

**Photocatalytic reduction of mono, di, and tri-nitrophenols over
 Bi_2MoO_6 /Carbon Nitride Heterojunction**

Phyu Phyu Cho,^a Phyu Phyu Mon,^a Devthade Vidyasagar,^a Giridhar Madras,^b Ch. Subrahmanyam^{a,*}

^a Department of Chemistry, Indian Institute of Technology Hyderabad, Kandi, Sangareddy, 502285, Telangana, India.

^b Department of Chemical Engineering, Indian Institute of Technology Hyderabad, Kandi, Sangareddy, 502285, Telangana, India.

* Corresponding author

Email: csubbu@iith.ac.in

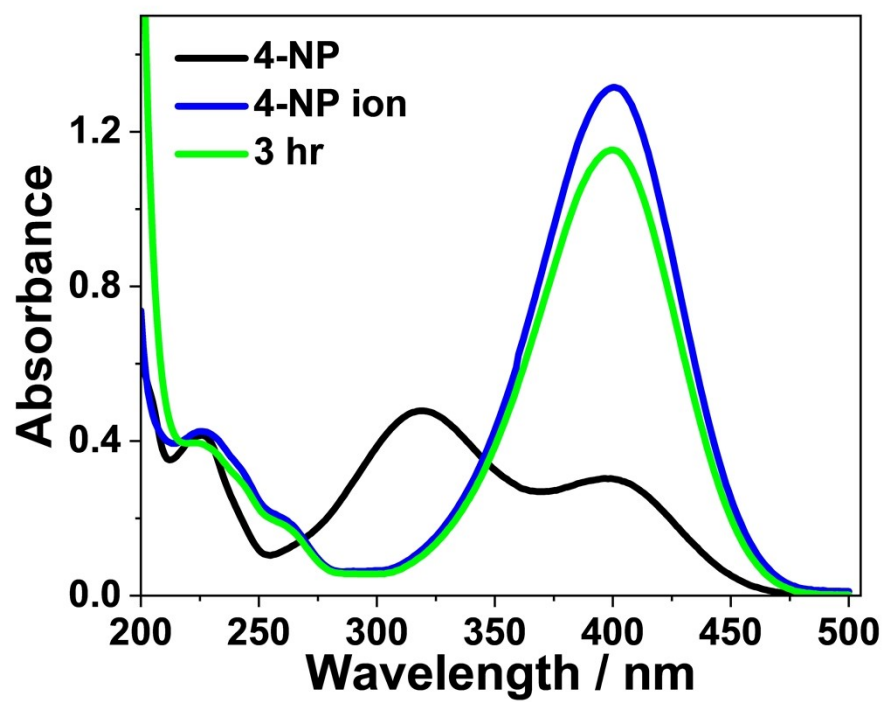


Figure S1. UV-visible absorption spectra of 4-NP in the presence and absence of NaBH₄.

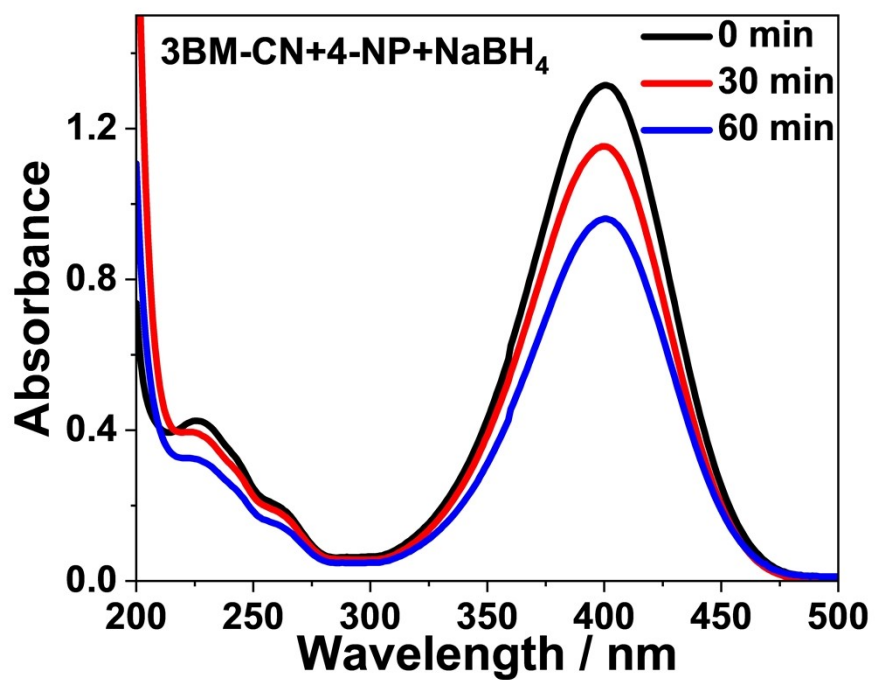


Figure S2. UV–visible absorption spectra of 4-NP with NaBH₄ in using 3BM-PCN catalyst.

