

Supporting information for

Rapid Synthesis of a BODIPY Derivative Serving as a Highly Selective and Sensitive Fluorescence Chemosensor for Hg²⁺ Ion Detection

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NMR spectra

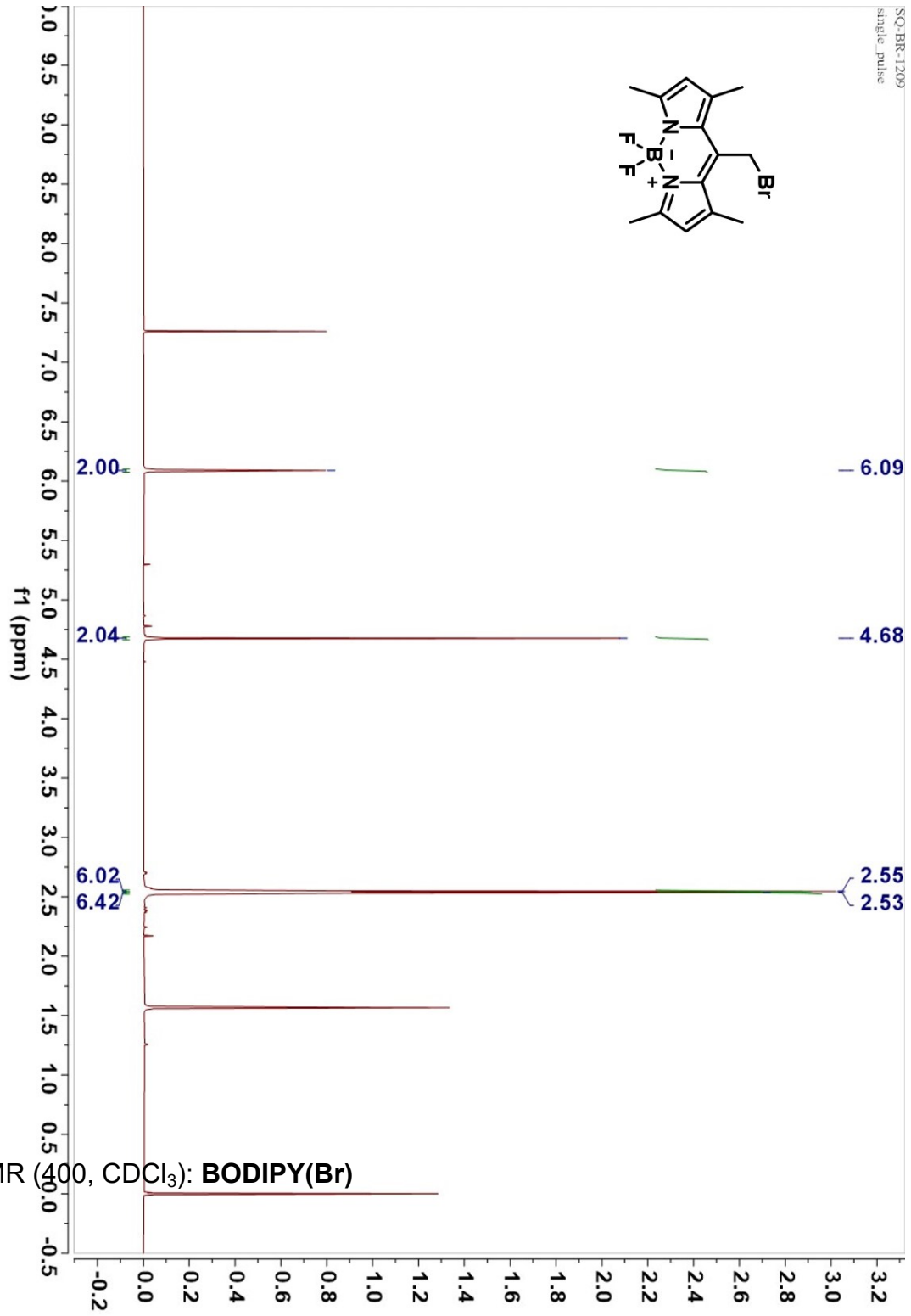


Figure S1. ¹H NMR (400, CDCl₃): BODIPY(Br)

