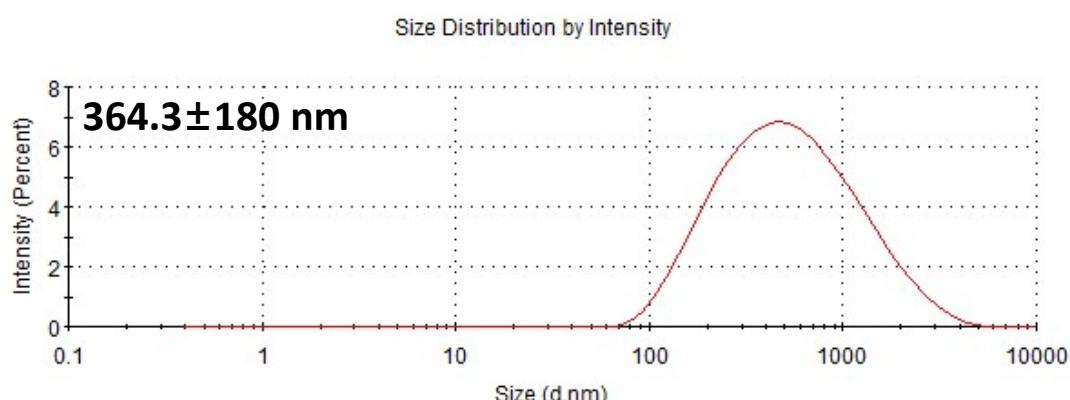


(a)



(b)

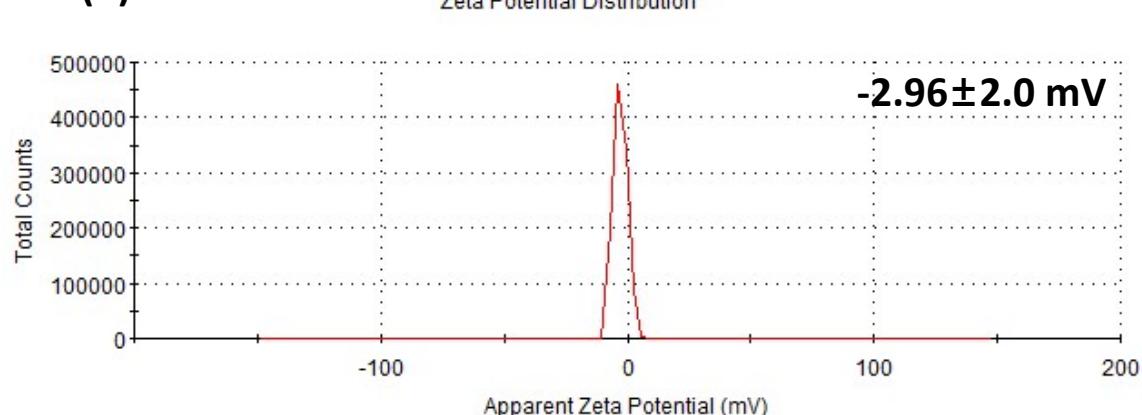


Figure S1 SQDs size distribution profile and zeta potential

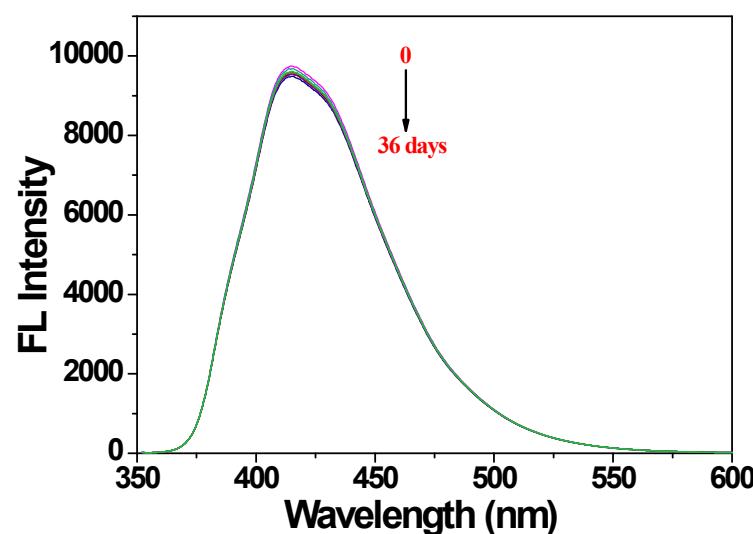


Figure S2. Emission changes of SQDs with an interval of 72 h for 36 days, after 9 months of preservation at 4 °C.

