

Electronic Supplementary Information

Amorphous Co_xS_y loaded $\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$ solid solution for effective visible-light photocatalytic H_2 generation

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Table S1 The elemental content ratio $\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$ and Co_xS_y for obtained from XPS test.

Sample	Element	Atomic ratio (%)
$\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$	Mn	14.82
	Cd	13.09
	S	23.95
Co_xS_y	Co	13.23
	S	17.42

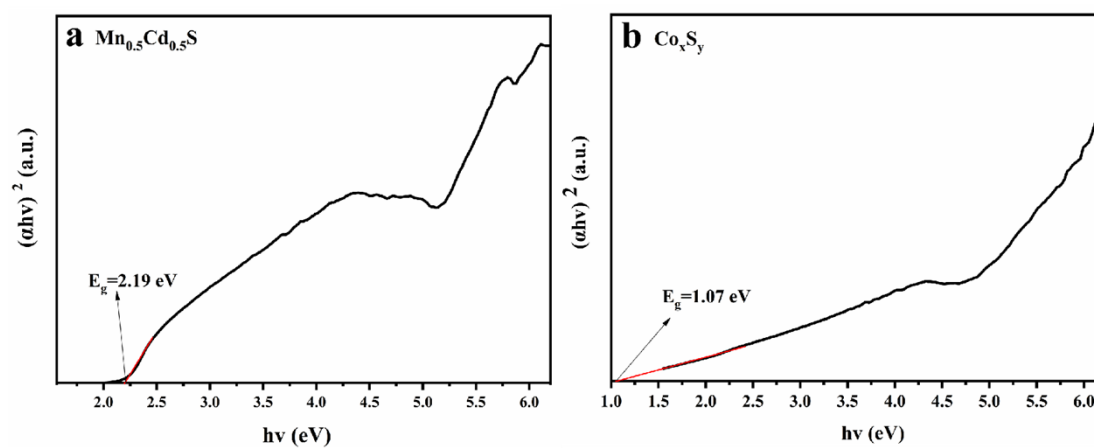


Figure S1 Tauc plot of (a) $\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$ and (b) Co_xS_y .

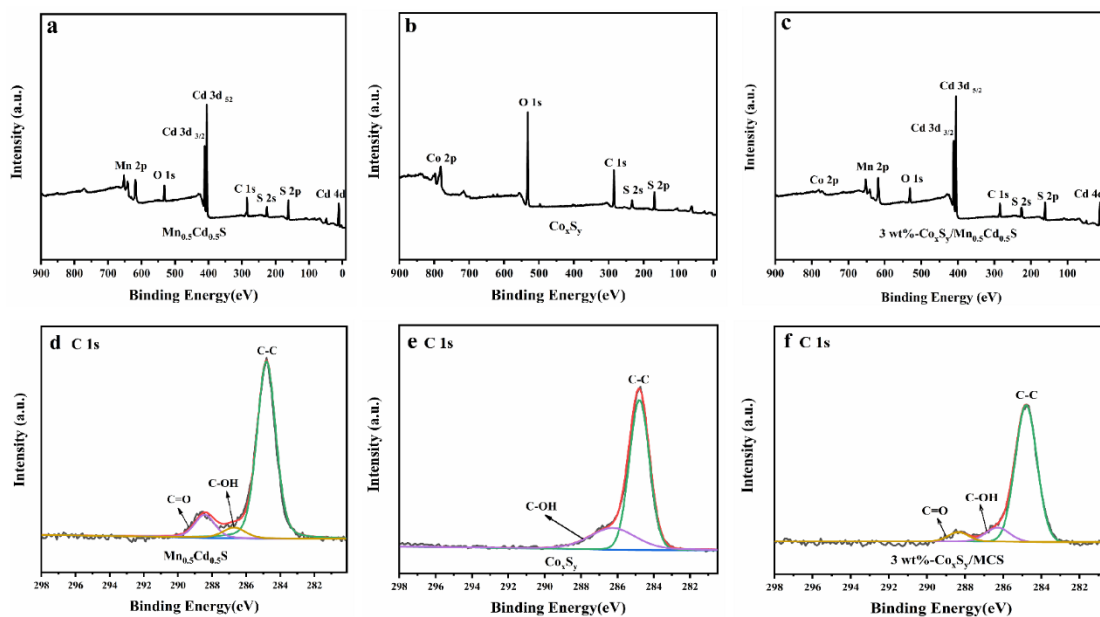


Figure S2 The survey spectrum and the high resolution XPS spectrum for C 1s spectrum of $Mn_{0.5}Cd_{0.5}S$, Co_xS_y , and 3 wt%- $Co_xS_y/Mn_{0.5}Cd_{0.5}S$.

