Electronic Supplementary Material (ESI) for New Journal of Chemistry. This journal is © The Royal Society of Chemistry and the Centre National de la Recherche Scientifique 2023

### SUPPLEMENTARY INFORMATION

# 4-Hydroxybenzoato-Rare Earth(III) Complexes – Syntheses, a Structural Goldmine from Coordination Diversity and Improved Corrosion Inhibition Behaviour

Joshua S. Moon<sup>a</sup>, Zhifang Guo<sup>b</sup>, Owen Beaumont<sup>a</sup>, Sophie Hamilton<sup>b</sup>, Guillame Bousrez<sup>b</sup>, Eleanor Mottram<sup>b</sup>, Jun Wang<sup>b</sup>, Anthony E. Somers<sup>c</sup>, Glen B. Deacon<sup>a</sup> and Peter C. Junk<sup>b\*</sup>

<sup>a</sup>School of Chemistry, Monash University, Clayton, Vic, 3800, Australia. <sup>b</sup>College of Science & Engineering, James Cook University, Townsville, Qld, 4811, Australia. <sup>c</sup>Institute for Frontier Materials, Deakin University, Burwood, 3125, VIC, Australia. \* Correspondence: peter.junk@jcu.edu.au.

### Contents

- 1. Preparation of 4-Hydroxybenzoato Rare Earth Aqua Complexes and Rare Earth 4hob Complexes with 2,2'-Bipyridine
- 2. Figure S1 PXRD traces of the bulk products for the aqua reactions.
- **3.** Figure S2 Comparison of the PXRD traces for types **a**, **b** and **c** dimers, generated from the crystal data for RE = Tb.
- 4. Figure S3 PXRD traces of (4-hydroxybenzoato)Ln(III) complexes (Ln = Pr and Y) at room temperature compared to their simulated patterns.
- **5.** Figure S4 PXRD traces of dried bulk Ho product at room temperature compared to it simulated patterns.
- Figure S5 PXRD traces for [RE(4hob)<sub>3</sub>(bpy)]<sub>2</sub> (RE = Pr, Nd, Sm, Eu, Gd, Tb, Dy, Er, Yb, Lu, Y) bulk products at room temperature.
- **7. Figure S6** Tafel plots for AS1020 mild steel after 24 h in control and inhibitor solutions.
- 8. Table S1 Hydrogen bonding Parameters [d/Å and </°] for the compounds
- **9. Table S2** Selected distances and bond lengths (Å) for the type b dimers,  $[RE_2(4hob)_6(H_2O)_4] \cdot 2H_2O$ .
- **10.Table S3** Selected distances and bond lengths (Å) for the type c dimers,  $[RE_2(4hob)_6(H_2O)_6] \cdot 10H_2O$ .
- 11.X-ray crystallography
- **12. Table S4** Structural and refinement data for RE-4hob species.
- 13. References

### 1. Preparation of 4-Hydroxybenzoato Rare Earth Aqua Complexes

The complexes were synthesised by reaction between sodium 4-hydroxybenzoate prepared *in situ* unless noted, and a rare earth salt (RECl<sub>3</sub>·*x*H<sub>2</sub>O; RE = Nd, Sm) in a 3:1 mole ratio, typically on a 0.5-1.5 mmol scale (by metal salt). Reactions were conducted in aqueous media at ambient temperature. Precipitates were collected for RE(4hob)<sub>3</sub> (RE = Nd, Sm).

# Characterization of the Aqua Complexes for which structures could not be obtained follow.

 $Md(4hob)_3(H_2O)_2$ . Fibrous bunches of pale purple needles (yield: 63%) dec. 268 °C.  $C_{21}H_{19}NdO_{11}$  (591.61) calculated: C 42.63, H 3.24, Nd 24.38%; found C 42.22, H 3.44, Nd 23.87%.

IR (cm<sup>-1</sup>): 3260s, v br, 1655vw, 1601s, 1509vs, 1413vs, 1252vs, 1176s, 1102m, 1015w, 914vw, 853m, 785vs, 699s.