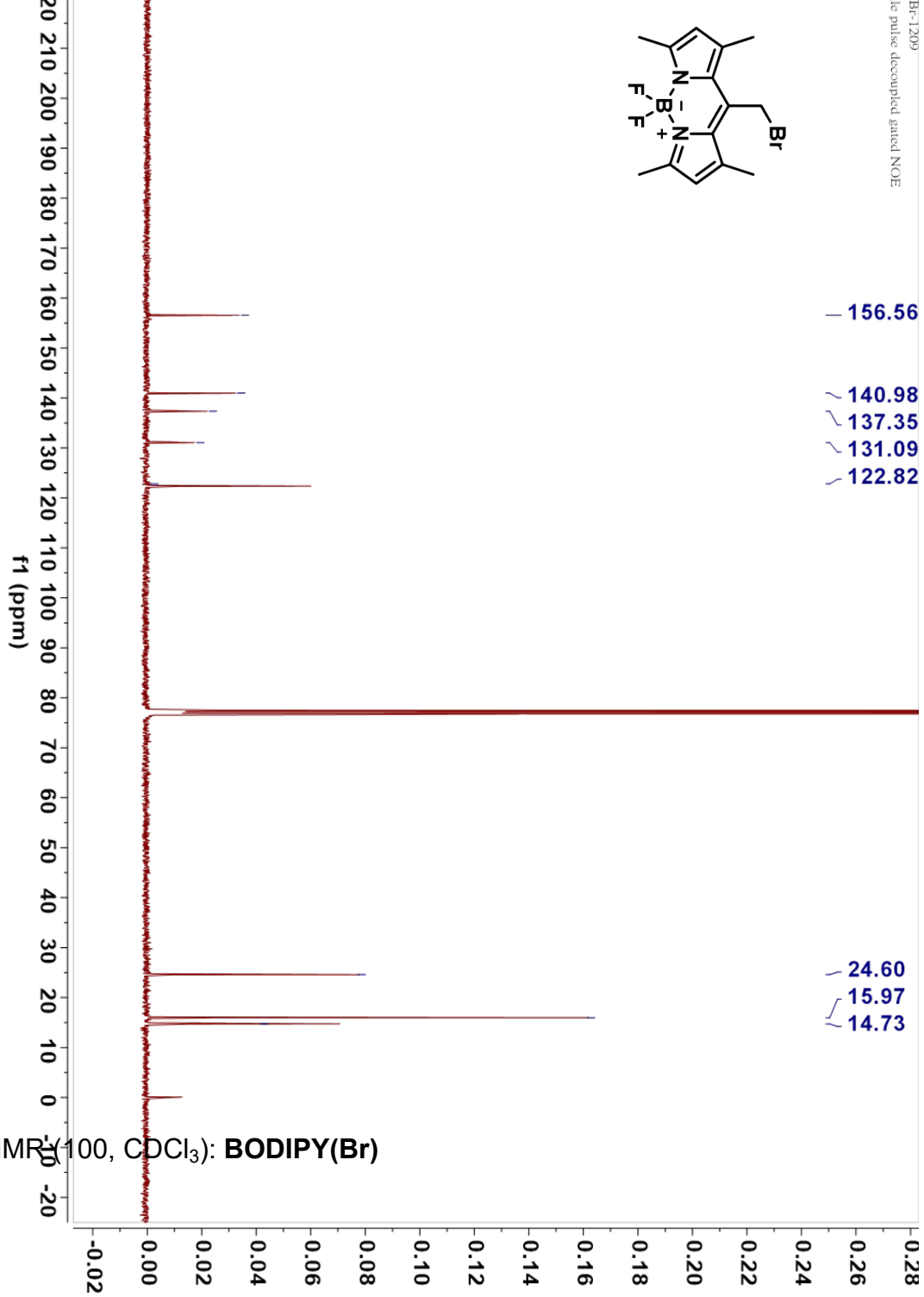
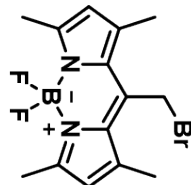