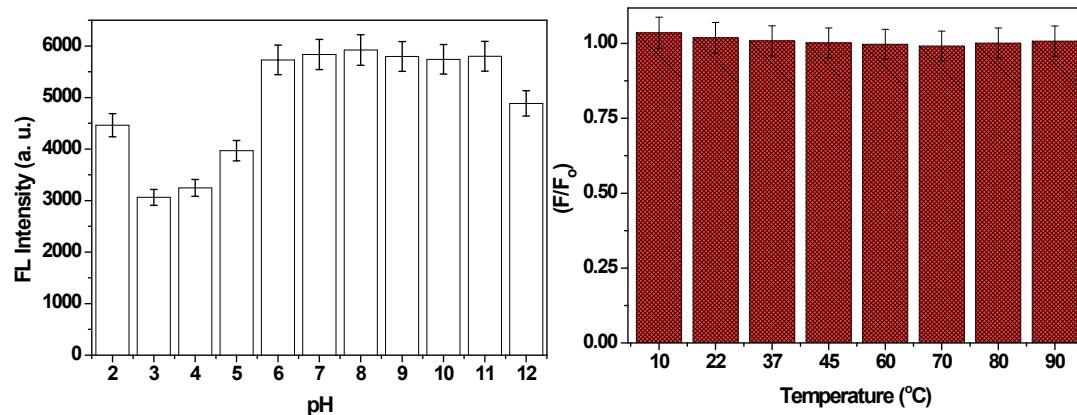


Figure S3. (a) Temperature and (b) pH influence on SQDs emission property.

Table S1. Comparison of SQDs LOD against 4-NP with the reported literature.

Material	Linear Rang(μM)	LOD(μM)	Ref.
B, N-CDs	0.5-60,60-200	0.2	1
CDs-BC	0.3-6.5,6.5-30	0.11	2
Hollow nanospheres	0-34.9	0.26	3
Au NCs	0.1-100	0.09	4
Au NCs	1-50	0.21	5
β -CD-CdTe QDs	20-100	0.3	6
N-CDs	0.25-7.5	0.05	7
N-CDs	0-250	0.4	8
Cr-CDs	0.8-150	0.27	9
B,N CDs	-	0.17	10
SQDs	0.2-30	0.07	11
SQDs	0.12-50	0.046	This work

Table S2. Comparison of SQDs LOD against 4-NP with the reported literature.

Material	Linear range(μM)	LOD(μM)	Ref.
Cu,I pellet	0-800	2.30	12
Curcumin	0.15-9.9	0.079	13
Si NPs	0.1-500	0.029	14
N-CDs	0.001-1	0.077	15
Poly(propyleneimine-Au NPs)	0.61-625	0.45	16
ZnCO ₂ O ₄ -GCE	1-4000	0.3	17
Polymethy red film	0.1-40	0.15	18
BaO nanorods	1.5-9	0.5	19
B-cyclodextrin	5-400	0.3	20
SQDs	0.12-50	0.171	This work