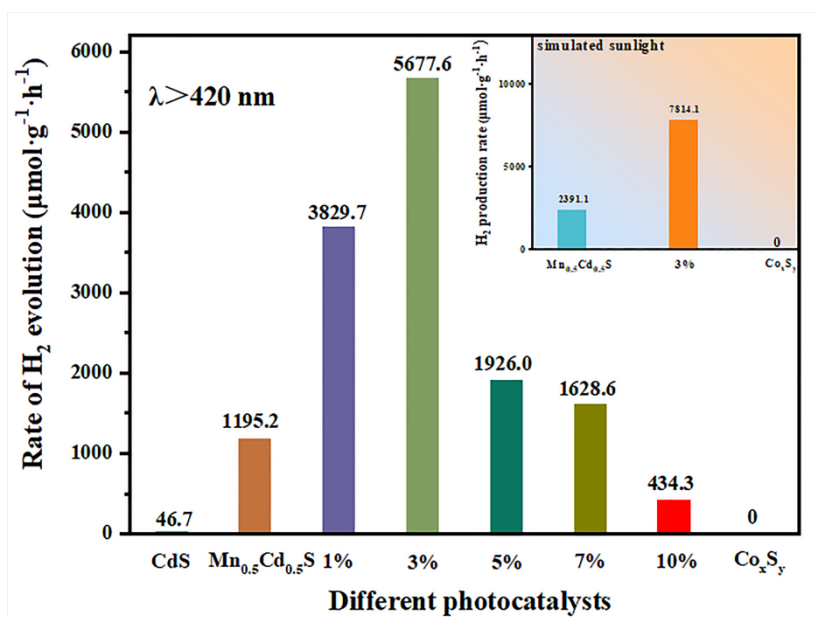


Figure S3 H_2 evolution rate of $Mn_{0.5}Cd_{0.5}S$ and composite catalysts under $\lambda > 420$ nm and simulated sunlight irradiation.

Table S1 Comparison of H₂ evolution performances between the as-obtained 3 wt%-Co_xS_y/Mn_{0.5}Cd_{0.5}S in this study and other photo catalytic systems in literatures.

Catalysts	Dosage	Light condition	Sacrificial agent	H ₂ production rate	Literature source
P/Mn _{0.25} Cd _{0.75} S	30 mg	300 W Xe lamp (λ>400 nm)	No sacrificial agents	863 μmol/h/g	[1]
MoS ₂ /Mn _{0.5} Cd _{0.5} S	50 mg	300 W Xe lamp (λ>420 nm)	0.35 M Na ₂ S and 0.25 M Na ₂ SO ₃	3951.2 μmol/h/g	[2]
Co _x S _y /Mn _{0.5} Cd _{0.5} S	30 mg	300 W Xe lamp (λ>420 nm)	0.5 M Na ₂ S and 0.5 Na ₂ SO ₃	5677.6 μmol/h/g	This work
NiCo ₂ S ₄ /Mn _{0.2} Cd _{0.8} S	10 mg	300 W Xe lamp	Na ₂ S/Na ₂ SO ₃	5677.8 μmol/h/g	[3]
CNTs/Mn _{0.5} Cd _{0.5} S	30 mg	300 W Xe lamp (λ≥420 nm)	TEOA	869 μmol/h/g	[4]
Mn _{0.2} Cd _{0.8} S/CoTiO ₃	10 mg	5 W light-emitting diode (λ≥420 nm)	Na ₂ S/Na ₂ SO ₃	2764 μmol/h/g	[5]
TiO ₂ /Mn _{0.2} Cd _{0.8} S	20 mg	300W Xe lamp	0.35 M Na ₂ S and 0.25M Na ₂ SO ₃	5822.94 μmol/h/g	[6]
Bi ₂ MoO ₆ /Mn _{0.2} Cd _{0.8} S	50 mg	300 W Xe lamp (λ>420 nm)	0.35 M Na ₂ S and 0.25 M Na ₂ SO ₃	1500 μmol/h/g	[7]