**Sm(4hob)<sub>3</sub>(H<sub>2</sub>O)<sub>2</sub>.** White precipitate (yield: 91%) dec. 276 °C. C<sub>21</sub>H<sub>19</sub>SmO<sub>11</sub> (597.73) calculated: C 42.20, H 3.20, Sm 25.16%; found C 42.47, H 3.28, Sm 24.83%.

IR (cm<sup>-1</sup>): 3299s, v br, 1698vw, 1599s, 1501s, 1397vs, 1238s, 1168vs, 1101m, 1013vw, 966vw, 854m,784vs, 699m.

#### Preparation of Rare Earth 4hob Complexes with 2,2'-Bipyridine

The complexes were synthesised by a reaction between sodium 4hydroxybenzoate prepared *in situ*, 2,2'-bipyridine (bpy), and a rare earth salt (REX<sub>3</sub>·xH<sub>2</sub>O; X = Cl<sup>-</sup> or NO<sub>3</sub><sup>-</sup>, or  $\frac{1}{2}$  SO<sub>4</sub><sup>2-</sup> (for Ce only) in a 3:1:1 mole ratio, on a 1 mmol scale (by metal salt). Reactions were conducted in aqueous medium at pH ~6. Precipitates were collected for complexes RE(4hob)<sub>3</sub>(bpy) (RE = La, Ce, Pr, Nd, Sm, Eu, Gd, Ho, Y, Er, Yb Lu).

# Characterization of the Complexes with 2,2'-Bipyridine for which structures could not be obtained follow.

**La(4hob)<sub>3</sub>(bpy).** White precipitate (yield: 12%) dec. 188 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>LaO<sub>9</sub> (706.43) calculated: C 52.71, H 3.28, N 3.97, La 19.66%; found C 53.09, H 3.37, N 4.03, La 19.38%.

IR (cm<sup>-1</sup>): 3076w, 1594m, 1525m, 1505m, 1381s, 1273m, 1239m, 1165m, 1098w, 1008w, 855m, 785m, 757m, 700w.

**Ce(4hob)<sub>3</sub>(bpy).** White precipitate (yield: 14%) dec. 152 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>CeO<sub>9</sub> (707.64) calculated: C 52.62, H 3.28, N 3.96, Ce 19.80%; found C 52.96, H 3.41, N 4.02, Ce 19.68%.

IR (cm<sup>-1</sup>): 3072w, 1594m, 1526m, 1505m, 1384s, 1274m, 1239m, 1165m, 1096m, 1009w, 855w, 785m, 757m, 701w.

**Pr(4hob)<sub>3</sub>(bpy).** Green precipitate (yield: 31%) dec. 181 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>PrO<sub>9</sub> (708.43) calculated: C 52.56, H 3.27, N 3.95, Pr 19.89%; found C 51.96, H 3.41, N 4.03, Pr 19.56%.

IR (cm<sup>-1</sup>): 3076w, 1596m, 1529m, 1505m, 1384s, 1275m, 1239m, 1165m, 1099w, 1062w, 1010w, 865w, 785m, 757m, 737w, 701w.

**Nd(4hob)<sub>3</sub>(bpy).** Blue precipitate (yield: 65%) dec. 165 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>NdO<sub>9</sub> (711.76) calculated: C 52.31, H 3.26, N 3.94, Nd 20.27%; found C 52.76, H 3.39, N 3.98, Nd 19.90%.

IR (cm<sup>-1</sup>): 3194w, 1595m, 1531m, 1505m, 1392s, 1247m, 1240m, 1167m, 1099w, 1011w, 856w, 786w, 753s, 686s.

Sm(4hob)<sub>3</sub>(bpy). White precipitate (yield: 72%) dec. 180 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>SmO<sub>9</sub> (717.88) calculated: C 51.87, H 3.23, N 3.90, Sm 20.94%; found C 52.34, H 3.38, N 3.95, Sm 20.67%. IR (cm<sup>-1</sup>): 3160w, 1595m, 1531m, 1505m, 1392s, 1273m, 1240m, 1166m, 1099w,

1063w. 1011w. 856m. 786m. 758m. 737w. 699w.