Figure S2. ¹³C NMR (100, CDCl₃): BODIPY(Br)

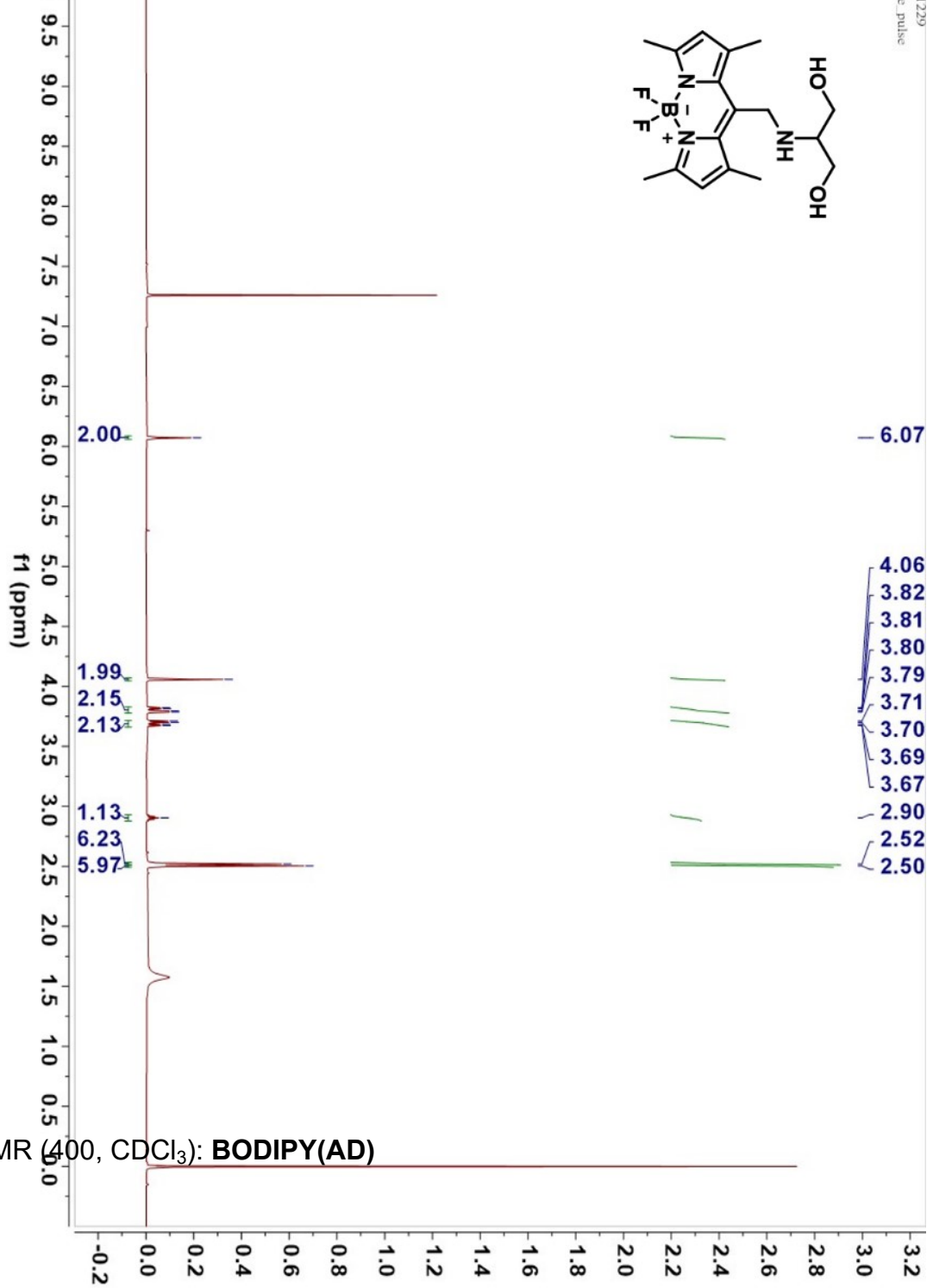


Figure S3. ¹H NMR (400, CDCl₃): BODIPY(AD)

