

# Unravelling the sensing efficacy of graphene oxide towards hazardous volatile organic compounds in polyurethane industry

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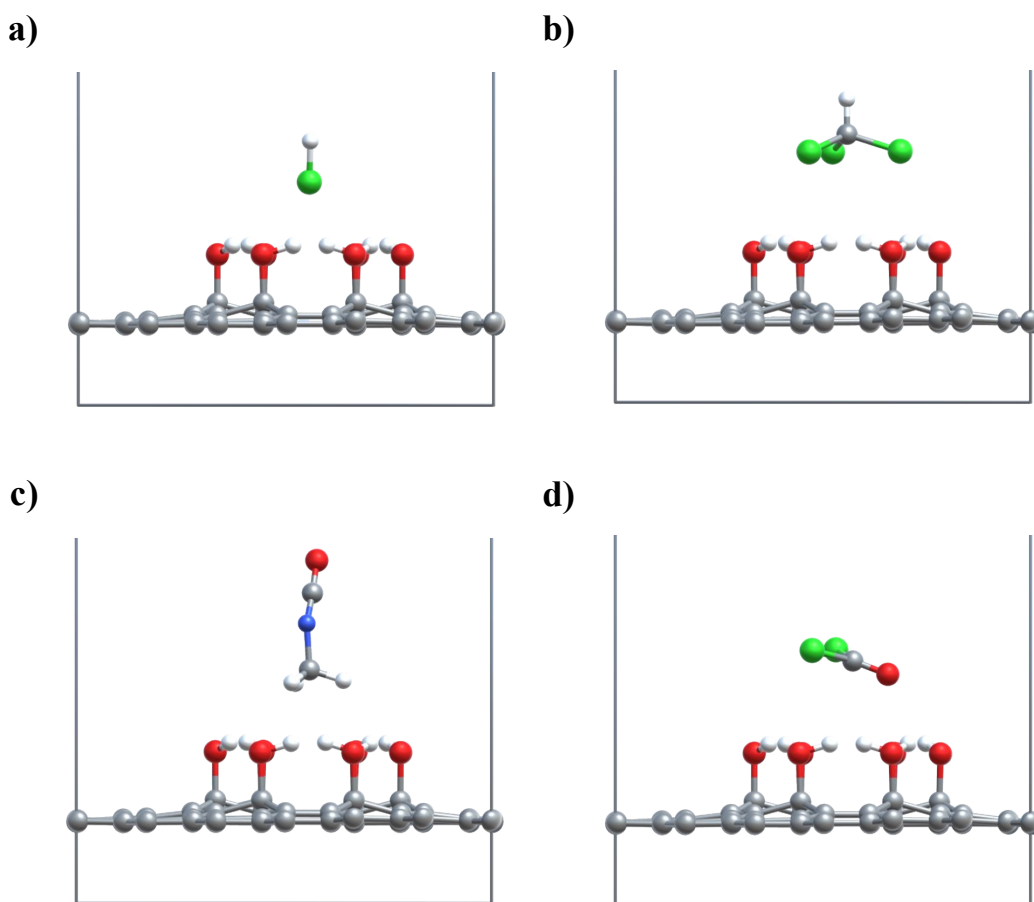
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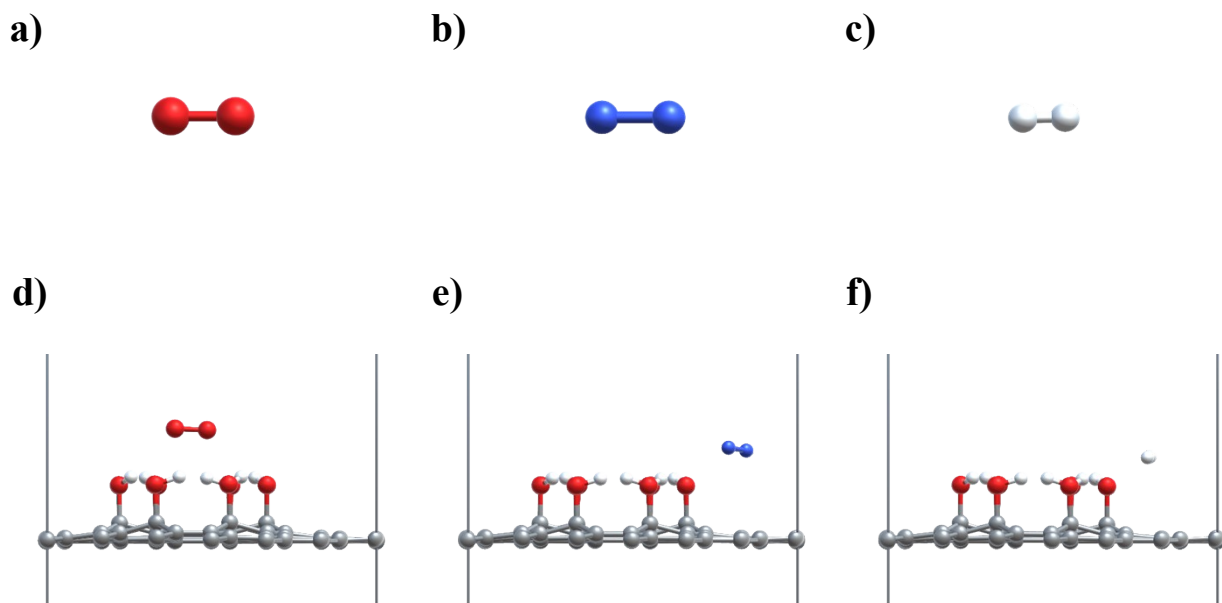
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## Supporting information



**Figure S1.** Geometry optimized adsorbed orientations of a) HCl, b) TCM, c) MIC, and d) PhG analytes placed above the hexahydroxyl ring system



**Figure S2.** Optimized structures of a) O<sub>2</sub>, b) N<sub>2</sub>, and c) H<sub>2</sub> and their most stable adsorbed conformations represented under d) GO-O<sub>2</sub>, e) GO-N<sub>2</sub>, and f) GO-H<sub>2</sub>, respectively

**Table S1.** Computed adsorption energies (in eV) of O<sub>2</sub>, N<sub>2</sub>, and H<sub>2</sub> adsorbed onto GO slab and its corresponding shortest interaction distance (in Å)

Adsorbed slab	$E_{ads}$	
	VOCs placed outside hydroxyl core	VOCs placed above hydroxyl core
GO-O <sub>2</sub>	-0.20	-0.22
GO-N <sub>2</sub>	-0.12	-0.10
GO-H <sub>2</sub>	-0.08	-0.06