Eu(4hob)<sub>3</sub>(bpy). White precipitate (yield: 80%) dec. 180 °C.  $C_{31}H_{23}N_2EuO_9$ (719.49) calculated: C 50.62, H 3.15, N 3.88, Eu 20.66%; found C 51.21, H 3.25, N 3.94, Eu 20.33%.

IR (cm<sup>-1</sup>): 3161w, 1595m, 1531m, 1505m, 1386s, 1240m, 1166m, 1098w, 1011w, 855w, 785m, 757m, 699m.

Gd(4hob)<sub>3</sub>(bpy). White precipitate (yield: 90%) dec. 205 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>GdO<sub>9</sub> (724.77) calculated: C 50.26, H 3.13, N 3.85, Gd 21.23%; found C 50.79, H 3.30, N 3.90, Gd 21.10%.

IR (cm<sup>-1</sup>): 3182w, 1595m, 1531m, 1393s, 1245m, 1166m, 1098w, 1012w, 856m, 785m, 757m, 700w.

**Ho(4hob)<sub>3</sub>(bpy).** Pink precipitate (yield: 55%) dec. 190 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>HoO<sub>9</sub> (732.45) calculated: C 50.83, H 3.17, N 3.82, Ho 22.52%; found C 51.31, H 3.39, N 3.97, Ho 21.70%.

IR (cm<sup>-1</sup>): 3535m, 3518m, 3015w, 2969w, 1735s, 1598w, 1538w, 1418w, 1370m, 1216m, 1092w, 744m.

**Y(4hob)<sub>3</sub>(bpy).** White precipitate (yield: 38%) dec. 162 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>YO<sub>9</sub> (656.43) calculated: C 56.72, H 3.53, N 4.27, Y 13.54%; found C 57.15, H 3.74, N 4.43, Y 13.21%.

IR (cm<sup>-1</sup>): 3075w, 1632w, 1595m, 1531w, 1506w, 1457w, 1393s, 1274m, 1240m, 1167m, 1098w, 1063w, 1013w, 856w, 786m, 758m, 737w, 701w.

Er(4hob)<sub>3</sub>(bpy). Pink microcrystals (yield: 86%) dec. 150 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>ErO<sub>9</sub> (734.78) calculated: C 50.67, H 3.16, N 3.81, Er 22.76%; found C 50.99, H 3.31, N 3.94, Er 22.43%.

IR (cm<sup>-1</sup>): 3533w, 3113w, 1596m, 1531m, 1505m, 1397s, 1276m, 1242m, 1167m, 1098w, 1014w, 856w, 815w, 786m, 757s, 701m.

Yb(4hob)<sub>3</sub>(bpy). White precipitate (yield: 86%) dec. 173 °C. C<sub>31</sub>H<sub>23</sub>N<sub>2</sub>YbO<sub>9</sub> (740.56) calculated: C 50.28, H 3.13, N 3.78, Yb 23.37%; found C 49.79, H 3.24, N 3.90, Yb 23.87%.

IR (cm<sup>-1</sup>): 3168w, 1596m, 1531m, 1508m, 1408s, 1275m, 1243m, 1167m, 1099w, 1063w, 1016w, 857w, 787m, 759m, 737w, 701w.

Lu(4hob)<sub>3</sub>(bpy). White precipitate (yield: 91%) dec. 160 °C.  $C_{31}H_{23}N_2LuO_9$ (742.49) calculated: C 50.15, H 3.12, N 3.77, Lu 23.56%; found C 50.59, H 3.90, N 3.87. Lu 23.18%.

IR (cm<sup>-1</sup>): 3188w, 2888w, 1596m, 1531m, 1402s, 1246m, 1167m, 1099w, 1017w, 856m, 786m, 759m, 700w.



**Figure S1.** PXRD traces of the bulk products for the aqua reactions ( $2\theta$  range: 5-55°). All traces are normalized to facilitate comparison.