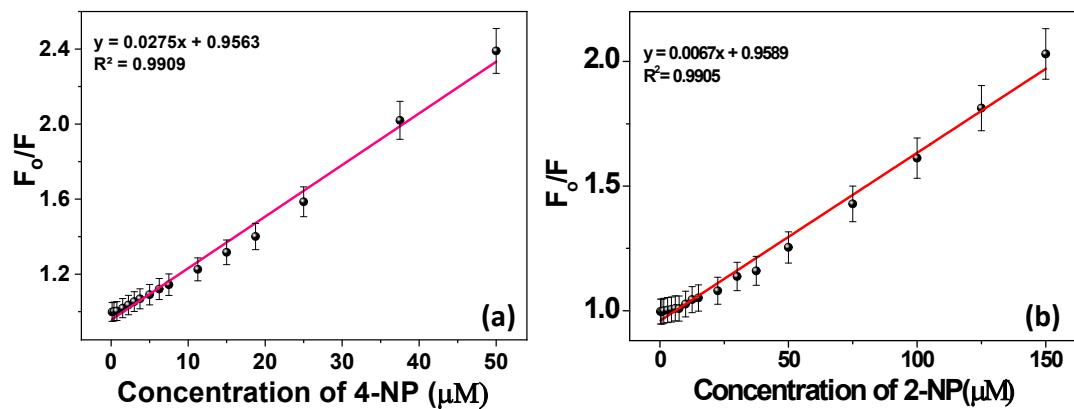


Figure S4. SQDs Stern-Volmer plots in the presence of nitrophenols: (a) 4-NP (b) 2-NP.

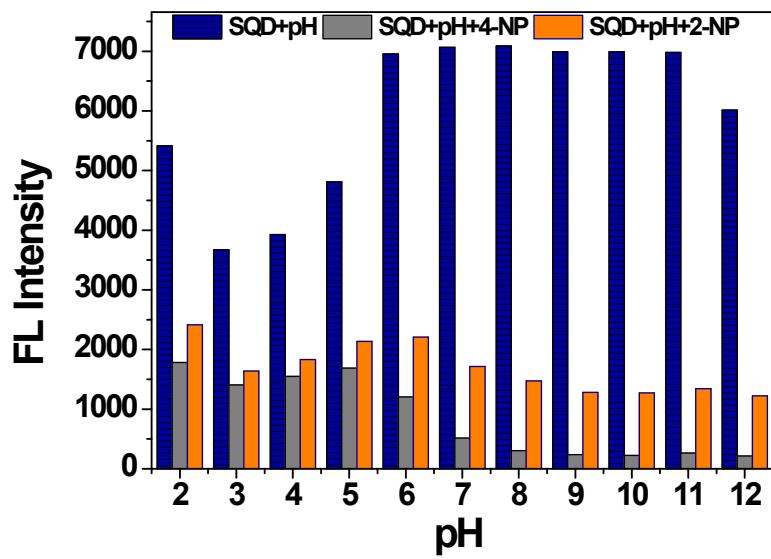


Figure S5. Influence of pH on emission and quenching property of SQDs in the presence of nitrophenols.