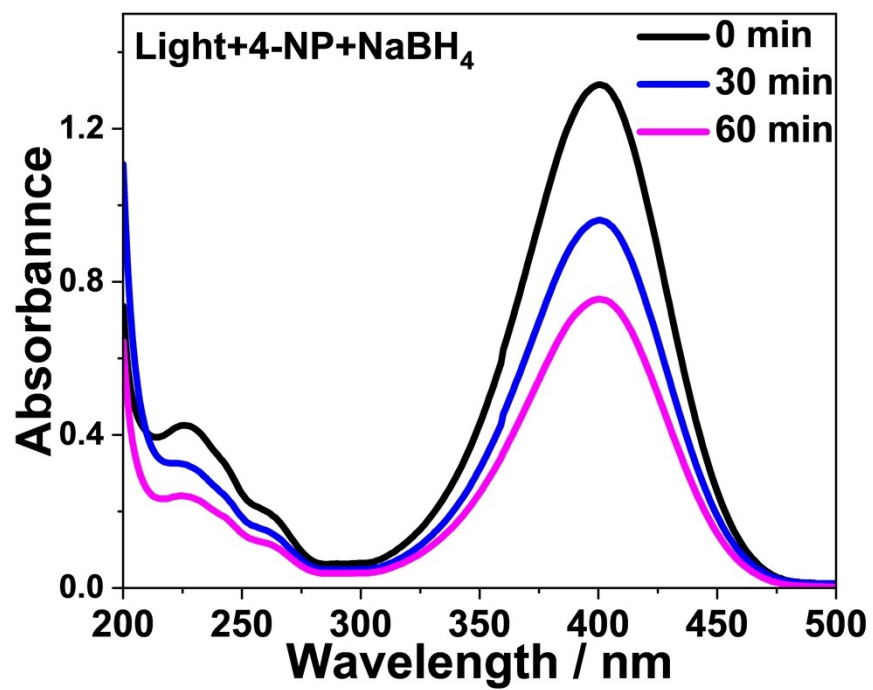


Figure S3. UV-visible absorption spectra of 4-NP with NaBH₄ under irradiation light.

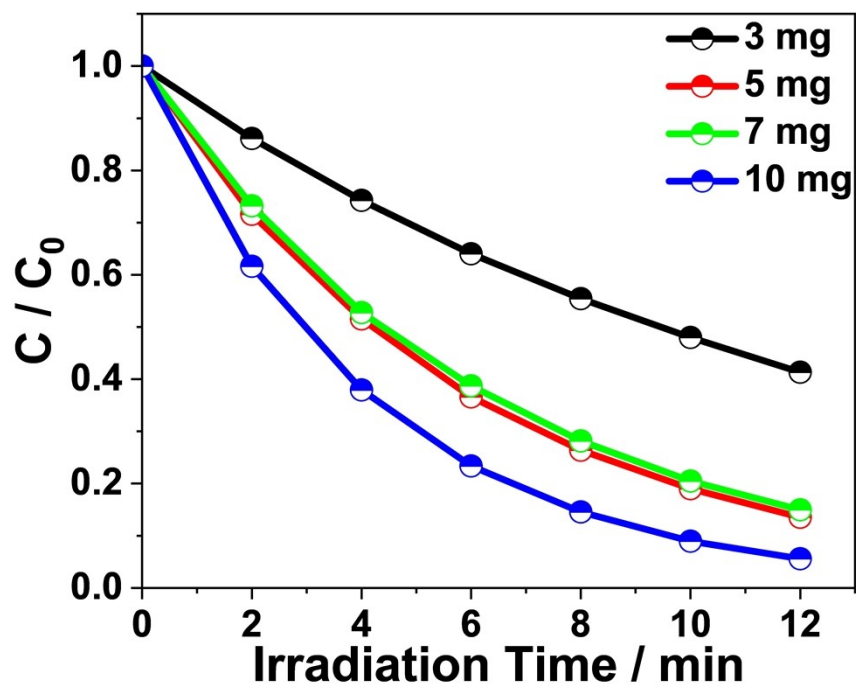


Figure S4. Plot of C/C_0 Vs reaction time of 4-NP in the various catalyst concentration. Reaction condition: catalyst (3BM-PCN); 50 mL of 4-NP (2 mM); 1 mL of NaBH_4 (20 mM).