**Figure S2.** Comparison of the PXRD traces for types **a**, **b** and **c** dimers, generated from the crystal data for RE = Tb ( $2\theta$  range 5-50°). The data for **c** was drawn from the CCDC (DUPWUH).



**Figure S3** PXRD traces of (4-hydroxybenzoato)Ln(III) complexes (Ln = Pr and Y) at room temperature compared to their simulated patterns ( $2\theta$  range 5-50°). The data for Y Simulated was drawn from the CCDC (BEWBAG).

![](_page_4_Figure_2.jpeg)

**Figure S4** PXRD traces of dried bulk Ho product at room temperature compared to it simulated patterns ( $2\theta$  range 5-50°): [Ho<sub>2</sub>(4hob)<sub>6</sub>(H<sub>2</sub>O)<sub>4</sub>]·2H<sub>2</sub>O (a Simulated) and [Ho<sub>2</sub>(4hob)<sub>6</sub>(H<sub>2</sub>O)<sub>4</sub>]·10H<sub>2</sub>O (b Simulated).

![](_page_5_Figure_0.jpeg)

**Figure S5.** PXRD traces for  $[RE(4hob)_3(bpy)]_2$  (RE = Pr, Nd, Sm, Eu, Gd, Tb, Dy, Er, Yb, Lu, Y) bulk products at room temperature (2 $\theta$  range: 5-55°).

![](_page_5_Figure_2.jpeg)

Figure S6. Tafel plots for AS1020 mild steel after 24 h in control and inhibitor solutions.

D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)		
a. [La(4hob) <sub>3</sub> (H <sub>2</sub> O) <sub>2</sub> ] <sub>n</sub>			-			
O2-H2AO7*	0.8710(15)	1.9584(15)	2.806(2)	164.2(1)		
O2-H2BO7#	0.8711(15)	2.0779(15)	2.873(2)	151.46(11)		
b. [Pr <sub>4</sub> (4hob) <sub>12</sub> (H <sub>2</sub> O) <sub>8</sub> ]·4H <sub>2</sub>	0					
O4-H4BO16*	0.9905(16)	1.7526(14)	2.738(2)	172.89(13)		
O3-H3AO5#	0.8506(16)	1.8291(16)	2.677(2)	174.61(14)		
O2#-H2B#O15	0.8539(16)	1.9799(16)	2.833(2)	177.52(13)		
C. [Tb <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>4</sub> ]·4H <sub>2</sub> (	0					
012-H12A03*	0.850(3)	1.8774(17)	2.721(3)	171.0(2)		
d. [Dy <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>4</sub> ]·2H <sub>2</sub> O						
O1-H1BO8	0.861(2)	1.934(2)	2.741(3)	155.53(18)		
e. [Y <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>6</sub> ] 10H <sub>2</sub> O						
O2-H2BO6'	0.850(2)	2.1701(18)	2.93(3)	149.30(13)		

**Table S1.** Hydrogen bonding Parameters [d/Å and </°] for the compounds

**Table S2.** Selected distances and bond lengths (Å) for the type **b** dimers,  $[RE_2(4hob)_6(H_2O)_4] \cdot 2H_2O$ .

RE:	Eu	Gd	Tb	Dy	Но	Y	Er	Yb	Lu
RE-RE	5.0393(3)	5.0331(9)	5.0167(3)	5.0067(6)	4.9931(3)	4.9876(3)	4.9718(5)	4.9402(4)	4.9358(3)
RE-01	2.359(2)	2.3607(17)	2.3397(17)	2.330(3)	2.325(2)	2.3179(12)	2.308(2)	2.283(3)	2.281(3)
RE-02	2.418(2)	2.4206(17)	2.4007(17)	2.406(3)	2.386(2)	2.3779(12)	2.373(3)	2.346(3)	2.347(3)
RE-03	2.412(2)	2.4114(16)	2.3986(17)	2.376(2)	2.369(2)	2.3668(12)	2.350(2)	2.332(3)	2.324(2)
RE-O4	2.444(2)	2.4492(18)	2.4281(17)	2.415(2)	2.402(2)	2.4063(12)	2.390(2)	2.365(3)	2.368(2)
RE-05	2.479(2)	2.4748(17)	2.4648(18)	2.451(2)	2.436(2)	2.4468(12)	2.426(3)	2.413(3)	2.414(2)
RE-06	2.393(2)	2.3987(17)	2.3730(18)	2.368(2)	2.355(2)	2.3532(12)	2.342(3)	2.317(3)	2.313(2)
RE-07	2.327(2)	2.3278(17)	2.3073(17)	2.292(2)	2.283(2)	2.2829(12)	2.267(2)	2.239(3)	2.243(2)
RE-08'	2.349(2)	2.3482(17)	2.3312(18)	2.316(2)	2.3045(19)	2.3006(12)	2.288(2)	2.273(3)	2.263(2)