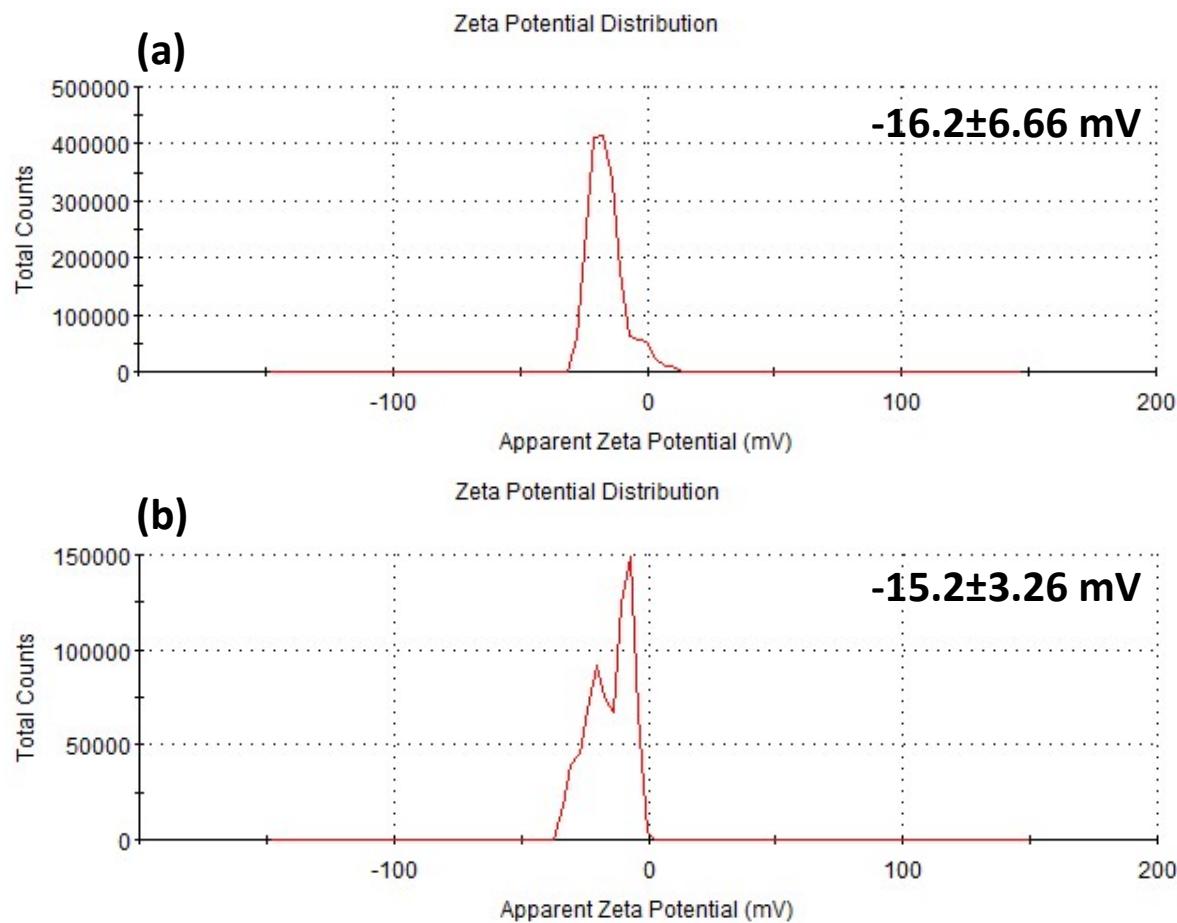


Figure S6. Zeta potential of SQDs in the presence of 4-NP(a) and 2-NP (b)