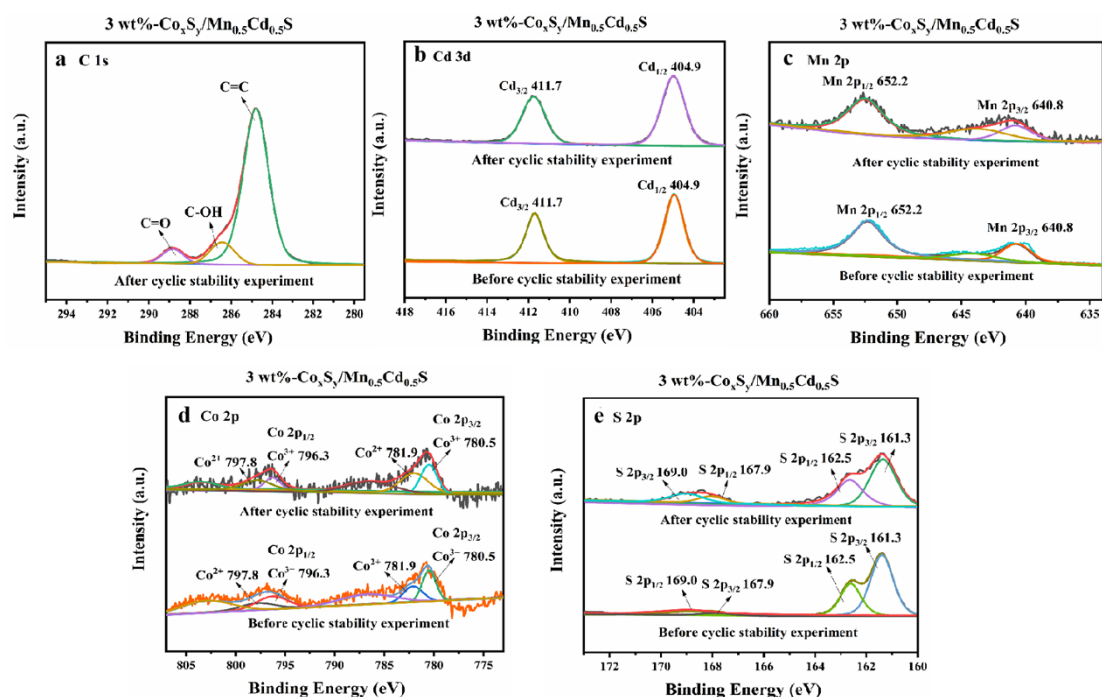


Figure S4 (a)-(e) The C 1s spectrum and the high resolution XPS spectrum for Cd 3d, Mn 2p, Co 2p, and S 2p spectrum of 3 wt%-Co_xS_y/Mn_{0.5}Cd_{0.5}S after cyclic stability experiment.

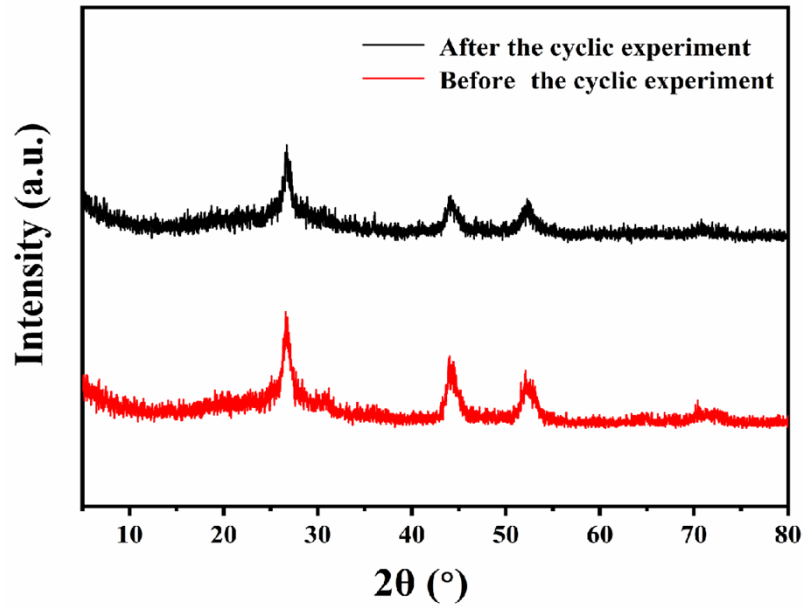


Figure S5 XRD of 3 wt%- $\text{Co}_x\text{S}_y/\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$ before and after the cyclic experiment.