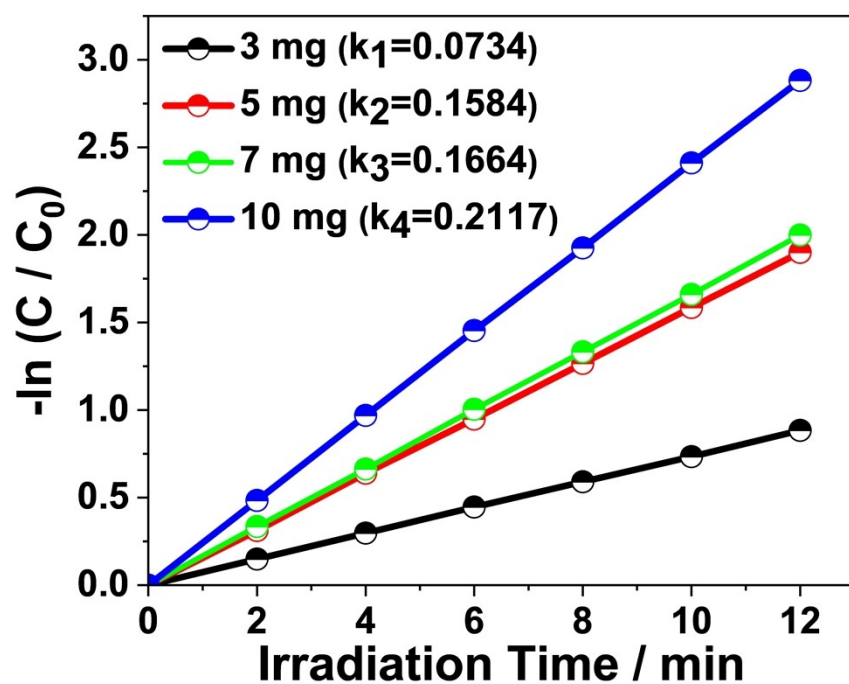


Figure S5. Pseudo first order kinetic plot for the photoreduction of 4-NP in the various catalyst concentration. Reaction condition: catalyst (3BM-PCN); 50 mL of 4-NP (2 mM); 1 mL of NaBH_4 (20 mM).

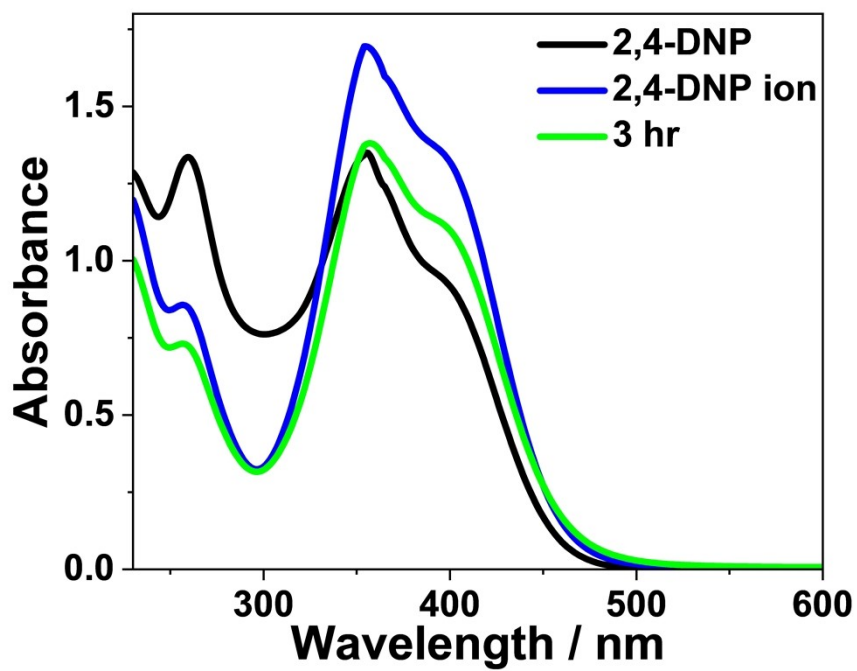


Figure S6. UV–visible absorption spectra in the presence and absence of NaBH_4 of the 2,4-DNP. Reaction condition: 10 mg of catalyst; 50 mL of 2,4-DNP (2 mM); 1 mL of NaBH_4 (20 mM).