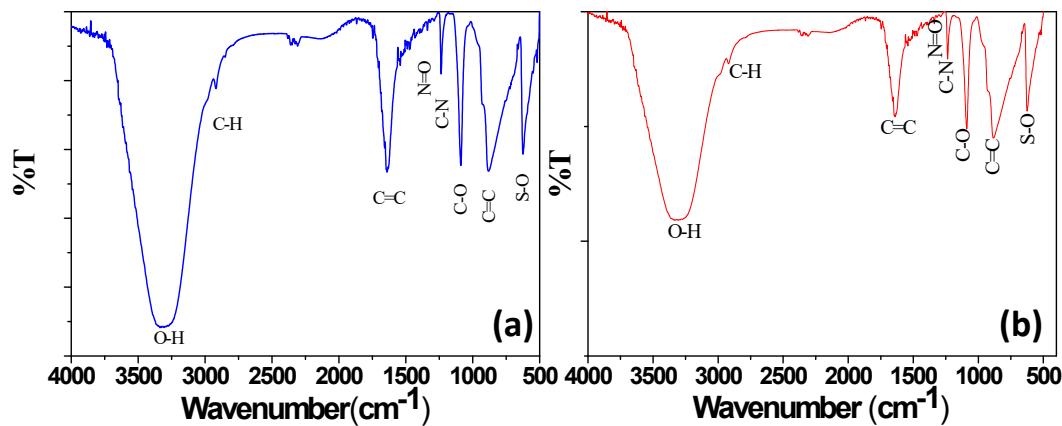


Figure S7. FT-IR spectra of SQDs in the presence of (a) 4-NP and (b) 2-NP

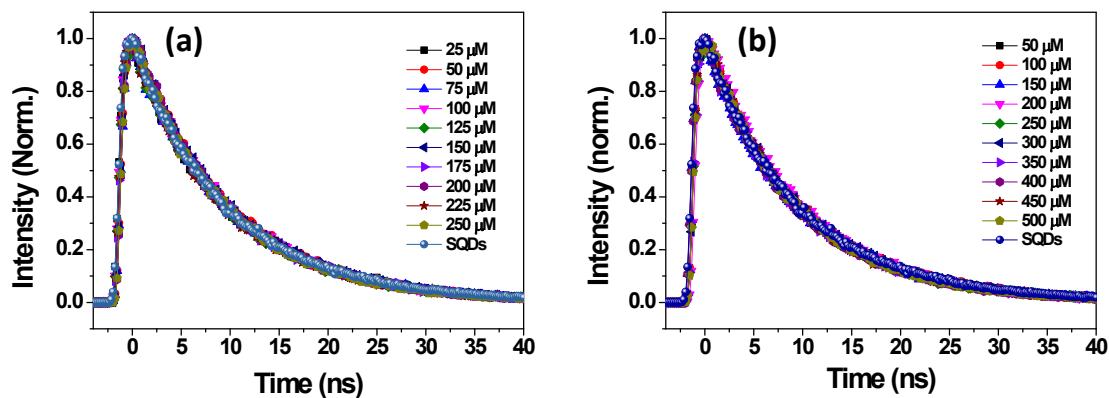


Figure S8. TRPL titration of SQDs in the presence of (a) 4-NP (25-250 μM) and (b) 2-NP (50-500 μM).

Table S3. Faster (τ_1 and τ_2), slower (τ_3) and average (τ_{avg}) decay values of SQDs in 4-NP solution

4-NP	τ_1 (ns)	τ_2 (ns)	τ_3 (ns)	τ_{avg} (ns)
SQDs	0.78	2.7	10.3	9.64
SQDs+25 μM	0.85	3.6	10.4	10.78
SQDs+50 μM	0.82	3.2	10.2	10.73

SQDs+75 μM	0.14	4.5	10.3	9.80
SQDs+100 μM	0.84	3.0	10.0	9.75
SQDs+125 μM	0.89	3.8	10.1	10.30
SQDs+150 μM	0.92	2.4	9.8	9.90
SQDs+175 μM	0.25	3.7	10.0	9.59
SQDs+200 μM	0.59	2.9	9.88	9.57
SQDs+225 μM	0.12	2.1	97.9	9.82
SQDs+250 μM	0.14	1.5	9.70	9.78

Table S4. Faster (τ_1 and τ_2), slower (τ_3) and average (τ_{avg}) decay values of SQDs in 2-NP solution

2-NP	$\tau_1(\text{ns})$	$\tau_2(\text{ns})$	$\tau_3(\text{ns})$	$\tau_{\text{avg}} (\text{ns})$
SQDs	0.78	2.7	10.3	9.64
SQDs+50 μM	0.96	4.6	10.3	9.84
SQDs+100 μM	0.36	3.3	10.1	9.75
SQDs+150 μM	0.143	3.5	10.2	9.72
SQDs+200 μM	0.84	3.2	10.0	9.75
SQDs+250 μM	0.93	4.0	10.1	9.67
SQDs+300 μM	0.44	3.8	10.0	9.64
SQDs+350 μM	0.24	2.9	98.7	9.57
SQDs+400 μM	0.36	3.6	98.0	9.38
SQDs+450 μM	0.15	4.6	98.3	9.33
SQDs+500 μM	0.11	3.8	98.0	9.37

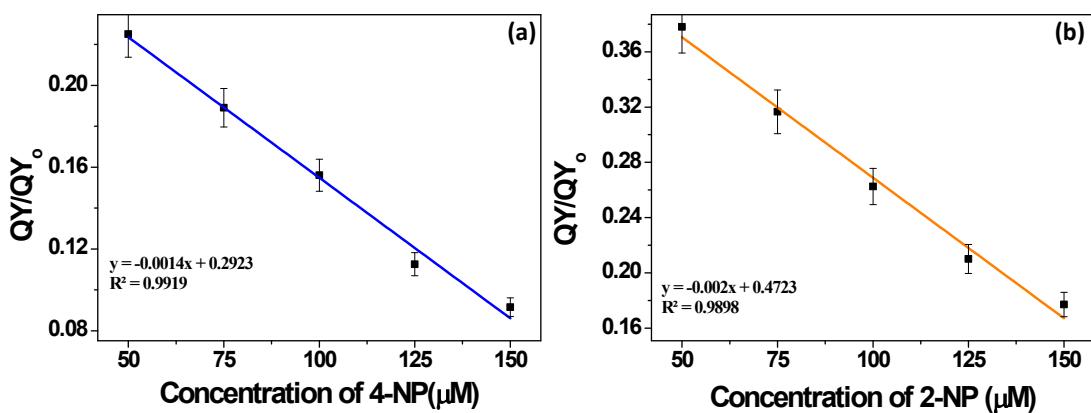


Figure S9. Influence of nitrophenols on SQDs PLQY (a) SQDs+4-NP (b) SQDs+2-NP.

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