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GALLIC ACID-BASED DENDRIMERS WITH THIACALIX[4]ARENE CORE: SYNTHESIS, AGGREGATION AND USE FOR Pd NP's STABILIZATION

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Figure S1. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound (4)*.





Figure S2. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound* (5).





Figure S3. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound* (6).





Figure S4. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound (12)*



(b)





Figure S5. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound (13)*









Figure S6. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound (14)*









Figure S7. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound (15)*



(b)





Figure S8. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound (16)*









Figure S9. NMR ¹H (a), ¹³C (b), and HRESI MS (c), FT IR (d) spectra of *compound (17)*



Figure S10. NMR ¹H of *compound* 10*a* (a) and of *mixture* 10*a*, *b* (b) (DMSO_{d6}, 400 MHz, 25 °C)



Figure S11 AFM evaluation of the dendrimers: A) **15**, B) **16**, C) **17**, where (a) AFM images and (b) cross-section view of dendrimers showing a diameters and heights; C(macrocycles) = 0.1 mM in 5% THF – water



Figure S12 UV-Vis spectra of $PdCl_4^{2-}$, dendrimer and double $PdCl_4^{2-}$ -dendrimer systems for A) **15**, B) **16** and C) **17**, where (a) $PdCl_4^{2-}$ (0.2 mM), (b) dendrimer (0.1 mM) and dendrimer (0.1 mM) in the presence of different amounts of palladium after reduction during 1 hour (0.1 mM) (c) 0.05mM, (d) 0.1 mM, (e) 0.15 mM, (f) 0.2 mM; water with 5% THF, $1 = 1 \text{ cm}^{-1}$.



Figure S13 Plots of $\ln(C_t/C_0)$ vs time in the presence of (a) 0.5Pd&15, (b) 1.5Pd&15, (c) 2Pd&15; (d) 0.5Pd&17; (e) 1Pd&17; (f) 1.5Pd&17 (g) 0.5Pd&16; (h) 1Pd&16 (i) 1.5Pd&16 (j) Pd⁰; (C(*p*-nitrophenol) = 0.1 mM, C(NaBH₄) = 5 mM, n(Pd) in metal-dendrimer = 5 nanomole, 5 % THF - water, 20 °C, l = 1 cm.