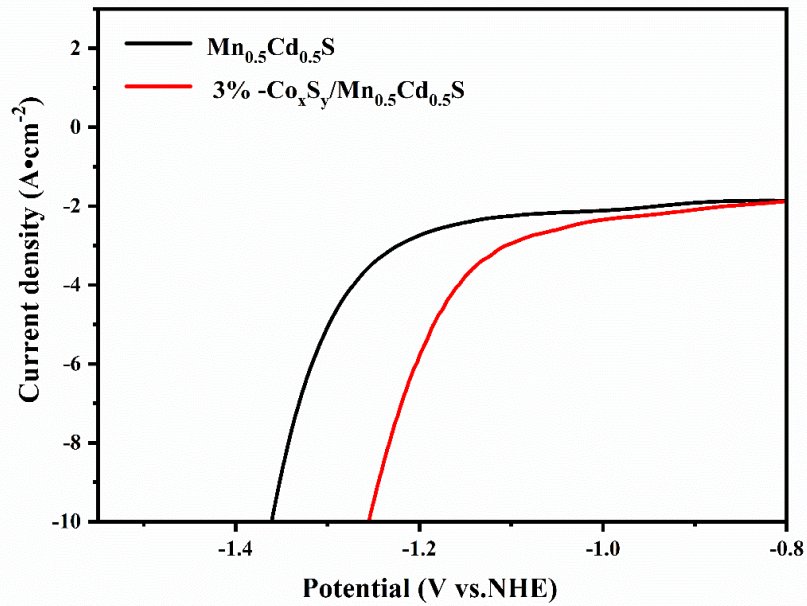


Figure S6 LSV plots for catalysts $\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$ and 3 wt%- $\text{Co}_x\text{S}_y/\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$.

Table S3 Energy band structure information of $\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$ and Co_xS_y .

Samples	$\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$	Co_xS_y
Bandgap energy (eV)	2.19	1.07
Conduction band vs.NHE (eV)	(-0.93)~(-1.13)	(-1.10)~(-1.30)
Valance band vs.NHE (eV)	1.06~1.26	(-0.03)~(-0.23)

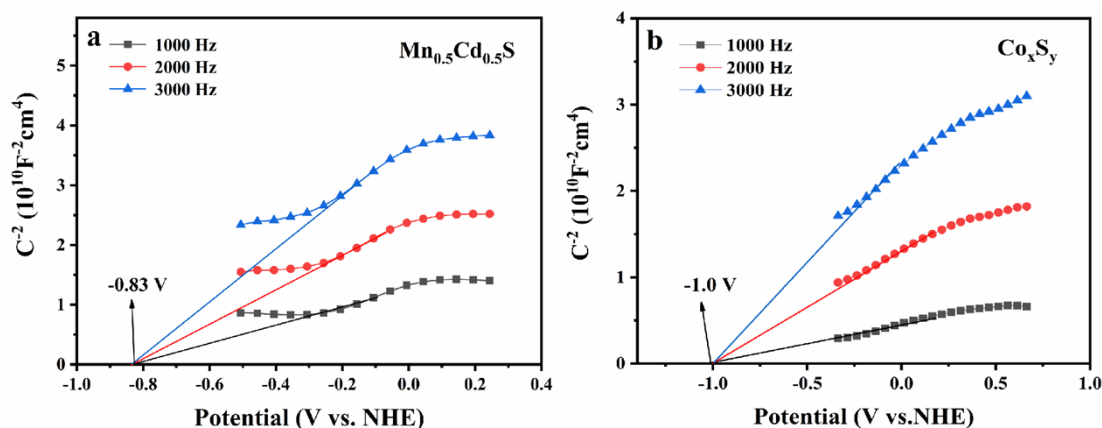


Figure S7 Mott-Schottky plots of (a) $\text{Mn}_{0.5}\text{Cd}_{0.5}\text{S}$ and (b) Co_xS_y .

References

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