**Table S3.** Selected distances and bond lengths (Å) for the type **c** dimers,  $[RE_2(4hob)_6(H_2O)_6] \cdot 10H_2O$ . Data for RE = Tb is drawn from CCDC code DUPWUH.

RE:	Tb	Но	Y	Er
RE-01	2.4482(19)	2.4205(17)	2.4316(17)	2.3132(14)
RE-02	2.423(2)	2.3991(16)	2.4118(16)	2.3923(14)

RE-03	2.4042(17)	2.3789(17)	2.3807(18)	2.3697(14)
RE-04	2.4276(19)	2.4026(17)	2.3974(19)	2.3908(14)
RE-05	2.4387(17)	2.4167(17)	2.4246(19)	2.4103(15)
RE-06	2.3161(14)	2.2930(16)	2.3004(15)	2.2848(13)
RE-07'	2.3276(18)	2.3012(16)	2.2985(19)	2.2973(14)
RE-08	2.2790(17)	2.2593(16)	2.2569(17)	2.2511(14)

### X-ray crystallography

[Tb<sub>2</sub>(4hob)<sub>6</sub>(H<sub>2</sub>O)<sub>4</sub>]·4H<sub>2</sub>O, Complexes  $[Pr_4(4hob)_{12}(H_2O)_8] \cdot 4H_2O$ ,  $[Dy_2(4hob)_6(H_2O)_4] \cdot 2H_2O$ ,  $[Y_2(4hob)_6(H_2O)_6] \cdot 10H_2O$ ,  $[Er_{2}(4hob)_{6}(H_{2}O)_{4}]\cdot 2H_{2}O$ ,  $[Yb_2(4hob)_6(H_2O)_4]\cdot 2H_2O$ ,  $[Lu_2(4hob)_6(H_2O)_4]\cdot 2H_2O$ ,  $[Tb_2(4hob)_6(bpy)_2]\cdot 2bpy\cdot 4H_2O$ and [Dy<sub>2</sub>(4hob)<sub>6</sub>(bpy)<sub>2</sub>]·2bpy·4H<sub>2</sub>O were measured on a Bruker APEX-II CCD diffractometer equipped with graphite-monochromated Mo-Ka radiation ( $\lambda = 0.71073$ Å) at 123 K. Absorption corrections were completed using Apex II program suite using SADABS.<sup>1</sup> Complexes  $[La(4hob)_3(H_2O)_2]_n$  and  $[Gd_2(4hob)_6(H_2O)_4] \cdot 2H_2O$  were measured at the Australian Synchrotron on the MX1 beamline, data integration was completed using Blue-ice<sup>2</sup> and XDS<sup>3</sup> software programs. Others were measured on a Rigaku SynergyS diffractometer at 123 K. The SynergyS operated using microsource Mo-K $\alpha$  radiation ( $\lambda = 0.71073$  Å) for complex [Er<sub>2</sub>(4hob)<sub>6</sub>(H<sub>2</sub>O)<sub>6</sub>]·10H<sub>2</sub>O or Cu-Kα radiation ( $\lambda = 1.54184$ Å) for complexes {[Ce(4hob)<sub>3</sub>(H<sub>2</sub>O)<sub>2</sub>]·3H<sub>2</sub>O}<sub>n</sub>, [{Y(4hob)<sub>3</sub>}·H<sub>2</sub>O]<sub>n</sub>,  $[Eu_2(4hob)_6(H_2O)_4] \cdot 2H_2O$ ,  $[Tb_2(4hob)_6(H_2O)_4] \cdot 2H_2O$ ,  $[Ho_2(4hob)_6(H_2O)_4] \cdot 2H_2O$ ,  $[Ho_2(4hob)_6(H_2O)_6] \cdot 10H_2O$ ,  $[Y_2(4hob)_6(H_2O)_4] \cdot 2H_2O$ . Data processing was conducted using CrysAlisPro.55 software suite.<sup>4</sup> Structural solutions were obtained by either direct methods <sup>5</sup> or charge flipping <sup>6</sup> methods and refined using full-matrix leastsquares methods against F<sup>2</sup> using SHELX2018,<sup>7</sup> in conjunction with the Olex2<sup>8</sup> graphical user interface. All hydrogen atoms were placed in calculated positions using the riding model. Crystal data and refinement details are given in **Table S3**.