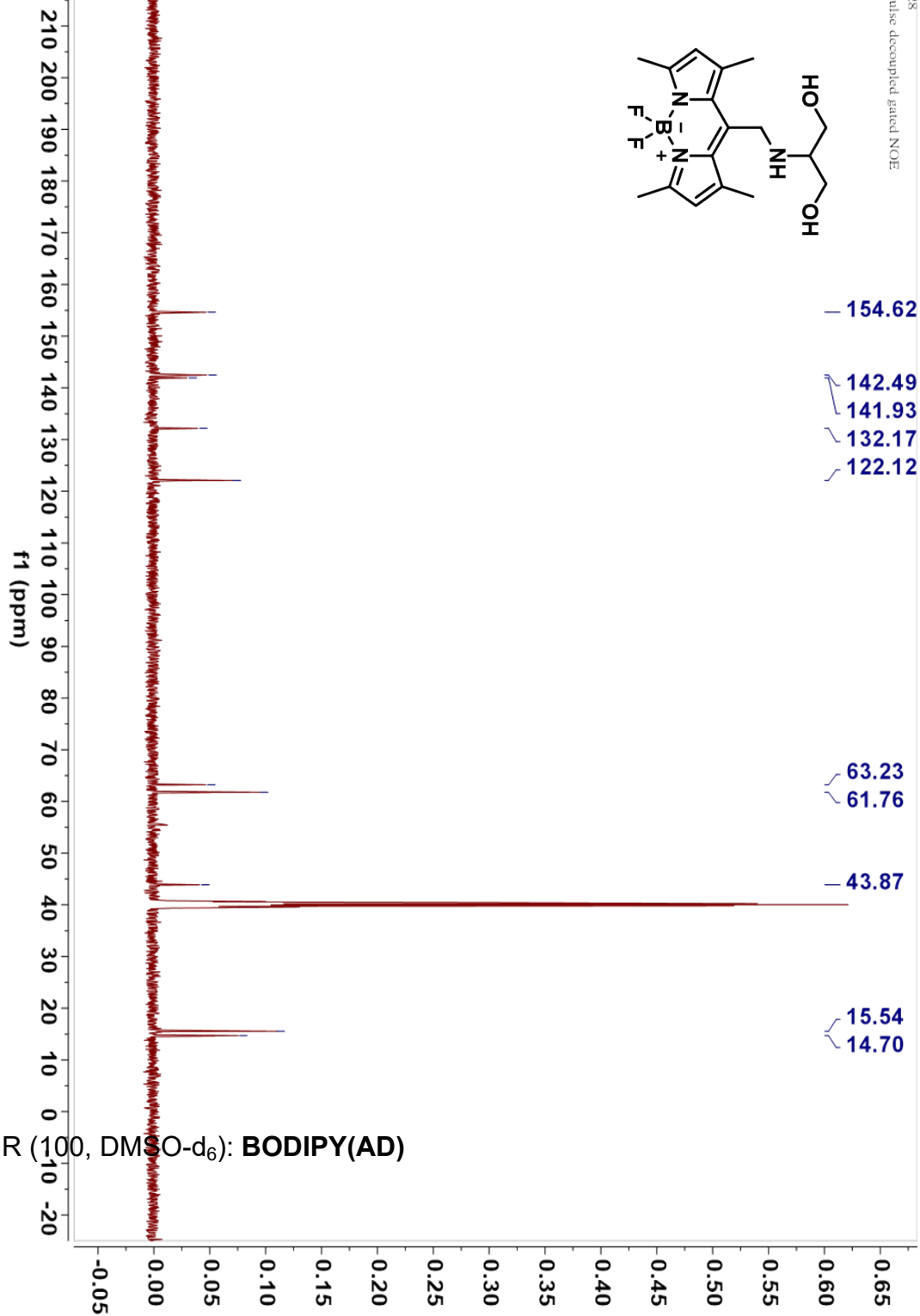
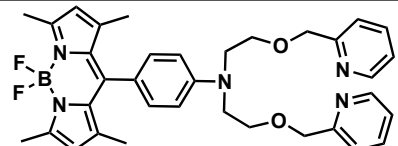
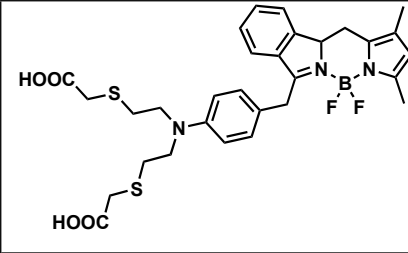
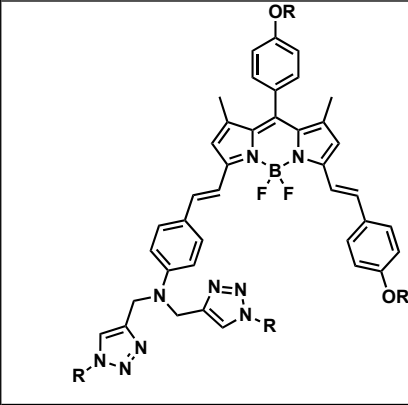