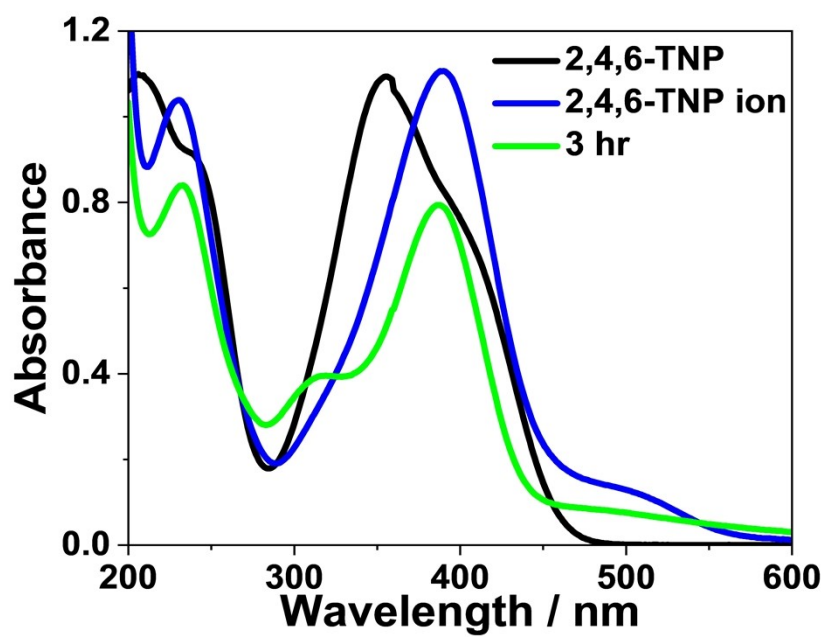


Figure S7. UV–visible absorption spectra in the presence and absence of NaBH_4 of the 2,4,6-TNP. Reaction condition: 10 mg of catalyst; 50 mL of 2,4,6-TNP (2 mM); 1 mL of NaBH_4 (20 mM).

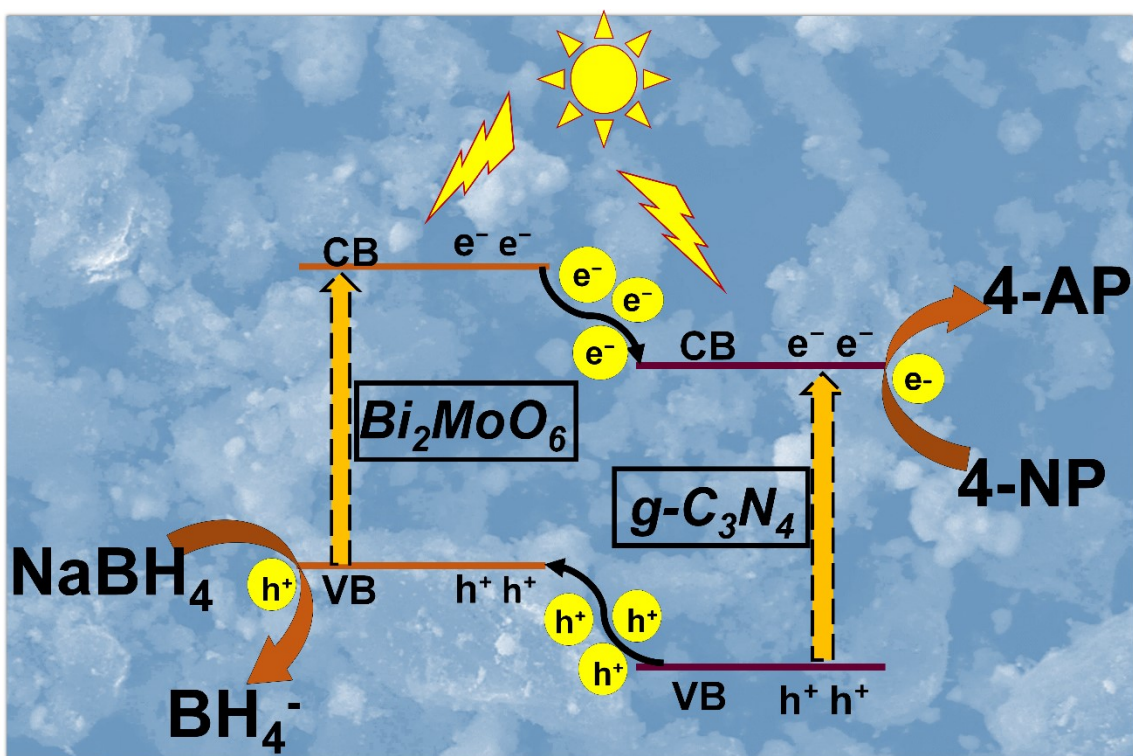


Figure S8. Schematic band diagram showing the photocatalytic charge transfer mechanism of the $\text{Bi}_2\text{MoO}_6/\text{g-C}_3\text{N}_4$ nanocomposite.