Table S4. Structural and refinement data for RE-4ho	species.
---	----------

	[La(4hob) <sub>3</sub> (H <sub>2</sub> O) <sub>2</sub> ] <sub>n</sub>	${[Ce(4hob)_{3}(H_{2}O)_{2}]\cdot 3H_{2}O}_{n}$	$[Pr_4(4hob)_{12}(H_2O)_8] \cdot 4H_2O$	$[Tb_2(4hob)_6(H_2O)_4]\cdot 4H_2O$	$[Eu_2(4hob)_6(H_2O)_4] \cdot 2H_2O$
Formula	C <sub>21</sub> H <sub>19</sub> LaO <sub>11</sub>	C <sub>21</sub> H <sub>25</sub> CeO <sub>14</sub>	C <sub>84</sub> H <sub>84</sub> O <sub>48</sub> Pr <sub>4</sub>	$C_{42}H_{46}O_{26}Tb_2$	C <sub>42</sub> H <sub>42</sub> Eu <sub>2</sub> O <sub>24</sub>
M <sub>r</sub>	586.27	641.53	2425.15	1284.63	1234.67
Space group	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	P21/c
a (Å)	16.215(3)	6.36120(10)	12.0467(3)	9.0067(7)	10.4168(3)
b (Å)	12.467(3)	10.0317(2)	14.1840(4)	11.0069(9)	14.9277(5)
c (Å)	10.233(2)	19.6952(9)	14.6821(4)	12.9695(10)	15.5044(5)
α (°)	90	104.267(3)	93.1860(10)	91.8950(10)	90
β (°)	94.01(3)	93.474(3)	102.7450(10)	108.879(2)	105.9590(10)
γ (°)	90	96.567(2)	105.5620(10)	93.5260(10)	90
V (Å <sup>3</sup> )	2063.6(7)	1204.91(7)	2339.57(11)	1212.36(17)	2318.00(13)
Ζ	4	2	1	1	2
$\rho_{\rm calc},  {\rm g}  {\rm cm}^{-3}$	1.887	1.768	1.721	1.760	1.769
$\mu$ , mm <sup>-1</sup>	2.134	15.248	2.145	2.982	19.940
Ν <sub>τ</sub>	26356	4963	51444	8275	24759
N (R <sub>int</sub> )	4444(0.0259)	4963	13334(0.0539)	5757(0.0230)	4842(0.0595)
$R_1(l > 2\sigma(l))$	0.0207	0.1128	0.0314	0.0278	0.0355
wR <sub>2</sub> (all data)	0.0540	0.2855	0.0785	0.0723	0.0988
GOF	1.096	1.358	1.027	1.071	1.060