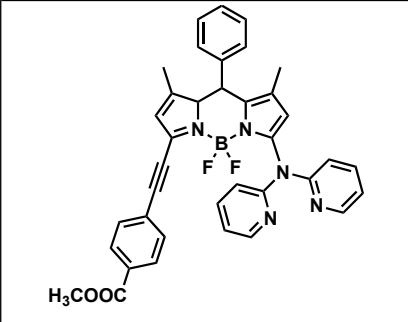
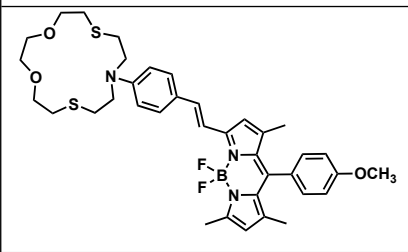
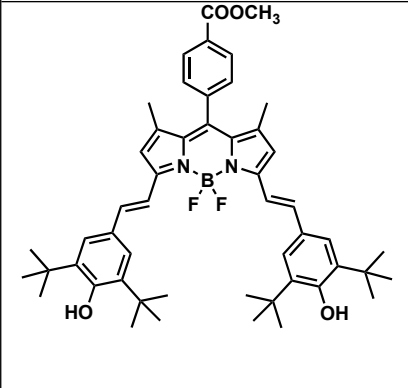


