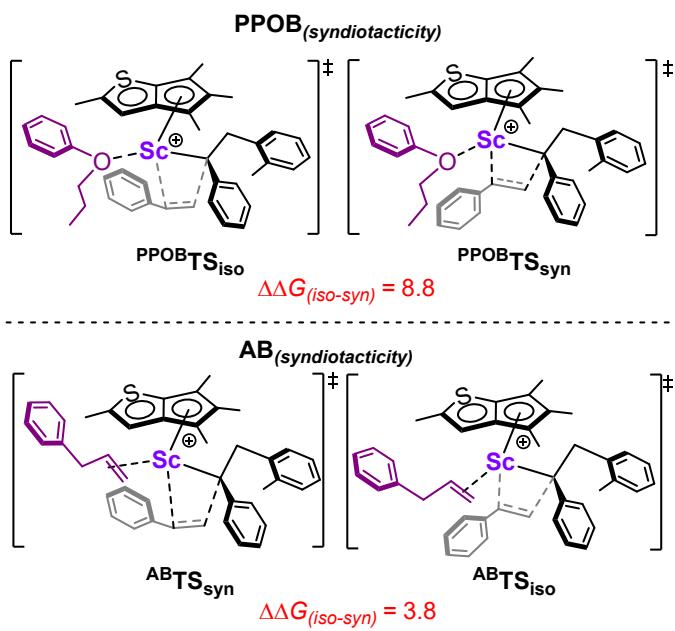


Fig. S7 The energy barrier difference between the isotactic TS and the syndiotactic TS at the chain propagation stage when the additives are PPOB and AB.