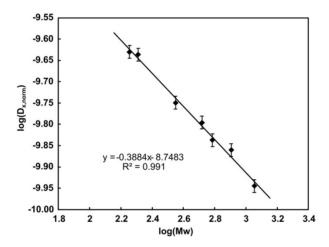
## Kinetics and possible mechanism of chlorogenic acid-water complexes formation

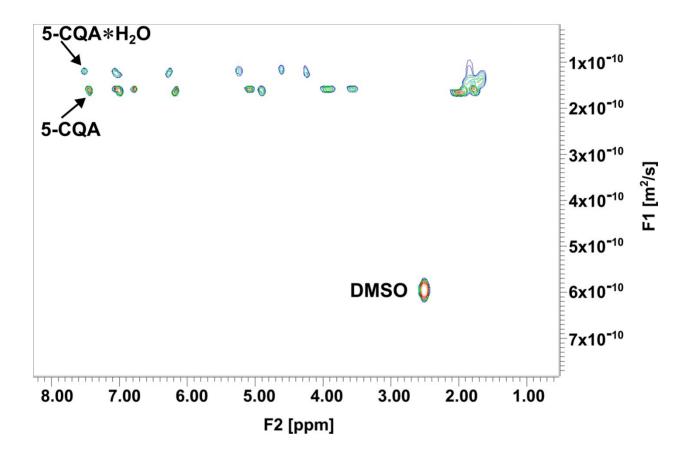
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## 1. External calibration for molecular weight determination in DMSO- $d_6$ using NMR DOSY



**Figure S1.** External calibration curve constructed for molecular weight estimation of polyphenolic compounds in DMSO-d<sub>6</sub>.

2. Determination of 5-CQA\*H<sub>2</sub>O molecular weight using external calibration curve in DMSO-d<sub>6</sub> and NMR DOSY



**Figure S2.** <sup>1</sup>H DOSY spectrum of dry residue obtained from incubated 5-CQA water solution dissolved in DMSO-d<sub>6</sub>.

**TABLE S1**. Results of the molecular weight estimation of 5-CQA\*H<sub>2</sub>O complexes in DMSO-d<sub>6</sub> using NMR DOSY measurements and external calibration method.

Average value of diffusion coefficient of 5-CQA*H <sub>2</sub> O <sup>a</sup> [m <sup>2</sup> /s]	1.20•10 <sup>-10</sup>
Logarithm of normalized diffusion coefficient of 5-CQA*H <sub>2</sub> O	-9.877
Determined molecular weight of 5-CQA*H <sub>2</sub> O [g/mol]	805.42
Theoretical molecular weight of 5-CQA*H <sub>2</sub> O [g/mol]	744.65

<sup>&</sup>lt;sup>a</sup> Average was calculated over all non-overlapping signals of 5-CQA\*H<sub>2</sub>O-1 and 5-CQA\*H<sub>2</sub>O-2 as the differences in diffusion coefficients of both 5-CQA\*H<sub>2</sub>O did not exceed experimental error for the employed method for determination of diffusion coefficients.