Figure S4. ^{13}C NMR (100, DMSO-d_6): **BODIPY(AD)**

Table S1. Recent BODIPY-based fluorophores for detecting Hg(II) ions.

Structures	Detection Limit	Number of Synthetic Steps	Reaction time and overall yield	References
	0.18 μM	4 steps	48 h, 17.8%	1

	5.7 nM	7 steps	76 h, 9.2%	2
	0.09 μM	4 steps	53 h, 1.5%	3
	0.07 μM	5 steps	70 h, 7.6%	4
	99 ppm	3 steps	6.5 h, 7.9%	5
	0.7 μM	2 steps	19 h, 16.8%	6

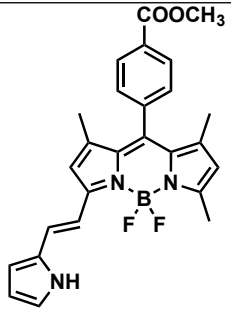
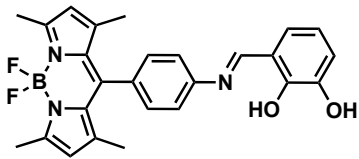
	0.05 μ M	2 steps	22 h, 3.4%	7
	0.9 μ M	3 steps	36 h, 24.0%	8

Table S2. Crystal data and structure refinement for BODIPY(AD)

Identification code	K10911-CHL-A(SQ-dioxide)
Empirical formula	$C_{34}H_{48}B_2F_4N_6O_4$
Formula weight	704.42
Temperature/K	113(2)
Crystal system	monoclinic
Space group	$P2_1/n$
$a/\text{\AA}$	9.00305(16)

b/Å	17.8739(4)
c/Å	21.0833(3)
$\alpha/^\circ$	90
$\beta/^\circ$	91.3815(15)
$\gamma/^\circ$	90
Volume/Å ³	3391.73(10)
Z	4
$\rho_{\text{calc}}/\text{mg}/\text{mm}^3$	1.379
μ/mm^{-1}	0.105
F(000)	1496.0
Crystal size/mm ³	0.15 × 0.15 × 0.13
2 θ range for data collection	4.558 to 50°
Index ranges	-10 ≤ h ≤ 10, -19 ≤ k ≤ 21, -25 ≤ l ≤ 25
Reflections collected	32271
Independent reflections	5960[R(int) = 0.0325]
Data/restraints/parameters	5960/0/475
Goodness-of-fit on F ²	1.045
Final R indexes [$I \geq 2\sigma(I)$]	R ₁ = 0.0435, wR ₂ = 0.1160
Final R indexes [all data]	R ₁ = 0.0537, wR ₂ = 0.1212
Largest diff. peak/hole / e Å ⁻³	0.73/-0.67
CCDC number	2272046

Table S3. Absorption and fluorescence data of BODIPY(AD).

Solvents	λ_{ab} (nm)	ϵ (L/mol·cm)	λ_{em} (nm)	Φ^a	Δ_{st} (cm ⁻¹) 1)	ϵ^b
Toluene	512	66000	522	84.5%	374	2.38
DCM	511	111800	520	51.2%	338	8.93
Acetone	506	86400	513	1.96%	307	20.7
MeOH	506	101600	515	6.85%	345	32.7
DMSO	508	78600	522	6.48%	528	46.68

DMSO (Hg ²⁺)	520	54900	536	118%	574	46.68
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a: The Φ of DCM was absolute quantum yield, and the Φ of the others were relative quantum yields by using the

