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Supporting Information

Preparation of phosphorus-doped copper-nickel electrode and its application in Electro-hydrogenation of 5-HMF

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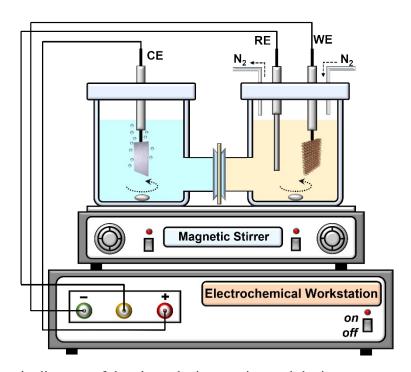


Fig. S1. Schematic diagram of the electrolysis experimental device.

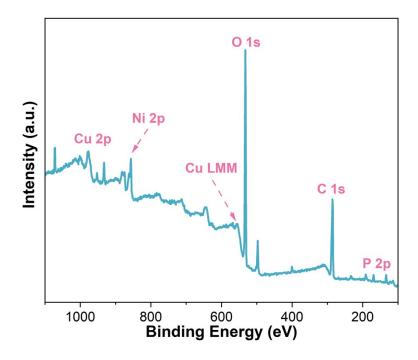


Fig. S2. XPS spectra of Cuhs-Ni-P/CF bimetallic electrodes

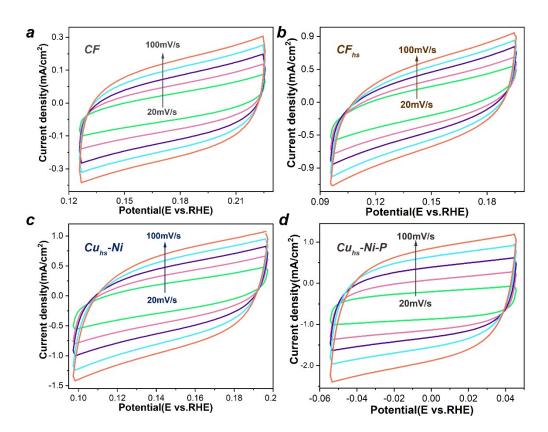


Fig. S3. Cyclic voltammetry curves of (a)CF, (b) high surface area CF, (c) Cu_{hs}-Ni/CF and (d) Cu_{hs}-Ni-P/CF electrodes. Conditions: 20mM FF, pH:8.0.

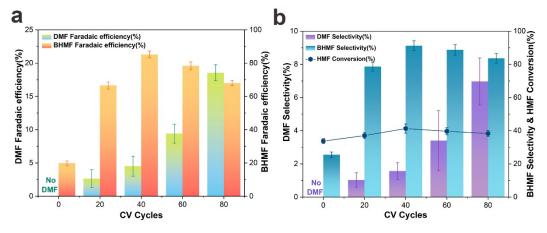


Fig.S4. (a) Faradaic efficiency and (b) selectivity of the electro-catalytic reduction of HMF on the electrode with different Ni deposition time. Conditions: 20mM HMF,pH=8.0,voltage: -1.0 V vs. Ag/AgCl, charge: 50 C

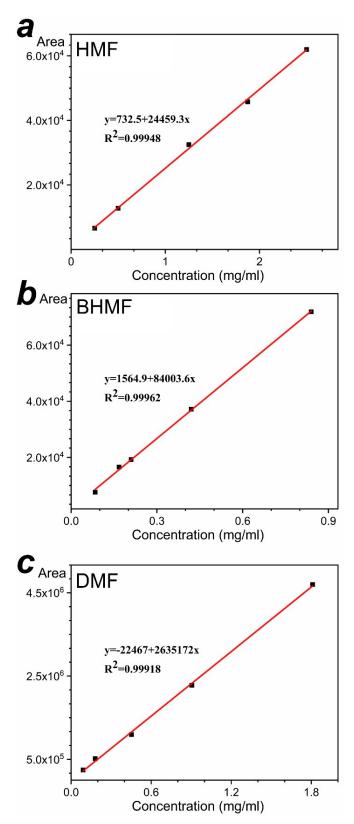


Fig.S5. Standard curve of (a) HMF; (b) BHMF and (c) DMF.



Fig.S6. SEM images of (a) Cu_{hs} /CF electrode; (b) Cu_{hs} -Ni /CF electrodes and (c) Cu_{hs} -Ni-P/CF electrodes

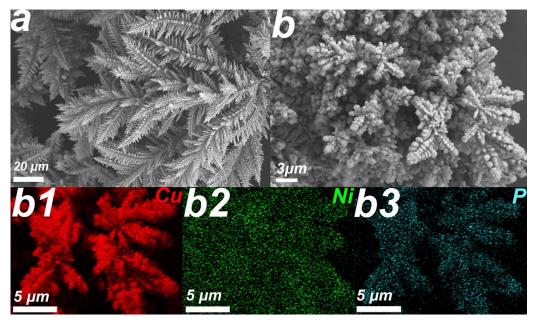


Fig.S7. (a) SEM images of the Cu_{hs} -Ni-P/CF; (b) SEM images of the post-electrolysis Cu_{hs} -Ni-P/CF electrodes; (b1-3) elemental mapping of the post-electrolysis Cu_{hs} -Ni-P/CF electrode.