	$[Gd_2(4hob)_6(H_2O)_4] \cdot 2H_2O$	[Tb <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>4</sub> ]·2H <sub>2</sub> O	$[Dy_2(4hob)_6(H_2O)_4]\cdot 2H_2O$	[Ho <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>4</sub> ]·2H <sub>2</sub> O	$[Y_2(4hob)_6(H_2O)_4] \cdot 2H_2O$
Formula	$C_{42}H_{42}Gd_2O_{24}$	C <sub>42</sub> H <sub>42</sub> O <sub>24</sub> Tb <sub>2</sub>	$C_{42}H_{42}Dy_2O_{24}$	C <sub>42</sub> H <sub>42</sub> Ho <sub>2</sub> O <sub>24</sub>	$C_{42}H_{42}O_{24}Y_2$
M <sub>r</sub>	1245.25	1248.59	1255.75	1260.61	1108.57
Space group	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> 2 <sub>1</sub> /c
a (Å)	10.4100(3)	10.39280(10)	10.4033(3)	10.3692(3)	10.37330(10)
b (Å)	14.9000(5)	14.9032(2)	14.8810(5)	14.8578(5)	14.88310(10)
c (Å)	15.5100(5)	15.4634(2)	15.5628(5)	15.4168(5)	15.43080(10)
α (°)	90	90	90	90	90
β (°)	105.9610(10)	105.8680(10)	105.9720(10)	105.8910(10)	105.8630(10)
γ (°)	90	90	90	90	90
V (Å <sup>3</sup> )	2313.00(13)	2303.80(5)	2316.29(13)	2284.40(13)	2291.59(3)
Ζ	2	2	2	2	2
$ ho_{ m calc}$ , g cm <sup>-3</sup>	1.788	1.800	1.800	1.833	1.607
μ, mm <sup>-1</sup>	2.930	15.670	3.289	7.048	4.163
Ν <sub>τ</sub>	38190	24578	34051	24375	42965
N (R <sub>int</sub> )	4213(0.0358)	4808(0.0437)	5313(0.0701)	4773(0.0397)	4773(0.0338)
$R_1(I > 2\sigma(I))$	0.0199	0.0256	0.0319	0.0277	0.0253
wR <sub>2</sub> (all data)	0.0520	0.0649	0.0820	0.0750	0.0702
GOF	1.138	1.050	1.119	1.120	1.106

	$[\text{Er}_2(4\text{hob})_6(\text{H}_2\text{O})_4]\cdot 2\text{H}_2\text{O}$	$[Yb_2(4hob)_6(H_2O)_4] \cdot 2H_2O$	[Lu <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>4</sub> ]·2H <sub>2</sub> O	[Ho <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>6</sub> ]·10H <sub>2</sub> O	[Y <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>6</sub> ]·10H <sub>2</sub> O
Formula	C <sub>42</sub> H <sub>42</sub> Er <sub>2</sub> O <sub>24</sub>	$C_{42}H_{42}O_{24}Yb_2$	$C_{42}H_{42}Lu_2O_{24}$	C <sub>42</sub> H <sub>62</sub> Ho <sub>2</sub> O <sub>34</sub>	$C_{42}H_{62}O_{34}Y_2$
M <sub>r</sub>	1265.27	1276.83	1280.69	1440.77	1288.73
Space group	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> 2 <sub>1</sub> /c	<i>P</i> -1	<i>P</i> -1
a (Å)	10.3698(3)	10.3616(3)	10.3683(3)	10.8167(2)	10.7986(9)
b (Å)	14.8378(5)	14.8065(5)	14.8139(5)	11.3009(2)	11.4256(9)
<i>c</i> (Å)	15.5102(5)	15.4569(5)	15.4882(5)	11.45320(10)	11.6045(9)
α (°)	90	90	90	90.5010(10)	90.629(2)
β (°)	106.0290(10)	105.8930(10)	105.9210(10)	101.7230(10)	100.564(2)
γ (°)	90	90	90	105.3020(10)	105.4750(10)
V (Å <sup>3</sup> )	2293.70(13)	2280.74(13)	2287.66(13)	1319.33(4)	1353.76(19)
Ζ	2	2	2	1	1
$ ho_{calc}$ , g cm <sup>-3</sup>	1.832	1.859	1.859	1.813	1.581
μ, mm <sup>-1</sup>	3.722	4.164	4.379	6.333	2.233
Ν <sub>τ</sub>	25669	24650	36496	26548	9360
N (R <sub>int</sub> )	5273(0.0675)	4017(0.0737)	5243(0.0661)	5499(0.0434)	6426(0.0153)
$R_1(l > 2\sigma(l))$	0.0325	0.0333	0.0305	0.0318	0.0351
wR <sub>2</sub> (all data)	0.0872	0.0874	0.0844	0.0826	0.0969
GOF	1.070	1.046	1.057	1.083	1.052