DCM sample as standard. *b*: Dielectric constant of each solvent

Computational data

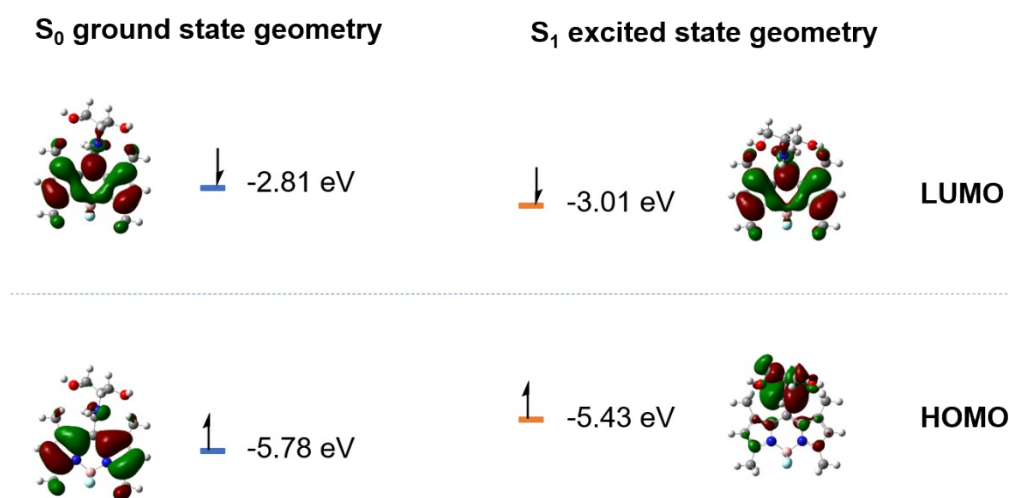


Figure S5. TD-DFT calculations for BODIPY(AD) in DMSO. The dominant configuration of S_1 excited states at S_0 ground state geometry (left) and at S_1 excited state geometry (right) showed that ICT process occurred when the molecule was excited. The poor overlap of wavefunctions between the excited state and the ground state at S_1 geometry caused the diminishment of fluorescence.

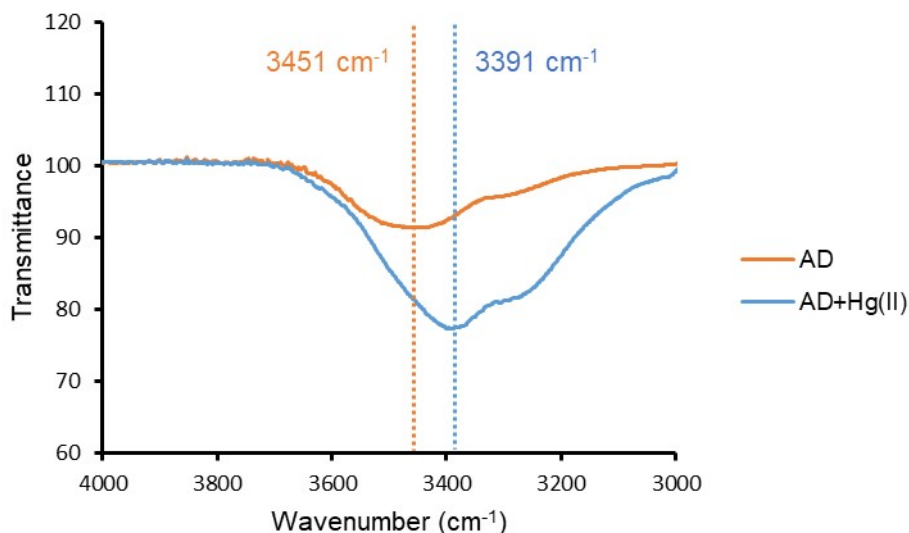


Figure S6. FI-IR spectra demonstrated that N-H and O-H peak shifted after adding Hg^{2+} ion to BODIPY(AD).

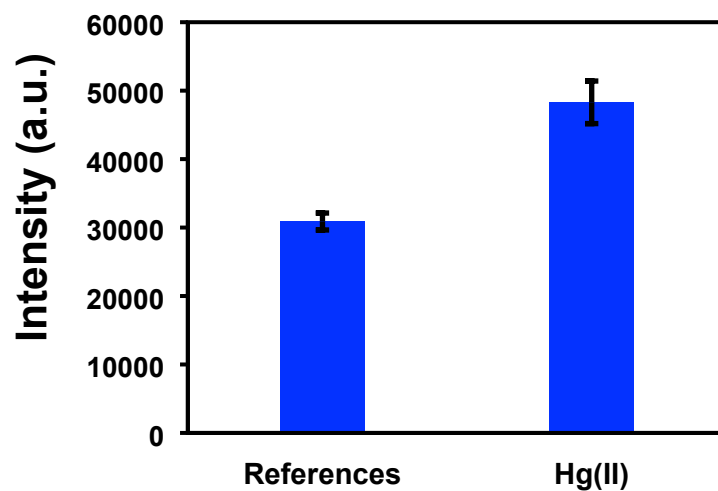


Figure S7. Fluorescence of BODIPY(AD) in the buffer solution with and without Hg^{2+} ions.

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