	[Er <sub>2</sub> (4hob) <sub>6</sub> (H <sub>2</sub> O) <sub>6</sub> ]·10H <sub>2</sub> O	$[{Y(4hob)_3} \cdot H_2O]_n$	[Tb <sub>2</sub> (4hob) <sub>6</sub> (bpy) <sub>2</sub> ]·2bpy·4H <sub>2</sub> O	[Dy <sub>2</sub> (4hob) <sub>6</sub> (bpy) <sub>2</sub> ]·2bpy·4H <sub>2</sub> O
Formula	$C_{42}H_{62}Er_2O_{34}$	C <sub>21</sub> H <sub>17</sub> O <sub>10</sub> Y	$C_{82}H_{70}N_8O_{22}Tb_2$	C <sub>82</sub> H <sub>70</sub> Dy <sub>2</sub> N <sub>8</sub> O <sub>22</sub>
M <sub>r</sub>	1445.43	518.26	1837.30	1844.46
Space	<i>P</i> -1	<i>P</i> -1	<i>P</i> 2 <sub>1</sub> /n	<i>P</i> -1
a (Å)	10.8172(2)	5.29940(10)	17.3213(6)	17.3138(5)
b (Å)	11.3109(2)	12.1464(2)	13.1760(4)	13.1899(4)
c (Å)	11.44570(10)	17.0480(3)	19.1799(6)	19.1424(6)
α (°)	90.5310(10)	92.421(2)	90	90
β (°)	101.7250(10)	97.081(2)	113.2410(10)	113.1660(10)
γ (°)	105.3210(10)	101.336(2)	90	90
V (Å <sup>3</sup> )	1319.51(4)	1065.25(3)	4022.1(2)	4019.0(2)
Ζ	1	2	2	2
$ ho_{\rm calc}$ , g cm <sup>-3</sup>	1.819	1.616	1.517	1.524
μ, mm <sup>-1</sup>	3.261	4.363	1.823	1.924
Ν <sub>τ</sub>	36333	20531	56091	48794
N (R <sub>int</sub> )	8024(0.0481)	4365(0.0573)	11752(0.0601)	9227(0.0723)
$R_1(l > 2\sigma(l))$	0.0242	0.0386	0.0350	0.0367
$wR_2$ (all data)	0.0511	0.1076	0.0836	0.1014
GOF	1.037	1.093	1.000	1.047

### References

- 1. Sheldrick, G. M. SADABS; University of Gottingen, Gcottingen (Germany), 1996.
- T. M. McPhillips, S. E. McPhillips, H. J. Chiu, A. E. Cohen, A. M. Deacon, P. J. Ellis, E. Garman, A. Gonzalez, N. K. Sauter, R. P. Phizackerley, S. M. Soltis, J. P. Kuhn, J. Synchrotron Radiat. 2002, 9, 401-406.
- 3. W. Kabsch, J. Appl. Crystallogr. 1993, 26, 795.
- 4. CrysAlisPRO v.39. Agilent Technologies Ltd., Yarnton, Oxfordshire, England.
- 5. G. M. Sheldrick, Acta Crystallogr. Sect. A. 2008, 64, 112-122.
- 6. O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard, H. Puschmann, J. Appl. Crystallogr., 2009, 42, 339-341.
- 7. G. M. Sheldrick, Acta Cryst. 2015, C71, 3-8.
- 8. L. L. J. Barbour, J. Supramol. Chem., 2001, 1, 189-191.